Formation of Four Different Aromatic Scaffolds from Nitriles through Tandem Divergent Catalysis**

Ju Hyun Kim, Jean Bouffard, and Sang-gi Lee*

Abstract: A zinc bromide complex, formed by the sequential reaction of nitriles with a Reformatsky reagent and terminal alkynes, is used as an intermediate for divergent palladiumcatalyzed reactions. The reaction pathway of the intermediate is precisely controlled by the choice of the reaction solvent or the palladium catalyst to quickly form four different aromatic scaffolds—arylamines, aminoindenes, pyrroles, and quinolines—starting from readily available nitriles.

Divergent catalytic reactions provide quick access to structurally different compounds from a common precursor through controlled reaction pathways, and are highly attractive tools in the discovery of drugs and functional materials.^[1,2] A more promising, yet challenging strategy that remains largely unexplored is tandem divergent catalysis, which combines the key advantages inherent to both tandem reactions^[3] and divergent catalysis to provide a rapid access to different structures from the same simple reagents while minimizing the generation of waste. In the course of our studies on the tandem use of the Blaise reaction in catalysis,^[4] we envisoned that the zinc bromide complex A, formed by the sequential reaction of nitriles with a Reformatsky reagent and 1-alkynes,^[5] may serve as a viable intermediate for divergent catalysis involving selective C-C and C-N bond-forming reactions. Herein, we report a strategy based on divergent palladium catalysis that provides four distinct compound classes-arylamines, aminoindenes, pyrroles, and quinolines—from simple nitriles (Scheme 1).^[6]

Our investigations began with the Pd-catalyzed intramolecular C–C bond-forming reactions of A1 ($R^1 = o$ -C₆H₄Br, $R^2 = Ph$), formed by the sequential reaction of 2-bromobenzonitrile, a Reformatsky reagent, and 1-phenylacetylene. When the intermediate A1 was reacted in the presence of a catalytic amount of [Pd(PPh₃)₄] in DMF, 1-aminonaphthalene **3a** was isolated in 22% yield (entry 1, Table S1).^[7,8] A cursory inspection of the skeletal framework of intermediate A1 and 1-aminonaphthalene **3a** showed that an unusual 1,2 migration of a C–C bond had occurred.

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Scheme 1. A tandem strategy for the divergent conversion of nitriles to arylamines, aminoindenes, pyrroles, and quinolines.

Optimization of the reaction parameters (Table S1) showed that the catalytically active Pd⁰ species is best obtained from Pd(OAc)₂ (5 mol%) and PPh₃ (15 mol%), and that the outcome of the tandem catalytic reaction is determined by the reaction solvent. Ultimately, the choice of *N*-methyl-2-pyrrolidinone (NMP) in the presence of 2.0 equiv of Bu₄NI at 135 °C for 6 h was optimal for the tandem synthesis of **3a** (78% overall yield, Scheme 2a). A significant finding is the formation of aminoindenene **4a** when the tandem reaction is conducted in DMF/H₂O. Rather than merely accelerating the reduction of Pd(OAc)₂ to Pd⁰,^[9] the tandem catalytic reaction pathways of **A1** in this solvent mixture are redirected to give the hydrodehalogenated aminoindene **4a** in 72% yield (Scheme 2b).^[10]

These results point toward a Heck-type 5-*exo*-trig carbopalladation, giving the σ -bonded complex **B** as a second



Scheme 2. Pd/solvent-controlled selective tandem synthesis of 1-aminonaphthalene 3 a and indenes 4 a and 5 a from a common nitrile 1 a.

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Scheme 3. Proposed reaction pathways for the formation of 3a in NMP and 4a in DMF/H₂O.

common intermediate,^[11] which may be in equilibrium with its imine tautomer C (Scheme 3). In the absence of β -hydrogen atoms, a β -carbon cleavage from **C** (C1–C2 bond cleavage), which affords the stabilized palladium enolate, is among plausible reaction pathways.^[12-13] The latter would undergo a subsequent 6-endo cyclization of **D**, forming the Cexo-C1 bond, and β-H elimination/aromatization would afford 3a-ZnBr and HPdBr. Closure of the catalytic cycle would then occur upon regeneration of Pd⁰ and liberation of aminonaphthalene 3a, as observed in NMP as solvent. The formation of aminoindene 4a in DMF/H2O is ascribed to the protonative and reductive trappings of alkylpalladium **B**.^[14] When the same reaction was conducted in deuteriumlabelled [D₇]DMF/H₂O, 75% deuterium was incorporated in place of one of the protons in the exocyclic methyl group, implying the intermediacy of a palladium formate such as **F**, which undergoes decarboxylation followed by reductive elimination to afford the deuterium labeled $[D_{75\%}]$ -4a. Reactions performed in DMF/D₂O and [D₇]DMF/D₂O gave 25% and 100% deuterium incorporation, respectively, in **4a**. These labeling experiments suggest that dimethylammonium formate, generated from DMF and H₂O, can effect either the protolysis of **B** or its reduction with a rate ratio of $k_1/k_2 = 3$.^[15]

The intermediacy of σ -carbopalladate **B** offers an additional opportunity for increasing the diversity of aminoindenes obtained in this tandem reaction if its interception with a suitable carbon nucleophile occurs at a rate competitive with the hydrodehalogenation sequence. To our delight, the tandem Pd-catalyzed reaction of **A1** with 2.0 equivalents of phenylacetylene in the presence of 3.0 equivalents K₃PO₄ in DMF/H₂O satisfied these requirements, affording **5a** in 79% yield within 1 h (Scheme 2 c).

Under these reaction conditions, various kinds of aminonaphthalenes **3**, aminoindenes **4**, and the alkynylated amino-





[a] For details on the reaction conditions, see the Supporting Information. Yields of isolated products are given.

indenes 5 can be selectively synthesized from nitriles (Table 1). The tandem reaction of A1, prepared starting from 2-bromobenzonitrile with phenylalkynes that bear either electron-donating or electron-withdrawing groups afforded the corresponding 1-aminonaththalenes 3b-3f in excellent yields. Alkyl alkynes such as 1-hexyne and 4-phenylbutyne were also successfully incorporated to provide 4-alkyl-substituted 1-aminonaphthalenes 3g and 3h, albeit with slightly lower yields. Variation of the nitrile moiety enabled the rapid synthesis of polysubstituted arylamines 3i-**30** in excellent yields in a tandem one-pot manner. Switching to the conditions optimized for the trapping of **B** gave aminoindenes with methyl (4a-4e) and phenyl propargyl (5a-5e) substituents from the same nitriles and alkynes. The incorporation of 3-ethynylpyridine (5 f) and cyclohexenyne (5g) groups highlights the rapid generation of molecular diversity through these tandem one-pot catalytic sequences.

We also anticipated that the intermediate A2, generated using 2-chloroaryl alkynes, may serve as a common precursor for catalyst-controlled selective C–N bond-forming reactions to provide both pyrroles **6** and quinolines **7** selectively,

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through oxidative intramolecular olefin amination^[16] and redox-neutral arylamination reactions,^[17] respectively. After screening different reaction conditions (see Table S2), we found that the treatment of intermediate A2, generated from benzonitrile and o-chlorophenylacetylene, with 10 mol% $Pd(OAc)_2$ in the presence of $Cu(OAc)_2$ (2.0 equiv) and AcOH (2.0 equiv) in DMF at 90 °C for 2 h afforded pyrrole 6a selectively in 62% yield.^[18,19] The catalytic reaction pathway of A2 can be redirected by using a Pd⁰ catalyst ligated with a sterically hindered electron-rich phosphine, $[Pd{P(tBu_3)}_2]$ (5 mol%), and tBuOK (1.1 equiv) at 100 °C in DMF to afford quinoline 7a in 60% yield.^[20] With a set of optimized reaction conditions in hand for the selective synthesis of pyrroles 6 and quinolines 7, several pairs of these heterocycles were successfully synthesized through tandem reactions of the intermediate A2, generated from various nitriles 1, a Reformatsky reagent, and 2-chloro arylacetylene 2 (Table 2).





[a] For details on the reaction conditions, see the Supporting Information. Yields of isolated products are given.

Mechanistic proposals for the formation of pyrroles **6** and quinolines **7** are outlined in Scheme 4. For the formation of pyrrole **6** (Scheme 4a), the transformation begins with a ligand substitution with Pd(OAc)₂, resulting in aminopalladate complex **C**. A 5-endo-trig cyclization^[21] through either a syn or anti aminopalladation of **D** gives the intermediate **E**, and subsequent β -H elimination affords



Scheme 4. Proposed mechanisms for the formation of a) pyrroles **6**, and b) quinolines **7**.

pyrroles **6**. The Pd^{II} species can be regenerated by oxidation of Pd⁰ with Cu(OAc)₂ and O₂/AcOH.^[22] Nucleophilic substitution pathways forming the γ -C-palladate **F**, followed by nucleophilic substitution, reductive elimination, and isomerization in sequence cannot be ruled out at present. For the formation of quinoline through a typical Buchwald–Hartwig catalytic cycle (Scheme 4b), oxidative addition of Pd⁰ to Ar-Cl first affords **I**. Deprotonation of the amine with *t*BuOK forms **J**, which leads to Pd⁰ and N-arylated **K** through reductive elimination. Finally, the latter undergoes aromatization to produce the targeted quinoline **7**.

In summary, we have developed tandem divergent catalytic methods for the selective conversion of simple nitriles to four distinct classes of aromatic compounds. The zinc bromide complexes of β -enaminoesters, generated by the sequential reaction of nitriles with a Reformatsky reagent and 1-alkynes, are used as common intermediates. The catalytic reaction pathways of the intermediates are precisely controlled by a simple change of the reaction solvents or palladium catalysts to selectively afford arylamines, amino-indenes, pyrroles, and quinolines. This tandem divergent catalytic approach is shown to be particularly useful to rapidly elaborate simple molecules into different complex structures.

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