

194. Regioselective Alkylation of the Polyfunctional Nucleophile 1-(Methylthio)-3-triethylsilyloxypentadienyllithium

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Dedicated to Prof. *Vladimir Prelog* on the occasion of his 75th birthday

(24. VIII. 81)

Summary

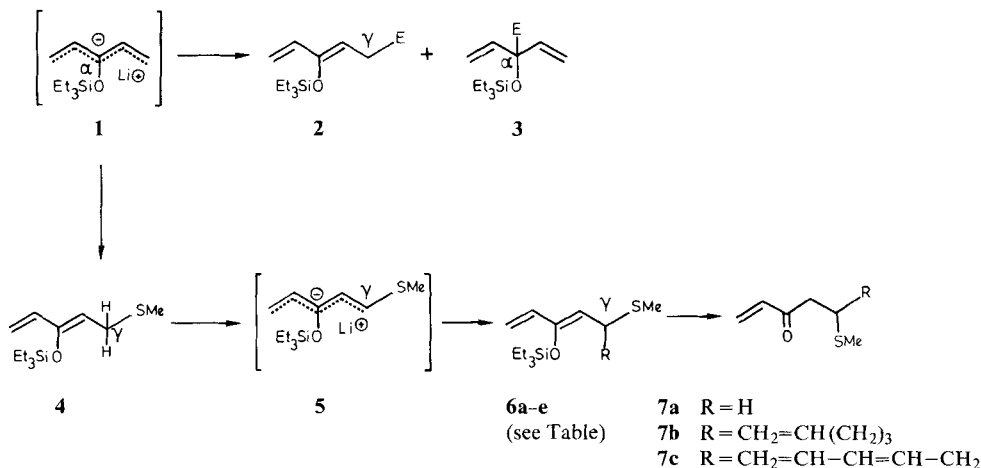
γ -Selective sulfenylation of the triethylsilyloxypentadienyllithium **1** gave the versatile alkylthiodiene **4** which on successive deprotonation and alkylation furnished with high regioselectivity the γ -products **6**. Fluoride-promoted silylether cleavage **6** $\rightarrow$ **7** may be followed by intramolecular [4+2]-addition **7c** $\rightarrow$ **8** and sulfoxide elimination **8** $\rightarrow$ **9**. The conversions **7b** $\rightarrow$ **12** and **7a** $\rightarrow$ **17** demonstrate the feasibility of **5** to serve as an equivalent of the hypothetical β -deprotonated divinylketone **13** whose two enone units may be unmasked separately.

Introduction. – In connection with our general interest in intramolecular *Diels-Alder* reactions we have reported the electrophilic substitution of the lithiated triethylsilyloxypentadiene **1** [1] [2]. The utility of **1** as a convenient C₅-unit for both the construction and attachment of the functionalized dienes **2** and dienophile units depends on a selective γ -substitution (**1** $\rightarrow$ **2**). This is generally achieved with carbonyl electrophiles. On alkylation of the metalated diene **1**, however, the undesired α -attack **1** $\rightarrow$ **3** was unsatisfactorily competitive (*Scheme 1*). As a solution to this problem we present here a reliable method for regioselective γ -alkylation of 3-silyloxydienes.

Preparation and Alkylation of the Methylthiosilyloxypentadienyllithium 5. – Determinative to this approach was the regioselective γ -sulfenylation of **1**: Dropwise addition of dimethyl disulfide (1 mol-equiv.) to a freshly prepared 2 M solution of **1** in THF at -78° under argon, stirring of the reaction mixture for 15 min at -78° , quenching with sat. aq. NH₄Cl-solution and distillation gave the methylthiodiene **4** as a stable colorless oil in 81% yield. Not even a trace of the α -sulfenylated isomer **3**, (E = SMe) was found in the reaction mixture. The methylthio substituent was expected not only to facilitate conveniently the deprotonation **4** $\rightarrow$ **5**, but, particularly, to direct the alkylation of **5** entirely towards the desired γ -products **6**¹⁾. Indeed, addition of **4** to LDA (1.1 mol-equiv.) in THF/HMPA at -78° gave a deep-red solution of

¹⁾ For the electrophilic substitutions of sulfur-substituted allyl anions see reviews [3] [4], as well as [5] [6]; for the preparation of 1,1-bis(methylthio)-3-trimethylsilyloxydiene see [7]. This work has been presented by one of us (*W.O.*) at the 6th International Symposium on Synthesis in Organic Chemistry, Cambridge (England), July 1979.

Scheme 1

Table. Alkylation products of the methylthiosilyloxypentadienyllithium **5** by halides R-X, and corresponding yields

X	R in 6	Solvent	Yield of 6 (%) ^{a)}
I	a CH ₃	THF/HMPA 5:1	91
I	b C ₂ H ₅	THF	78
Br	c CH ₂ =CH(CH ₂) ₂	THF/HMPA 10:1	87
Br	d CH ₂ =CH(CH ₂) ₃	THF/HMPA 5:1	89
Br	e CH ₂ =CH-CH=CH-CH ₂	THF	82

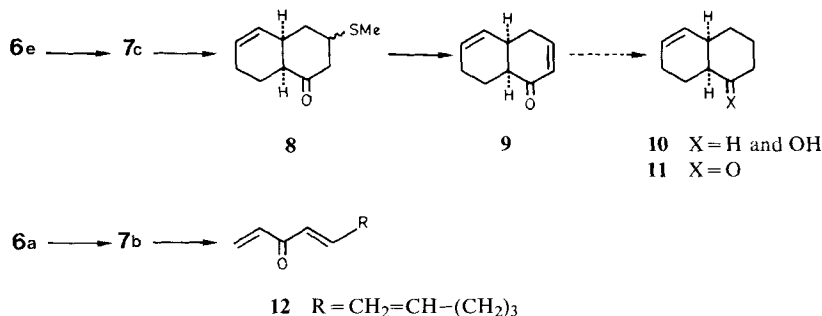
^{a)} Yields are based on the diene **4**.

5 which was treated with a series of alkyl or alkenyl halides (1.2 to 2.0 mol-equiv.) at -78° . Quenching of the decolorized reaction mixtures with sat. aq. NH₄Cl-solution and work-up furnished the single γ -products **6a-e** in high yields as indicated in the *Table*. Deprotonation of **4** and subsequent alkylation may be also carried out in THF without HMPA as demonstrated by the conversions **4** $\rightarrow$ **6b** and **4** $\rightarrow$ **6c**.

Further reactions of the substituted dienes 4 and 6. - This easy access to **6** containing olefinic substituents R permits their efficient use in intramolecular *Diels-Alder* reactions. For example, silyl ether cleavage of the tetraene **6e** with KF/MeOH at -10 to 0° for 1 h, subsequent aq. work-up and chromatography furnished directly the *cis*-fused bicyclic decenones **8** (C(3)-epimer-mixture) in 63% yield (*Scheme 2*).

Hence the initially formed enone **7** undergoes a rapid *endo*-controlled intramolecular [4+2]-addition to the diene moiety. After cycloaddition, the methylthio group could be in principle reductively removed [8] or, exploited for further functionalization. Thus, oxidation of the epimers **8** with NaIO₄ in aq. MeOH-solution at -25° followed by sulfoxide elimination [9] in boiling CCl₄ gave the dienone **9** in 62% yield. The depicted *cis*-fusion of **9** (and of **8**), indicated by ¹H-NMR-evidence ($J_{A,B} = 9$ Hz), was confirmed by correlation with the previously

Scheme 2

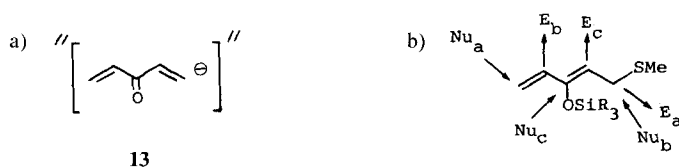


described bicyclic enone **11** [2]: Reduction of **9** with NaBH_4 in EtOH furnished the alcohol **10** which on oxidation with PCC [10] (1.6 mol-equiv.)/NaOAc/ CH_2Cl_2 gave **11** which proved to be identical with an authentic sample.

The enones **7** may also serve as precursors to the cross-conjugated dienones **12**. This is illustrated by the conversion **7b** $\rightarrow$ **12** via the oxidation/sulfoxide-elimination sequence (66% yield) (Scheme 2). Thus introduction of the sulfur moiety into **1** has further increased the functionality of this C_5 -unit so that **5** is a practical equivalent of the hypothetical β -deprotonated divinyl ketone **13** (Scheme 3a)²⁾.

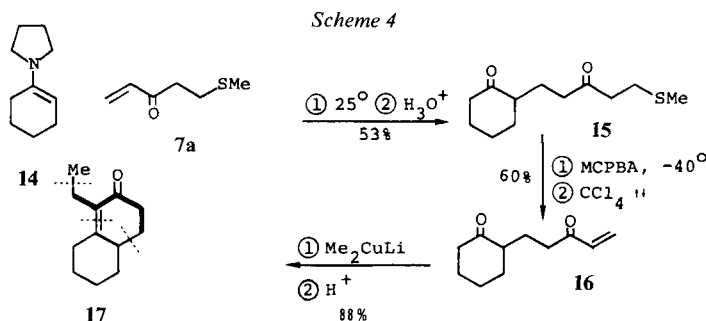
A further advantage is that both latent enone functions can be liberated separately, thus permitting the regioselective polysubstitution of **4** by a range of different electrophiles and nucleophiles (Scheme 3b). For example this stepwise functionalization of **4** was demonstrated by its conversion to the bicyclodecenone **17** (Scheme 4). The enone **7a**, prepared by fluoride-promoted cleavage of **4**, underwent a Michael reaction with the enamine **14** to give, after aq. acidic work-up, the diketone **15**. Subsequent unmaking of the second enone by oxidation/sulfoxide elimination furnished **16** in 60% yield. Treatment of **16** with lithium dimethylcuprate and aq. HCl-solution gave directly the annelated product **17** in 88% yield (which arises from conjugate addition and trapping of the enolate by an intramolecular aldolisation)³⁾. Thus, among the possibilities indicated in Scheme 3 the conversion **7a** $\rightarrow$ **17** illustrates the formation of three C, C-bonds by successive attack of the nucleophiles Nu_a , Nu_b and the electrophile E_c .

Scheme 3



²⁾ For an alternative equivalent of **13** see [11].

³⁾ For a combination of conjugate addition with subsequent intramolecular aldolisation see [12].



Accordingly, the versatile and selective reactivity of the C₅-building block 4 may prove of value in the synthesis of natural products.

Financial support of this work by the Schweizerischer Nationalfonds zur Förderung der wissenschaftlichen Forschung, Sandoz Ltd., Basle and Givaudan SA, Vernier, is gratefully acknowledged. We also thank Mr. J. P. Saulnier and Mrs. D. Clement for the NMR. and MS. measurements.

Experimental Part

General. - See [2].

Preparation and Alkylation of the Methylthiosilyloxypentadienyllithium (5) (Scheme 1) (→ 6a-e). - *Preparation of (3Z)-5-(methylthio)-3-triethylsilyloxy-1,3-pentadiene (4).* A solution of *sec*-BuLi in cyclohexane (3 mmol) was added dropwise to a stirred solution of 3-triethylsilyloxy-1,3-pentadiene (396 mg, 2 mmol) in dry THF (6 ml) at -78° under Ar. After 30 min at -78° dimethyldisulfide (freshly distilled, 310 mg, 3.3 mmol) was added dropwise to the stirred solution of the anion 1. After 15 min at -78° the decolorized reaction mixture was poured into sat. aq. NH₄Cl-solution. Extraction with pentane, work-up and distillation gave the thioether 4 (oil, 590 mg, 81%), b.p. 134°/12 Torr or 108-109°/1.5 Torr. Rf (hexane) 0.14, GC. (160°): 16.17. - UV.: 239 (4.22). - IR.: 2950, 2905, 2870, 1642, 1605, 1362, 1050, 910. - ¹H-NMR.: 0.5-1.2 (15 H); 2.07 (s, 3 H); 3.24 (d, J = 8, 2 H); 4.90 (t, J = 8, 1 H); 5.06 (d×d, J = 10 and 2, 1 H); 5.38 (d×d, J = 17 and 2, 1 H); 6.22 (d×d, J = 17 and 10, 1 H). - MS.: 244 (4, C₁₂H₂₄OSSi⁺), 197 (100), 169 (7), 133 (7), 115 (70), 87 (78).

Preparation of (3Z)-5-(methylthio)-3-triethylsilyloxy-1,3-hexadiene (6a). A solution of *n*-BuLi in hexane (0.55 mmol) was added dropwise to a stirred solution of diisopropylamine (60.5 mg, 0.6 mmol) in THF (3 ml) at -78° under Ar. After 10 min at -78° a solution of 4 (122 mg, 0.5 mmol) in THF (1 ml) was added slowly to give a clear yellow solution. After a further 20 min at -78° addition of HMPA (0.6 ml) changed the color of the mixture to deep-red. Then methyl iodide (142 mg, 1 mmol) was added to the mixture which then decolorized within 1 h at -78°. Pouring the reaction mixture into cold water, extraction with ether, work-up, chromatography (benzene) and distillation at 120-130° (bath)/1 Torr furnished 6a (oil, 142 mg 91%), Rf (toluene) 0.67. - UV.: 240 (4.33). - IR.: 2960, 2880, 1601, 1365, 1055. - ¹H-NMR.: 0.5-1.2 (15 H); 1.3. (d, J = 7, 3 H); 2.04 (s, 3 H); 3.80 (m; irradiation at 1.31 → d, J = 10.5, 1 H); 4.70 (d, J = 10.5, 1 H); 5.06 (d×d, J = 10 and 2, 1 H); 5.37 (d×d, J = 17 and 2, 1 H); 6.21 (d×d, J = 17 and 10, 1 H). - MS.: 258 (6, C₁₃H₂₆OSSi⁺), 211 (100), 115 (75), 105 (8), 103 (5), 87 (82).

Preparation of (3Z)-5-methylthio-3-triethylsilyloxy-1,3-heptadiene (6b). A solution of 4 (122 mg, 0.5 mmol) in THF (1 ml) was added dropwise to a stirred solution of lithium diisopropylamide (LDA, prepared as describe above) in THF (1.5 ml) at -78° under Ar. After 10 min at -78° ethyl iodide (156 mg, 1 mmol) was added slowly (no addition of HMPA) and the reaction mixture was left at -78° for a further 10 min. The reaction mixture changed its color from orange to pale-yellow. Pouring the mixture into cold water, extraction with ether, work-up, chromatography (hexane/toluene 1:4) and distillation at 120-130° (bath)/1 Torr furnished 6b (oil, 105 mg, 78%), Rf (toluene) 0.69. - IR.: 2960, 2880, 1602, 1368, 1058. - ¹H-

NMR.: 0.5–1.2 (18 H); 1.63 (*m*, 2 H); 2.03 (*s*, 3 H); 3.65 ($d \times t$, $J = 10.5$ and 7, 1 H); 4.72 (*d*, $J = 10.5$, 1 H); 5.07 ($d \times d$, $J = 10$ and 2, 1 H); 5.38 ($d \times d$, $J = 17$ and 2, 1 H); 6.24 ($d \times d$, $J = 17$ and 10, 1 H). – MS.: (M^+ not observed), 121 (52), 119 (85), 117 (88), 103 (92), 87 (54), 75 (100).

(3*Z*)-5-(Methylthio)-3-triethylsilyloxy-1,3,8-nonatriene (**6c**). A solution of **4** (488 mg, 2 mmol) in THF (2 ml) was added slowly to a stirred solution of LDA (2.2 mmol) in THF (10 ml) at -78° under Ar. After 20 min at -78° HMPA (1 ml) was added followed by the addition of 4-bromo-1-butene (338 mg, 2.5 mmol). After further 30 min at -78° the decolorized reaction mixture was poured into cold water and extracted with ether. Work-up, chromatography (hexane/toluene 1:9) and distillation at 130 – 145° (bath)/1 Torr gave the triene **6c** (oil, 519 mg, 87%), Rf (toluene) 0.70. – IR.: 2960, 2880, 1644, 1605, 1365, 1057, 915. – $^1\text{H-NMR}$.: 0.5–1.2 (15 H); 1.70 (*m*, 2 H); 2.03 (*s*, 3 H); 2.17 (*m*, 2 H); 3.71 ($d \times t$, $J = 10.5$ and 7, 1 H); 4.71 (*d*, $J = 10.5$, 1 H); 4.9–5.2 (3 H); 5.37 ($d \times d$, $J = 17$ and 2, 1 H); 5.84 (*m*, 1 H); 6.22 ($d \times d$, $J = 17$ and 10, 1 H). – MS.: (M^+ not observed), 251 (66), 197 (24), 121 (55), 115 (100), 103 (76), 87 (73).

5-(Methylthio)-3-triethylsilyloxy-1,3,9-decatriene (**6d**). A solution of **4** (784 mg, 3.21 mmol) in THF (1 ml) was added slowly to a solution of LDA (3.8 mmol) in THF (8 ml) at -78° under Ar. After 20 min at -78° HMPA (1.5 ml), followed by 5-bromo-1-pentene (600 mg, 4 mmol) was added. After 1 h at -78° the reaction mixture was poured into cold water. Extraction with ether, work-up and distillation *i.v.* gave the triene **6d** (oil, 890 mg, 89%), b.p. 132 – $134^\circ/0.7$ Torr, Rf (toluene) 0.71. – IR.: 2955, 1642, 1601, 1368, 1055, 915. – $^1\text{H-NMR}$.: 0.5–1.2 (15 H); 1.2–1.8 (4 H); 2.04 (*s*, 3 H); 2.07 (*m*, 2 H); 3.71 (*m*, irradiation at $1.57 \rightarrow d$, $J = 10.5$, 1 H); 4.70 (*d*, $J = 10.5$, 1 H); 4.9–5.2 (3 H); 5.37 ($d \times d$, $J = 17$ and 2, 1 H); 5.83 (*m*, 1 H); 6.22 ($d \times d$, $J = 17$ and 10, 1 H). – MS.: 312 (4, $\text{C}_{17}\text{H}_{32}\text{OSSi}^+$), 265 (100), 180 (6), 121 (23), 115 (44), 87 (58).

5-(Methylthio)-3-triethylsilyloxy-1,3,7,9-decatetraene (**6e**). A solution of **4** (488 mg, 2 mmol) in THF (1 ml) was added dropwise to a stirred solution of LDA (2.2 mmol) in THF (3 ml) at -78° under Ar. After 15 min at -78° a solution of 5-bromo-1,3-pentadiene [**14**] (323 mg, 2.2 mmol) in THF (1 ml) was added (no addition of HMPA). The instantaneously decolorized reaction mixture was poured into cold water. Extraction with ether, work-up, chromatography (hexane/toluene 1:4) and distillation at 150 – 160° (bath)/1 Torr gave the tetraene **6e** (oil, 508 mg, 82%), Rf (hexane) 0.23. – IR. (film): 2960, 2915, 2875, 1600, 1420, 1365, 1245, 1050, 1005, 750. – $^1\text{H-NMR}$.: 0.5–1.2 (15 H); 2.02 (*s*, 3 H); 2.39 (*m*, 2 H); 3.77 ($d \times t$, $J = 10.5$ and 7; irradiation at $2.39 \rightarrow d$, $J = 10.5$, 1 H); 4.71 (*d*, $J = 10.5$, 1 H); 4.8–5.4 (2 H); 5.16 ($d \times d$, $J = 10$ and 2, 1 H); 5.37 ($d \times d$, $J = 17$ and 2, 1 H); 5.71 (*m*, irradiation at $2.39 \rightarrow d$, $J = 15$, 1 H); 5.9–6.6 (3 H). – MS.: (M^+ not observed), 263 (5), 251 (6), 197 (45), 133 (8), 115 (100), 105 (14), 103 (10), 97 (95), 79 (36).

Reactions of the Methylthiosilyloxydienes 4 and 6 (Schemes 2 and 4). – *cis*-4-(Methylthio)bicyclo[4.4.0]dec-7-en-2-one (**8**). KF (174 mg, 3 mmol) was added during 5 min to a stirred solution of crude **6e** (733 mg, 2.36 mmol) in methanol (25 ml) at -10° under Ar. After 1 h at $-10^\circ \rightarrow 0^\circ$ the reaction mixture was poured into cold water. Extraction with ether, work-up and chromatography (CH_2Cl_2) gave the bicyclic ketone **8** (oil, mixture of stereoisomers, 292 mg, 63%), Rf (CH_2Cl_2) 0.23, GC. (180°): 10.19. – IR.: 2934, 1710, 1264, 1235, 944. – $^1\text{H-NMR}$.: 1.4–3.0 (14 H); 5.72 (*m*, 2 H). – MS.: (M^+ not observed), 119 (90), 117 (93), 103 (100), 86 (57), 84 (100), 75 (97).

cis-Bicyclo[4.4.0]-3,7-decadien-2-one (**9**). A solution of NaIO_4 (214 mg, 1 mmol) in water (1 ml) was added slowly to a stirred solution of **8** (178 mg, 0.91 mmol) in methanol (10 ml) at -25° under Ar. After 3 h at -25° the reaction mixture was poured into cold water. Extraction with CH_2Cl_2 and work-up gave a crude sulfoxide which was heated in CCl_4 under reflux for 2.5 h. Evaporation of the solution and chromatography (CH_2Cl_2) gave the dienone **9** (oil, 82 mg, 62%), Rf (CH_2Cl_2) 0.26 GC. (180°): 9.9. – IR.: 3020, 2920, 1676, 1390, 1257, 1130. – $^1\text{H-NMR}$.: 1.5–3.0 (8 H); 5.64 ($d \times m$, $J = 10$, 1 H); 5.68 ($d \times m$, $J = 10$, 1 H); 6.05 ($d \times t$, $J = 10$ and 2 H); 6.91 ($d \times t$, $J = 10$ and 4, 1 H). Addition of varying amounts of $\text{Eu}(\text{FOD})_3$, combined with decoupling experiments, indicates a vicinal coupling of the angular protons $J \approx 9$ Hz in agreement with the assigned *cis*-fusion. – MS.: 148 (80, $\text{C}_{10}\text{H}_{12}\text{O}^+$), 133 (38), 107 (28), 94 (35), 80 (100), 79 (65).

All-*cis*-bicyclo[4.4.0]dec-7-en-2-ol (**10**). NaBH_4 (10.3 mg, 0.27 mmol) was added portionwise to a stirred solution of the bicyclic dienone **9** (20 mg, 0.135 mmol) in ethanol (4 ml) at 25° . After 20 min at 25° the mixture was heated under reflux for 15 min and then was poured into cold water. Extraction with CH_2Cl_2 gave the all-*cis*-alcohol **10** identical (TLC., GC., $^1\text{H-NMR}$., IR.) to a sample, previously prepared [2].

cis-Bicyclo[4.4.0]dec-7-en-2-one (**11**). A solution of **10** (30 mg, 0.197 mmol) in CH_2Cl_2 (4 ml) was added in one portion to a rapidly stirred slurry of pyridinium chlorochromate (63 mg, 0.31 mmol) in dry CH_2Cl_2 (10 ml) containing dry NaOAc (41 mg, 0.5 mmol) at 25° under Ar. After 1 h at 25° the reaction mixture was poured into ether. Filtration through *Celite*, evaporation and chromatography gave the bicyclic enone **11** (oil, 27 mg, 92%) identical to a sample, prepared previously [2].

5-(Methylthio)-1,9-decadien-3-one (7b). A solution of **6d** (1.39 g, 4.46 mmol) in methanol (10 ml) was added slowly to a stirred solution of **KF** (290 mg, 5 mmol) in methanol (15 ml) at -10° under Ar. The reaction mixture was allowed to attain -5° during 30 min. Evaporation and chromatography (toluene/EtOAc 19:1) furnished **7b** (oil, 630 mg, 71%), Rf (toluene/EtOAc 19:1) 0.43. – IR.: 2930, 1683, 1405, 960, 920. – $^1\text{H-NMR.}$: 1.58 (*m*, 4 H); 2.08 (*s*, 3 H); 2.12 (*m*, 2 H); 2.5–3.3 (3 H); 4.9–5.2 (2 H); 5.6–6.5 (4 H). – MS.: 198 (19, $\text{C}_{11}\text{H}_{18}\text{OS}^+$), 183 (61), 151 (66), 150 (68), 107 (51), 95 (100).

(4E)-1,4,9-Decatrien-3-one (12). A solution of *m*-chloroperbenzoic acid (80%, 258 mg, 1.2 mmol) in CH_2Cl_2 (9 ml) was added dropwise to a stirred solution of **7b**, (198 mg, 1 mmol) in CH_2Cl_2 (15 ml) at -78° under N_2 . After 1 h at -78° the reaction mixture was poured into 10% aq. Na_2SO_3 -solution. Extraction with ether, work-up and distillation at 120° (bath)/11 Torr afforded the dienone **12** (pale yellow oil, 96 mg, 66%), Rf (toluene/EtOAc 19:1) 0.40. – GC. (150°): 8.17. – IR.: 1670, 1635, 1617, 1409, 988. – $^1\text{H-NMR.}$: 1.64 (*qa*, $J = 7$, 2 H); 2.21 (*m*, 4 H); 4.9–5.2 (2 H); 5.6–6.1 (2 H); 6.1–7.2 (4 H). – MS.: (M^+ not observed), 136 (9), 122 (20), 108 (59), 96 (50), 82 (41), 55 (100).

5-(Methylthio)-1-penten-3-one (7a). **KF** (700 mg, 12 mmol) was added to a stirred solution of crude **4** (obtained from 1.6 g (8.1 mmol) of 3-triethylsilyloxy-1,4-pentadiene) in methanol (20 ml) at 0° . Subsequent stirring of the reaction mixture at 0° for 1 h, work-up and chromatography (CH_2Cl_2) gave the enone **7a**, (oil, 584 mg, 56% yield from 3-triethylsilyloxy-1,4-pentadiene), Rf (CH_2Cl_2) 0.39, GC. (SE-30, 140°): 3.17. – IR.: 2910, 1680, 1620, 1405, 1095, 985, 965, 915. – $^1\text{H-NMR.}$: 2.12 (*s*, 3 H); 2.65–3.10 (4 H); 5.70–6.70 (3 H). – MS.: 130 (20, $\text{C}_6\text{H}_{10}\text{OS}^+$), 83 (46), 82 (57), 75 (33), 74 (10), 61 (30), 55 (100), 47 (21).

2-(5-Methylthio-3-oxopentyl)cyclohexanone (15). A mixture of the enone **7a** (584 mg, 4.5 mmol) and the enamine **14** [15] (820 mg, 5.4 mmol) in dry dioxane was stirred at 25° for 3 h. After addition of 2N aq. **HCl** (3 ml) the mixture was stirred for 15 h. Work-up and chromatography (toluene/EtOAc 9:1 $\rightarrow$ 3:1) gave **15** (oil, 544 mg, 53%), Rf (toluene/EtOAc 3:1) 0.42. – IR. (film): 2920, 2855, 1708, 1130. – $^1\text{H-NMR.}$: 1.1–2.9 (17 H); 2.10 (*s*, 3 H). – MS.: 228 (8, $\text{C}_{12}\text{H}_{20}\text{O}_2\text{S}^+$), 178 (35), 86 (60), 85 (45), 84 (100), 83 (75), 55 (80), 47 (50).

2-(3-Oxo-4-pentenyl)cyclohexanone (16). A solution of *m*-chloroperbenzoic acid (90%, 440 mg, 2.4 mmol) in CH_2Cl_2 (5 ml) was added slowly over 15 min to a stirred solution of **15** (540 mg, 2.4 mmol) in CH_2Cl_2 (30 ml) at -30° . Work-up (CH_2Cl_2) gave a crude sulfoxide (551 mg, 96% yield. – $^1\text{H-NMR.}$: 1.2–2.9 (13 H); 2.60 (*s*, 3 H); 2.97 (*m*, 4 H)). A solution of this crude sulfoxide (80 mg, 0.33 mmol) in CCl_4 (6 ml) was heated under reflux for 4 h. Evaporation and chromatography (toluene/EtOAc 3:1) gave **16** (oil, 36 mg, 60%), Rf (toluene/EtOAc 3:1) 0.35. – IR. (film): 2930, 2857, 1705, 1684, 1620, 1452, 1407, 1132, 969. – $^1\text{H-NMR.}$: 1.2–2.9 (13 H); 5.83 ($d \times d$, $J = 8.5$ and 4, 1 H); 6.1–6.6 (2 H). – MS.: 180 (2, $\text{C}_{11}\text{H}_{16}\text{O}_2^+$), 151 (24), 111 (23), 84 (20), 83 (18), 70 (24), 55 (100), 41 (24).

2-Ethylbicyclo[4.4.0]dec-1-en-3-one (17). Methylolithium (1.9M in ether, 1.69 mmol) was added to a suspension of **CuI** (dried by heating at $140^{\circ}/0.1$ Torr for 1 h, 172 mg, 0.9 mmol) in dry ether (7 ml) at -10° under Ar. To the clear solution, **16** (135 mg, 0.75 mmol) in dry ether (5 ml) was added dropwise with stirring at -78° . Successive stirring of the mixture at -60° for 90 min, at 0° for 1 h and work-up gave a crude product which was heated in THF (6 ml)/conc. aq. **HCl** (0.1 ml) for 2 h under reflux. Work-up and chromatography (pentane/ether 2:1) furnished the bicyclic enone **17** (oil, 117 mg, 88%), Rf (toluene/EtOAc 3:1) 0.48, GC. (SE-30, 180°): 6.8. – IR.: 2918, 2850, 1670, 1616, 1450, 1364, 1196. – $^1\text{H-NMR.}$: 0.91 (*t*, $J = 7$, 3 H); 1.1–2.7 (14 H); 2.91 (*m*, 1 H). – MS.: (M^+ not observed), 178 (78), 150 (41), 149 (100), 121 (31), 107 (31), 93 (28), 91 (21), 79 (38).

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