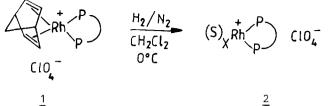
RHODTUM-DIPHOSPHINE COMPLEXES AS CATALYSTS IN ALDOL ADDITIONS

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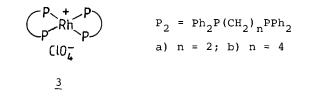
Summary: Rhodium(I)-diphosphine complexes catalyze the aldol addition of enolsilanes to aldehydes, an enantioselective version being possible if chiral phosphines are used.

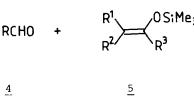
The recent report by Sato, Matsuda and Izumi¹⁾ concerning the first example of rhodium-catalyzed crossed aldol additions of enolsilanes to aldehydes prompts us to report our results in this area. Using known procedures $^{2-4)}$, we prepared the complexes 1, 2^{5} and 3 and discovered that they catalyze the aldol addition of enolsilanes 5 to aldehydes 4 in dichloromethane (Table 1).

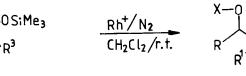


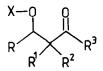
 $P_2 = Ph_2P(CH_2)_nPPh_2$ S = Solvent

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a)	x=SiMe ₃
	x=H

Table 1. Rh(I)-catalyzed aldol addition at 22°C in CH₂Cl₂

Entry	R	R ¹	R ²	R ³	cata (moi	alyst ^a 1-%)	reaction time (h)	isol. yield ^b (%)
1	Ph	CH3	CH3	OMe	<u>1a</u>	(5)	2	75
2	Ph	СНЗ	СНЗ	OMe	<u>2a</u>	(3)	18	63
3	Ph	СНЗ	CH3	OMe	<u>3a</u>	(10)	18	57
4	^{n-C} 6 ^H 13	СНЗ	СНЗ	OMe	<u>1a</u>	(10)	6	71
5	Ph	СНЗ	СНЗ	OMe	<u>1b</u>	(5)	2	81
6	Ph	СН3	CH ₃	OMe	<u>2b</u>	(5)	18	62
7	Ph	OEt	OSiMe ₃	OEt	<u>1a</u>	(10)	1	92
8	n-C ₃ H ₇	OEt	OSiMe ₃	OEt	<u>1a</u>	(10)	24	91
9	n-C ₆ H ₁₃	OEt	OSiMe ₃	OEt	<u>1a</u>	(10)	6	82
10	CH2=CH (CH2)8	OEt	OSiMe ₃	OEt	<u>1a</u>	(10)	2	74
11	Ph	Н	Н	Ph	<u>1a</u>	(10)	48	35

a) In the generation of <u>2a</u> there is always a minor amount of <u>3a</u> present as observed by ³¹P NMR spectroscopy.

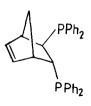
b) Products following chromatography on SiO₂ (pet.-ether 40-60/ethyl acetate 10:1). In case of entries 1-6 mixtures of <u>6a</u> and <u>6b</u> are formed, the former predominating. Complete desilylation using HF/H₂O/THF is possible. Products in entries 7-10 occur as <u>6a</u> (diastereomeric mixtures). Yields were not optimized.

None of the above reactions⁶⁾ proceed in strongly coordinating solvents such as THF, acetonitrile or methanol, which are known to form Lewis acid/Lewis base adducts with rhodium⁷⁾. This may be an indication that the mechanism of aldol

addition in CH_2Cl_2 involves coordination of one or more aldehyde molecules at the metal according to <u>7</u>, leading to an activation of the carbonyl function⁸. Control experiments show that R_3SiClO_4 (which could be formed under the reaction conditions) does <u>not</u> catalyze the aldol addition. Also, $[Ph_2P(CH_2)_2PPh_2]_2RhCl$ is a much less effective catalyst, i.e., the aldol addition requires considerably longer reactions times. Alternatively, intermediate rhodium enolates may be formed which in turn undergo aldol additions⁹)

$$[Rh]^{\textcircled{}} + RCHO \longrightarrow \left[\begin{array}{c} 0 \\ Rh \\ \end{array} \right]^{\textcircled{}}$$

Preliminary experiments on asymmetric induction were also performed using commercially available (R,R)-NORPHOS $(\underline{8})^{10}$ and (S,S)-CHIRAPHOS $(\underline{9})^{11}$ in the reaction of 5 (R¹=R²=CH₃; R³=OCH₃) with benzaldehyde (Table 2). These are the first examples of enantioselective aldol additions using chiral rhodium catalysts¹².



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Table 2. Catalytic Asymmetric Aldol Additions of $5(R^{1}=R^{2}=CH_{3}; R^{3}=OCH_{3})$ to Benzaldehyde

	Catalyst	Mol-8	of catalyst	ee (%) ^a	Configuration ^b
<u>1</u>	(CHIRAPHOS)		5	5	S
<u>2</u>	(NORPHOS)		5	7	R
3	(NORPHOS)		5	12	R

 a) Aldolization proceeded to >75% in all cases. The crude product was desilylated and the ee value of the aldol determined by NMR shift experiments (Eu(tfc)₂).

b) Absolute configuration: Lit. 13).

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In summary, rhodium-diphosphine complexes are useful catalysts for aldol additions of enolsilanes to aldehydes, as little as 3-5 mol-% often being sufficient. It remains to be seen whether more efficient chiral ligands for asymmetric catalytic C-C bond formation can be developed. They could also be useful in the Rh-catalyzed cyanohydrin-forming addition of Me,SiCN to aldehydes¹⁴⁾.

This work was supported by the Fonds der Chemischen Industrie and BAYER AG, (Leverkusen).

References and Notes:

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- 2) H.B. Kagan, "Asymmetric Synthesis using Organometallic Catalysts", in G. Wilkinson, (ed.), "Comprehensive Organometallic Chemistry", Pergamon Wilkinson, (ed.), "Comprehensive Organometallic Chemistry", Pergamon Oxford, 1982, vol.8, p. 463 ff; J.M. Brown and P.A. Chaloner, "Asymmetric Hydrogenation Reactions Using Chiral Diphosphine Complexes of Rhodium", in L. Pignolet, (ed.), "Homogeneous Catalysis with Metal Phosphine Complexes", Plenum Press, N.Y., 1983, chapter 4.
 K. Tani, K. Suwa, T. Yamagata and S. Otsuka, <u>Chem. Lett.</u> 265 (1982).
 B.R. James and D. Mahajan, <u>Can. J. Chem.</u>, <u>57</u>, <u>180</u> (1979).
 Hydrogenation of 1 to form <u>2</u> was carried out in CH₂Cl₂, not in donor solvents as previously reported²). <u>2</u> is therefore only a formal representation and may be covalent. See: J. Peone jr. and L. Vaska, <u>Angew. Chem.</u> Int. Ed. Engl. 10, <u>511</u> (1971).

- Int. Ed. Engl. 10, 511 (1971).
- 6) A typical procedure involves the solution of an aldehyde (2 mmol) and an enolsilane (2 mmol) in CH_2Cl_2 (2 ml) in the presence of freshly prepared Rh-catalyst (5 mol-%). After the given amount of time the solvent was stripped off and the residue extracted with pet.-ether/ethyl acetate (10:1) followed by chromatography on SiO2.
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- 8) Concerning the structure and electronic nature of the PhCHO/BF3 adduct, see M.T. Reetz, M. Hüllmann, W. Massa, S. Berger, P. Rademacher and P. Heymanns, J. Am. Chem. Soc. 108, 2405 (1986). 9) The authors in lit¹⁾ also speculate on the intermediacy of Rh-enolates.

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- 13) M. Guette, J. Capillon and J.P. Guette, Tetrahedron 29, 3659 (1973).
- 14) We have observed that Me₃SiCN adds smoothly to aldehydes at room temperature in the presence of catalytic amounts (5%) of <u>1</u> or <u>2</u>, conversion being 90%.

(Received in Germany 2 December 1986)