Concise Enantiodivergent Synthesis of Eutypoxide B

Seiichi Takano,* Minoru Moriya and Kunio Ogasawara

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan

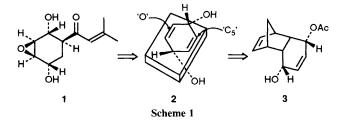
The first enantiodivergent synthesis of eutypoxide B 1, a metabolite of the fungus *Eutypa lata*, has been accomplished in a stereo- and regio-controlled manner by using the single chiral building block **3** as a chiral equivalent of (*Z*)-cyclohex-2,5-dien-1,4-diol **2**.

The fungus *Eutypa lata*, the pathogen responsible for vineyard dieback, produces a secondary metabolite eutypoxide $B^{1,2}$ 1 having five stereogenic centres on a cyclohexane ring. The racemic total synthesis of 1 has been reported by Tabacchi and coworkers² by employing the Diels–Alder reaction as the key step. However, the key reaction directed the diastereofacial selection in such a way so as to give the diastereoisomeric mixture containing the desired epimer only in a ratio of 1:10; fortunately the epimers could be separated in the later stages.² We report here the enantiodivergent and stereocontrolled approach to both enantiomers of eutypoxide B 1 starting from the single chiral building block³ 3 which we have devised.

Because the synthesis of 1 could formally be achieved by regio- and enantio-selective addition of the C₅ subunit and the epoxide oxygen to *meso-(Z)-1*,4-dihydroxycyclohexa-2,5-diene 2 from the *anti*-face to the hydroxy groups, we began the synthesis using the optically pure 3, obtained by the lipase mediated asymmetrization of a *meso*-precursor,³ as a chiral equivalent of 2 (Scheme 1). Thus, 3 was first oxidized to the ketone $4, \dagger [\alpha]_D{}^{30} - 219.0$ (*c* 1.29, CHCl₃), \ddagger in 91% yield to carry out regiospecific 1,4-addition. Owing to its biased structure, 4 allowed the 1,4-addition of vinylmagnesium bromide⁴ stereospecifically from the less hindered *exo*-face to give the vinyl ketone 5, m.p. 73–74 °C, $[\alpha]_D{}^{32} + 20.9$ (*c* 0.94, CHCl₃), exclusively, in 81% yield. Reduction of 5 with sodium borohydride again occurred stereospecifically at the *exo*-face

to afford the *endo*-alcohol **6** whose stereochemistry was confirmed by the bromo-ether formation and the reductive reversion by sequential treatment with *N*-bromosuccinimide (NBS) and zinc.^{3.5} Practically, **5** was sequentially reduced with sodium borohydride and deacylated with potassium carbonate in methanol in the same flask to furnish the diol **7**, m.p. 131–132 °C, $[\alpha]_D^{27}$ +72.2 (*c* 1.16, MeOH), in 92% yield.

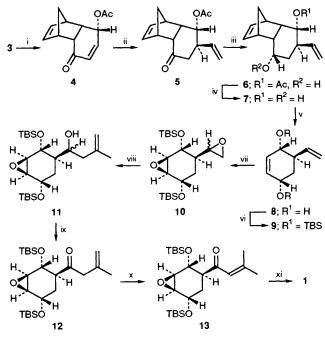
Upon thermolysis in refluxing diphenyl ether³ (ca. 280 °C, 1 h), 7 gave the subsituted cyclohexenediol 8 in 73% yield by retro-Diels-Alder reaction with removal of cyclopentadiene. To install the requisite functional groups in the most efficient way, 8 was first transformed into the disilyl ether 9, $[\alpha]_D^{26}$ -9.4 (c 1.02, CHCl₃), in 94% yield, which was then oxidized with an excess amount of *m*-chloroperbenzoic acid (MCPBA) in the presence of radical inhibitor^{2,6} to give rise to the diepoxide 10 in 74% yield as a mixture of diastereoisomers being epimeric at the stereogenic centre on the side chain epoxide. The reaction of the mixture with prop-2-enylmagnesium bromide in the presence of copper(1) iodide⁷ proceeded chemoselectively at the terminal of the side chain epoxide to give a mixture of the secondary alcohols 11 which, without separation, was oxidized with pyridinium chlorochromate (PCC) to give the single β , γ -enone 12, m.p. 78–79 °C, $[\alpha]_D^{29}$



 $^{^{+}}$ All new isolable compounds showed satisfactory analytical (combustion and/or high resolution mass) and spectral (IR, 1 H NMR and mass) data.

[‡] Optical purity was determined to be >99.5% enantiomeric excess by HPLC (Chiralcel OD, 2% PrⁱOH-hexane.

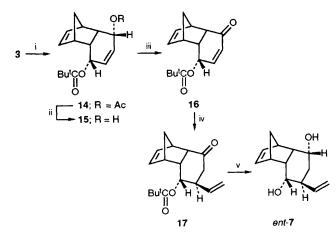
J. CHEM. SOC., CHEM. COMMUN., 1993



Scheme 2 Reagents and conditions: i, PCC, CH_2Cl_2 , room temp.; ii, vinylmagnesium bromide, trimethylsilyl chloride (TMSCl), $CuBr \cdot Me_2S$. tetrahydrofuran (THF)-hexamethylphosphoramide (HMPA), -78 °C, then 5% HCl; iii, NaBH4, MeOH, 0 °C; iv, K_2CO_3 , room temp.; v, diphenyl ether, reflux, 45 min: vi, tert-butyldimethylsilyl chloride (TBSCl), imidazole, dimethylformamide (DMF), room temp.; vii, MCPBA (3 equiv.), 4,4'-thiobis(6-tertbutyl-*m*-cresol) (10 mol%), (CH₂Cl₂, reflux, 1 h; viii, prop-2enylmagnesium bromide, CuI, THF, -25 °C; ix, PCC, CH₂Cl₂, room temp.; x, DBU (1 equiv.), CH₂Cl₂, room temp.; xi, Bu₄NF, THF, room temp.

+18.3 (c 0.71, CHCl₃), in 89% overall yield. On exposure to 1,8-diazabicyclo[5.4.0]undecene (DBU) in dichloromethane, 12 afforded the α , β -enone 13, $[\alpha]_D^{28}$ +8.0 (c 0.46, CHCl₃), in 91% yield by facile isomerization. Finally, desilylation of 13 gave (-)-eutypoxide B 1, $[\alpha]_D^{23}$ -56.6 (c 0.68, CHCl₃), in 63% yield, whose spectral data were identical with those reported² (Scheme 2).

In order to obtain enantiomeric (+)-eutypoxide B (*ent*-1), **3** was first transformed into the mixed diester³ 14 which was then treated with methanolic potassium carbonate to give the monoester 15 in 85% overall yield. Oxidation of 15 with PCC, followed by treating the resulting enone 16, $[\alpha]_D{}^{30} + 233.0$ (*c* 1.08, CHCl₃),‡ obtained in 94% yield, with vinylmagnesium bromide in the presence of copper(t) bromide and trimethylsilyl chloride⁴ allowed stereoselective 1,4-addition to



Scheme 3 Reagents and conditions: i, $(Bu^{t}CO)_{2}O, 4-N, N$ -dimethylaminopyridine, Et₃N, CH₂Cl₂, room temp.; ii, K₂CO₃, MeOH, room temp.; iii, PCC, CH₂Cl₂, room temp.; iv, vinylmagnesium bromide, TMSCl, CuBr·Me₂S, THF-HMPA, -78 °C, then 5% HCl; v, LiAlH₄, THF, 0 °C

give the single ketone 17. On sequential reduction with sodium borohydride and lithium aluminium hydride 17 gave the *ent*-7, m.p. 132–133 °C, $[\alpha]_D^{30}$ –72.5 (*c* 0.48, MeOH), in satisfactory overall yield. This constitutes the synthesis of *ent*-(+)-eutypoxide B (*ent*-1) in the formal sense.

In summary, the present enantiodivergent approach is compatible with the synthesis of both enantiomers of eutypoxide B 1 though the absolute structure of the natural product has yet to be reported.

Received, 18th January 1993; Com. 3/00295K

References

- 1 J.-M. Renaud, G. Tsoupras, H. Stoeckli-Evans and R. Tabacchi, *Helv. Chim. Acta*, 1989, **72**, 1262.
- 2 E. Defrancq, J. Gordon, A. Brodard and R. Tabacchi, Helv. Chim. Acta, 1992, 75, 276.
- 3 S. Takano, M. Moriya, Y. Higashi and K. Ogasawara, J. Chem. Soc., Chem. Commun., 1993, 177.
- 4 Cf. Y. Horiguchi, S. Matsuzawa, E. Nakamura and I. Kuwajima, Tetrahedron Lett., 1986, 27, 4025.
- 5 Cf. S. Takano, K. Inomata and K. Ogasawara, Chem. Lett., 1989, 359.
- 6 Y. Kishi, M. Aratani, H. Tanino, T. Fukuyama, T. Goto, S. Inoue, S. Sugiura and H. Kakoi, J. Chem. Soc., Chem. Commun., 1972, 64.
- 7 Cf. C. Huynh, F. Derguini-Boumechal and G. Linstrumelle, Tetrahedron Lett., 1979, 1503.

615