A NEW REACTION OF $\alpha-\text{EPOXIDES}$: RING OPENING WITH THE FORMATION OF NITRATOPERCHLORATES

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The direction of the ring opening of α -epoxides catalyzed by protic acids is affected by the addition of perchlorate salts [1-3]. In a study of tricyclodecane α -epoxides, we found that lithium perchlorate may act as a reagent leading to the formation of covalent perchlorates [4]. In the present communication, we report a new reaction involving the ring opening of epoxides by nitronium tetrafluoroboride in the presence of perchlorate salts. The study was carried out on the α -epoxides of ethylene, propylene, and cyclohexene which were treated with a twofold excess of NO₂BF₄ and LiClO₄ or Bu₄NClO₄ in either ethyl acetate or methylene chloride at 0-25°C. In all cases, chromatography of the reaction mixture gave the corresponding nitratoperchlorates (I)-(III) in 60-90% yield. These products are stable at 25°C but decompose explosively upon heating above 100°C.

$$R^{1}HC \xrightarrow{NO_{2}BF_{4}} R^{1}CH \xrightarrow{CHR^{2}} CHR^{2}$$

$$0ClO_{3} \quad 0NO_{2}$$

$$M = Li, Bu_{4}N; R^{1} = R^{2} = H (I); R^{1} = CH_{3}, R^{2} = H(II); R^{1} + R^{2} = (CH_{2})_{4}(III).$$

The structures of (I)-(III) were shown by PMR and IR spectroscopy as well as by chemical transformations. In the case of propylene oxide, isomeric 1-perchloryloxy-2-propyl nitrate is formed in 30% yield as indicated by PMR spectroscopy in addition to (II) (obtained in 60% yield). The ring opening of cyclohexene oxide proceeds with steric specificity to yield the product with trans configuration.

This is a new reaction of α -epoxides and serves as a convenient, efficient, and safe method for the preparation of a new class of compounds, namely, vicinal nitratoperchlorates.

LITERATURE CITED

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