

Efficient Synthesis of 3-Aryl(hetaryl)-1,5,3-dioxazepanes Involving Catalysts Containing Sm and Co

N. N. Makhmudiyarova, K. I. Prokof'ev, L. V. Mudarisova,
A. G. Ibragimov, and U. M. Dzhemilev

Institute of Petrochemistry and Catalysis, Russian Academy of Sciences, Ufa, 450075 Russia
e-mail: ink@anrb.ru

Received July 4, 2012

Abstract—Efficient synthetic procedure was developed for 3-aryl(hetaryl)-1,5,3-dioxazepanes consisting in the transamination of 3-*tert*-butyl-1,5,3-dioxazepane with arylamines and also by the reaction of 1,2-ethanediol with *N,N*-bis(methoxymethyl)aryl(hetaryl)amines in the presence of catalytic amounts of Sm and Co compounds.

DOI: 10.1134/S1070428013050217

Oxazepane derivatives are widely used in the medicine as substances exhibiting analgesic, antipyretic, sedative [1, 2], anticancer [3], and fungicidal [4, 5] action. Quite a number of papers treats the problems of the synthesis of *O,N*-containing five- and six-membered heterocycles [6–8], whereas scarce information exist on the selective synthesis of *N*-substituted 1,5,3-dioxazepanes [9–11]. *N*-Alkyl(cycloalkyl, benzyl)-1,5,3-dioxazepanes are obtained by heating alkyl(cycloalkyl, benzyl)amines with paraformaldehyde and 1,2-ethylene glycol in aromatic solvents [9–11]. To the time of the start of our study no literature data existed on the selective synthesis of 3-aryl(hetaryl)-1,5,3-dioxazepanes.

By the analogy of the previously investigated transamination of 3-methyl-1,3,5-dithiazinane with aniline giving 3-phenyl-1,3,5-dithiazinane [12], we carried out the reaction of primary arylamines with 3-*tert*-butyl-1,5,3-dioxazepane in the presence of catalysts based on transition and rare earth metals presuming the possibility of the replacement of the nitrogen atom in the molecule of 3-*tert*-butyl-1,5,3-dioxazepane by the transamination leading to the formation of the corresponding 3-aryl-1,5,3-dioxazepanes in the presence of catalytic amounts of Sm and Co complexes.

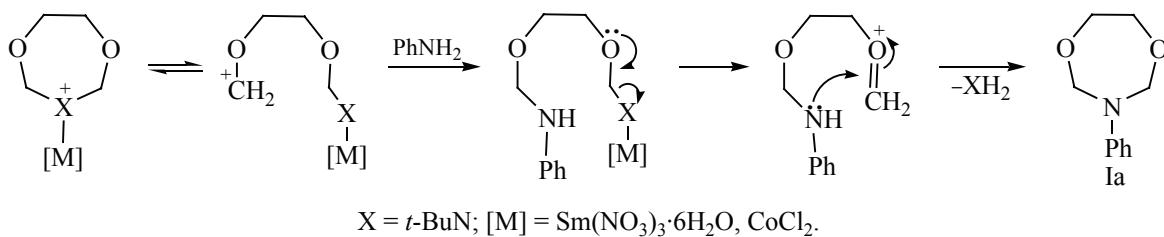
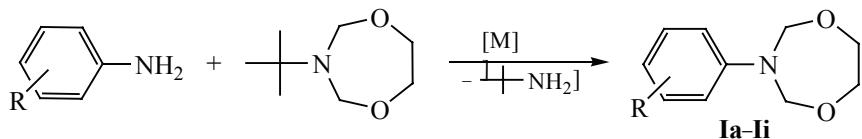
Preliminary experiments have shown that noncatalytic reaction of aniline with an equimolar quantity of 3-*tert*-butyl-1,5,3-dioxazepane (~20°C, 3 h, solvent

chloroform) afforded 3-phenyl-1,5,3-dioxazepane (**Ia**) in less than 8% yield. The yield of compound **Ia** may be increased if the transamination is carried out in the presence of catalysts based on the salts and complexes of Cu, Pd, Co, Zr, Ti, Hf, V, Fe, Sm [12], where the most active are Sm(NO₃)₃·6H₂O and CoCl₂. Under the action of these catalysts (5 mol%, ~20°C, 3 h, solvent CHCl₃) the transamination proceeds highly selectively (~100%) with the formation of *N*-phenyl-1,5,3-dioxazepane (**Ia**) in 70 and 65% yields respectively. The significant acceleration of the reaction is apparently due to the coordination of the nitrogen atom of the molecule of the initial 3-*tert*-butyl-1,5,3-dioxazepane to the central atom of the catalyst [13].

The probable route [14, 15] of the catalytic transamination of 3-*tert*-butyl-1,5,3-dioxazepane under the action of aniline involves a stage of the opening of the initial heterocycle, the nucleophilic addition of aniline to the carbocation and the intramolecular cyclization affording 3-phenyl-1,5,3-dioxazepane (**Ia**) (Scheme 1).

In the ¹H NMR spectrum of compound **Ia** the proton signals appear as narrow singlets in the region of 4.80 (H^{2,4}) and 3.08 ppm (H^{6,7}) indicating the fast ring inversion in the NMR time scale. In the ¹³C NMR spectrum the dioxazepane carbon atoms give rise to signals δ 54.92 and 35.78 ppm.

The structure of the initial arylamine does not considerably affect the yield of compound **I**. In the reaction of

Scheme 1.**Scheme 2.**

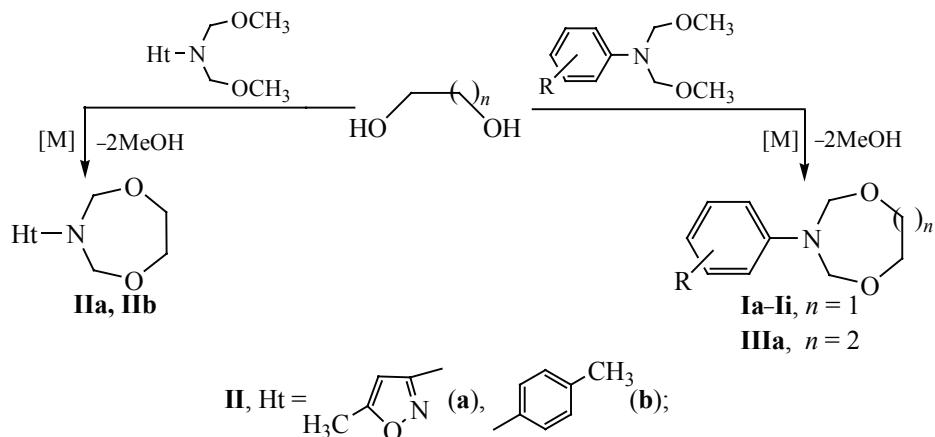
$\text{R} = \text{H}$ (**a**), *m*-CH₃ (**b**), *p*-CH₃ (**c**), *o*-OCH₃ (**d**), *m*-OCH₃ (**e**), *p*-OCH₃ (**f**), *o*-NO₂ (**g**), *m*-NO₂ (**h**), *p*-NO₂ (**i**); $[\text{M}] = \text{Sm}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}, \text{CoCl}_2$.

o-, *m*-, and *p*-nitro-, *o*-, *m*-, and *p*-methoxy-, and also *m*- and *p*-methylanilines with equimolar quantity of 3-*tert*-butyl-1,5,3-dioxazepane (5 mol% of Sm(NO_3)₃·6H₂O, ~20°C, 3 h) the corresponding 3-aryl-1,5,3-dioxazepanes **Ia–Ii** were formed in 60–70% yields (Scheme 2).

The attempt to transaminate the 3-*tert*-butyl-1,5,3-dioxazepane with hetaryl amines (3-amino-5-methylisoxazole, 2-amino-, 3-amino-, 2-amino-4-methyl-, 2-amino-5-methylpyridine) failed.

To develop an efficient synthetic procedure for 3-hetaryl-1,5,3-dioxazepanes we examined the reaction of 1,2-ethanediol with *N,N*-bis(methoxymethyl)-*N*-hetaryl amines. Whereas the aminomethylation of

C,H- and S,H-acids with alkoxymethylamines results in acyclic amines and aminosulfides due to the formation of C–C and C–S bonds [16], it is expectable that the aminomethylation of 1,2-ethanediol with *N,N*-bis(alkoxymethyl)amines would lead to the corresponding dioxazepanes by the simultaneous formation of two C–O bonds. The performed experiments showed that under the chosen conditions [5 mol% of Sm(NO_3)₃·6H₂O, 20°C, 0.5 h] only *N,N*-bis(methoxymethyl)-*N*-hetaryl amines prepared from 3-amino-5-methylisoxazole and 2-amino-5-methylpyridine readily reacted with 1,2-ethanediol to give selectively 3-hetaryl-1,5,3-dioxazepanes **IIa, IIb** in 60 and 64% yields (Scheme 3). Compounds **IIa, IIb** are

Scheme 3.

$\text{R} = \text{H}$ (**a**), *m*-CH₃ (**b**), *p*-CH₃ (**c**), *o*-OCH₃ (**d**), *m*-OCH₃ (**e**), *p*-OCH₃ (**f**), *o*-NO₂ (**g**), *m*-NO₂ (**h**), *p*-NO₂ (**i**); $[\text{M}] = \text{Sm}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ for **Ia–Ii, IIa, IIb**, zeolites 0.96 HY-BS for **IIIa**.

not formed without catalyst.

The developed procedure of the cycloaminomethylation of ethylene glycol with *N,N*-bis(methoxy-methyl) hetaryl amines made it possible to extend this reaction to *N,N*-bis(methoxymethyl)-*N*-arylamines (aryl = Ph, *m*- and *p*-methyl-, *o*-, *m*-, and *p*-methoxy-, *o*-, *m*-, and *p*-nitrophenyl) and to obtain *N*-aryl-1,5,3-dioxazepanes **Ia–Ii** in 80–89% yields (Scheme 3). We attempted to try this procedure for a selective synthesis of dioxazocanes and dioxazonanes by the reaction of *N,N*-bis(methoxymethyl)-*N*-arylamines with 1,3-propanediol and 1,4-butanediol, and only in the presence of zeolites of the grade 0.96HY-BS (5 wt%) *N,N*-bis(methoxymethyl)-*N*-phenylamine reacted with 1,3-propanediol in CHCl₃ at ~60°C furnishing 3-phenyl-1,5,3-dioxazocane (**IIIa**) in 40% yield (Scheme 3).

Thus our research resulted in the development of an efficient synthesis of 3-aryl(hetaryl)-1,5,3-dioxazepanes by transamination of 3-*tert*-butyl-1,5,3-dioxazepane with aryl amines and also by cycloaminomethylation of 1,2-ethanediol with *N,N*-bis(methoxy-methyl) aryl(hetaryl)amines under mild conditions in high yield and selectivity in the presence of catalysts based on Sm and Co compounds.

EXPERIMENTAL

The reaction progress was monitored by TLC on Silufol W-254 plates, eluent petroleum–ether EtOAc–CHCl₃, 5 : 1. The reaction products were analyzed by HPLC on a chromatograph Altex-330 (USA) equipped with a UV detector at the wavelength 340 nm. ¹H and ¹³C NMR spectra were registered on a spectrometer Bruker Avance 400 (100.62 and 400.13 MHz respectively), solvent CDCl₃. GC-MS measurements were carried out on instruments Finnigan-4021 (glass capillary column 5000 × 0.25 mm, stationary phase HP-5, carrier gas helium, ramp from 50 to 300°C at a rate 5 deg/min, vaporizer temperature 280°C, ion source temperature 250°C, 70 eV) and Shimadzu QP-2010 Plus (capillary column Supelco PTE-5 30 m × 0.25 mm). For column chromatography silica gel KSK (100–200 μm) was utilized.

3-Aryl-1,5,3-dioxazepanes. a. Transamination of 3-*tert*-butyl-1,5,3-dioxazepane with aryl amines. Into a Schlenk vessel placed on a magnetic stirrer was charged in an argon flow 10 mmol of 3-*tert*-butyl-1,5,3-dioxazepane [9] and 0.5 mmol of catalyst Sm(NO₃)₃·6H₂O or CoCl₂, the mixture was stirred for 30 min, 5 ml of chloroform

and 10 mmol of an appropriate arylamine was added, and the stirring was continued for 3 h at room temperature (~20°C).

b. Reaction of *N,N*-bis(methoxymethyl)-*N*-arylamines with 1,2-ethanediol. Into a Schlenk vessel placed on a magnetic stirrer was charged in an argon flow 10 mmol of *N,N*-bis(methoxymethyl)-*N*-aryl-amine [17] and 10 mmol of 1,2-ethanediol at room temperature (~20°C), 5 ml of chloroform, 0.5 mmol of catalyst Sm(NO₃)₃·6H₂O was added, the mixture was stirred for 30 min.

3-Phenyl-1,5,3-dioxazepane (Ia). Yield 70% (*a*), 80% (*b*), mp 177–178°C. ¹H NMR spectrum, δ, ppm: 3.08 s (4H, CH₂), 4.80 s (4H, CH₂), 6.91–7.35 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 35.78 (C^{6,7}), 54.92 (C^{2,4}), 115.93 (C^{9,13}), 119.95 (C¹¹), 129.30 (C^{10,12}), 145.83 (C⁸). Mass spectrum, *m/z* (*I*_{rel}, %): 179 [M]⁺ (100), 159 [C₉H₁₃NO]⁺ (30), 104 [C₄H₁₀O₂]⁺ (15), 92 [C₆H₆N]⁺ (45), 77 [C₆H₅]⁺ (95).

3-(3-Methylphenyl)-1,5,3-dioxazepane (Ib). Yield 64% (*a*), 61% (*b*), mp 173–174°C. ¹H NMR spectrum, δ, ppm: 2.65 s (3H, CH₃), 3.88 s (4H, CH₂), 4.89 s (4H, CH₂), 6.70–7.28 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 21.33 (C¹⁴), 62.29 (C^{6,7}), 82.90 (C^{2,4}), 108.98 (C¹³), 113.66 (C⁹), 120.56 (C¹¹), 129.97 (C¹⁰), 147.48 (C¹²), 148.93 (C⁸). Mass spectrum, *m/z* (*I*_{rel}, %): 193 [M]⁺ (40), 180 [C₁₀H₁₅NO₂]⁺ (15), 136 [C₈H₁₀NO]⁺ (37), 106 [C₇H₈N]⁺ (100), 104 [C₄H₁₀O₂]⁺ (15), 92 [C₆H₇]⁺ (32), 77 [C₆H₅]⁺ (54).

3-(4-Methylphenyl)-1,5,3-dioxazepane (Ic). Yield 60% (*a*), 65% (*b*), mp 138–139°C. ¹H NMR spectrum, δ, ppm: 2.36 s (3H, CH₃), 3.87 s (4H, CH₂), 4.44 s (4H, CH₂), 6.75–7.23 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 21.34 (C¹⁴), 67.34 (C^{6,7}), 83.22 (C^{2,4}), 114.22 (C^{9,13}), 120.18 (C^{10,12}), 141.46 (C¹¹), 147.55 (C⁸).

3-(2-Methoxyphenyl)-1,5,3-dioxazepane (Id). Yield 63% (*a*), mp 118–119°C. ¹H NMR spectrum, δ, ppm: 3.53–3.79 m (7H, CH₂, CH₃), 4.94 s (4H, CH₂), 6.37–7.23 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 55.23 (C¹⁵), 68.77 (C^{6,7}), 80.88 (C^{2,4}), 102.01 (C¹⁰), 104.98 (C¹³), 109.21 (C¹²), 130.06 (C¹¹), 146.93 (C⁸), 160.61 (C⁹). Mass spectrum, *m/z* (*I*_{rel}, %): 209 [M]⁺ (5), 196 [C₁₀H₁₄NO₃]⁺ (5), 166 [C₉H₁₂NO₂]⁺ (21), 122 [C₇H₈NO]⁺ (31), 107 [C₇H₁₄NO]⁺ (23), 104 [C₄H₁₀O₂]⁺ (100), 77 [C₆H₅]⁺ (58).

3-(3-Methoxyphenyl)-1,5,3-dioxazepane (Ie). Yield 67 (*a*), 81% (*b*), mp 122–124°C. ¹H NMR spectrum, δ,

ppm: 3.47–3.82 m (7H, CH₂, CH₃), 4.87 s (4H, CH₂), 6.28–7.15 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 55.14 (C¹⁵), 69.40 (C^{6,7}), 80.66 (C^{2,4}), 102.18 (C⁹), 105.46 (C¹¹), 108.33 (C¹³), 129.78 (C¹²), 147.80 (C⁸), 158.26 (C¹⁰).

3-(4-Methoxyphenyl)-1,5,3-dioxazepane (If). Yield 62% (a), 80% (b), mp 129–130°C. ¹H NMR spectrum, δ, ppm: 3.43–3.70 m (7H, CH₂, CH₃), 4.87 s (4H, CH₂), 6.40–7.28 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 59.89 (C¹⁵), 73.83 (C^{6,7}), 81.07 (C^{2,4}), 112.66 (C^{10,12}), 129.92 (C^{9,13}), 146.74 (C⁸), 160.69 (C¹¹).

3-(2-Nitrophenyl)-1,5,3-dioxazepane (Ig). Yield 64% (a), 60% (b), mp 140–143°C. ¹H NMR spectrum, δ, ppm: 3.69 m (4H, CH₂), 5.27 s (4H, CH₂), 7.31–7.83 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 68.27 (C^{6,7}), 79.30 (C^{2,4}), 109.10 (C¹⁰), 114.07 (C¹³), 120.56 (C¹²), 129.75 (C¹¹), 147.02 (C⁸), 149.09 (C⁹). Mass spectrum, *m/z* (*I*_{rel}, %): 224 [M]⁺ (60), 195 [C₉H₁₁N₂O₃]⁺ (50), 151 [C₇H₇N₂O₂]⁺ (100), 137 [C₆H₅N₂O₂]⁺ (15), 104 [C₄H₁₀O₂]⁺ (35), 77 [C₆H₅]⁺ (23).

3-(3-Nitrophenyl)-1,5,3-dioxazepane (Ih). Yield 69% (a), 77% (b), mp 133–134°C. ¹H NMR spectrum, δ, ppm: 3.78 m (4H, CH₂), 5.04 s (4H, CH₂), 7.24–7.90 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 69.52 (C^{6,7}), 80.16 (C^{2,4}), 109.95 (C⁹), 114.35 (C¹¹), 121.17 (C¹³), 129.02 (C¹²), 147.48 (C⁸), 148.51 (C¹⁰).

3-(4-Nitrophenyl)-1,5,3-dioxazepane (Ii). Yield 66% (a), 59% (b), mp 156–157°C. ¹H NMR spectrum, δ, ppm: 3.54 m (4H, CH₂), 4.97 s (4H, CH₂), 7.38–7.90 m (4H, CH). ¹³C NMR spectrum, δ, ppm: 71.05 (C^{6,7}), 82.62 (C^{2,4}), 109.60 (C^{10,12}), 128.93 (C^{9,13}), 146.98 (C⁸), 149.20 (C⁹).

3-(5-Methylisoxazol-3-yl)-1,5,3-dioxazepane (IIa). Yield 60% (b). ¹H NMR spectrum, δ, ppm: 2.14 m (3H, CH₃), 3.54 d (4H, CH₂, *J* 4 Hz), 5.43 s (4H, CH₂), 7.36 s (1H, CH). ¹³C NMR spectrum, δ, ppm: 15.13 (C¹³), 68.25 (C^{6,7}), 86.30 (C^{2,4}), 89.17 (C¹²), 140.72 (C⁸), 165.60 (C¹¹).

3-(5-Methylpyridin-2-yl)-1,5,3-dioxazepane (IIb). Yield 64% (b). ¹H NMR spectrum, δ, ppm: 2.18 s (3H, CH₃), 3.22 d (4H, CH₂, *J* 4.4 Hz), 5.35 s (4H, CH₂), 7.16–7.20 m (3H, CH). ¹³C NMR spectrum, δ, ppm: 20.17 (C¹⁴), 71.03 (C^{6,7}), 87.15 (C^{2,4}), 101.30 (C¹³), 119.20 (C¹¹), 128.26 (C¹²), 140.40 (C¹⁰), 151.91 (C⁸).

ACKNOWLEDGMENTS

The study was carried out under the financial support of the Russian Foundation for Basic Research (grants nos. 11-03-00101-a, 11-03-97011-p_Volga region_a).

REFERENCES

- Schmidt, G., US Patent 3546214, 1970.
- Nagarajan, K. and Shah, R.K., *Ind. J. Exp. Biol.*, 1974, vol. 12, pp. 217, 229.
- Grandolini, G., Perioli, L., and Ambrogi, V., *Eur. J. Med. Chem.*, 1999, vol. 34, p. 701.
- Levy, O., Erez, M., Varon, D., and Keinan, E., *Bioorg. Med. Chem. Lett.*, 2001, vol. 11, p. 2921.
- Serrano-Wu, M.H., St. Laurent, D.R., Chen, Y., Huang, S., Lam, K.R., Matson, J.A., Mazzucco, C.E., Stickle, T.M., Tully, T.P., Wong, H.S., Vyas, D.M., and Balasubramanian, B.N., *Bioorg. Med. Chem. Lett.*, 2002, vol. 12, p. 2757.
- Ivanskii, V.I., *Khimiya heterocycleicheskikh soedinenii* (Chemistry of Heterocyclic Compounds), Moscow: Vysshaya Shkola, 1978, vol. 185, p. 371.
- Gilchrist, T.L., *Heterocyclic Chemistry*, London: Marshfield, Mass.: Pitman, 1985.
- Eicher, T. and Hauptmann, S., *The Chemistry of Heterocycles*, Weinheim: Wiley VCH, 2003.
- Kapnang, H. and Charles, G., *Tetrahedron Lett.*, 1980, vol. 21, p. 2949.
- Shibata, Y. and Imagawa, T., US Patent 2009096744, 2009.
- Matyushov, V.F. and Gritsenko, T.M., *Khim. Geterotsikl. Soedin.*, 1971, p. 25.
- Niatshina, Z.T., Murzakova, N.N., Vasilieva, I.V., Rakhimova, E.B., Akhmetova, V.R., and Ibragimov, A.G., *Arkivoc.*, 2011, vol. 8, p. 141.
- Kukushkin, Yu.N., *Reaktsionnaya sposobnost' koordinatsionnykh soedinenii* (Reactivity of Coordination Compounds), Leningrad: Khimiya, 1987, p. 228.
- Mokrov, G.V., Likhosherstov, A.M., Lezina, V.P., Gudasheva, T.A., Bushmarinov, I.S., and Antipin, M.Yu., *Izv. Akad. Nauk, Ser. Khim.*, 2010, p. 1228.
- Krohn, K. and Cladius-Brandt, S., *Synthesis*, 2010, p. 1344.
- Shikaliev, Sh.M., *Azerb. Khim. Zh.*, 2010, no. 1, p. 205.
- Borch, R.F. and Hassid, A.I., *J. Org. Chem.*, 1972, vol. 37, p. 1673.