

SYNTHESIS OF THE ISOINDOLOBENZAZEPINE ALKALOID LENNOXAMINE

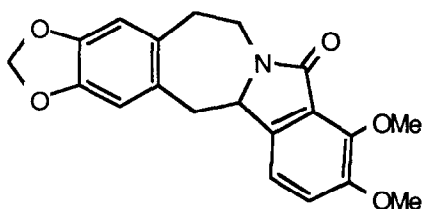
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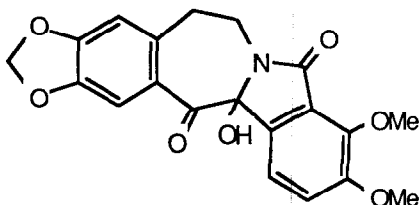
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Summary: The naturally occurring isoindolobenzazepine lennoxamine (1) has been synthesised from 6-bromopiperonal in 9 steps by a route which involves formation of the 2-arylbenzazepine (10) from the azide (9) as the key step.

Barberries are a rich source of alkaloids, and recent studies of Chilean Berberidaceae have provided the first members of a new class of alkaloid, the isoindolobenzazepines,¹⁻³ exemplified by lennoxamine (1) and chilene (2). The ring system, which is accessible *in vivo* and *in vitro* by oxidation of berberine alkaloids³⁻⁵ and from phthalide-isoquinoline alkaloids,⁶ has recently been the subject of several synthetic approaches.⁷⁻¹⁰ We now report a total synthesis of the alkaloid lennoxamine (1), isolated from *Berberis darwinii* Hook,² by a route based on vinyl azide chemistry.¹¹⁻¹³

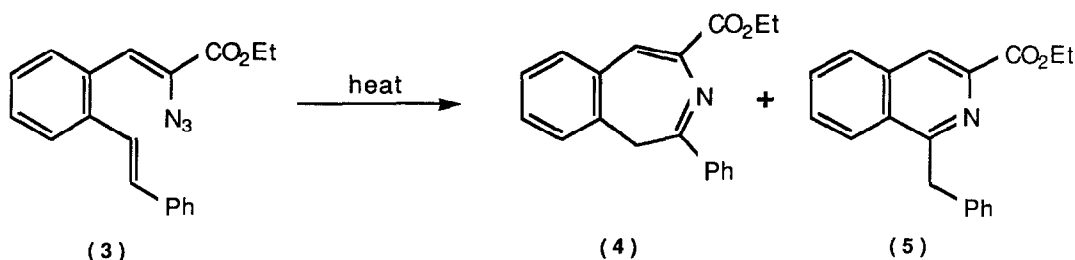


(1)



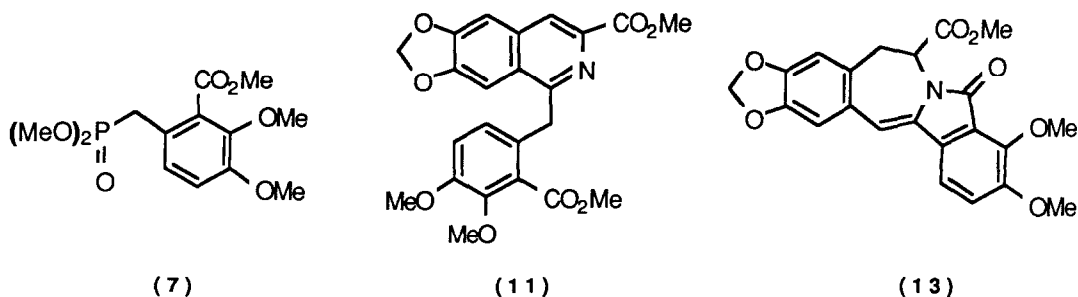
(2)

We have recently shown that in the thermolysis of azidocinnamates bearing *ortho*-alkenyl side chains, the major reaction pathway involves interaction of the azide/nitrene with the substituent double bond rather than with the aromatic ring. In particular, the *ortho*-styryl azide (3) gave the 3-benzazepine (4) (37%) and the 1-benzylisoquinoline (5) (36%) as the major products (Scheme 1),¹³ and therefore we investigated the use of this type of reaction as a route to the 3-benzazepine portion of lennoxamine (1).

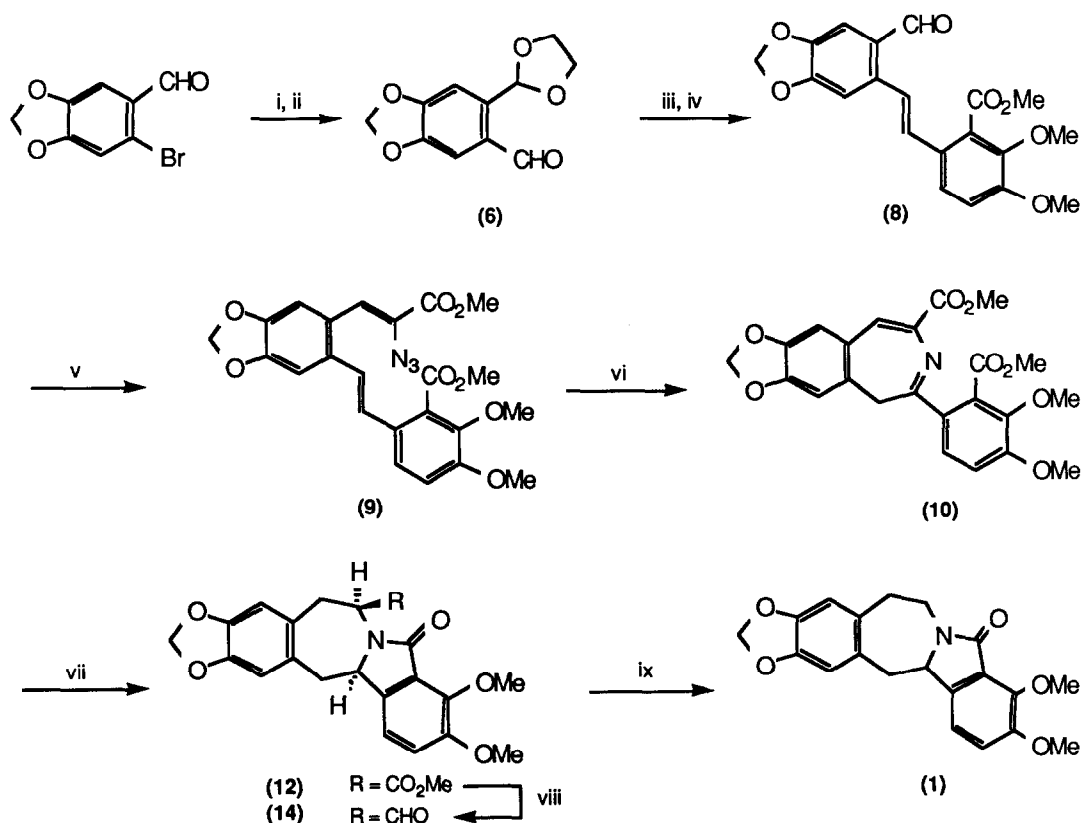


Scheme 1

The precursor to the required azidocinnamate (9) is the highly substituted (E)-stilbene aldehyde (8), and this was prepared from 6-bromopiperonal as shown in Scheme 2. Thus the ethylene acetal derived from 6-bromopiperonal¹⁴ was converted into the mono-protected dialdehyde (6) (88%), which underwent Wadsworth-Emmons reaction with the known⁹ phosphonate (7) to give, after acid hydrolysis of the acetal, the (E)-stilbene aldehyde (8) (74% over 2 steps). Condensation of the aldehyde (8) with methyl azidoacetate gave the unstable azide (9) in 66% yield. On heating in boiling xylene, the azide (9) rapidly decomposed to give the required 2-aryl-3-benzazepine (10) as the major product (55%). The benzazepine (10) was accompanied by the benzyloquinoline (11) (29%) and the corresponding 4-styryl indole-2-carboxylate (4%). Treatment of the 3-benzazepine (10) with sodium cyanoborohydride in acetic acid resulted in reduction of both double bonds, and concomitant cyclisation of the NH onto the ester group of the 2-substituent to give the *cis*-tetrahydro isoindolo[1,2-*b*][3]benzazepine (12) in 80% yield.¹⁵ The *cis*-diastereomer of the benzazepine (12) was accompanied by small amounts of the *trans*-isomer (epimeric at C-6) (4%) and the dihydro compound (13) (3%). It is not clear whether the *trans*-tetrahydro isoindolobenzazepine is formed in the reduction or during isolation, because it was subsequently found that the *cis*-isomer rapidly epimerised to the more stable *trans*-isomer on treatment with base.



The total synthesis of lennoxamine (1) now simply required the removal of the unwanted ester group at C-6, and this was achieved by reduction of the *cis*-ester (12) to the *cis*-aldehyde¹⁵ (14) (70%) using di-isobutylaluminium hydride in toluene, followed by decarbonylation using a rhodium (I) catalyst¹⁶ to give lennoxamine (1) (51%). Since the stereochemical centre at C-6 is removed in the final step, separation of the diastereomers of the tetrahydro isoindolobenzazepine formed in the cyanoborohydride reduction step is unnecessary, and without this separation, the 2-aryl-3-benzazepine (10) was converted into lennoxamine (1) in 43% overall yield. The sample of lennoxamine was identical (m.p., t.l.c., 200 MHz ¹H n.m.r.) to material from natural sources.¹⁷



Scheme 2. Reagents: i, ethylene glycol, TsOH (cat), toluene, reflux; ii, *n*-BuLi, -60°C, ether, then DMF; iii, KO^tBu/phosphonate (7), THF; iv, dilute HCl, CH₂Cl₂; v, MeO₂CCH₂N₃, NaOMe, MeOH, THF, 5°C; vi, xylene, reflux; vii, NaBH₃CN, AcOH; viii, DIBAL, toluene, -70°C; ix, (Ph₃P)₂Rh(CO)Cl, Ph₂P(CH₂)₃PPh₂, xylene, reflux.

Acknowledgements

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References and Notes

1. V. Fajardo, V. Elango, B. K. Cassels, and M. Shamma, *Tetrahedron Lett.*, 1982, **23**, 39.
2. E. Valencia, A. J. Freyer, M. Shamma, and V. Fajardo, *Tetrahedron Lett.*, 1984, **25**, 599.
3. E. Valencia, I. Weiss, S. Firdous, A. J. Freyer, M. Shamma, A. Urzúa, and V. Fajardo, *Tetrahedron*, 1984, **40**, 3957.
4. J. L. Moniot, D. M. Hindenlang, and M. Shamma, *J. Org. Chem.*, 1979, **44**, 4347.
5. C. R. Dorn, F. J. Koszyk, and G. R. Lenz, *J. Org. Chem.*, 1984, **49**, 2642.
6. S. Teitel, W. Klotzer, J. Borgese, and A. Brossi, *Can. J. Chem.*, 1972, **50**, 2022.
7. S. Ruchirawat, W. Lertwanawatana, S. Thianpatanagul, J. L. Cashaw, and V. E. Davis, *Tetrahedron Lett.*, 1984, **25**, 3485.
8. J. Chiefari, W. Janowski, and R. Prager, *Tetrahedron Lett.*, 1986, **27**, 6119.
9. E. Napolitano, G. Spinelli, R. Fiaschi, and A. Marsili, *J. Chem. Soc., Perkin Trans. 1*, 1986, 785.
10. P. H. Mazzocchi, C. R. King, and H. L. Ammon, *Tetrahedron Lett.*, 1987, **28**, 2473.
11. C. J. Moody and G. J. Warrellow, *J. Chem. Soc., Perkin Trans. 1*, 1987, 913.
12. C. J. Moody and G. J. Warrellow, *J. Chem. Soc., Perkin Trans. 1*, 1986, 1123.
13. D. M. B. Hickey, C. J. Moody, and C. W. Rees, *J. Chem. Soc., Perkin Trans. 1*, 1986, 1113.
14. A. H. Parijs, *Rec. Trav. Chim. Pays-Bas.*, 1930, **49**, 17.
15. The stereochemistry of the *cis*-ester (**12**) was assigned by correlation with the *cis*-aldehyde (**14**), the stereochemistry of which was confirmed by n.O.e. difference spectra. In common with the *cis*-ester (**12**), the *cis*-aldehyde (**14**) readily epimerises to the more stable *trans*-isomer on treatment with base.
16. M. D. Meyer and L. I. Kruse, *J. Org. Chem.*, 1984, **49**, 3195.
17. Supplied by Professor M. Shamma, Pennsylvania State University.

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