TOTAL SYNTHESES OF 1,5-DIDEOXY-1,5-IMINO-D-erythro-L-allo-OCTITOL AND (1S,2R,6R,7R,8S,8aS)-PENTAHYDROXYINDOLIZIDINE

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Summary: The title compounds (-)-1 and (-)-2 were derived from the optically pure Diels-Alder adduct (-)-3 of furan to 1-cyanovinyl (1R')-camphanate.

Polyhydroxyindolizidines belong to an important class of alkaloids which includes (-)-swainsonine and (+)-castanospermine, two potent inhibitors of mannosidases and glucosidases, respectively.¹ We report the total, asymmetric synthesis of the new 1,5-dideoxy-1,5-iminooctitol (-)-1 and its conversion to the corresponding pentahydroxyindolizidine (-)-2.



The "naked sugar" (-)-3 (Diels-Alder adduct of furan to 1-cyanovinyl (1R')-camphanate)² was converted to the branched uronolactone (+)-4 with high stereoselectivity as already described.³ Treatment of (+)-4 with benzylic alcohol (BnOH) in THF with K_2CO_3 led to mixtures of the equilibrating benzyl uronates (+)-5 and (-)-6. When run in DMSO (6 equiv. of BnOH) in the presence of 3 equivalents of anhydrous CsF (20°C) (-)-6 (76%, isolated)⁴ was the major epimer (>10:1). In pure MeOH and in the presence of anhydrous K_2CO_3 , (+)-4 led to 7, the product favored under kinetically controlled conditions. On staying at 20°C with K_2CO_3 in MeOH, it was slowly equilibrated with 8. Hydrogenolysis of (-)-6, which could be isolated pure by flash chromatography on silica gel, gave the corresponding uronic acid that was esterified into 8 with CH_2N_2 in CHCl₃. While 7 and 8 had similar stabilities in MeOH, (-)-6 appears to be more stable than (+)-5 in DMSO.⁵ This phenomenon is not yet understood.

Silylation of (-)-6 with (t-Bu)Me₂SiOSO₂CF₃ and 2,6-lutidine (CH₂Cl₂, 0°C) gave (+)-9 (mostly β -anomer).⁶ Hydrogenolysis (H₂/10% Pd-C, EtOAc, 20°C) of (+)-9 followed by treatment with N₃PO(OPh)₂ in toluene containing 5 equiv. of Et₃N induced a Curtius rearrangement⁷ providing the corresponding isocyanate that was quenched with BnOH (100°C, 15 h) to give (+)-10⁸ in 92% yield. Desilylation with Bu₄N⁺F⁻ (1 M in THF, 0-20°C), followed by hydrogenolysis (H₂, 10% Pd-C, EtOAc, 20°C, 24 h) furnished (-)-12 (83%). This process implies the intermediacy of sugar 11 which generates the corresponding cyclic imine upon debenzylation of the NHCbz moiety. Hydrolysis of (-)-12 (CF₃COOH/H₂O 8:1, 20°C, 5 h) followed by purification by chromatography on Dowex 50 Wx8 (H⁺ form) gave



1,5-dideoxy-1,5-imino-D-*erythro*-L-*allo*-octitol ((-)-1) in 96% yield.⁹ Treatment of (-)-1 with pyridine, Ph₃P, CCl₄, Et₃N (20°C, 15 h) and then quenching with MeOH,¹⁰ provided the pentahydroxyindolizidine (-)-2 (82%) which was characterized in the form of the corresponding pentaacetate (-)-13¹¹ obtained by treatment with Ac₂O, pyridine and 4-(Me₂N)-pyridine.

The structures of the new compounds described above were given by their mode of formation, their reactivity, their elemental analyses and their spectral data.¹² The ¹C₄ chair conformation of (-)-1 as well as the ⁴C₇-chair conformation of the six-membered ring in (-)-13 were confirmed by the 360 MHz ¹H-NMR spectra with the help of double irradiation experiments including NOE measurements. In particular, irradiation of proton H-C(8a) in (-)-13 ($\delta_{\rm H}$ = 2.47 ppm, ³J(H-C(8),H-C(8a)) = 10.0 Hz, ³J(H-C(1),H-C(8a)) = 4.5 Hz) led to significant NOE's at signals $\delta_{\rm H}$ = 2.14 and 2.17 ppm attributed to H_β-C(3) and H_β-C(5), respectively.

This work demonstrates the potential of the "naked sugars" in the total, asymmetric synthesis of complicated sugars and alkaloids. The biological properties of (-)-1 and (-)-2 and their derivatives are under scrutinity.

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- 4. Data of (-)-6: m.p. 100-101°C. $[\alpha]^{25}_{589} = -12$ (c = 0.91, CHCl₃). ¹H-NMR (250 MHz, CDCl₃) $\delta_{\rm H}$: 7.37 (m, Ph); 5.47 (d, J = 3.5 Hz, H-C(1)); 5.28, 5.30 (2d, ²J = 12.5 Hz); 4.73 (dd, J = 6.0, 1.0 Hz, H-C(3)); 4.63 (d, J = 6.0 Hz, H-C(2)); 4.62 (dd, J = 11.0, 1.0 Hz, H-C(4)); 4.10-3.90 (m, 4 H, H-C(6), H-C(7), H'-C(8), H-C(8)); 3.19 (dd, J = 11.0, 2.0 Hz, H-C(5)); 3.14 (d, J = 3.5 Hz, HO-C(1)); 3.04 (br.s, HO-C(6)); 1.49, 1.37, 1.30, 1.28 (4s, 2 Me₂C). IR (KBr) v: 3600-3300, 3020, 2980, 1710 (COO), 1370 cm⁻¹. Data of (+)-5: oil; $[\alpha]^{25}_{589} = +16$ (c = 0.74, CHCl₃); IR (film) v: 1730 cm⁻¹ (COO). ¹H-NMR (250 MHz, CDCl₃) $\delta_{\rm H}$: 7.38 (m, Ph); 5.37 (d, J = 8.0 Hz, H-C(1)); 5.27, 5.13 (2d, ²J = 12.0 Hz); 4.86 (dd, J = 6.0, 1.0 Hz, H-C(3)); 4.64 (d, J = 6.0 Hz, H-C(2)); 4.65 (dd, J = 6.0, 1.0 Hz, H-C(4)); 4.21 (d, J = 8.0 Hz, HO-C(1)); 4.04 (m, H-C(6), H-C(7), H₂-C(8)); 3.23 (d, J = 6.0 Hz, HO-C(6)); 3.07 (dd, J = 6.0, 6.0 Hz, H-C(5)), 1.49, 1.38, 1.31, 1.30 (4s, 2 Me₂C).
- In DMSO/BnOH (6 equiv.) and CsF (3 equiv.), pure (+)-5 was converted to (-)-6 at 20°C in 2 h. Under these conditions H₂O elimination was a slow process.
- 6. Data of (+)-9: oil; $[\alpha]^{25}_{589} = +27$ (c = 1.065, CHCl₃). IR (CH₂Cl₂) v: 3030, 2950, 2930, 2890, 1730, 1460 cm⁻¹. ¹H-NMR (250 MHz, CDCl₃) $\delta_{\rm H}$: 7.38 (m, C₆H₅); 5.40 (s, H-C(1)); 5.22, 5.12 (2d, ²J = 12.0 Hz); 4.57 (dd, J = 10.5, 1.5 Hz, H-C(4)); 4.50 (dd, J = 6.0, 1.5 Hz, H-C(3)); 4.47 (d, J = 6.0 Hz, H-C(2)); 4.40 (dd, J = 4.0, 1.8 Hz, H-C(6)); 4.12 (ddd, J = 8.0, 6.5, 4.0 Hz, H-C(7)); 3.90 (dd, J = 8.0, 6.5 Hz, H-C(8)); 3.67 (t, J = 8.0 Hz, H'-C(8)); 2.84 (dd, J = 10.5, 1.8 Hz, H-C(5)); 1.49, 1.39, 1.30, 1.26 (4s, 2 Me₂C); 0.89, 0.85 (2s, 2 t-Bu); 0.15, 0.14, 0.13, 0.08 (4s, 2 Me₂Si).
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- 8. Data of (+)-10: oil, $[\alpha]^{25}_{589} = +46$ (c = 0.99, CHCl₃). IR (CH₂Cl₂) v : 3430, 3030, 2950, 2930, 2890,

2850, 1720, 1500, 1470 cm⁻¹. ¹H-NMR (250 MHz, CDCl₃) δ_{H} : 7.35 (m, C₆H₅); 5.42 (s, H-C(1)); 5.18, 5.10 (2d, ²J = 12.0 Hz); 5.0 (d, J = 9.5 Hz, NH); 4.82 (d, J = 6.0 Hz, H-C(2)); 4.58 (d, J = 6.0 Hz, H-C(3)); 4.48 (d, J = 2.2 Hz, H-C(6)); 3.96 (ddd, J = 8.5, 6.5, 2.2 Hz, H-C(7)); 3.88 (d, J = 0.6 Hz, H-C(4)); 3.82 (dd, J = 9.5 Hz, 0.6, H-C(5)); 3.72 (m, H₂-C(8)); 1.48, 1.40, 1.32, 1.25 (4s, 2 Me₂C); 0.92, 0.90 (2, 2 t-Bu); 0.12, 0.15 (2s, 2 Me₂Si).

- 9. Data of (-)-1: colourless crystals, m.p. $205^{\circ}C$ (dec.). $[\alpha]^{25}_{589} = -27$ (c = 0.3, H₂O). IR (KBr) v : 3700 3100, 2920, 1630, 1560, 1400, 1070, 1020, 900 cm⁻¹. ¹H-NMR (360 MHz, D₂O, CH₃CN as internal standard) δ_{H} : 4.17 (t, J = 2.5 Hz, H-C(3)); 3.92 (dd, J = 8.0, 1.0 Hz, H-C(6)); 3.84 (dd, J = 10.5, 3.5 Hz, H-C(8)); 3.78 (m, H-C(7)); 3.70 (ddd, J = 10.5, 5.0, 2.5 Hz, H-C(2)); 3.63 (dd, J = 10.5, 6.0 Hz, H'-C(8)); 3.62 (dd, J = 10.5, 2.5 Hz, H-C(4)); 2.93 (dd, J = 10.5, 1.0 Hz, H-C(5)); 2.86 (dd, J = 12.5, 5.0 Hz, H-C(1)); 2.68 (dd, J = 12.5, 10.5 Hz, H'-C(1)). ¹³C-NMR (62.9 MHz, pyridine-D₅) δ_{C} : 75.9, 73.9, 71.0, 70.5, 70.45 (5d, ¹J(C,H) = 140-150 Hz, C(2), C(3), C(4), C(6), C(7)); 64.7 (t, ¹J(C,H) = 140 Hz, C(8)); 56.3 (d, ¹J(C,H) = 135 Hz, C(5)); 46.3 (t, ¹J(C,H) = 135 Hz, C(1)). MS-CI (NH₃) m/z : 224 ([M + 1]⁺, 59), 207 (3), 204 (3), 203 (7), 192 (10), 162 (9), 145 (27), 132 (100).
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- 11. Data of (-)-2: colourless crystals, m.p. 175°C (dec). $[\alpha]^{25}_{589} = -70$ (c = 0.85, H₂O). ¹H-NMR (360 6.0, 3.5 Hz, H-C(1)); 4.14 (dd, J = 3.0, 3.0 Hz, H-C(7)); 3.89 (dd, J = 10.5, 3.0 Hz, H-C(8)); 3.84 (ddd, J = 10.0, 5.0, 3.0 Hz, H-C(6)); 2.92 (dd, J = 11.0, 2.5 Hz, H_{α}-C(3)); 2.90 (dd, J = 11.0, 5.0 Hz, H_{α} -C(5)); 2.69 (dd, J = 11.0, 8.0 Hz, H_{β} -C(3)); 2.46 (dd, J = 10.5, 3.5 Hz, H-C(8a)); 2.32 (dd, J = 10.5, 3.5 Hz, H-C(8a)); 3.5 Hz, H-C(8a)); 3.5 11.0, 10.0 Hz, H₈-C(5)); ¹³C-NMR (62.9 MHz, D₂O, pyridine-D₅ as internal standard) δ_{C} : 73.0, 71.0, 70.7, 69.1, 68.3 (5d, ${}^{1}J(C,H) = 145-150$ Hz, C(1), C(2), C(6), C(7), C(8)); 66.4 (d, ${}^{1}J(C,H) = 140$ Hz, C(8a)); 61.1, 52.4 (2t, ${}^{1}J(C,H) = 140$ Hz, C(3), C(5)). MS-CI (NH₃) m/z: 224 (8), 223 (M⁺ + NH₄⁺, 1.2), 206 (M⁺ + 1, 100), 205 (M⁺, 26), 188 (20), 187 (13), 170 (9), 145 (41), 116 (12), 102 (25), 98 (20), 80 (19). Data of (-)-13: colourless crystal, m.p. $150-151^{\circ}$ C; $[\alpha]^{25}_{589} = -19$ (c = 0.2, CHCl₃). ¹H-NMR (360 MHz, C_7D_8) δ_{H} : 5.76 (t, J = 3.0 Hz, H-C(7)); 5.41 (dd, J = 6.5, 4.5 Hz, H-C(1)); 5.27 (dd, J = 10.0, 3.0 Hz, H-C(8)); 4.91 (ddd, J = 10.0, 5.0, 3.0 Hz, H-C(6)); 4.90 (ddd, J = 8.0, 6.5, 2.5)Hz, H-C(2)); 2.79 (dd, J = 10.5, 2.5 Hz, H_a-C(3)); 2.75 (dd, J = 10.0, 5.0 Hz, H_a-C(5)); 2.47 (dd, J = 10.0, J = 110.0, 4.5 Hz, H-C(8a)); 2.17 (t, J = 10.0 Hz, H₈-C(5)); 2.14 (dd, J = 10.5, 8.0 Hz, H₈-C(3)); 1.85, 1.83, 1.75, 1.68, 1.61 (5s, 5 Ac). ¹³C-NMR (62.9 MHz, CDCl₃) δ_C : 170.1, 169.9, 169.8, 169.5, 169.4 (5s), **70.5**, **69.2**, **68.2**, **67.6**, **66.3** (**5d**, ${}^{1}J(C,H) = 150-160$ Hz, C(1), C(2), C(6), C(7), C(8)); **62.3** (d, ${}^{1}J(C,H)$ = 135 Hz, C(8a)); 57.9, 48.9 (2t, ${}^{1}J(C,H)$ = 140 Hz, C(3), C(5)); 20.74, 20.7, 20.5, 20.4, 20.3 (5q, 1 J(C,H)=130 Hz). MS-EI m/z: 416 (M⁺ + 1, 3), 356 (M⁺ - OAc, 3); 355 (M⁺ - HOAc, 24), 296 (55), 295 (9), 294 (26), 252 (24), 236 (100), 194 (10), 176 (7), 134 (34), 126 (4), 96 (13), 84 (12). Calc. Anal. for C₁₈H₂₅NO₁₀ (415.401): C 52.05, H 6.07, N 3.37; found: C 52.02, H 6.02, N 3.45.
- 12. Details will be given in a forthcoming full-paper.

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