

Die Spaltungsuntersuchungen wurden unter „Standardbedingungen“<sup>(1)</sup> mit Piperidin, Benzylamin und Benzylmercaptan vorgenommen. Die Ausbeuten beziehen sich auf die nach Abzug des nicht gespaltenen Anteils maximal mögliche Umsetzung als 100 %-Wert.

Anmerkung zu den Tabellen:

IR = angegeben sind NH und Amid Banden; Et = absol. Ester; Pe = absol. Petrolether; Ee = Essigsäureethylester; Di = Dichlormethan; Cyclo = Cyclohexan; t-But = tert. Butyl-Rest; i-Prop = Isopropyliden-Rest.

## Literatur

- 1 H. G. Schweim, Arch. Pharm. (Weinheim) 319, 814 (1986).
- 2 H. G. Schweim, Arch. Pharm. (Weinheim) 320, 430 (1987).
- 3a B. S. Drach, J. Y. Dolgushina, A. D. Sinita and A. V. Kirisanov, Zh. Obshch. Khim. 42, 1240 (1972); C. A. 77, 100361b (1972).
- 3b P. Grammaticakis, Bull. Soc. Chim. France 1947, 664.
- 4 T. Kibbel, Dissertation Hamburg 1986.
- 5 H. Poisel, Chem. Ber. 110, 948 (1977).
- 6 I. Malassa, Dissertation Hamburg 1985.
- 7 A. J. Speziale und L. R. Smith, J. Org. Chem. 27, 3742 (1962).
- 8 M. Kilner und J. N. Pinkney, J. Chem. Soc. A, 1971, 2891.

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## Studies with Heterocyclic Hydrazidic Halides: New Pyrazolo[5,1-c]-1,2,4-triazoles and Pyrazolo[5,1-c][1,2,4]triazines

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Syntheses of the title compounds using the hydrazone **3** as starting material are reported.

### Untersuchungen mit Heterocyclischen Hydrazidhalogeniden: Neue Pyrazolo[5,1-c]-1,2,4-triazole und Pyrazolo[5,1-c][1,2,4]triazine

Synthesen der Titelverbindungen aus dem Hydrazone **3** werden beschrieben.

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Interest in condensed pyrazoles has been recently revived<sup>(1)</sup>. The considerable biological activities of these compounds are most likely beyond this interest<sup>(2)</sup>. In previous work we have reported several new approaches for the synthesis of fused pyrazoles utilizing readily obtainable functionally substituted pyrazo-

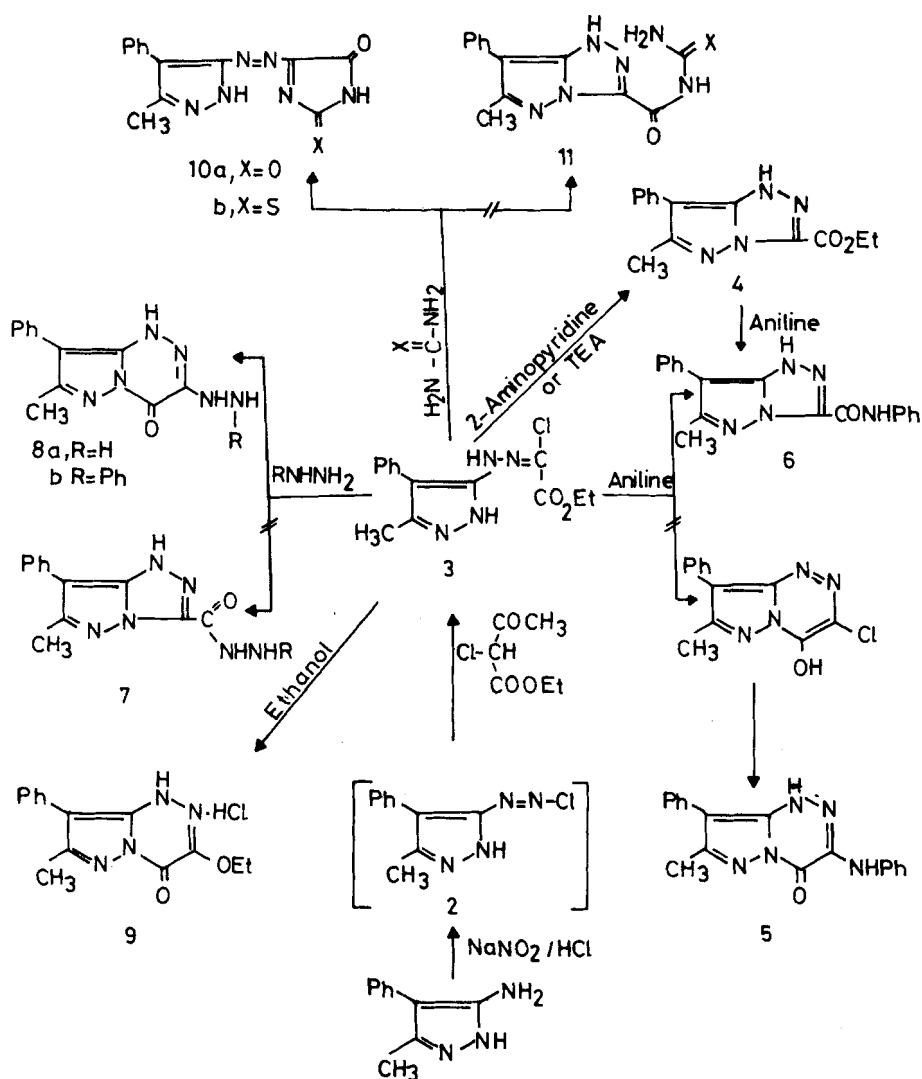
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les as starting materials<sup>1, 3</sup>). Although the ethoxycarbonylmethano-(3-methyl-4-phenylpyrazol-5-yl)hydrazonyl chloride (**3**) is an interesting starting material for the synthesis of fused pyrazoles<sup>4, 5</sup>), the chemistry of these new pyrazoles has received little attention compared to the broad synthetic utilities of its aryl counter analogues.

Here we report on the synthesis of ethoxycarbonylmethano-(3-methyl-4-phenylpyrazol-5-yl)hydrazonyl chloride (**3**) and its utility in synthesis of fused pyrazoles. Thus, diazotized 5-amino-3-methyl-4-phenylpyrazole (**1**) couples with ethyl  $\alpha$ -chloroacetate to yield **3**, formed via coupling and *Japp-Klingemann* acyl group cleavage.

The pyrazolo[5,1-*c*]-1,2,4-triazole **4** was obtained by cyclization of **3** under basic conditions.



When **3** was heated with aniline, a product which may be formulated as **5** or its isomer **6** was isolated. Structure **6** was established for this product based on its identity with a specimen prepared from **4** with aniline.

**3** reacted with hydrazine hydrate and phenylhydrazine to yield products that can be formulated as **7** or its isomer **8**. Structure **8** is assumed for the reaction product as it could not be obtained from the ester **4**. A similar reaction of hydrazines with pyrazol-5-ylhydrazonyl derivatives leading to pyrazolo[5,1-*c*]1,2,4-triazines has been reported<sup>4)</sup>.

**3** was converted into a product  $C_{14}H_{15}N_4O_2 \cdot Cl$  on long reflux in EtOH. The  $^1H$ -NMR spectrum revealed a pattern very similar to that of the starting hydrazonyl chloride. So, structure **9** was suggested.

**3** reacted with urea and thiourea to yield a product that can be formulated as **10** or **11**. Structure **11** was eliminated as urea failed to react with **4** under similar conditions. The proposed structures are in agreement with analytical and spectra data.

## Experimental Part

Melting points: uncorrected. – IR spectra (KBr): Pye-Unicam SP 1000. –  $^1H$ -NMR spectra: EM-390 90 MHz spectrometer,  $Me_4Si$  as internal standard, chemical shifts as  $\delta$  (ppm). Analytical data: Analytical Data Unit at Cairo University.

### *Ethoxycarbonylmethano-(3-methyl-4-phenylpyrazol-5-yl)hydrazonyl chloride (3)*

Ethyl  $\alpha$ -chloroacetoacetate (0.01 mole) in EtOH (100 ml) was treated with anhydrous sodium acetate (0.02 mole) and then cooled to 0°. To this solution a solution of diazotized **1** (prepared at 0° from 0.01 mole of **1**, 25 ml of HCl and the appropriate quantity of  $NaNO_2$ ) was added. The mixture was stirred at 0° for 1 h. The solid yellow product, so formed, was collected and crystallized from water; yield 80 %, m.p. 100°. –  $C_{14}H_{15}ClN_4O_2$  (306.5) Calc. C 54.7 H 4.93 N 18.2 Cl 11.5 Found C 54.2 H 4.91 N 18.0 Cl 11.3. – IR: 3560, 3180 (NH), 1700 (CO), 1635  $cm^{-1}$  (C=N). –  $^1H$ -NMR:  $\delta$  (ppm) = 1.2 (t, 3H,  $CH_3$ ); 3.3 (s, 3H,  $CH_3$ ), 4.2 (q, 2H,  $CH_2$ ), 7.4 (m, 7H,  $C_6H_5$  and 2NH).

### *6-Ethoxycarbonyl-2-methyl-3-phenylpyrazolo[5,1-*c*]1,2,4-triazole (4)*

**3** (0.01 mole) in EtOH (50 ml) and either 2-aminopyridine (0.01 mole) or triethylamine (1 ml) were refluxed for 2 h, evaporated in vacuo and then triturated with water. The resulting solid yellow product was crystallized from EtOH; yield 70 %, m.p. 154°. –  $C_{14}H_{13}N_4O_2$  (269.1) Calc. C 62.4 H 4.86 N 20.8 Found C 62.0 H 4.93 N 20.5. – IR: 3460, 3340 (NH), 1710 (CO), 1640  $cm^{-1}$  (C=N). –  $^1H$ -NMR:  $\delta$  (ppm) = 1.4 (t, 3H,  $CH_3$ ), 3.3 (s, 3H,  $CH_3$ ), 4.5 (q, 2H,  $CH_2$ ), 7.2–7.5 (m, 6H,  $C_6H_5$  and NH).

#### *Reaction of 3 with aniline:*

### *6-Benzamido-2-methyl-3-phenylpyrazolo[5,1-*c*]1,2,4-triazole (6)*

**3** (0.01 mole) and aniline (0.01 mole) were heated at 100° (bath temp.) for 3 h, then triturated with EtOH. The solid yellow product was crystallized from EtOH/DMF; yield 70 %, m.p. > 300°. –  $C_{18}H_{15}N_5O$  (317.2) Calc. C 68.1 H 4.76 N 22.0 Found C 67.8 H 5.02 N 21.9. – IR: 3335, 3140 (NH), 1680 (CO), 1620  $cm^{-1}$  (C=N).

The same product **6** (m.p. and mixed m.p.) was isolated by refluxing **4** (0.01 mole) with aniline (0.01 mole) in EtOH (50 ml) for 5 h.

*Reaction of 3 with hydrazines, urea and thiourea*

**3** reacted with hydrazine hydrate, phenylhydrazine, urea and thiourea following the procedure described above for the reaction of **3** with aniline.

*6-Hydrazino-2-methyl-3-phenylpyrazolo[5,1-c][1,2,4]triazin-7-one (8a)*

Brown crystals from EtOH/H<sub>2</sub>O; yield 65 %, m.p. 215°. – C<sub>12</sub>H<sub>12</sub>N<sub>6</sub>O (256.1) Calc. C 56.2 H 4.72 N 32.8 Found C 56.8 H 4.46 N 32.5. – IR: br (3300–3160) (NH), 1710 (CO), 1620 cm<sup>-1</sup> (C=N).

*6-Phenylhydrazino-2-methyl-3-phenylpyrazolo[5,1-c][1,2,4]triazin-7-one (8b)*

Yellow crystals from DMF/H<sub>2</sub>O; yield 80 %, m.p. > 300°. – C<sub>18</sub>H<sub>16</sub>N<sub>6</sub>O (332.2) Calc. 65.0 H 4.85 N 25.3 Found C 65.6 H 4.51 N 25.0. – IR: 3300, 3120 (NH), 1670 (CO), 1620 cm<sup>-1</sup> (C=N).

*5-(2,5-Dioxo-imidazol-4-yl)azo-2-methyl-3-phenylpyrazole (19a)*

Yellow crystals from EtOH; yield 60 %, m.p. > 300°. – C<sub>13</sub>H<sub>10</sub>N<sub>6</sub>O<sub>2</sub> (282.1) Calc. C 55.3 H 3.57 N 29.7 Found C 56.1 H 3.28 N 29.5. – IR: 1620 (C=N), 1700 (CO), 3020, 3200, 3175 cm<sup>-1</sup> (NH).

*5-(5-Oxo-imidazol-2-thion-4-yl)azo-2-methyl-3-phenylpyrazole (10b)*

Brown crystals from EtOH; yield 65 %, m.p. 205°. – C<sub>13</sub>H<sub>10</sub>N<sub>6</sub>OS (298.2) Calc. C 52.3 H 3.38 S 10.7 Found C 52.0 H 3.07 S 10.5. – IR: 3200 (NH), 1700 (CO), 1600 cm<sup>-1</sup> (C=N).

*6-Ethoxy-2-methyl-3-phenylpyrazolo[5,1-c][1,2,4]triazin-4-one hydrochloride (9)*

**3** (1 g) in 50 ml EtOH was refluxed for 2 h and the mixture was evaporated in vacuo. The solid colourless product, so formed, was crystallized from EtOH; yield 70 %, m.p. 140°. – C<sub>14</sub>H<sub>15</sub>N<sub>4</sub>O<sub>2</sub> · Cl (306.6) Calc. C 54.7 H 4.93 Cl 11.5 Found C 54.3 H 5.22 Cl 11.6. – IR: 3200, 3080 (NH), 1720 (CO), 1650 cm<sup>-1</sup> (C=N). – <sup>1</sup>H-NMR: δ (ppm) = 1.2 (t, 3H, CH<sub>3</sub>), 3.3 (s, 3H, CH<sub>3</sub>), 4.2 (q, 2H, CH<sub>2</sub>), 7.3 (m, 6H, C<sub>6</sub>H<sub>5</sub> and NH).

**References**

- 1 E. M. Zayed, S. A. S. Ghozlan, and A. A. H. Ibrahim, *Monatsh. Chem.* **115**, 431 (1984).
- 2 R. H. Springer, M. B. Seholten, D. E. O'Brien, T. Novinson, J. P. Miller, and R. K. Robins, *J. Med. Chem.* **25**, 235 (1982).
- 3 S. A. S. Ghozlan, E. M. Zayed, and M. H. Elnagdi, *Gazz. Chim. Ital.* **113**, 219 (1983).
- 4 M. H. Elnagdi, E. M. Zayed, M. A. E. Khalifa, and S. A. S. Ghozlan, *Monatsh. Chem.* **112**, 245 (1981).
- 5 E. A. A. Hafez, N. M. Abed, I. A. El Sakka, and M. H. Elnagdi, *J. Heterocyclic Chem.* **20**, 285 (1983).