

Alkylation Studies; Part II¹: Bis-alkylation of Diethyl Cyanomethanephosphonate

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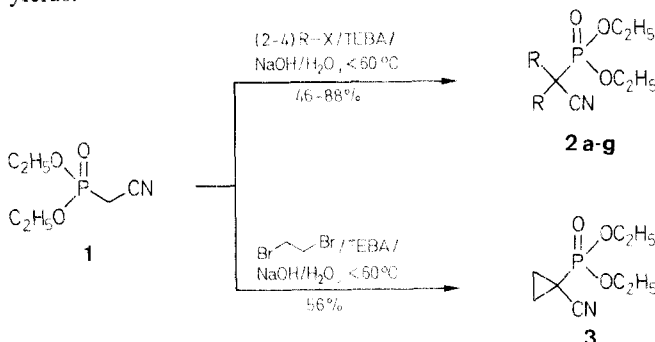
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A convenient high-yield method for the geminal dialkylation of diethyl cyanomethanephosphonate with alkyl halides under ion-pair extraction conditions is described.

The monoalkylation of diethyl cyanomethanephosphonate (**1**) using the ion-pair extraction technique³ has been described² and it has been shown^{4,5} that cyanomethanephosphonic tetramethyldiamide can be monoalkylated under phase-transfer and under ion-pair extraction conditions. The synthesis of differently dialkylated cyanomethanephosphonic tetramethylamides has also been reported^{4,5}. A patent⁶ describes the synthesis of dibenzyl diethylcyanomethanephosphonate using DBU as a base. However, no other examples of bis-alkylation of diethylcyanomethanephosphonate (**1**) have thus far been reported.

We have already reported^{1,7} that diethyl malonate can be cleanly bisalkylated under ion-pair extraction conditions. In order to define the scope and reactivity of phosphonate-activated methylene compounds, the bis-alkylation of diethyl cyanomethanephosphonate (**1**) was studied. Thus, it

was found that compound **1** can be cleanly bis-alkylated by using primary, allylic, and benzylic halides under ion-pair extraction conditions. The reaction is complete in 1–3 hours and the usual work-up gives 46–88% yields of bis-alkylated phosphonates **2**. No attempt was made to optimize the yields.

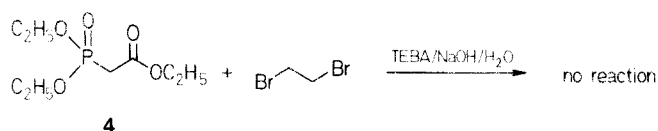


The reaction of diethyl cyanomethanephosphonate (**1**) with 1,2-dibromoethane gave diethyl 1-cyanocyclopropane-phosphonate (**3**) in 56% yield. When cyclopropanation of **1** was attempted using a catalytic amount of TEBA (~ 5%) or without catalyst, no appreciable amount of **3** was formed. Contrary to the cyano derivative, triethylphosphonoacetate (**4**) could not be cyclopropanated under these conditions. The starting material was not recovered. The ester group presumably hydrolyzed before the alkylation step.

Table. Bis-alkylation of Diethyl Cyanomethanephosphonate (**1**)

Alkylating Agent R—X	Ratio 1: R—X	Reaction Time [h]	Product	Yield [%]	b. p. [°C]/torr	Molecular Formula ^a	¹ H-NMR (CDCl ₃) δ [ppm]
H ₃ C—I	1:4	3	2a	80	100/0.8	C ₈ H ₁₆ NO ₃ P (205.2)	1.38 (t, 6H, J _{H,H} = 7 Hz); 1.51 (d, 6H, J _{P,CH} = 15 Hz); 4.25 (2q, 4H, J _{H,H} = 7 Hz)
C ₂ H ₅ —I	1:3	1	2b	71	112/1.0	C ₁₀ H ₂₀ NO ₃ P (233.2)	1.0–1.4 (2t, 12H); 1.6–2.05 (2m, 4H); 4.25 (2q, 4H, J _{H,H} = 7 Hz)
H ₂ C=CH—CH ₂ —Br	1:3	1	2c	88	130–132/1.2	C ₁₂ H ₂₀ NO ₃ P (257.3)	1.36 (t, 6H, J _{H,H} = 7 Hz); 2.4–2.78 (d of d, 4H, J _{H,H} = 7 Hz, J _{P,CH} = 14 Hz); 4.25 (2q, 4H, J _{H,H} = 7 Hz); 5.1–6.3 (m, 6H)
H ₂ C=C(CH ₃)—CH ₂ —Cl	1:2.5	1	2d	72	121–125/0.8	C ₁₄ H ₂₄ NO ₃ P (285.3)	1.36 (t, 6H, J _{H,H} = 7 Hz); 1.92 (s, 6H); 2.5–2.7 (ABX, d, 2H, J _{H,H} = 4 Hz, s, 2H, J _{P,CH} = 14 Hz); 4.25 (2q, 4H, J _{H,H} = 7 Hz); 5.98 (s, 4H)
(CH ₃) ₂ C=CH—CH ₂ —Cl	1:2.5	1	2e	83	146–149/1.2	C ₁₆ H ₂₈ NO ₃ P (313.4)	1.36 (t, 6H, J _{H,H} = 7 Hz); 1.65 (s, 6H); 1.75 (s, 6H); 2.38–2.75 (2d, 4H, J _{P,CH} = 14 Hz, J _{H,H} = 7 Hz); 4.25 (2q, 4H, J _{H,H} = 7 Hz); 5.30 (t, 2H, J = 7 Hz)
C ₆ H ₅ —CH ₂ —Cl	1:2	1	2f	46	208/0.8	C ₂₀ H ₂₄ NO ₃ P (357.4)	1.25 (t, 6H, J _{H,H} = 7 Hz); 2.9–3.25 (m, 4H); 4.00 (2q, 4H, J _{H,H} = 7 Hz); 7.25 (br. s, 10H)
4-F—C ₆ H ₄ —CH ₂ —Cl	1:2	1	2g	70	198/0.8	C ₂₀ H ₂₂ F ₂ NO ₃ P (393.4)	1.20 (t, 6H, J _{H,H} = 7 Hz); 2.8–3.5 (m, 4H); 4.05 (2q, 4H, J _{H,H} = 7 Hz); 6.8–7.2 (m, 8H)
Br—CH ₂ —CH ₂ —Br	1:2	1	3	56	100–104/0.6	C ₈ H ₁₄ NO ₃ P (203.2)	1.40 (t, 6H, J _{H,H} = 7 Hz); 1.60 (m, 4H); 4.25 (2q, 4H, J _{H,H} = 7 Hz)

^a The microanalyses were in good agreement with the calculated values: C ± 0.25, H ± 0.07, N ± 0.15.



Further reactions of diethyl 1-cyanocyclopropanephosphonate (**3**) toward nucleophilic ring opening are in progress.

Microanalyses were performed by Atlantic Microlab, Inc., Atlanta, Georgia. NMR spectra were recorded on Varian T-60.

Diethyl 4-Cyano-1,6-heptadiene-4-phosphonate (2c); Typical Procedure:

To a 12.5 normal solution (100 ml) of sodium hydroxide (prepared by dissolving 50 g sodium hydroxide in water and diluting to a total volume of 100 ml), benzyltriethylammonium chloride (TEBA; 22.75 g, 0.1 mol) is added with vigorous stirring. To this solution, a mixture of diethyl cyanomethanephosphonate (**1**; 17.7 g, 0.1 mol) and 3-bromopropene (36.3 g, 0.3 mol) is added. When the reaction temperature has reached $\sim 60^\circ\text{C}$, water-bath cooling is applied. The stirring is continued for 1 h, and the mixture is then diluted with water (100 ml) and extracted with ether (4×100 ml). The combined ether solution is washed with water (3×50 ml) and with saturated sodium chloride solution (50 ml), and dried with magnesium sulfate. The solvent is removed, and the residue distilled in vacuum; yield: 22.53 g (88%); b.p. $130\text{--}132^\circ\text{C}/1.3$ torr.

$\text{C}_{12}\text{H}_{20}\text{NO}_3\text{P}$	calc.	C 56.02	H 7.84	N 5.44
(257.3)	found	56.27	7.89	5.39

Received: June 21, 1985

(Revised form: December 18, 1985)

¹ For Part I, see: Singh, R. K. *Synthesis* **1985**, 54.

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⁶ Oediger, H. *German Patent (DOS) 2 206 778* (1973). Bayer AG; *C. A.* **1973**, 79, 115163.

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