## Synthesis of 2-Aminotropone Oximes and 2-Alkoxytropone Imines<sup>†</sup>

J. Chem. Research (S), 1997, 362–363†

Tetsuo Nozoe,<sup>1</sup>‡ Lung Ching Lin,<sup>\*ab</sup> Chih-Hsien Hsu,<sup>a</sup> Shwu-Chen Tsay,<sup>b</sup> Gholam H. Hakimelahi<sup>b</sup> and Jih Ru Hwu<sup>\*b,c</sup>

<sup>a</sup>Department of Chemistry, National Taiwan University, Taipei, Taiwan 10671, Republic of China

<sup>b</sup>Institute of Chemistry, Academia Sinica, Nankang, Taipei, Taiwan 11529, Republic of China <sup>c</sup>Department of Chemistry, National Tsing Hua University, Hsinchu, Taiwan 30043, Republic of China

A synthetic route was developed for preparation of 2-aminotropone oximes **10–14**, a new class of compounds, from tropolone **2**; 2-methoxytropone imines **7–9** and tropylium salts **17** were generated as the key intermediates.

*cis*-Diamminedichloroplatinum analogues 1 are used as drugs for clinical cancer chemotherapy.<sup>2,3</sup> In order for 1 to exhibit significant biological activity, the two amino groups must be in a *cis* configuration in  $1.^{4,5}$  Here we report the preparation of 2-aminotropone oximes (*e.g.*, **10–14**), in which the two adjacent nitrogen atoms are attached to a planar nucleus and could coordinate to Pt to form *cis*-platinum complexes.

$$\begin{array}{c} CI & NH_2R \\ CI & Pt & M_2R \end{array}$$
 1 R = alkyl, aryl

Reaction of tropone with NH<sub>2</sub>OH·HCl and pyridine in methanol generates tropone oxime and 2-aminotropone.<sup>6</sup> Under the same conditions, 2-alkyltropones can also be converted to 2-alkyltropone oximes and 2-alkyl-7-aminotropones,<sup>7</sup> yet tropolone **2** remains intact.<sup>8</sup> By replacement of pyridine with various bases, including NaOH, NaOMe, NaOAc, Na<sub>2</sub>CO<sub>3</sub> and Et<sub>3</sub>N, we were also unable to convert tropolone to the corresponding oximes by using NH<sub>2</sub>OH·HCl. The unusual resonance phenomenon associated with tropolone and an inherent intramolecular hydrogen bonding between the OH and the C=O groups<sup>9</sup> may decrease its reactivity towards oxime formation. Further-



p-MeC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>Cl, pyridine, 0 °C (96%); ii, for  $3 \rightarrow 4$ : *p*-MeC<sub>6</sub>H<sub>4</sub>NH<sub>2</sub>, BuOH, reflux (72%); for  $3 \rightarrow 5$ : PhCH<sub>2</sub>NH<sub>2</sub>, BuOH, reflux (76%); for  $3 \rightarrow 6$ : cyclohexylamine, BuOH, reflux (65%); iii, for  $4-6 \rightarrow 7-9$ : Me<sub>2</sub>SO<sub>4</sub>, toluene, reflux, NaHCO<sub>3</sub> (aq) (60-65%); iv, for  $7 \rightarrow 10+15$  and  $8 \rightarrow 11+15$ : NH<sub>2</sub>OH·HCl, NaOMe, MeOH; for  $7-9 \rightarrow 12-14+16$ : NH<sub>2</sub>OMe·HCl, NaOMe, MeOH

\*To receive any correspondence.

more, our attempts to oximate 2-(p-tolylsulfonyl)tropone and 2-aminotropone were also unsuccessful. Herein we report an indirect way to convert tropolone 2 to the desired 2-aminotropone oximes 10-14. To the best of our knowledge, this provides the only available up-to-date route for the preparation of 2-aminotropone oximes, an unprecedented class of compounds.

After tosylation of  $2^{10}$  the resultant toluene-*p*-sulfonate 3 was treated with various amines,11 including p-toluidine, benzylamine and cyclohexylamine, in BuOH under reflux to give the corresponding 2-aminotropones 4-6 in 65-76% yields (Scheme 1). Methylation<sup>11</sup> of 4-6 with dimethyl sulfate in toluene under reflux followed by treatment with NaHCO<sub>3</sub> afforded the 2-methoxytropone imines 7-9 in 60-65% overall yields. Upon reaction with NH<sub>2</sub>OR·HCl (R = H or Me) and NaOMe in MeOH, 7-9 were converted to the desired oximes 10-14 in 24-54% yields. The spectroscopic data are summarized in Table 1. In these reactions, an unexpected by-product (i.e., 2-methoxytropone oxime 15 or 16) was generated and its structure was determined with the aid of single-crystal X-ray diffraction analysis.§ Formation of oximes 10-14 came from a nucleophilic attack of 7-9 by  $NH_2OR$  (R = H or Me) at the C-2 position and formation of oximes 15 and 16 came from an attack at the C-1 position.

Furthermore, we found that reactions of 2-aminotropones **4–6** with methyl fluorosulfonate gave the corresponding isolable tropylium salts **17** (Scheme 2).<sup>12</sup> The desired oximes **10–14** can be obtained by treatment of **17** with NH<sub>2</sub>OH·HCl or NH<sub>2</sub>OMe·HCl. Moreover, we were able to synthesize 2-aminotropone imines **18–20** in high yields (>80%) by treating tropylium salts **17** with primary amines, including *p*-toluidine and benzylamine.<sup>13</sup> These 2-aminotropone imines were also obtained in excellent yields (91–98%) by the reactions of 2-methoxytropone imines **7** and **8** with amines (Scheme 2 and Table 1).<sup>11</sup>

In conclusion, 2-methoxytropone imines 7–9 and tropylium salts 17 were prepared readily from tropolone 2 *via* toluene-*p*-sulfonate 3 and 2-aminotropones 4–6. These key intermediates (*i.e.*, 7–9 and 17) were successfully converted to the 2-aminotropone oximes (10 and 11) and oxime methyl ethers (12–14) upon treatment with NH<sub>2</sub>OR·HCl (R = H or Me), and to 2-aminotropone imines 18–20 with primary amines. Use of the resultant new nitrogen-containing planar compounds to form *cis*-diamminedichloroplatinum analogues and to test their biological activity are under investigation in our laboratory.

## Experimental

General Procedure for the Conversion of 2-Methoxytropone Imines (7–9) to 2-Aminotropone Oximes 10–14.—To a clear solution of NH<sub>2</sub>OH·HCl or NH<sub>2</sub>OMe·HCl (2.30 mmol) and NaOMe (2.30 mmol) in MeOH (20 mL) was added a 2-methoxytropone imine (7–9, 2.22 mmol). The reaction mixture was stirred at room temperature for 4.0 h. After the solvent was removed under reduced

<sup>†</sup>This is a **Short Paper** as defined in the Instructions for Authors, Section 5.0 [see *J. Chem. Research (S)*, 1997, Issue 1]; there is therefore no corresponding material in *J. Chem. Research (M)*. ‡Deceased April 4, 1996.

<sup>\$</sup>Details of the X-ray crystal-structure determination will be reported elsewhere.

Table 1 Conversion of 2-methoxytropone imines 7–9 to a mixture of oximes 10–16 or to 2-aminotropone imines 18–20 by the use of various nitrogen-containing reagents

Imine	Reagent	Products (% in yield) <sup>ª</sup>	$\delta_{\rm H}~({\rm CDCI_3})$	$\delta_{\rm c}$ (CDCl <sub>3</sub> )	v <sub>max</sub> /cm <sup>-1</sup>	Мр ( <i>T/</i> °С)	Found (calcd) (%)		
							С	н	Ν
7	NH₂OH∙HCI	<b>10</b> (28) + <b>15</b> (50)	<b>10</b> : 2.31 (s, 3 H), 6.12–7.20 (m, 10 H), 7.46 (br s, 1 H); <b>15</b> : 3.72 (s, 3 H), 5.80–7.11 (m, 5 H), 9.83 (br s, 1 H)	<b>10</b> : 21.9, 106.3, 117.6, 121.6, 125.8, 130.9, 132.9, 133.1, 135.8, 137.6, 145.8, 151.0; <b>15</b> : 55.6, 104.9, 120.7, 124.2, 129.0, 131.1, 149.9, 157.0	10: 3315 (OH, NH), 1583 (C=N); 15: 3298 (OH), 1599 (C==N)	<b>10</b> : liq.; <b>15</b> : 132–133	10: 74.26 (74.31) 15: 63.48 (63.57)	6.55 (6.24) 5.99 (6.00)	12.26 (12.38) 9.38 (9.27)
8	NH₂OH∙HCI	<b>11</b> (24) + <b>15</b> (50)	11: 4.43 (d, 2 H), 5.82–7.33 (m, 12 H)	<b>11</b> : 47.4, 103.4, 115.6, 119.1, 127.2, 127.4, 128.7, 132.0, 132.4, 137.4, 146.2, 150.1	11: 3352 (OH, NH), 1584 (C≕N)	<b>11</b> : liq.	<b>11</b> : 74.18 (74.31)	6.31 (6.24)	12.49 (12.38)
7	NH₂OMe∙HCI	<b>12</b> (54) + <b>16</b> (30)	<b>12</b> : 2.33 (s, 3 H), 4.00 (s, 3 H), 6.11–7.28 (m, 9 H), 7.62 (br s, 1 H); <b>16</b> : 3.84 (s, 3 H), 4.01 (s, 3 H), 5.82–6.94 (m, 5 H)	12: 20.9, 62.1, 105.3, 117.3, 120.7, 124.9, 129.9, 132.0, 132.2, 134.8, 136.6, 144.8, 148.4; 16: 560, 62.0, 105.2, 120.9, 124.3, 129.0, 131.0, 151.0, 151.2	12: 3152 (NH), 1583 (C—N); 16: 1590 (C—N)	12: liq.; 16: liq.	<b>12</b> : 74.88 (74.97) <b>16</b> : 65.45 (65.44)	6.73 (6.71) 6.84 (6.71)	11.57 (11.66) 8.40 (8.48)
8	NH₂OMe∙HCl	<b>13</b> (47) + <b>16</b> (25)	<b>13</b> : 4.02 (s, 3 H), 4.49 (d, 2 H), 5.82–7.43 (m, 10 H), 7.87 (br s, 1 H)	<b>13</b> : 47.2, 62.0, 103.3, 116.5, 119.2, 127.2, 127.4, 128.8, 132.0, 132.5, 134.9, 137.8, 146.3, 148.6	13: 3360 (NH), 1584 (C──N)	<b>13</b> : liq.	<b>13</b> : 75.09 (74.97)	6.63 (6.71)	11.77 (11.66)
9	NH₂OMe∙HCl	<b>14</b> (51) + <b>16</b> (31)	<b>14</b> : 0.92–2.00 (m, 11 H), 3.43 (br s, 1 H), 3.96 (s, 3 H), 5.81–7.06 (m, 5 H)	<b>14</b> : 24.7, 25.7, 32.4, 50.8, 61.9, 115.4, 118.0, 131.8, 132.5, 145.4, 148.5	<b>14</b> : 3338 (NH), 1583 (C <del>—</del> N)	<b>14</b> : liq.	<b>14</b> : 72.44 (72.38)	8.63 (8.68)	12.15 (12.06)
7	p-MeC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	<b>18</b> (98)	<b>18</b> : 2.41 (s, 3 H), 2.49 (s, 3 H), 6.19–7.32 (m, 14 H)	<b>18</b> : 20.9, 114.5, 121.5, 122.6, 130.0, 133.2, 133.4, 142.5, 152.1	<b>18</b> : 3059 (NH), 1587 (C──N)	<b>18</b> : 143–144	<b>18</b> : 83.83 (83.96)	6.71 (6.71)	9.25 (9.33)
7	PhCH <sub>2</sub> NH <sub>2</sub>	<b>19</b> (91)	<b>19</b> : 2.32 (s, 3 H), 4.54 (s, 2 H), 6.10–7.32 (m, 15 H)	<b>19</b> : 20.8, 47.2, 105.4, 119.9, 120.7, 120.9, 127.2, 127.4, 130.0, 131.9, 133.0, 133.7, 137.4, 148.4, 150.8, 155.1	<b>19</b> : 3265 (NH), 1582 (C──N)	<b>19</b> : 113–114	<b>19</b> : 84.09 (83.96)	6.80 (6.71)	9.24 (9.33)
8	$PhCH_2NH_2$	<b>20</b> (95)	<b>20</b> : 4.62 (br s, 4 H), 6.27–7.33 (m, 16 H)	<b>20</b> : 50.0, 111.6, 119.1, 126.8, 127.2, 128.4, 133.8, 139.3, 153.2	<b>20</b> : 3055 (NH), 1591 (C──N)	<b>20</b> : 81–82	<b>20</b> : 84.03 (83.96)	6.56 (6.71)	9.38 (9.33)

<sup>a</sup>All data are satisfactory for the high- and low-resolution mass spectroscopy.





pressure, the residue was purified by use of column chromatography (EtOAc/hexanes = 1:5 as eluent) to give a 2-aminotropone oxime (10-14) and a 2-methoxytropone oxime (15 or 16)in a pure form. The yields and spectroscopic data are listed in Table 1.

This work was supported by the National Science Council of the Republic of China and Academia Sinica.

Received, 7th March 1997; Accepted, 5th June 1997 Paper E/7/01621B

## **References and notes**

- 1 811-2-5-1, Kami-Yoga, Setagaya-Ku, Tokyo 158, Japan
- 2 B. Rosenberg, L. VanCamp, J. E. Trosko and V. H. Mansour, *Nature (London)*, 1969, 222, 385.
- 3 K. R. Harrap, Cancer Treat. Rev., 1985, 12, 21.
- 4 S. J. Lippard, H. M. Ushay, C. M. Merkel and M. C. Poirier, *Biochemistry*, 1983, 22, 5165.
- 5 M. V. Keck and S. J. Lippard, J. Am. Chem. Soc., 1992, 114, 3386.
- T. Machiguchi, T. Hasegawa, M. Ohno, Y. Kitahara, M. Funamizu and T. Nozoe, J. Chem. Soc., Chem. Commun., 1988, 838.
  T. Nozoe, T. Mukai and I. Murata, Proc. Jpn. Acad., 1953, 29,
- 169.
- 8 T. Nozoe, T. Mukai, K. Takase and T. Nagase, Proc. Jpn. Acad., 1952, 28, 477.
- 9 W. von É. Doering and L. H. Knox, J. Am. Chem. Soc., 1951, 73, 828.
- 10 W. von E. Doering and C. F. Hiskey, J. Am. Chem. Soc., 1952, 74, 5688.
- A. Zask, N. Gonnella, K. Nakanishi, C. J. Turner, S. Imajo and T. Nozoe, *Inorg. Chem.*, 1986, **25**, 3400.
  P. Beak, J.-K. Lee and B. G. McKinnie, *J. Org. Chem.*, 1978, **43**,
- 12 P. Beak, J.-K. Lee and B. G. McKinnie, J. Org. Chem., 1978, 43, 1367.
- 13 cf. W. R. Brasen, H. E. Holmquist and R. E. Benson, J. Am. Chem. Soc., 1961, 83, 3125; K. Kikuchi, Y. Maki and K. Sato, Bull. Chem. Soc. Jpn., 1978, 51, 2338.