



0040-4039(95)00111-5

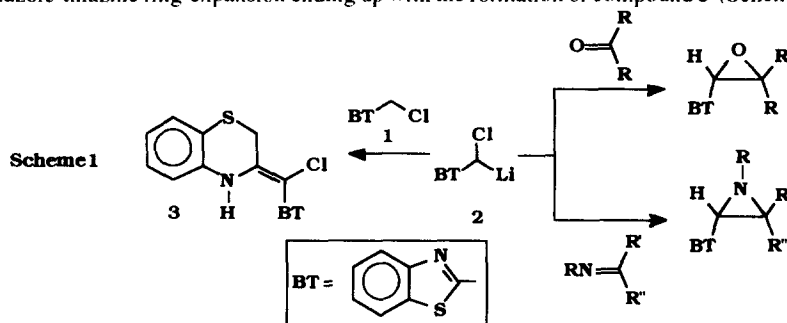
RING EXPANSION OF 2-CHLOROMETHYLBENZOTHAZOLE : SYNTHESIS OF HETEROARYLALKYLIDENE 1,4-BENZOTHAZINES

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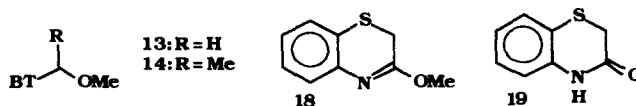
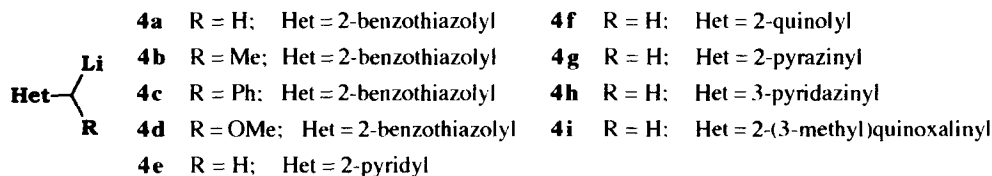
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Abstract: the title heteroarylalkylidene dihydro-1,4-benzothiazines **7** have been prepared from heteroarylalkyllithiums **4** and 2-chloromethylbenzothiazole **1**.

We have recently reported that 2-benzothiazolylchloromethylithium **2**, promptly available by lithiation of 2-chloromethylbenzothiazole (BTCH₂Cl) **1**, behaves as a halocarbenoid. Indeed, **2** adds to carbonyl compounds and imines to give benzothiazolyl oxiranes¹ and aziridines² respectively, and, in the absence of an external electrophile, adds to the C-N double bond of its precursor **1** causing an unusual thiazole-thiazine ring-expansion ending up with the formation of compound **3** (Scheme 1).³

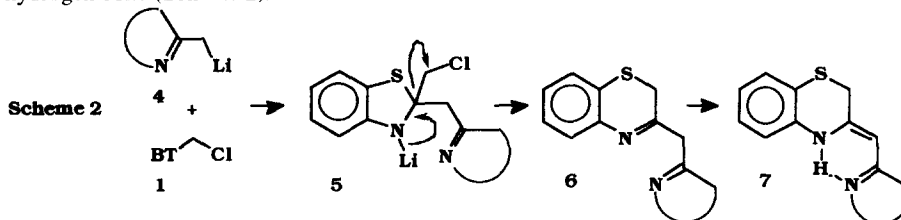


In the present communication we report on the synthetic exploitation of such a thiazole-thiazine ring-enlargement for the preparation of novel benzothiazine derivatives, a group of heterocyclic compounds which are especially interesting owing to their occurrence in nature as mammalian pigments and to their extensive pharmacological activity.⁴⁻¹⁰

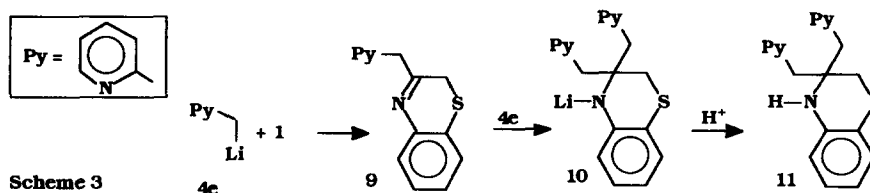


Treatment of 2-benzothiazolylmethylithium **4a**, generated by lithiation (LDA) of commercial 2-methylbenzothiazole, with 2-chloromethylbenzothiazole **1** gave a very high yield of benzothiazine **7a** together with a small amount (~ 7%) of the substitution product **8a**. Similarly, the reaction of **1** with benzothiazolylethyllithium **4b** and benzothiazolylbenzylithium **4c** furnished benzothiazines **7b** and **7c** accompanied by the substitution products **8b** and **8c**. Comparable results were obtained when **1** was treated with organolithiums prepared by lithiation of some other heteroaryl methanes. Indeed, treatment of **1** with pyridyl- **4e**, and quinolyl-methylithium **4f** provided benzothiazines **7d** and **7e** and reactions with pyrazinyl- **4g**, pyridazinyl- **4h** and quinoxalinyllithium **4i** produced good yields of benzothiazines **7f**, **7g** and **7h** respectively (See Table).

The formation of benzothiazines **7** could be rationalized by assuming that organolithiums **4** add to the C-N double bond of **1** to give thiazoline intermediates **5** which undergo ring-enlargement to the unstable dihydrobenzothiazines **6** and then to tautomers **7**, which are stabilized by the intramolecular hydrogen bond (Scheme 2).



In the reaction of **1** with picolylithium **4e** the formation of the dihydrobenzothiazine **11** was observed together with benzothiazine **7d**. A possible explanation is that organolithium **4e** adds to the ring-expanded compound **9** (or to its tautomer **7d**) leading to **10** and then to **11** (Scheme 3). Similarly, dihydrobenzothiazine **12** formed in the reaction of **1** with **4g**.



Lithiation of 2-methoxymethylbenzothiazole **13** with LDA furnished **4d** that could be trapped with excess MeI yielding methoxyethylbenzothiazole **14**. Moreover, **4d** reacted with **13** and then with **1** to give a product which had the structure of **17**. For its formation, we suggest that lithiated ether **4d** adds to **13** giving thiazoline **15**, which undergoes ring opening with the leaving thiofinane being trapped by **1** to furnish **16**, which equilibrates to **17** (Scheme 4).

It was interesting to observe that also MeONa in methanol was capable of addition to the C-N double bond of **1**, thus initiating the thiazole-thiazine ring-enlargement. Indeed, when MeONa (5 eq.) was added to a methanol solution of **1** a 3.5:1:1 mixture (90% overall yield) of the substitution product **13** and the ring-expanded compounds **18** and **19** formed. Benzothiazinone **19** likely arises by the demethylation of methoxybenzothiazine **18**. It must be emphasized the unusual stability of **18**; indeed, this kind of benzothiazines are unstable and tend to dimerize.¹¹

Scheme 4

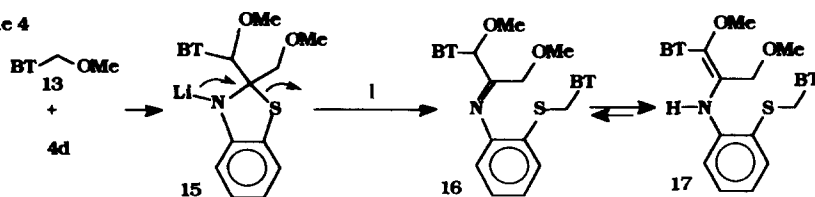


Table. Reaction of heteroarylalkyllithiums **2**, **4a-i** (RLi) with 2-chloromethylbenzothiazole **1** in THF at -78°C .

| RLi | Reaction Products (% yield) ^{a, b} | RLi | Reaction Products (% yield) ^{a, b} |
|-----------|--|-----------|--|
| 2 | 3 (85%) | 4f | 7e (60%) |
| 4a | 7a (88%) + 8a (7%) | 4g | 7f (67%) + 12 (23%) |
| 4b | 7b (80%) + 8b (12%) | 4h | 7g (70%) |
| 4c | 7c (30%) + 8c (30%) | 4i | 7h (40%) |
| 4e | 7d (56%) + 11 (12%) | | |

a) Yields calculated on isolated, purified products. **b)** All the new benzothiazines showed consistent IR, ^1H NMR, ^{13}C NMR and MS data.

Alkylidene 1,4-benzothiazines **7a-h** appear to be quite stable.¹² This is likely due to the strong intramolecular hydrogen bond between the aza groups of the two heterocyclic systems. The low field resonance of the N-H proton in the ^1H NMR (δ : 12-13) is also consistent with a Z configuration at the

C-C double bond of the enamine system.¹³ Support to this comes also from the value (5.3 Hz) of the $^3J(\text{CH}_2\text{CNH})$ coupling constant ($^{13}\text{CNMR}$) in the thiazine system.¹⁴ The chelation properties of benzothiazines **7** are worthy to be evaluated as well as their potential in terms of pharmacological activity.

Acknowledgements: We thank italian CNR and MURST (Rome) for financial support.

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12. **General Procedure:** To diisopropylamine (2.4 mmole) in 10 ml of THF was added at 0°C 1ml of 2.4N n-BuLi. To this solution, cooled at -78°C, was added dropwise a solution of **4a** (2.0 mmole) in 10 ml of THF. After 15 min at -78°C a THF solution of **1** (2.0 mmole) was added dropwise and then the reaction mixture was allowed to warm to RT and quenched with aqueous NH_4Cl after 6h. Extraction with ether (3x25 ml), drying over Na_2SO_4 and evaporation of the solvent under reduced pressure left a residue that was column chromatographed (silica gel, E.P./E.E.:7/3 as eluent) to give benzothiazine **7a**.
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(Received in UK 22 December 1994; accepted 13 January 1995)