

Synthesis of Some Naturally-occurring Styrylamides

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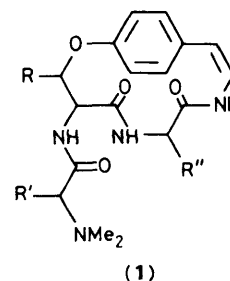
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Thermal elimination from *N*-(β -phenyl- β -phenylsulphinylolethyl)amides yielded styrylamides (2)–(4)

In connection with our programme of synthesis of cyclic peptides based on the route developed by Rapoport,¹ we needed a simple, efficient modification to permit introduction of the styrylamide group in such peptides, (1).² To explore such a synthetic goal, we selected as targets three naturally-occurring styrylamides (2)–(4), isolated from *Pleiospermium alatum*. (Wight and Arn) Swingle,³ *Aegle marmelos* Corr.,⁴ and *Amyris plumeri* D.C.,^{5a} respectively. Several new styrylamides, including Tunichrome B-1, a V-complexing agent,⁶ and Amathamide A and B, from a marine Bryozoa,⁷ prompt us to report one successful route to the styrylamide group (Scheme 1).

Beginning with the appropriate aldehyde (5), the thio-phenol present in the alkaline solution of (5) and nitro-methane added to the intermediate β -nitrostyrene to give (6) in a one-pot sequence. Following reduction of (6) to (7), acylation with the appropriate acid derivative gave the three

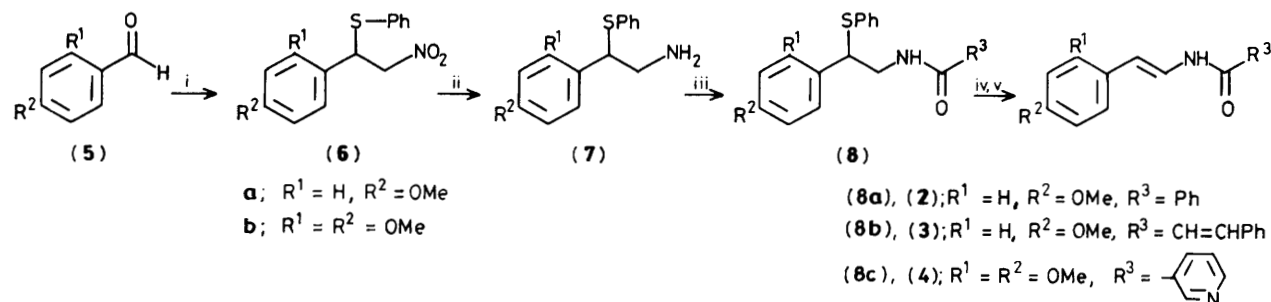
amides (8). Oxidation of (8a–c) with *m*-chloroperbenzoic acid (MCPBA) at -50°C gave sulfoxides that under reflux



Nummularine-M; R = Ph, R' = R'' = Bu^s (ref. 2a)

Frangulanin; R = Prⁱ, R' = Bu^s, R'' = Buⁱ (ref. 2c)

Sativanine-A; R = Ph, R' = Bu^s, R'' = Prⁱ (ref. 2d)



Scheme 1. Reagents and conditions: i, MeNO₂, MeNH₂Cl-Na₂CO₃, PhSH, 25 °C, 3 days [(6a) 86%, (6b) pure oil, t.l.c. quantitative]; ii, Zn, HCl-HOAc, heat, 2 h (pure oils, t.l.c., 84–90%); iii, CH₂Cl₂, a: (7a), PhCOCl-NaHCO₃, 25 °C (83% MeOH), b: (7a), (PhCH=CHCO)₂O, NaHCO₃, 25 °C (66% MeOH), c: (7b), 3-pyCOOCOEt (from 3-pyCO₂H, ClCO₂Et, Et₃N), 0 °C (82% crude, glass solid); iv, MCPBA-CH₂Cl₂-Et₂O, -50 to 0 °C, 6 h (quantitative, pure by t.l.c., solids); v, toluene, CaCO₃, reflux, 5 h [(2) and (3) 65% from CHCl₃; (4) 29%, recrystallised poorly from C₆H₆].

yielded (2)–(4) in 20 [for (4)]–40% overall yields.†‡ Presumably then, ring closure of appropriate substituents at

R² and R³ of (8) followed by oxidation–thermal elimination as above would lead to the desired cyclic peptide having the styrylamide function.

† The alkene of the styrylamide in Zizyphin-A, a cyclic peptide, was introduced *via* oxidative elimination on a selenide formed by substitution; U. Schmidt, A. Lieferknecht, H. Bokens, and H. Griesser, *Angew. Chem.*, 1981, **93**, 1121.

‡ Satisfactory C, H analysis were obtained for (2)–(4).

(2) M.p. 192–193 °C (lit.³ m.p. 178–180 °C); i.r. (KBr) 3325, 1655 cm⁻¹; u.v. (EtOH) 221 (log ε 4.21), 311 (4.39), 317 (4.39) nm; *m/z* 253 (*M*⁺); ¹H n.m.r. (200 MHz) [(CD₃)₂CO] δ 3.79 (3H, s, OMe), 6.42 (1H, d, *J* 15 Hz, =CH), 6.89 (2H, d, *J* 8.8 Hz, *o*-HArOMe), 7.35 (2H, d, *J* 8.8 Hz, *m*-HArOMe), 7.52 (3H, m, ArH), 7.58 (1H, d, *J* 15 Hz, =CHN), 8.00 (2H, dd, *J* 7.7, 1.8 Hz, *o*-HArCO).

(3) M.p. 206.5–207.5 °C (lit.⁴ m.p. 191 °C); i.r. 3462, 1640 cm⁻¹; u.v. 284 (log ε 4.47), 300 (4.49), 338 (4.52) nm; *m/z* 279 (*M*⁺); ¹H n.m.r. [(CD₃)₂CO] δ 3.78 (3H, s, OMe), 6.25 (1H, d, *J* 14.7 Hz, =CHArOMe), 6.74 (1H, d, *J* 15.4 Hz, =CHCO-), 6.88 (2H, d, *J* 8.8 Hz, *o*-HArOMe), 7.32 (2H, d, *J* 8.8 Hz, *m*-HArOMe), 7.45 (3H, m, ArH), 7.55 (1H, d, *J* 14.7 Hz, =CHN), 7.61 (2H, m, ArH), 7.65 (1H, d, *J* 15.4 Hz, =CHAr).

(4) M.p. 160–161.5 °C (lit.⁵ m.p. 159–160 °C); i.r. 3250, 1655 cm⁻¹; u.v. 220 (log ε 4.40), 262 (4.20) 331 (4.38) nm; *m/z* 284 (*M*⁺); ¹H n.m.r. (CDCl₃) δ 3.82 (3H, s, OMe), 3.85 (3H, s, OMe), 6.46 [1H, m, 3-HAr(OMe)₂], 6.49 [1H, dd, *J* 8, 2 Hz, 5-HAr(OMe)₂], 6.51 [1H, d, *J* 14.7 Hz, =CHAr(OMe)₂], 7.33 [1H, d, *J* 8 Hz, 6-HAr(OMe)₂], 7.43 [1H, dd, *J* 8, 7.8 Hz, 5H-3 pyridyl (py)], 7.67 (1H, dd, *J* 14.7, 14.3 Hz, =CHN-), 8.06 (1H, br. d, NH), 8.19 (1H, dt, *J* 7.8 Hz, 4H-3 py), 8.76 (1H, d, *J* 4.3 Hz, 6H-3 py), 9.06 (1H, br. s, 2H-3 py); shaking with D₂O removes the signal at δ 8.06 and reduces δ 7.67 to a doublet.

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References

- 1 J. C. Lagarias, R. A. Houghten, and H. Rapoport, *J. Am. Chem. Soc.*, 1978, **100**, 8202.
- 2 (a) V. B. Pandey, J. P. Singh, K. K. Seth, A. H. Shah, and G. Eckhardt, *Phytochemistry*, 1984, **23**, 2118; (b) A. H. Shah, V. B. Pandey, J. P. Singh, K. N. Singh, and G. Eckhardt, *ibid.*, p. 2120; (c) E. Haslinger and W. Robien, *Monaish. Chem.*, 1982, **113**, 95; (d) R. Tschesche, A. H. Shah, and G. Eckhardt, *Phytochemistry*, 1979, **18**, 702.
- 3 A. Chatterjee, M. Chakrabarty, and A. B. Kundu, *Aust. J. Chem.*, 1975, **28**, 457.
- 4 T. R. Govindachari and M. S. Premila, *Phytochemistry*, 1983, **22**, 755.
- 5 (a) B. A. Burke and H. Parkins, *Tetrahedron Lett.*, 1978, 2723; (b) the *cis* isomer of (4) has been isolated: B. A. Burke and S. Philip, *Heterocycles*, 1985, **23**, 257.
- 6 R. C. Bruening, E. M. Oltz, J. Furukawa, K. Nakanishi, and K. Kustin, *J. Am. Chem. Soc.*, 1985, **107**, 5298.
- 7 A. J. Blackman and D. J. Matthews, *Heterocycles*, 1985, **23**, 2829.