

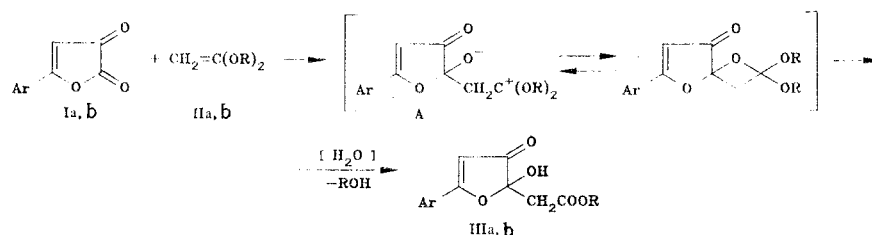
REACTION OF 5-ARYL-2,3-DIHYDROFURAN-2,3-DIONES
WITH KETENE ACETALS

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It is known that ketene acetals react with N-substituted isatins to give [2 + 2]-cycloadducts involving the ketone carbonyl group [1]. Other five-membered 2,3-dioxo heterocycles have not been previously investigated in this sort of reaction.

We have established that 5-aryl-2,3-dihydrofuran-2,3-diones Ia, b react with ketene diethylacetal (IIa) and ketene di-n-butylacetal (IIb) to give alkyl 5-aryl-2-hydroxy-3-oxo-2,3-dihydrofuran-2-ylethanoates IIIa, b.



I, III a Ar=C₆H₅, b Ar=*p*-CH₃C₆H₄; a R=C₂H₅, b R=*n*-C₄H₉

A possible scheme for the formation of esters III includes attack on the lactone carbonyl group of I by ketene acetal II, the formation of a zwitter ion (A) or a spiro compound (B), which is characteristic for aldehydes or ketones [2, 3], and their hydrolysis under the experimental conditions.

Ethyl 2-Hydroxy-3-oxo-5-phenyl-2,3-dihydrofuran-2-ylethanoate (IIIa). A mixture of 1.74 g (0.01 mole) of furandione Ia and 1.74 g (0.015 mole) of acetal IIa was refluxed for 1.5 h in 30 ml of methylene chloride, after which the solvent was evaporated, and the residue was recrystallized from acetonitrile to give 1.70 g (65%) of a product with mp 112-114°C. IR spectrum: 3120 (O—H), 1730 (COOC₂H₅), 1683 cm⁻¹ [C₍₃₎=O]. PMR spectrum (CDCl₃): 1.22 (3H, t, CH₃CH₂O); 2.63, 3.05 (2H, dd, J ≈ 16 Hz, CH₂); 4.18 (2H, q, CH₃CH₂O); 5.88 (1H, s, 4-H); 6.28 (1H, broad s, OH); 7.52 ppm (5H, m, C₆H₅). Mass spectrum, m/z (I, %): 262 (18) [M]⁺, 234 (36) [M - CO]⁺, 217 (24) [M - OC₂H₅]⁺, 189 (33) [M - COOC₂H₅]⁺, 147 (42) [C₆H₅COCH₂CO]⁺, 120 (3) [C₆H₅COCH₃]⁺, 115 (50) [OC—CH₂COOC₂H₅]⁺, 105 (96) [C₆H₅CO]⁺, 102 (100) [C₆H₅C≡CH]⁺, 87 (70) [CH₂COOC₂H₅]⁺.

n-Butyl 2-Hydroxy-3-oxo-5-(*p*-tolyl)-2,3-dihydrofuran-2-ylethanoate (IIIb). This compound was similarly obtained in 30 ml of chloroform and had mp 120-122°C. IR spectrum: 3110 (O—H), 1732 (COOC₄H₉), 1670 cm⁻¹ [C₍₃₎=O]. PMR spectrum (CDCl₃): 0.85 (3H, t, CH₃CH₂CH₂CH₂O); 1.42 (4H, m, CH₃CH₂CH₂CH₂O); 2.32 (3H, s, CH₃); 2.55, 2.98 (2H, dd, J ≈ 16 Hz, CH₂); 4.05 (2H, t, CH₃CH₂CH₂CH₂O); 5.78 (1H, s, 4-H); 6.22 (1H, broad s, OH); 7.30 ppm (4H, q, C₆H₄). The yield was 60%.

The results of elementary analysis were in agreement with the calculated values.

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