Conjugate Addition of Tris(phenylthio)methyl-lithium to $\alpha\beta$ -Unsaturated Ketones. Synthesis of γ -Keto-esters

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Summary Conjugate addition of (1) to unhindered $\alpha\beta$ unsaturated ketones proceeds in good yield to produce γ -keto-orthothioesters.

Carbanions, stabilized by adjacent sulphur atoms, have been extensively used in organic synthesis. Their reaction with $\alpha\beta$ -unsaturated ketones generally results in carbonyl (1,2) addition rather than conjugate (1,4) addition. We report that tris(phenylthio)methyl-lithium (1) reacts in a conjugate fashion with unhindered $\alpha\beta$ -unsaturated ketones producing γ -keto-orthothioesters which are in turn readily hydrolysed to γ -keto-esters. Thus the anion of reagent (1) can be considered as an ester carbanion equivalent (i.e. ROC=0).

Triphenyl orthothioformate $(2)^3$ is rapidly deprotonated to the anion $(1)^{3-5}$ by treatment with n-butyl-lithium in tetrahydrofuran at -78 °C under nitrogen. Addition of cyclohex-2-enone (1 equiv.) to this solution followed by hydrolytic workup affords compound (3) in good yield

(Table) which is converted into the ester (4)⁶ (95%) by Hg²⁺ catalysed methanolysis⁷ followed by acid treatment.⁸

TABLE

Reaction of $\alpha\beta$ -unsaturated ketones with (1).

Substrate		Yield (%)
Cyclohex-2-enone		95, 50b,c
2-Methylcyclohex-2-enone	٠.	85
5,5-Dimethylcyclohexenone		65
PhCH=CHCOMe		60
PhCH=CHCOPh		95
$PhCH = CHCOCMe_3$		85
MeCH=CHCOMe		65
3-Methylcyclohex-2-enone		$<$ 5 $^{\mathrm{b}}$
$Me_2C = CHCOMe$		$<$ 5 $^{\mathrm{b}}$

^a Isolated yield of γ -keto-orthothioester unless otherwise stated. ^b Determined by n.m.r. analysis of the crude reaction product mixture. ^c Using the sodium salt (5).

(PhS)3 CX

- (1) X = Li
- X = H(2)
- (5) X = Na

Reduction of (3) with Raney nickel gives 3-methylcyclohexanone (70%).

Results of the reaction of (1) with various $\alpha\beta$ -unsaturated ketones are listed in the Table. The yields of γ-ketoorthothioesters are satisfactory except for hindered $\beta\beta$ disubstituted enones. The use of the sodium salt (5), prepared from (2) and sodium bistrimethylsilylamide,9 gives significantly lower yields of orthothioester. Reaction of (1) with unsaturated aldehydes gives products resulting from 1,2 addition.10

The ready availability of (2) combined with the variety of possible transformations¹¹ of the orthothioester unit make (1) and related compounds potentially useful synthetic reagents.

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- 10 For the reaction of (1) with saturated aliphatic and aromatic aldehydes see ref. 4.
 11 E.g. R. H. DeWolfe, 'Carboxylic Ortho Acid Derivatives,' Academic Press, New York, 1970, ch. 6.