6-O-α-L-(2"-O- AND 3"-O-ISOFERULOYL) RHAMNOPYRANOSYLCATALPOLS FROM *PREMNA JAPONICA*

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Key Word Index—*Premna japonica*; Verbenaceae; iridoid; $6-O-\alpha-L$ -rhamnopyranosylcatalpol; $6-O-\alpha-L-(2''-O-isoferuloyl)$ rhamnopyranosylcatalpol; $6-O-\alpha-L-(3''-O-isoferuloyl)$ rhamnopyranosylcatalpol.

Abstract—From the methanol extract of the leaves of *Premna japonica*, two new monoacyl $6-O-\alpha-L$ -rhamnopyranosylcatalpols were isolated. The structures of these compounds were determined to be $6-O-\alpha-L-(2''-O-isoferuloy)$ and 3''-O-isoferuloy|rhamnopyranosylcatalpols by NMR spectroscopy and chemical conversions.

INTRODUCTION

In the course of investigation of the Philippine medicinal plant, *Premna odorata* Blanco, several new acylated rhamnopyranosylcatalpols have been isolated [1, 2]. The related plant, *P. japonica* Miq. (= *P. microphylla* Turcz.) (Verbenaceae) (Japanese name: Hamakusagi) is used in China as an antipyretic, hemostasis and for snake venom [3]. This paper describes the isolation and structure determination of new monoacylated iridoid derivatives from this plant, collected at the south eastern part of Tokushima Prefecture, Shikoku Island, Japan.

RESULTS AND DISCUSSION

The methanol extract of the dried leaves of *P. japonica* was separated by a combination of highly porous polymer, Diaion HP-20 and silica gel column chromatography, and droplet counter-current chromatography (DCCC). Compound 1, $C_{31}H_{40}O_{17}$, was obtained as a colourless amorphous powder, whose *M*, was determined by the observation of ions at m/z 685 [MH]⁺, 707 [M + Na]⁺ and 723 [M + K]⁺ in the FAB-MS on addition of sodium and potassium iodide, respectively. Its IR spectrum showed the presence of a conjugated ester (1700 and 1625 cm⁻¹) and an aromatic ring (1610 and 1510 cm⁻¹). In the ¹³C NMR spectrum, a total of 31 signals were observed (Table 1). Comparison of these signals with those of 2"-caffeoylrhamnopyranosylcatal-pol (5) [1] showed that 1 was most probably a derivative of this compound.

Mild alkaline hydrolysis of 1 in methanol gave a methyl ester of an acyl moiety and $6-O-\alpha_{-}$ -rhamnopy-ranosylcatalpol (7), which was identified with an authentic sample on TLC. The ¹H NMR spectrum of the methyl ester of the acyl moiety showed the presence of methoxyl and carbomethoxyl groups, *trans*-olefinic protons and three aromatic protons, coupled in an ABX system. These data suggested that this acyl moiety is

ferulic acid or isoferulic acid. By comparing the chemical shifts of these aromatic protons with authentic samples of methyl ferulate and methyl isoferulate the acyl portion was identified as isoferulic acid.

Thus the structure of compound 1 was determined to be $6-O-\alpha-L-(2''-O-isoferuloyl)$ rhamnopyranosylcatalpol. Acetylation of 1 gave octaacetate (2), whose spectroscopic data also supported the structure.

Compound 3, $C_{31}H_{40}O_{17}$, was obtained as a colourless powder and had similar physical properties to compound 1. Since differences were observed in the rhamnosyl signals in the ¹³C NMR spectra (Table 1), compound 3 is an isomer of the esterified position of 1. This position was determined to be C-3" by comparison with the ¹³C NMR data of 3"-caffeoyl-rhamnopyranosylcatalpol (6) [1]. The acyl portion was also obtained by mild alkaline hydrolysis in methanol as its methyl ester. The comparison of the ¹H NMR spectral data with that of authentic samples revealed that the acyl moiety is also isoferulic acid. Thus compound 3 is determined to be 6- $O-\alpha-L-(3"-O-isoferuloyl)$ rhamnopyranosylcatalpol. The spectroscopic data of its octaacetate (4) also supported the structure.

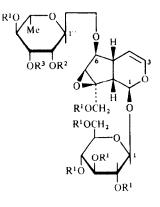
Although the isolation of mono-, di- and triacyl-6-O- α -L-rhamnopyranosylcatalpols has been reported from several plants of the Verbenaceae and Scrophulariaceae [2], isoferuloyl esters of 6-O- α -L-rhamnopyranosylcatalpol were obtained for the first time from natural source.

EXPERIMENTAL

¹H and ¹³C NMR were measured at 100 and 25 MHz, respectively, except as otherwise stated. EIMS: 70 eV. Ferulic and isoferulic acids were purchased from Tokyo Kasei Chemical Co. Ltd (Tokyo). 6-0- α -L-Rhamnopyranosylcatalpol was from our previous experiment [1].

Plant material. The leaves of P. japonica were collected in May (1988) at the south eastern part of Tokushima Prefecture (Shikoku Island). A voucher specimen was deposited at the Department of Pharmacognosy, Institute of Pharmaceutical Science, Hiroshima University School of Medicine (88-PJ-Tokushima-1).

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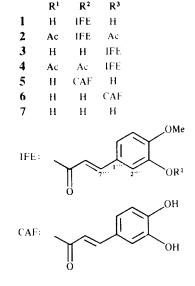


Table 1. ${}^{13}CNMR$ data of compounds 1, 3, 5 and 6 (CD₃OD)

С	1	5*	3	6*
1	95.3	95.2	95.2	95.2
3	142.3	142.3	142.2	142.2
4	103.5	103.5	103.6	103.6
5	37.3	37.2	37.2	37.2
6	84.5	84.4	83.8	83.8
7	59.6	59.6	59.4	59.3
8	66.6	66.5	66.6	66.6
9	43.4	43.3	43.3	43.2
10	61.5	61.5	61.5	61.5
1′	99.8	99.7	99.7	99.7
2′	74.9	74.8	74.8	74.8
3'	78.7	78.6	78.6	78.5
4'	71.8	71.7	71.7	71.7
5'	77.8	77.6	77.7	77.6
6′	63.0	62.9	62.9	62.9
1″	97.9	97.8	100.2	100.2
2‴	74.3	74.1	70.3	70.3
3‴	70.6	70.5	75.4	75.3
4‴	74.3	74.2	71.4	71.7
5″	70.4	70.3	70.3	70.3
6″	18.1	18.1	18.0	18.0
1‴	129.0	127.7	129.0	127.5
2‴	112.7	114.9	112.5	115.2
3‴	151.7	149.6	151.5	149.5
4‴	148.1	146.7	147.9	146.7
5'''	115.0	116.5	114.8	116.5
6'''	122.9	123.2	122.8	123.0
7′′′	147.2	147.6	146.7	147.1
8'''	116.0	115.3	116.4	115.3
9′″	168.5	168.7	168.7	168.9
$-\mathrm{OCH}_3$	56.5		56.4	

*Data taken from ref. [1].

Isolation procedure. The dried leaves of P. japonica (1.45 kg) were extracted with MeOH (285.0 g). The MeOH extract was dissolved in 95% aq. MeOH and extracted with *n*-hexane. The MeOH layer was evapd and suspended in H₂O, and then extracted with EtOAc and *n*-BuOH, successively. The *n*-BuOH extract (84.38 g) was separated by Diaion HP-20 CC with stepwise increases of MeOH in H₂O (20, 40, 60, 80 and 100%). The 40% MeOH eluent (24.08 g) was chromatographed twice over silica gel (CHCl₃-MeOH-H₂O and CHCl₃-MeOH) to give 662 and 608 mg of compound 1 and 3 rich fractions, respectively. The final purification of these fractions by droplet countercurrent chromatography with the solvent system of CHCl₃-MeOH-H₂O-*n*-PrOH (45:60:40:10, ascending method) afforded 297 mg of compound 1 and 237 mg of compound 3.

Compound 1. Amorphous white powder; $[\alpha]_D - 115.5^{\circ}$ (MeOH; *c* 0.32); IR v_{max}^{KBr} cm⁻¹: 3375, 2900, 1700, 1625, 1610, 1510, 1440, 1265, 1125, 1065, 915, 830, 805, UV λ_{max}^{McOH} nm (log ε): 217 (4.24), 235 (4.09) sh, 243 (4.14), 298 (4.27), 310 (4.27), 326 (4.35); FABMS *m/z*: 685 [MH]⁺, 707 [M + Na]⁺ (+ Na], 723 [M + K]⁺ (+ KI); ¹H NMR (CD₃OD, 400 MHz); δ 1.30 (3H, *d*, *J* = 6 Hz), 2.42 (H, *m*), 2.57 (H, *dd*, *J* = 8, 10 Hz), 3.40 (H, *t*, *J* = 9 Hz), 3.49 (H, *t*, *J* = 9 Hz), 3.79 (H, *d*, *J* = 13 Hz), 4.77 (H, *d*, *J* = 8 Hz), 5.02 (H, *d*, *J* = 2 Hz), 5.07 (H, *dd*, *J* = 4, 6 Hz), 5.09 (H, *d*, *J* = 2 Hz), 5.15 (H, *dd*, *J* = 8 Hz), 7.06 (H, *dd*, *J* = 2, 8 Hz), 7.10 (H, *d*, *J* = 2 Hz), 7.62 (H, *d*, *J* = 16 Hz); ¹³C NMR: see Table 1. (Found: C, 53.07; H, 6.17. C₃₁H₄₀O₁₇ · H₂O requires: C, 52.99; H, 6.02%).

Compound 1 octaacetate (2). Compound 1 (37 mg) was treated with a mixture of Ac₂O and pyridine at 25° overnight. Usual work-up gave 48 mg of colourless powder. $[\alpha]_D - 31.7^\circ$ (CHCl₃; c 0.46); IR v^{KBr}_{max} cm⁻¹: 1745, 1630, 1605, 1510, 1430, 1365, 1220, 1120, 1040, 905; UV λ^{MeOH}_{max} nm (log ε): 227 (4.22), 294 (4.42) inf, 309 (4.50); EIMS m/z: 331, 177, 169, 110; FABMS m/z: 1021 [MH]⁺, 1043 [M+Na]⁺ (+NaI), 1059 [M+K]⁺ (+KI); ¹H NMR (CDCl₃): δ 1.26 (3H, d, J = 6 Hz), 1.99, 2.02, 2.03, 2.05, 2.06, 2.10, 2.13 (3H ea., s, alcoholic Ac × 7), 2.33 (3H, s, phenolic Ac), 3.88 (3H, s), 6.40 (H, d, J = 16 Hz), 6.92 (H, d, J = 8 Hz), 7.66 (H, d, J = 16 Hz); ¹³C NMR (CDCl₃): δ 17.4, 20.6 (Ac × 7), 35.5, 41.7, 56.0, 58.0, 61.1, 62.2, 62.4, 67.0, 68.3, 68.9, 70.0, 70.6, 71.2, 72.3, 72.6, 83.5, 94.3, 96.6 (\times 2), 102.4, 112.4, 115.7, 122.3, 127.3, 127.9, 140.1, 141.1, 145.0, 153.3, 166.0, 168.7, 169.0, 169.3, 169.9 (\times 2), 170.2, 170.6 (\times 2). (Found: C, 54.69; H, 5.47. C₄₇H₅₆O₂₅ · 1/2 H₂O requires: C, 54.81; H, 5.57%).

Compound 3. Amorphous white powder; $[\alpha]_D - 117.9^{\circ}$ (MeOH; c 0.30); IR ν_{max}^{KBr} cm⁻¹: 3375, 2900, 1690, 1625, 1610, 1510, 1265, 1125, 1055, 920, 840, 810; UV λ_{max}^{mcOH} nm (log ε): 217 (4.27), 234 (4.11) sh, 242 (4.15), 296 (4.31), 311 (4.31), 325 (4.37); FABMS *m*/*z*: 685 [MH]⁺, 707 [M+Na]⁺ (+NaI), 723 [M +K]⁺ (+KI); ¹H NMR (CD₃OD, 400 MHz): δ 1.31 (3H, *d*, *J* = 6 Hz), 2.45 (H, *m*), 2.57 (H, *dd*, *J* = 8, 10 Hz), 3.14 (H, *t*, *J* = 9 Hz), 3.81 (H, *d*, *J* = 13 Hz), 3.89 (3H, s), 4.04 (H, *dd*, *J* = 1, 8 Hz), 4.09 (H, *dd*, *J* = 2, 7 Hz), 5.10 (H, *d*, *J* = 10 Hz), 5.12 (H, *dd*, *J* = 4, 6 Hz), 6.38 (H, *dd*, *J* = 1, 6 Hz), 6.41 (H, *d*, *J* = 16 Hz), 6.94 (H, *d*, *J* = 16 Hz); ¹³C NMR: see Table 1. (Found: C, 53.07; H, 6.13. C₃₁H₄₀O₁₇ · H₂O requires: C, 52.99; H, 6.02%).

Compound 3 octaacetate (4). Compound 3 (45 mg) was treated with a mixture of Ac₂O and pyridine at 25° overnight. Usual work-up gave 58 mg of colourless powder (4). $[\alpha]_{\rm D} = -56.8^{\circ}$ $(CHCl_3; c 0.44)$, IR ν_{max}^{KBr} cm⁻¹: 1750, 1630, 1610, 1510, 1430, 1365, 1220, 1120, 1040, 900; UV λ^{MeOH} nm (log ε): 227 (4.24), 294 (4.44) inf, 309 (4.50); EIMS m/z: 331, 177, 169, 110; FABMS m/z: 1021 $[MH]^+$, 1043 $[M+Na]^+$ (+NaI), 1059 $[M+K]^+$ (+KI); ¹H NMR (CDCl₃): δ 1.24 (3H, d, J = 6 Hz), 2.02 (3H, s), 2.03 (6H, s), 2.05 (3H, s), 2.11 (3H, s), 2.13 (3H, s), 2.17 (3H, s) (alcoholic Ac ×7), 2.32 (3H, s, phenolic Ac), 3.87 (3H, s), 6.20 (H, d, J = 16 Hz), 6.32 (H, br. d, J = 6 Hz), 6.97 (H, d, J = 9 Hz), 7.56 (H, d, J = 16 Hz); 13 C NMR (CDCl₃): δ 17.4, 20.6 (Ac × 6), 20.8 (Ac), 20.9 (Ac), 35.4, 41.7, 56.0, 58.0, 61.1, 62.1, 62.3, 67.0, 68.3, 68.9, 70.2, 70.7, 71.1, 72.3, 72.6, 83.6, 94.3, 96.5 (×2), 102.4, 112.3, 115.6, 122.2, 127.4, 127.9, 140.1, 141.1, 144.8, 153.2, 165.7, 169.0, 169.2, 169.9, 170.0 (×2), 170.2, 170.6 (×2). (Found: C, 54.69; H, 5.48. C47H56O25 · 1/2 H2O requires: C, 54.81; H, 5.57%).

Alkaline hydrolysis of compounds 1 and 3. Compounds 1 (50 mg) and 3 (50 mg) were treated with 0.2 M methanolic. NaOH and stirred for 2 hr at 15° under a N₂ atmosphere. The solns were neutralized with Dowex $50W \times 8$. The ion exchange resin was removed by filtration, and the filtrates were evapd. The residues were partitioned between EtOAc (50 ml) and H₂O (50 ml). The H_2O layers were evapd to give residues. The EtOAc layers were evapd and purified by prep. TLC (silica gel, 0.5 mm, developed with CHCl₃-MeOH, 15:1 and eluted with CHCl₃-MeOH, 9:1) to give residues (12 mg, 79% and 11 mg, 73%, respectively). The compounds from the H₂O layers were identified with an authentic sample of 6-O-a-L-rhamnopyranosylcatalpol (7) by co-chromatography on TLC (silica gel, CHCl₃-MeOH-H₂O, 15:6:1 and EtOAc-EtOH-H₂O, 8:2:1). The compounds from the EtOAc layers were also identified with an authentic sample of methyl isoferulate on TLC (silica gel, CHCl₃-MeOH, 10:1) and comparison of IR and ¹HNMR

spectra with those of the authentic samples. Methyl isoferulate from compound 1, IR v_{max}^{KBr} cm⁻¹: 3350, 1715, 1630, 1605, 1580, 1505, 1440, 1350, 1310, 1260, 1165, 1125, 1020, 975, 915, 845, 800, 765; ¹H NMR (CD₃OD, 400 MHz): δ3.76 (3H, s), 3.88 (3H, s), 6.31 (H, d, J = 16 Hz, H-8), 6.93 (H, d, J = 8 Hz, H-5), 7.04 (H, dd, J = 2, 8 Hz, H-6), 7.06 (H, d, J = 2 Hz), 7.56 (H, d, J = 16 Hz, H-7); HR-EIMS m/z: 208.0732 [M]⁺, 177.0576 [M-Me]⁺. Methyl isoferulate from compound 3, IR ν_{max}^{KBr} cm⁻¹: 3350, 1715, 1635, 1610, 1580, 1505, 1440, 1355, 1310, 1265, 1165, 1125, 1020, 975, 920, 850, 805, 765; ¹H NMR (CD₂OD, 400 MHz); δ3.76 (3H, s), 3.88 (3H, s), 6.30 (H, d, J = 16 Hz, H-8), 6.93 (H, d, J = 8 Hz, H-5), 7.04 (H, dd, J = 2, 8 Hz, H-6), 7.06 (H, d, J = 2 Hz), 7.56 (H, d, J= 16 Hz, H-7); HR-EIMS m/z: 208.0774 [M]⁺, 177.0568 [M $-OMe]^+$, (208.0735, calcd. for $C_{11}H_{12}O_4$, 177.0553, calcd. for C10H9O3). Authentic methyl ferulate and isoferulate were obtained by refluxing the corresponding acids in MeOH containing a few drops of conc HCl. Authentic methyl isoferulate ¹H NMR (CD₃OD, 400 MHz): δ 3.76 (3H, s), 3.88 (3H, s), 6.31 (H, d, J = 16 Hz, H-8), 6.93 (H, d, J = 8 Hz, H-5), 7.04 (H, dd, J = 2, 8 Hz, H-6), 7.06 (H, d, J = 2 Hz), 7.56 (H, d, J = 16 Hz, H-7). Authentic methyl ferulate ¹H NMR (CD₃OD, 400 MHz): δ3.76 (3H, s), 3.88 (3H, s), 6.34 (H, d, J = 16 Hz, H-8), 6.80 (H, d, J = 8 Hz, H-5), 7.05 (H, dd, J = 2, 8 Hz, H-6), 7.16 (H, d, J = 2 Hz), 7.59 (H, d, J = 16 Hz. H-7).

Detection of sugar portion. Ca 2 mg of each compound (1 and 3) was heated in 2 ml HCl in dry MeOH at 95° for 3 hr. The reaction mixture was neutralized with addition of Ag₂CO₃ and then filtered. After the solvent was removed, several drops of TMS-imidazole soln were added and kept at 50° for 15 min. The reaction mixture was partitioned between 1 ml of H₂O and 2 ml of *n*-hexane. The hexane layer was concd and subjected to GLC analysis. Column: 1.5% OV-1 (2 mm × 2 m); Temp: 180° (isothermal); N₂: 40 ml/min. R_i: rham, 2.81 min, gluc, 9.08 and 9.98 min; Compound 1: rham, 2.81 min, gluc, 9.07 (overlapped by methyl isoferulate-3-OTMS) and 9.97 min; Compound 3: rham, 2.80 min. gluc, 9.07 (overlapped by methyl isoferulate-3-OTMS) and 10.00 min.

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