Note

Partial tosylation of methyl *α*-D-mannopyranoside

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Selective esterification provides a facile synthetic route for the preparation of important intermediates in carbohydrate chemistry, and the order of relative reactivity of the hydroxyl groups in sugars is closely correlated to their stereo-chemistry^{1,2}. This communication describes partial tosylation of methyl α -D-manno-pyranoside (1).

Treatment of 1 with 2 molar equivalents of *p*-toluenesulfonyl chloride in pyridine gave a mixture which was fractionated by column chromatography on silica gel to afford the 2,3,6-trisulfonate 2 (3%), the 3,4,6-trisulfonate 3 (1%), the 3,6-di-sulfonate 4 (35%), a mixture of the 4,6- (5) and 2,6- (6) disulfonate (14%), and the 6-sulfonate 7 (35%).

The structure of 2 was demonstrated by preparing the monobenzoate, which was shown to be the 4-benzoate 8 by ¹H-n.m.r. spectroscopy. The signal of H-4 of 8 appeared to the lowest field of the ring-proton resonances, because of the deshielding effect of benzoyloxy groups on the same methine proton at C-4. The structure of 3 was also confirmed by comparison of the ¹H-n.m.r. spectra of 3 and its benzoate 9. The signal of H-2 in the spectrum of 9 appeared in lower field than that of H-2 in the spectrum of 3. This finding indicates that C-2 of 3 and 9 is substituted with a hydroxyl and a benzoyloxy group, respectively.

The structure of 4 was established by preparing the dibenzoate 10. The signal of H-3 in the ¹H-n.m.r. spectrum of 4 appeared at lower field than those of H-2 and H-4, whereas the signals of H-2 and H-4 in the ¹H-n.m.r. spectrum of 10 appeared at the lowest field of the ring-proton resonances. Compounds 5 and 6 showed the same R_F value on t.l.c., and could not be crystallized. Therefore, in order to determine the molar ratio of the disulfonates 5 and 6, a mixture of the two was benzoylated and the product analyzed by ¹H-n.m.r. spectroscopy. An authentic specimen of 11 was prepared by acetalation of 7, followed by tosylation, deacetalation, and benzoylation. The remaining disulfonate 12 was isolated by fractional crystallization from the benzoylated mixture of 11 and 12. The ¹H-n.m.r. spectrum of the resulting benzoylated mixture indicated the ratio of 11 and 12 in the mixture to be 3:7, based on the relative proton-intensities of the tosyloxy methyl group in the spectrum.

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The structure of 7 was confirmed by 1 H-n.m.r. spectroscopy. The signals of H-2, H-3, and H-4 in the spectrum of its benzoylated derivative 13 appeared at the lowest field of the ring-proton resonances. This finding indicates that C-6 of 7 is substituted with a tosyloxy group.

Trimolar tosylation of 1 gave a product mixture from which 2 (5%), 3 (2%), 4 (40%), a mixture of 5 and 6 (15%), and 7 (25%) were isolated, after column chromatography. The mixture of 5 and 6 was converted into the corresponding benzoyl derivatives 11 and 12 in the ratio of 1:3 (shown by ¹H-n.m.r. spectroscopy as described for dimolar tosylation).

In di- and tri-tosylation of 1, the highest yield of 4 in the disulfonates and the preponderance of 6 over 5 showed that the realtive reactivity of the hydroxyl groups in 1 towards *p*-toluenesulfonyl chloride is OH-6 > OH-3 > OH-2 > OH-4.

The highest reactivity among the secondary hydroxyl groups in 1, that of O-3, may be explained by intramolecular hydrogen-bonding between OH-3 and the (axial) OH-2 group. Interestingly, OH-2, which is axial, is more reactive than the equatorial OH-4 group. This finding is in good agreement with the foregoing observations³⁻⁵ that the hydroxyl groups participating in intramolecular hydrogen-bonding activate each other. The lowest reactivity that of the OH-4 group, is consistent with the results⁶ of trimolar methanesulfonylation of 1, which affords the 2,3,6-trimethanesulfonate as the major product in 41% yield.

Partial tosylation of the 3,6-disulfonate 4 gave further information on the relative reactivity of the hydroxyl groups in 1. No reaction occurred when 4 was subjected to mono- and di-tosylation; however, addition of a large excess of *p*-toluenesulfonyl chloride resulted in formation of the 2,3,4,6-tetrasulfonate, the 2,3,6-trisulfonate, the 3,4,6-trisulfonate, and the starting material in the molar ratios of 3:5:2:1. This result indicates that the low yields of the trisulfonates (7%) in tritosylation of 1 may be due to the retardation effect as described^{4,7} previously in tosylation. However, it seems that isolated hydroxyl groups in the trisulfonates are not so sterically hindered, judging from the yield of the tetrasulfonate in tosylation of 4.

ç	H ₂ OTs
	0
	R'O
R ³ O	OMe

	R ¹	R ²	R ³
2	Ts	Ts	н
3	н	Ts	Ts
4	Н	Ts	Н
5	Н	н	Ts
6	Ts	н	н
7	н	н	Н
8	Ts	Ts	Bz
9	Bz	Ts	Ts
10	Bz	Ts	Bz
11	Bz	Bz	Ts
12	Ts	Bz	Bz
13	Bz	Bz	Bz

EXPERIMENTAL

General methods. — Melting points were determined on a Yanagimoto hotstage microscope and are uncorrected. Optical rotations were measured with a Jasco DIP-181 digital polarimeter for solutions in chloroform, and ¹H-n.m.r. spectra were recorded with a Hitachi R-24 60 MHz instrument for solutions in chloroform-d with tetramethylsilane as the internal reference. T.l.c. was performed on TLC silica gel 60 G (Merck), and column chromatography on silica gel 60 (70-230 mesh, Merck).

Partial tosylation of methyl α -D-mannopyranoside (1). — (a) With two molar equivalents. p-Toluenesulfonyl chloride (2.16 g, 2.2 mol, equiv.) was added portionwise to a stirred solution of 1 (1 g) in dry pyridine (40 mL) at 0°. The mixture was kept for 24 h at 0°, stirred for 40 h at 5°, and then for 48 h at room temperature. The mixture was extracted with dichloromethane, and the extract was successively washed with dilute hydrochloric acid, saturated sodium hydrogencarbonate and water, dried (sodium sulfate), and evaporated to a syrup which was chromatographed on silica gel (250 g).

Elution with 9:1 benzene-acetone gave the syrupy 2,3,6-trisulfonate 2 (90 mg, 3%); $[\alpha]_D^{2^4} + 4.6^\circ$ (c 1.7) and the syrupy 3,4,6-trisulfonate 3 (38 mg, 1%); $[\alpha]_D^{2^2} + 59.2^\circ$ (c 2.6); ¹H-n.m.r.: δ 4.82 (t, 1 H, $J_{3,4} = J_{4,5} = 9$ Hz, H-4), 4.60 (q, 1 H, $J_{2,3}$ 3 Hz, H-3), 3.25 (s, 3 H, CH₃O), and 2.40 (s, 9 H, 3 C₆H₄CH₃).

Elution with 4:1 benzene-acetone afforded the syrupy 3,6-disulfonate 4 (893 mg, 35%); $[\alpha]_D^{21}$ +37.0° (c 0.7); ¹H-n.m.r.: δ 4.57 (d, 1 H, $J_{1,2}$ 2 Hz, H-1), 4.53 (q, 1 H, $J_{2,3}$ 4, $J_{3,4}$ 9 Hz, H-3), 3.22 (s, 3 H, CH₃O), and 2.38 (s, 6 H, 2 C₆H₄CH₃). Further elution with 4:1 benzene-acetone gave a mixture of the 4,6- (5) and 2,6- (6) disulfonate (349 mg, 14%) as a syrup.

Final elution with 2:1 benzene-acetone gave the foamy 6-*p*-toluenesulfonate 7 (626 mg, 35%); $[\alpha]_D^{24}$ +37.4° (*c* 1.1); ¹H-n.m.r.: δ 4.60 (s, 1 H, $J_{1,2}$ 0 Hz, H-1), 3.23 (s, 3 H, CH₃O), and 2.38 (s, 3 H, C₆H₄CH₃).

(b) With three molar equivalents. Treatment of 1 (1 g) with p-toluenesulfonyl chloride (2.945 g, 3.0 mol. equiv.) in pyridine (40 mL), followed by chromatographic fractionation of the product on silica gel as described in (a), afforded five fractions: 2 (176 mg, 5%), 3 (76 mg, 2%), 4 (1.032 g, 40%), 5 + 6 (388 mg, 15%), and 7 (449 mg, 25%).

Methyl 4-O-benzoyl-2,3,6-tri-O-p-tolylsulfonyl- α -D-mannopyranoside (8). — Benzoylation of the trisulfonate 2 (90 mg) with benzoyl chloride (0.1 mL) in pyridine (2 mL) at 0° gave a crystalline solid (95 mg, 91%) which, on recrystallization from ethanol, yielded the title compound 8; m.p. 150–151°, $[\alpha]_D^{24}$ +8.4° (*c* 0.9); ¹H-n.m.r.: δ 5.10 (t, 1 H, $J_{3,4} = J_{4,5} = 9$ Hz, H-4), 3.38 (s, 3 H, CH₃O), and 2.42, 2.33, and 2.18 (3 s, 9 H, 3 C₆H₄CH₃).

Anal. Calc. for C₃₅H₃₆O₁₃S₃: C, 55.25; H, 4.77; S, 12.64. Found: C, 55.11; H, 4.70; S, 12.44.

Methyl 2-O-benzoyl-3,4,6-tri-O-p-tolylsulfonyl- α -D-mannopyranoside (9). —

The trisulfonate **3** (77 mg) was treated with benzoyl chloride (0.1 mL) in pyridine (2 mL) to afford the title compound **9** (88 mg, 99%). Attempts to crystallize it failed; $[\alpha]_{D}^{24}$ -14.8° (c 0.2); ¹H-n.m.r.: δ 5.03 (q, 1 H, $J_{2,3}$ 2, $J_{3,4}$ 8 Hz, H-3), 4.92 (broad s, 1 H, H-2), 4.62 (d, 1 H, $J_{1,2}$ 2 Hz, H-1), 3.27 (s, 3 H, CH₃O), and 2.40, 2.37, and 2.32 (3 s, 9 H, 3 C₆H₄CH₃).

Methyl 2,4-di-O-*benzoyl-3,6-di*-O-p-*tolylsulfonyl-α*-D-mannopyranoside (10). — Treatment of the disulfonate **4** (183 mg) with benzoyl chloride followed by crystallization from ethanol gave the title compound **10** (154 mg, 60%); m.p. 85°, $[\alpha]_D^{26}$ -31.5° (c 0.9); ¹H-n.m.r.: δ 5.58 (t, 1 H, $J_{3,4} = J_{4,5} = 9$ Hz, H-4), 5.37 (q, 1 H, $J_{1,2}$ 2, $J_{2,3}$ 4 Hz, H-2), 5.17 (q, 1 H, H-3), 4.80 (d, 1 H, H-1), 6.38 (s, 3 H, CH₃O), and 5.28, and 5.18 (2 s, 6 H, 2 C₆H₄CH₃).

Anal. Calc. for $C_{35}H_{34}O_{12}S_2$: C, 59.15; H, 4.82; S, 9.02. Found: C, 58.98; H, 4.74; S, 9.16.

Methyl 2,3-di-O-benzoyl-4,6-di-O-p-tolylsulfonyl- α -D-mannopyranoside (11). — A solution of the 6-sulfonate 7 (100 mg) in N,N-dimethylformamide (1 mL) containing p-toluenesulfonic acid (4 mg) was heated for 1.5 h at 70° with 2,2-dimethoxypropane (1 mL). The mixture was extracted with dichloromethane and the extract was evaporated to a syrup. O-Deacetonation in 70% acetic acid (10 mL), followed by column-chromatographic purification of the product afforded the syrupy 4,6-disulfonate 5 (73 mg, 51%); $[\alpha]_D^{22}$ +53.3° (c 2.1). Benzoylation of 5 (62 mg) with benzoyl chloride in pyridine yielded the title compound 11 (100 mg, 97%), which could not be crystallized; $[\alpha]_D^{25}$ -79.7° (c 2.4); ¹H-n.m.r.: δ 5.73 (q, 1 H, $J_{2,3}$ 3, $J_{3,4}$ 9 Hz, H-3), 5.53 (q, 1 H, $J_{1,2}$ 1 Hz, H-2), 5.23 (t, $J_{4,5}$ 9 Hz, H-4), 4.80 (d, 1 H, H-1), 3.42 (s, 3 H, CH₃O), and 2.42 ad 2.15 (2 s, 6 H, C₆H₄CH₃).

Methyl 3,4-di-O-benzoyl-2,6-di-O-p-tolylsulfonyl- α -D-mannopyranoside (12). — A mixture of 5 and 6 (87 mg) in pyridine was treated with benzoyl chloride. Conventional processing, followed by two recrystallizations from chloroformethanol, gave the title compound 12 (27 mg, 22%); m.p. 186–187°, $[\alpha]_D^{25}$ +21.3° (c 0.9); ¹H-n.m.r.: δ 3.40 (s, 3 H, CH₃O), and 2.35 and 2.08 (2 s, 6 H, 2 C₆H₄CH₃).

Anal. Calc. for $C_{35}H_{34}O_{12}S_2$: C, 59.15; H, 4.82; S, 9.02. Found: C, 59.03; H, 4.68; S, 9.17.

Methyl 2,3,4-tri-O-benzoyl-6-O-p-tolylsulfonyl- α -D-mannopyranoside (13). — Treatment of the 6-sulfonate 7 (100 mg) with benzoyl chloride in pyridine yielded a solid which was recrystallized from ethanol to give the title compound 13 (163 mg, 86%); m.p. 205–206°, $[\alpha]_D^{18}$ –105.8° (c 2.8); ¹H-n.m.r.: δ 6.0–5.5 (m, 2 H, H-3,4), 5.57 (broad q, 1 H, H-2), 4.83 (d, 1 H, $J_{1,2}$ 1 Hz, H-1), 3.43 (s, 3 H, CH₃O), and 2.28 (s, 3 H, C₆H₄CH₃).

Anal. Calc. for C₃₅H₃₂O₁₂S: C, 63.63; H, 4.88; S, 4.85. Found: C, 63.42; H, 4.86; S, 5.03.

Partial tosylation of methyl 3,6-di-O-p-tolylsulfonyl- α -D-mannopyranoside (4). — Compound 4 (103 mg) was tosylated for 4 days at room temperature by daily addition of 48 mg (1.2 mol. equiv.) of p-toluenesulfonyl chloride. After conventional processing, column-chromatographic fractionation gave the syrupy 2,3,4,6-tetrasulfonate (37 mg, 23%); $[\alpha]_{D}^{28}$ +20.5° (c 1.9), the 2,3,6-trisulfonate 2 (50 mg, 37%), the 3,4,6-trisulfonate 3 (19 mg, 14%), and the starting material 4 (8 mg, 8%).

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