J. CHEM. SOC., CHEM. COMMUN., 1985

Acyclic Stereoselection using 1,2-Asymmetric Induction. The First Total Synthesis of (+)-Corynomycolic Acid

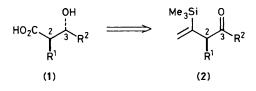
Yasunori Kitano, Yuichi Kobayashi, and Fumie Sato*

Department of Chemical Engineering, Tokyo Institute of Technology, Meguro, Tokyo 152, Japan

The first total synthesis of (+)-corynomycolic acid (1), which includes as the key step, diastereoselective reduction of the optically active carbonyl compounds (8) is described.

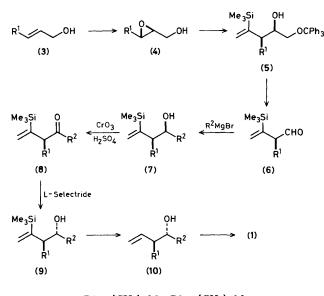
Mycolic acids, mostly found in nature as their ester with carbohydrates, are high molecular weight fatty acids having 28 to 90 carbon atoms and have the basic structure $R^{2}CH(OH)CH(R^{1})CO_{2}H^{1}$ The stereochemistry of the mycolic acids so far examined has been shown to be 2R, 3R as shown in formula (1)² Although these acids have interesting biological properties,¹ only a few syntheses of the racemic form of (1) have been reported where 2,3-anti stereochemistry was achieved with rather low stereoselectivity.³ Furthermore, these methods do not seem to be applicable to the synthesis of enantiomerically pure (1). In this communication, we report a general synthetic method for the preparation of optically active (1). The key feature of our approach to the synthesis of (1) is based on our earlier work and involves the stereoselective reduction of optically active (2) with a metal hydride reagent to achieve the required stereochemistry at C-3 by 1,2-asymmetric induction relative to the pre-existing C-2 chiral centre.^{4,5} In this synthesis, two alkyl chains R¹ and R² are incorporated individually at different stages. The effectiveness of our method is demonstrated by synthesising (+)-corynomycolic acid {(1) $R^1 = [CH_2]_{13}Me$, $R^2 =$ $[CH_2]_{14}Me$, one of the mycolic acids produced by Corynebacteria.2,6

Sharpless asymmetric epoxidation⁷ of the (E)-heptadec-2en-1-ol (3), obtained from but-2-yn-1-ol and n-tetradecyl iodide,⁸ using L-(+)-di-isopropyl tartrate produced the epoxide (4)† { $[\alpha]_D^{25} - 27.0^\circ$ (c 0.869, C₆H₆)} in 77% yield. Protection of the hydroxy group of (4) as a trityl ether (Ph₃CCl, 4-*N*,*N*-dimethylaminopyridine, NEt₃) and subsequent regioselective epoxide ring opening with 1-trimethylsilylvinylmagnesium bromide afforded (5) as the sole product. Compound (5) was converted into the aldehyde (6) { $[\alpha]_D^{25}$ -13.3° (c 0.900, CHCl₃)} by successive treatments with aqueous Cl₂CHCO₂H and NaIO₄.⁸ The yield of (6) from (4) was 47%. The reaction of n-pentadecylmagnesium bromide with (6) gave (7)‡ exclusively { $[\alpha]_D^{25} - 0.78^\circ$ (c 1.025,



 $[\]dagger$ Since asymmetric epoxidation of this type of (*E*)-alk-2-en-1-ols is known to proceed with high enantiomeric excess (over 95%), the optical purities of (4) and other intermediates used here were not confirmed.

 \ddagger ^{13}C N.m.r. (CDCl₃) (7) δ 73.4, 125.6, and 153.9; (9) δ 73.7, 127.9, and 153.9.



 $R^1 = [CH_2]_{13}Me, R^2 = [CH_2]_{14}Me$

CHCl₃)} in 90% yield.⁴ Jones oxidation of (7) followed by reduction with L-Selectride gave (9) (84%) as the sole product⁴ {[α]_D²⁵ +9.25° (*c* 1.06, CHCl₃)}: the stereoisomer (7) was not detected by ¹³C n.m.r. spectroscopy. Protodesilylation of (9) (NaH, hexamethylphosphoramide)⁹ gave the alcohol (10) in 86% yield. Finally (10) was transformed into (+)-(1) in 90% yield through the following sequence of reactions: conversion of (10) into its correponding benzyl ether (PhCH₂Br, KH, tetrahydrofuran), ozonolysis, Me₂S work up, oxidation (CrO₃, H₂SO₄), and debenzylation (H₂, Pd–C). The physical data $\{[\alpha]_D^{25} + 7.34^\circ (c \ 1.90, \text{CHCl}_3); m.p. 70.0-71.5 °C\}$ of (+)-(1) thus prepared were in good agreement with the literature values reported by Pudles *et al.* $\{[\alpha]_D + 7.5^\circ (c \ 1.64, \text{CHCl}_3); m.p. 70.0 °C\}$.^{6a} This is the first total synthesis of (+)-corynomycolic acid achieved in a stereo-and enantio-selective manner and it is clear that this method provides a general synthetic method for various mycolic acids.

Received, 11th December 1984; Com. 1749

References

- 1 For reviews see, E. Lederer, A. Adam, R. Ciorbaru, J.-F. Petit, and J. Wietzerbin, *Mol. Cell. Biochem.*, 1975, 7, 87; E. Lederer, *Chem. Phys. Lipids*, 1976, 16, 91.
- J.-F. Tocanne and C. Asselineau, *Bull. Soc. Chim. Fr.*, 1968, 4519;
 C. Asselineau, G. Tocanne, and J.-F. Tocanne, *ibid.*, 1970, 1455.
- 3 E. Lederer, V. Portelance, and K. Serck-Hanssen, Bull. Soc. Chim. Fr., 1952, 413; M. J. Polonsky and E. Lederer, *ibid.*, 1954, 504;
 C. H. Heathcock, M. C. Pirrung, S. H. Montgomery, and J. Lampe, *Tetrahedron*, 1981, 37, 4087; C. H. Heathcock and J. Lampe, J. Org. Chem., 1983, 48, 4330.
- 4 F. Sato, M. Kusakabe, and Y. Kobayashi, J. Chem. Soc., Chem. Commun., 1984, 1130.
- 5 Preparation of optically active (2, R¹ = Me) was reported: Y. Kobayashi, Y. Kitano, and F. Sato, J. Chem. Soc., Chem. Commun., 1984, 1329.
- 6 (a) A. Diara and J. Pudles, *Bull. Soc. Chim. Biol.*, 1959, 41, 481;
 (b) T. Ioneda, M. Lenz, and J. Pudles, *Biochem. Biophys. Res. Commun.*, 1963, 13, 110.
- 7 B. E. Rossiter, T. Katsuki, and K. B. Sharpless, J. Am. Chem. Soc., 1981, 103, 464.
- 8 D. E. Ames, A. N. Covell, and T. G. Goodburn, J. Chem. Soc., 1963, 5889.
- 9 F. Sato, Y. Tanaka, and M. Sato, J. Chem. Soc., Chem. Commun., 1983, 165.