

Letter

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# Iridium-Catalyzed Synthesis of Substituted Indanones from Aromatic Carboxylates and Unsaturated Ketones

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**ABSTRACT:** A catalytic annulation is presented that provides straightforward, modular synthetic access to 3-substituted indanones from benzoic acids and  $\alpha$ , $\beta$ -unsaturated ketones. It is catalyzed by a bimetallic Ir/In system and proceeds via hydroarylation followed by Claisen condensation and optional retro-Claisen deacylation. The annulation may be combined into a one-pot procedure with the synthesis of the unsaturated ketone substrates from aldehydes and acetone. Two complementary reaction protocols are provided that are applicable to diversely functionalized electron-rich and electron-poor substrates.

KEYWORDS: iridium; C-H activation; benzoic acids; substituted indanones; diketo compounds

Substituted indanones have distinct biological activity,<sup>1</sup> and are thus common substructures in pharmaceuticals.<sup>2</sup> Moreover, they are valuable synthetic intermediates.<sup>3</sup> Traditional synthetic entries to substituted indanones, such as inter- or intramolecular Friedel-Crafts reactions of cinnamic acid derivatives,<sup>4</sup> suffer from rather harsh, acidic conditions and/or low regioselectivities.

# Scheme 1. Strategies for the synthesis of 3-substituted indanones.



Transition metal-catalyzed reactions aiming at the regioselective construction of substituted indanones (Scheme 1),<sup>5</sup> including Pd-catalyzed carbonylative cyclisations of iodosubstituted 1,1-diaryl alkenes (a)<sup>6</sup> and Rh- or Co-catalyzed hydroacylations (b) have opened up regioselective entries to this substrate class,<sup>7</sup> but depend on the availability of elaborate substrates. The same limitation applies to the reductive cyclisation of 2-substitued chalcones (c)<sup>8</sup> and Aucatalyzed C–H coupling of acylsilanes with TMSarylacetylenes (d).<sup>9</sup> The CsF-promoted reductive crosscoupling of 1,3-indanedione monotosylhydrazone with an arylboronic acid exemplifies a synthetic strategy in which an organic residue is introduced selectively in the 3-position of a pre-formed indanone ring (e).<sup>10</sup>

As an advantageous alternative to these methods, we herein disclose a modular iridium-catalysed synthesis of indanones from widely available benzoic acid, aldehyde, and methyl ketone building blocks.<sup>n</sup>

# Scheme 2. Mechanistic blueprint of the targeted annelation.



In continuation of research performed in the groups of Yu,<sup>12</sup> Li,<sup>13</sup> Daugulis,<sup>14</sup> Larrosa,<sup>15</sup> Ackermann,<sup>16</sup> Su,<sup>17</sup> our group,<sup>18</sup> and others<sup>19</sup> on carboxylic acids as directing groups in C–H functionalizations, we recently discovered a new reaction mode of benzoic acids with  $\alpha$ , $\beta$ -unsaturated ketones (Scheme 2).<sup>20</sup> In the presence of a Rh catalyst, the benzoic acid 1 undergoes *ortho*-metalation to intermediate

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 $I_{1}^{21}$  which in turn carbometalates the  $\alpha_{1}\beta$ -unsaturated ketone 2 to give III. Rather than continuing along known Heck-type or decarboxylative hydroarylation pathways,<sup>22</sup> the Rh enolate then substitutes the unprotected carboxylate OH group to give the Claisen product 3. Under Lewisacidic reaction conditions, this step is inevitably followed by a retro-Claisen deacylation,<sup>23</sup> furnishing the indanone 4. The new reaction concept promised to open up a window of opportunities for the synthesis of indanones, but unfortunately, the scope of the prototypical Rh/In catalyst was rather limited. Best results were obtained with cyclic ketone substrates, whereas the yields were only moderate for simple alkenyl methyl ketones, and all test reactions had failed for 3-aryl alkenones. Moreover, it was not possible to stop the reaction at the acyl intermediate III. With Ru, similar substrates were reported to enter complex reaction cascades with double incorporation of the enone substrate into 2-substituted indanones and spirocyclic derivatives.24

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#### Table 1. Optimization of the reaction conditions<sup>a</sup>

	OH + Ph.	Catalyst base, additive			>
	-	2a	3aa <sup>Ph</sup>	4aa <sup>F</sup>	'n
	catalyst	base, additive	solvent	<b>3aa</b> (%)	<b>4aa</b> (%)
1	$[RhCp^*Cl_2]_2$	$K_2CO_3$ , $In(OTf)_3$	<sup>t</sup> AmOH	N.D	35
	$[RhCp^*Cl_2]_2$	K <sub>2</sub> CO <sub>3</sub>	-	20	2
3	$[\operatorname{Ru}(cy)\operatorname{Cl}_2]_2$	K <sub>2</sub> CO <sub>3</sub>	-	N.D	N.D
4	[IrCODCl] <sub>2</sub>	K <sub>2</sub> CO <sub>3</sub>	-	N.D	N.D
5	$[IrCp*Cl_2]_2$	K <sub>2</sub> CO <sub>3</sub>	-	39	25
7	$[IrCp*Cl_2]_2$	K <sub>3</sub> PO <sub>4</sub>	-	35	8
8	$[IrCp*Cl_2]_2$	Na <sub>2</sub> CO <sub>3</sub>	-	37	21
6	[IrCp*Cl <sub>2</sub> ] <sub>2</sub>	NaOAc	-	49	N.D
9	$[IrCp*Cl_2]_2$	KOAc	-	55	N.D
10	$[IrCp*Cl_2]_2$	KOAc <sup>b</sup>	-	63	3
11 <sup>c</sup>	$[IrCp^*Cl_2]_2$	KOAc <sup>b</sup>	-	72	3
12	$[IrCp*Cl_2]_2$	K <sub>2</sub> CO <sub>3</sub>	dioxane	28	20
13	$[IrCp*Cl_2]_2$	K <sub>2</sub> CO <sub>3</sub>	NMP	N.D	21
14	$[IrCp*Cl_2]_2$	K <sub>2</sub> CO <sub>3</sub>	<sup>t</sup> AmOH	24	44
15	$[IrCp*Cl_2]_2$	$K_2CO_3$ , Fe(OTf) <sub>3</sub>	<sup>t</sup> AmOH	N.D	60
16	$[IrCp*Cl_2]_2$	K <sub>2</sub> CO <sub>3</sub> , Zn(OTf) <sub>2</sub>	<sup>t</sup> AmOH	N.D	71
17	$[IrCp^*Cl_2]_2$	$K_2CO_3$ , Sc(OTf) <sub>3</sub>	<sup>t</sup> AmOH	N.D	80
18	[IrCp*Cl <sub>2</sub> ] <sub>2</sub>	$K_2CO_3$ , $In(OTf)_3$	<sup>t</sup> AmOH	N.D	84

<sup>a</sup>Conditions: 0.5 mmol of **1a**, 0.8 mmol of **2a**, 3 mol% of catalyst, 7.5 mol% of additive, 0.4 equiv. of base, 0.2 mL of H2O or of 3/1 solvent/H2O, 140 °C, 16 h. Yields determined by <sup>1</sup>H NMR using 1,3,5-trimethoxybenzene as the internal standard. <sup>*b*</sup>1.0 equiv. <sup>c</sup>150 °C. <sup>*t*</sup>AmOH = *tert*-amyl alcohol, cy = *p*-cymene.

We started our search for an effective and tuneable catalyst using the reaction of 2-methylbenzoic acid (1a) and  $\beta$ -phenyl vinyl ketone (2a) as a model (Table 1). As expected, unsatisfactory conversion mostly to the deacylated product **4aa** was observed in the presence of the Rh/In system (entry 1). In order to slow down deacylation, the Lewis acid InOTf was left out, and the reaction mixture was buffered with aqueous  $K_2CO_3$ . Under these conditions, neither ruthenium nor rhodium systems gave satisfactory conversion (entries 2-3).

In an extensive screening of various catalyst metals, solely  $[IrCp^*Cl_2]_2$  gave significant amounts of the desired product **3aa**, along with the deacylation product **4aa** (entry 4-5).<sup>25</sup> Further investigations revealed that the base has a decisive influence on the ratio between these products (entries 6-9). Only with KOAc or NaOAc, **3aa** was obtained selectively, all other bases tested led to deacylation. After further adjustments (entries 10-11), the highest yield of **3aa** was obtained with 3 mol%  $[IrCp^*Cl_2]_2$  as the catalyst in the presence of 2.5 M aqueous KOAc (1.0 equiv.) and 1.6 equiv. of the alkenone at 150 °C without solvent (Protocol A).

In the presence of  $K_2CO_3$ , the deacylated indanone **4aa** was formed in significant amounts (entry 5). Its fraction increased in aqueous polar organic solvents, particularly in <sup>t</sup>AmylOH/H<sub>2</sub>O, where it was the major product (entries 12-14). Adding catalytic amounts of Lewis acids further promoted the deacylation step, and best results were obtained with In(OTf)<sub>3</sub> (entries 15–18).<sup>23f</sup> The highest selectivity for **4aa** was obtained when stirring **1a** with 1.6 equiv. of the alkenone in the presence of 3 mol% [IrCp\*Cl<sub>2</sub>]<sub>2</sub> and 7.5% In(OTf)<sub>3</sub> with 0.4 equiv. of a 1 M solution of K<sub>2</sub>CO<sub>3</sub> in a 3:1 <sup>t</sup>AmylOH/water mixture at 140 °C (Protocol B).

#### Table 2. Substrate scope of 2-acyl-3-arylindanones<sup>a</sup>



<sup>a</sup>Conditions: 0.5 mmol of **1**, 0.8 mmol of **2**, 3 mol% of [IrCp\*Cl2]2, 0.5 mmol of KOAc, 0.2 mL H2O, 150 °C, 16 h.

Having thus found two complementary protocols for the non-deacylative (A) and deacylative annulation of benzoic 1

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acids (B), we next investigated their preparative applicability to the synthesis of indanones. The scope of the nondeacylative protocol (A) was investigated by coupling representative aromatic carboxylic acids with several unsaturated ketones. As shown in Table 2, aliphatic as well as aromatic substituents bearing electron-donating or electronwithdrawing groups are smoothly converted (**3aa-3at**). Both aliphatic and aromatic acyl groups can be introduced in the 2-position (**3au-3aw**).



<sup>a</sup>Reaction conditions: 0.5 mmol of 1, 0.8 mmol of 2, 3 mol% of [IrCp\*Cl2]2, 7.5 mol% of In(OTf)3, 0.2 mmol of K2CO3, 0.2 mL of <sup>t</sup>AmOH/H2O (3:1), 140 °C, 16 h. <sup>b</sup>10 mmol scale. <sup>c</sup>using 5 mol% of [IrCp\*Cl2]2. <sup>d</sup>using 0.6 mmol of 2.

The functional group tolerance was not tested for this protocol separately, since it is safe to assume that all compounds that were successfully converted in the deacylative reaction cascade (protocol B) will also survive this one-step process. The scope of the deacylative reaction variant (B)

with regard to the aromatic carboxylic acid was tested using  $\beta$ -phenyl vinyl ketone (2a) as the coupling partner (Table 3). Benzoic acids bearing electron-donating or electron-withdrawing groups were smoothly coupled to give moderate to good product yields. For meta-substituted benzoic acids, C-H activation occurred selectively at the less hindered ortho-position (4ga-4ja). Multisubstituted benzoic and naphthyl carboxylic acids also gave high yields (**4ka-40a**). When starting with unsubstituted benzoic acid (**1p**), the double hydroarylation product **4pa**' was obtained. Since the carbonyl group of indanones has been shown not to be an effective directing group for C-H activation,<sup>26</sup> the second hydroarylation must have occurred while the carboxylate group was still intact. Thus, one can conclude that the Ir enolate formation must be reversible (see Scheme 2, III  $\rightleftharpoons$  IV) and the hydroarylation step is faster than Claisen condensation. After double hydroarylation of 1p, only one of the ketone side chains can undergo the subsequent Claisen steps leading to 4pa'.

The scope with regard to the  $\alpha$ , $\beta$ -unsaturated ketones was explored using o-toluic acid as coupling partner.  $\beta$ -Arylated ketones bearing electron-rich and electron-poor substituents in their o-, *m*-, or *p*-position reacted similarly well, and various functional groups, including CF<sub>3</sub>, halo, CN, keto, ester, amide, pyridyl, and sulfonyl moieties were tolerated. The new catalyst system gives higher yields for most substrates that can also be converted with the Rhbased catalyst, but interestingly, cyclic alkenones, which are optimal for the Rh protocol,<sup>20</sup> give unsatisfactory results (**4as-4at**).

The deacylative reaction variant (B) is not limited to methyl ketones but can be conducted with various long-chain alkyl, aryl, and even divinyl ketones. At this stage, other types of electron-deficient alkenes, including acid derivatives or aldehydes, did not give the desired product in either protocol (Scheme S6). The reaction was successfully scaled up to 10 mmol for the example of  $\beta$ -phenyl vinyl ketone (**4aa**). The key advantage of this synthetic entry to indanones is that the  $\alpha$ , $\beta$ -unsaturated ketones are easily accessible by aldol condensation of methyl ketones with aldehydes. This condensation step can even be performed in situ, which was demonstrated for the example of indanone **4aa** (Scheme 3).

Scheme 3. One-pot process from benzaldehyde and acetone.



The reaction mechanism was elucidated in a series of experiments (Scheme S1). In the presence of  $[IrCp*Cl_2]_2$ , o-toluic acid was swiftly deuterated in *ortho*-position, indicating rapid and reversible C–H bond cleavage. Deuterium-labelling experiments revealed that the 3-arylindanone  $\alpha$ -hydrogen originates from the solvent (Scheme S2).





Chart 1. Yield-time plots for A) acylindanone protocol; B) deacylative protocol.



No deacylation of pre-formed **3aa** takes place without  $In(OTf)_3$  (Scheme 4a). However, starting from the hydroarylation product **5aa**, Claisen reaction with partial deacylation is promoted even by  $K_2CO_3$ , which explains why strong bases must be avoided when aiming to form product **4aa**. [IrCp\*Cl<sub>2</sub>]<sub>2</sub> and In(OTf)<sub>3</sub> both facilitate this step (Scheme 4b). A control reaction, in which excess **3aa** 

was added, proceeded much more slowly, indicating that **3aa** interferes with the first reaction step, possibly by its chelating coordination to the iridium catalyst (Scheme 4c).

These experiments suggest the presence of two competing pathways for the deacylative process, one proceeding via intermediate **5aa** directly to product **4aa**, the other via formation of **3aa**. The conversion vs time plots in Chart 1 illustrate how the intermediates **5aa** (protocol A) or **3aa** and **4aa** (protocol B) temporarily build up in the non-deacylative and the deacylative reaction protocols, respectively. All findings are in excellent agreement with the proposed mechanism (Scheme 2).

In conclusion, a newly developed Ir-based catalyst system enables the modular construction of 3-substituted indanone moieties from simple benzoic acids, aldehydes, and alkyl ketones. It fulfills the high expectations raised by the prototype Rh-based catalyst generation and leads the hydroarylation/ Claisen approach to synthetic maturity. Two complementary reaction protocols provide selective entries to either 2-acyl indanones or their deacylated derivatives.

#### ASSOCIATED CONTENT

#### **Supporting Information**

The Supporting Information is available free of charge on the ACS Publications website at DOI:

General methods, detailed mechanistic studies, experimental procedures, synthesis and characterization of products, and NMR spectra (PDF).

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#### Notes

The authors declare no competing financial interest.

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