

Synthesis of Enantiomerically Pure Δ^2 -Isoxazolines[†] via Sulphinyl Derivatives

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Optically active 3-sulphinylmethyl- Δ^2 -isoxazolines[†] allow an easy entry to enantiomerically pure Δ^2 -isoxazolines and to the corresponding β -hydroxy ketones.

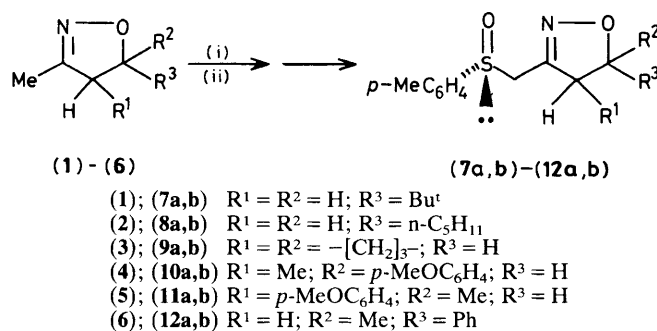
The control of relative and absolute stereochemistry in the construction of a sequence of asymmetrically substituted carbon atoms in acyclic molecules is still a challenging topic in synthesis.¹

In this context Δ^2 -isoxazolines[†] have recently received a great deal of interest. These compounds, which can be obtained by regioselective and stereocontrolled cycloadditions of nitrile oxides to olefins,² were shown to represent an easy entry to diastereoisomerically homogeneous β -ketols³ and γ -amino alcohols.⁴ Therefore, the development of a successful strategy for the synthesis of optically active Δ^2 -isoxazolines was of interest to us.[‡]

We report here that diastereoisomerically and/or enantiomerically pure Δ^2 -isoxazolines can be easily prepared via a new class of sulphinyl derivatives *i.e.* 3-(tolylsulphinylmethyl)- Δ^2 -isoxazolines.

Indeed, *exo*-metallation⁶ of racemic 3-methyl- Δ^2 -isoxazolines⁷ (1)–(6) and subsequent reaction with (–)-(*S*)-menthyl toluene-*p*-sulphinate affords in excellent yields compounds (7a,b)–(12a,b) as mixtures of diastereoisomers (Scheme 1).

The extent of chiral discrimination in this reaction, not unexpectedly, is low, ranging from 8 to 20%. However, the



Scheme 1. Reagents: (i), lithium di-isopropylamide (LDA); (ii), menthyl toluene-*p*-sulphinate.

individual stereoisomers can be separated by gravity or flash chromatography and the diastereoisomeric purity at all their stereocentres easily checked by ¹H n.m.r. spectroscopy. Yields, diastereoisomeric ratios, melting points, and optical rotations of compounds (7)–(12) are reported in Table 1.

The relative stereochemistry at C-4 and C-5 of the isoxazoline ring is pre-determined by the configuration of the olefin which undergoes the cycloaddition. The absolute configuration of the sulfoxide in (7)–(12) can be inferred as (*R*) from that of the starting sulphinate ester, as commonly accepted for a number of related Andersen-type syntheses which are known to proceed with complete inversion of chirality at sulphur.⁸ Therefore our reaction generates only two dia-

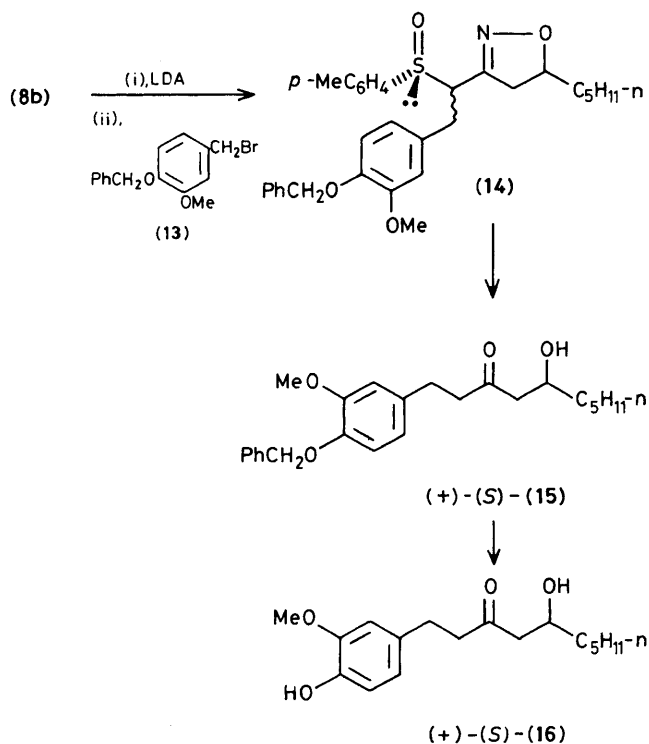
[†] Now 4,5-dihydroisoxazolines, see *Pure Appl. Chem.*, 1983, **55**, 409.

[‡] During the completion of this work an application of chiral isoxazolines to the asymmetric synthesis of β -hydroxy acids was published.⁵

Table 1. Synthesis of 3-(tolylsulphinylmethyl)- Δ^2 -isoxazolines (**7a,b**)—(**12a,b**).^a

Compound	Yields (%)	Diastereoisomeric ^b ratios a : b	Diastereoisomer a ^c		Diastereoisomer b ^c	
			$[\alpha]_D^{23}$ ^d	M.p./°C	$[\alpha]_D^{23}$ ^d	M.p./°C
(7a,b)	67	46 : 54	+337.5	94—96	+140.3	79—81
(8a,b)	80	45 : 55	+297.5	70—72	+134.4	63—65
(9a,b) ^e	84	40 : 60	+279.0	78—81	+79.8	82—84
(10a,b) ^f	75	55 : 45	+328.3	130—131	— ^g	—
(11a,b) ^f	76	46 : 54	+83.2	118—120	+262.2	89—91
(12a,b)	95	42 : 58	+263.8	102—104	+167.5	100—102

^a Reaction carried at -90°C under argon in tetrahydrofuran with 2 mol. equiv. of metallated (**1**)—(**6**) and 1 mol. equiv. of sulphinate ester. All new compounds gave analytical and spectral data in agreement with the proposed structures. ^b As determined by ^1H n.m.r. spectroscopy. ^c Diastereoisomer **a** is the one eluted first and diastereoisomer **b** is the one eluted second in column chromatography. ^d *c* 1 in CHCl_3 . ^e *cis* relative stereochemistry at C-4 and C-5 of the isoxazoline ring. ^f *trans* relative stereochemistry at C-4 and C-5 of the isoxazoline ring. ^g Compound (**10b**) could not be isolated free of (**10a**).



stereoisomers, the separation of which yields enantiomerically pure compounds. Conversion of compounds (**7**)—(**12**) into optically pure Δ^2 -isoxazolines is cleanly performed in nearly quantitative yields by reductive desulphurization with Na—Hg in dry methanol in the presence of NaH_2PO_4 .[§]

Using this method, (**10a**), the dextrorotatory enantiomer of (**4**), $[\alpha]_D^{23} + 208.2^\circ$ (*c* 1 in CHCl_3), was obtained in 90% yield. Analogously, both enantiomers of (**5**) were synthesized from (**11a**) and (**11b**): they had $[\alpha]_D^{23} - 269.5^\circ$ (*c* 0.2 in CHCl_3) and $+270.6^\circ$ (*c* 0.25 in CHCl_3), respectively. It should be noted that neither epimerization on the C-4 position of the isoxazoline nor reductive opening of the N—O bond¹⁰ was observed under these conditions.

[§] 1.5 g of 8% Na—Hg, 1.2 g of dry NaH_2PO_4 , and 15 ml of methanol per 1.0 mmol of substrate at 0°C ; reaction time 30 min.

[¶] Enantiomerically pure (**4**) can be a useful intermediate for the synthesis of nikkomycin B.⁹

Finally we note that both desulphurization and unmasking of the ketol moiety embedded in the heterocyclic ring could be simultaneously performed by Raney nickel-catalysed reactions of (**7**)—(**12**). This allowed the direct conversion (80% yield) of compound (**12b**) into 4-hydroxy-4-phenylpentan-2-one,¹¹ $[\alpha]_D^{23} - 5.1^\circ$; $[\alpha]_{365}^{23} + 37.8^\circ$ (*c* 1 in CHCl_3). This same compound was obtained via Na—Hg desulphurization (98% yield) of isoxazoline (–)-(6), $[\alpha]_D^{23} - 40.0^\circ$ (*c* 1 in CHCl_3) and subsequent Raney nickel-catalysed ring opening.³

(+)-(S)-Gingerol (**16**)¹² was synthesised in three steps starting from (**8b**) using this method. Alkylation of (**8b**) with the bromide (**13**) in the presence of hexamethylphosphoramide yielded sulphoxide (**14**) in 80% yield as a 6 : 4 mixture of epimers at the carbon α to the sulphinyl group. This mixture was converted (94% yield) by Raney nickel-catalysed hydrogenation in acidic medium³ into (+)-(S)-(15), m.p. 70°C , $[\alpha]_D^{23} + 19.6^\circ$ (*c* 0.4 in CHCl_3), enantiomeric excess (e.e.) $>96\%$.¹² This compound can then be de-benzylated to give (+)-(S)-(16), $[\alpha]_D^{23} + 25.7^\circ$ (*c* 1 in CHCl_3), with unchanged e.e.* as described elsewhere.¹²

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* Maximum reported rotation $[\alpha]_D^{24} + 25.1^\circ$ (*c* 1 in CHCl_3).¹²