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SYNTHESIS AND ACTIVITY OF C2-SUBSTITUTED ANALOGS OF INFLUENZA NEURAMINIDASE INHIBITOR GS 4071

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Abstract: The influence of C₂-substitution of GS 4071 on the influenza neuraminidase inhibitory activity was investigated. The introduction of lipophilic substituents (chloro, methyl, and methylthio) at the C₂ position resulted in a significant decrease of the activity. This result indicates that at the enzyme active site there is limited hydrophobic pocket for a group at the C₂ position of GS 4071. © 1997 Elsevier Science Ltd.

Influenza infection is the most serious respiratory illness both in term of morbidity and mortality.¹ In the search for a new anti-influenza agent, a potent influenza neuraminidase (NA) inhibitor GS 4071 (1) has recently been identified in our laboratory.² In addition, on the basis of in vitro and in vivo results, the ethyl ester of GS 4071 has been selected as a clinical trial candidate for the oral treatment and propylaxis of influenza infection.³

$$R$$
 CO_2H
 CO_2H
 $R = H (GS 4071)$
 $R = CI$
 $R = CI$
 $R = CI$
 $R = CI$
 $R = CH_3$
 $R = CH_3$

As part of the study of the structure activity relationship of this series, C2-substituted analogs of GS 4071 were investigated. Despite the presence of the highly lipophilic 3-pentyloxy group at the C3 position, GS 4071 has a negative partition coefficient value (log P = -1.1), indicating that it is still a rather polar molecule. Therefore, we investigated the possibility of making the molecule less polar and then improving its pharmacological properties by introducing lipophilic functional groups at the C2 position. To this end, C2-chloro (2), C2-methylthio (3), and C2-methyl (4) analogs were prepared.

The synthesis of the C2-chloro analog 2 was accomplished starting from the C2-chloro shikimic acid derivative 6, which was prepared from shikimic acid 5 by literature method⁴ (Scheme 1). Tetraester 6 was treated with a catalytic amount of sodium methoxide in methanol to give triol 7 in quantitative yield. Although the 3-pentyl ketal could be prepared directly from triol 7 in one step, a two-step procedure via acetonide 8 gave a better yield. Thus, 7 was first treated with 2,2-dimethoxypropane in acetone in the presence of toluenesulfonic acid to give acetonide 8. Mesylation and transketalization of 8 in the presence of a catalytic amount of perchloric

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Scheme 1a

^aReagents: (a) CH₃ONa (5 mol %), CH₃OH, rt; (b) 2,2-dimethoxypropane, aceton, p-TsOH (2 mol %), reflux; (c) MsCl, Et₃N, CH₂Cl₂, rt; 3-pentanone, HClO₄ (2 mol %), 40 °C; (d) BH₃·Me₂S, TMSOTf, CH₂Cl₂, -20-0 °C; (e) KHCO₃, CH₃OH/H₂O, 50 °C; (f) NaN₃, NH₄Cl, CH₃OH/H₂O, reflux; (g) Ph₃P, THF, rt; Et₃N/H₂O, rt; (h) NaN₃, NH₄Cl, DMF, 70 °C; (i) Ac₂O, pyridine, rt; (j) Ph₃P, THF/H₂O, 50 °C; KOH; Dowex (H⁺); (k) CH₃SNa, CH₃CN, rt; (l) Me₂CuLi, Et₂O, -78 °C.

acid gave 3-pentyl ketal 9 in 94% yield. Reductive acetal cleavage of 9 resulted in a mixture of 10 and its regio-isomer in 70% yield (3:2 ratio), which could not be separated by chromatography on silica gel. When the mixture of 10 and its regio-isomer was treated with potassium hydrogen carbonate in methanol-water, the regio-isomer remained unreacted, and 10 reacted to give epoxide 11, which was isolated by silica gel chromatography. Nucleophilic ring opening of epoxide 11 with sodium azide resulted in azido alcohol 12. Mesylation of 12 gave 13, which was then treated with triphenylphosphine in anhydrous tetrahydrofuran followed by triethylamine and water to give aziridine 14 in 75% yield. The ring opening of aziridine 14 with sodium azide gave 15 in 54% yield. Acetylation of 15 followed by reduction of the azido group and saponification of the methyl ester gave the C2-chloro analog 2.

The preparation of C₂-methylthio analog 3 was accomplished starting with intermediate 16. Michael addition of the methylthio group to 16 followed by elimination of the chloro group resulted in 17. Reduction of 17 followed by the saponification gave C₂-methylthio analog 3.

The synthesis of C₂-methyl analog was also accomplished starting from intermediate 8.⁵ Acetylation of 8 gave acetonide 18, which was then treated with methyl cuprate⁶ to give a mixture (1:1 ratio) of 19 and the reduced product in 70% yield. After removal of the acetyl group, 19 was converted to final product 4 through the same procedures used for the preparation of compound 2.

Table 1. Influenza Neuraminidase Inhibition⁷

Compound	R	IC ₅₀ (nM)
1	Н	1.4
2	Cl	3100
3	SCH ₃	3400
4	CH ₃	2300
		*

The enzymatic neuraminidase inhibitory activity of the three compounds is shown in Table. The results show that the introduction of a lipophilic chloro, methyl or methylthio group at the C₂ position of GS4 071 resulted in an approximately 2000-fold decrease in the inhibitory activity. This result indicates that the enzyme active site contains a limited hydrophobic pocket capable of accommodating modifications at the C₂ position of GS 4071.

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Neuraminidase enzyme assay: Neuraminidase enzyme activity was determined using minor modifications to the literature method. Influenza A/PR/8/34 (H1N1), purified on sucrose density gradients, was used as the source of enzyme and 2'-(4-methylumbelliferyl)- α -D-N-acetylneuraminic acid was used as substrate in a reaction buffer containing 33 mM MES, pH 6.5, and 4 mM calcium chloride. Virus was mixed with various inhibitor concentrations and incubated at room temperature for 30 min. before substrate was added to a final concentration of 10 μ M. Reactions were stopped after 8 min. at 37 °C with the addition of 1.5 volumes of 14 mM sodium hydroxide in 83% ethanol. Fluorescence was quantitated in a Perkin-Elmer fluorimeter (Model LS50B) with an excitation wavelength of 360 nm, emission wavelength of 448 nm, and slit widths of 2.5 nm.

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References and Notes

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