

An Improved Synthesis of 17α -Acyloxy-21-chloro-substituted Corticosteroids using Xanthene-9-carbonyl Chloride

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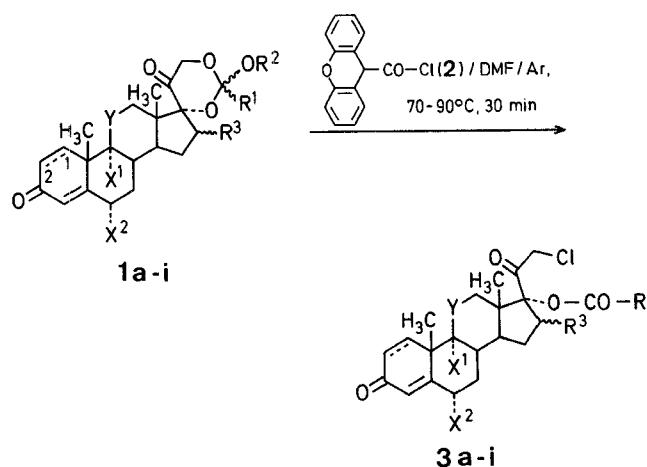
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Certain 17α -acyloxy-21-chloro corticosteroids **3** are known to be clinically useful as a potent topical antiinflammatory agent¹. These corticosteroids **3** have been hitherto synthesized from the corresponding $17\alpha,21$ -cyclic ortho esters **1** in 3 or 5 steps^{2,3,4}. Recently, a direct conversion from **1** to **3** by reaction with acid chlorides has been reported in a patent⁵.

However, the yields of compounds **3** are poor (0–28%) since the reaction has to be carried out with reactive acid chlorides such as acetyl chloride at room temperature. With these reagents, decomposition of **1** may occur by the action of hydrochloric acid and/or the carboxylic acid resulting from hydrolysis of the acid chloride. Thus, we found that compounds **1** are decomposed completely on reaction with dry hydrogen chloride. Furthermore, it was stated in the patent that the presence of dimethyl sulfoxide and/or a tertiary amide such

as dimethylformamide are essential for the preparation of products **3**.

Because of these practical limitations, we have developed an improved synthesis of products **3** using moisture-stable, crystalline acid chlorides. Thus, xanthene-9-carbonyl chloride (**2**)⁶ was found to convert **1** to **3** in good yields within 30 min at 70–90 °C. The presence of a tertiary amide or dimethyl sulfoxide is not necessary. Good yields (40–60%) could also be obtained using 1,2-dichloroethane, benzene, or acetonitrile as solvent.



1,3	C(1)-C(2)	X ¹	X ²	Y	R ¹	R ²	R ³
a	Δ^1	H	α -CH ₃	CH \rightarrow OH	C ₂ H ₅	C ₂ H ₅	H
b	Δ^1	H	α -CH ₃	CH \rightarrow OH	n-C ₃ H ₇	C ₂ H ₅	H
c	sat.	H	H	CH \rightarrow OH	C ₂ H ₅	C ₂ H ₅	H
d	Δ^1	H	H	CH \rightarrow OH	C ₂ H ₅	C ₂ H ₅	H
e	Δ^1	H	α -CH ₃	CH \rightarrow OH	C ₆ H ₅	CH ₃	H
f	Δ^1	F	H	CH \rightarrow OH	C ₂ H ₅	C ₂ H ₅	α -CH ₃
g	Δ^1	Cl	H	CH \rightarrow OH	C ₂ H ₅	C ₂ H ₅	β -CH ₃
h	sat.	H	H	CH ₂	C ₂ H ₅	C ₂ H ₅	H
i	Δ^1	H	H	C=O	C ₂ H ₅	C ₂ H ₅	H

Table 1. Corticosteroid $17\alpha,21$ -Cyclic Ortho Esters **1a-i** prepared

No.	Name	Yield [%]	m.p. [°C]	Molecular formula ^a or Lit. m.p.	I.R. (KBr) ν [cm ⁻¹]	M.S. m/e
1a	6 α -methylprednisolone- 17 α ,21-ethyl orthopropanoate	91	160–164° (dec)	C ₂₇ H ₃₈ O ₆ (458.6)	3350 (OH); 1720, 1645 (C=O)	459 (M ⁺ + 1); 458 (M ⁺); 441, 136, 135 (100%)
1b	6 α -methylprednisolone- 17 α ,21-ethyl orthobutanoate	91	164–166° (dec)	C ₂₈ H ₄₀ O ₆ (472.6)	3340 (OH); 1720, 1645 (C=O)	473 (M ⁺ + 1); 472 (M ⁺); 427, 356, 136, 135 (100%)
1c	hydrocortisone- 17 α ,21-ethyl orthopropanoate	90	186–190°	183–185 ^{o5}	—	—
1d	prednisolone- 17 α ,21-ethyl orthopropanoate	93	187–189°	180–184 ^{o5}	—	—
1e	6 α -methylprednisolone- 17 α ,21-methyl orthobenzoate	84	amorphous	C ₃₀ H ₃₆ O ₆ (492.6)	3350 (OH); 1715, 1645 (C=O)	493 (M ⁺ + 1); 492 (M ⁺); 475; 356; 136; 135; 105; 77
1f	dexamethasone- 17 α ,21-ethyl orthopropanoate	85	218–220°	219–221 ^{o5}	—	—
1g	beclomethasone- 17 α ,21-ethyl orthopropanoate	84	146–147°	C ₂₇ H ₃₇ ClO ₆ (493.0)	3340 (OH); 1715, 1655 (C=O)	449 (M ⁺ + 2 – OC ₂ H ₅); 447 (M ⁺ – OC ₂ H ₅); 410; 359; 331 (100%)
1h	11-deoxyhydrocortisone- 17 α ,21-ethyl orthopropanoate	88	180–182°	C ₂₆ H ₃₈ O ₅ (430.6)	1723, 1658 (C=O)	431 (M ⁺ + 1); 430 (M ⁺); 402; 242; 227; 124; 57 (100%)
1i	prednisone- 17 α ,21-ethyl orthopropanoate	92	188–191°	C ₂₆ H ₃₄ O ₆ (442.6)	1723, 1698, 1658 (C=O)	443 (M ⁺ + 1); 442 (M ⁺); 414; 368; 242; 121; 57 (100%)

^a Satisfactory microanalyses obtained: C ± 0.29, H ± 0.18, Cl + 0.06.

Table 2. 17 α -Acyloxy-21-chloro-corticosteroids 3a-i

Prod- uct	Yield ^a [%]	m.p. [°C]	[α] _D ²³ (C ₂ H ₅ OH)	Molecular formula ^b or Lit. m.p. [°C]	I.R. (KBr) ν [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃ /TMS) δ [ppm]	M.S. (30 eV) <i>m/e</i>
3a	88 ^c	147–149°	+63°	C ₂₅ H ₃₃ ClO ₅ (449.0)	3360 (OH); 1720, 1710, 1645 (C=O)	1.0–1.2 (m, 9 H); 1.47 (s, 3 H); 4.5 (br. s, 1 H); 6.0 (br. s, 1 H); 6.25 (d, 1 H, <i>J</i> =10 Hz); 7.37 (d, 1 H, <i>J</i> =10 Hz)	450 (M ⁺ +2); 449, 448 (M ⁺)
3b	57	120–123°	+50°	C ₂₆ H ₃₅ ClO ₅ (463.0)	3400 (OH); 1720, 1715, 1645 (C=O)	0.9 (m, 3 H); 1.00 (s, 3 H); 1.13 (d, 3 H, <i>J</i> =8 Hz); 1.46 (s, 3 H); 4.15 (d, 2 H, <i>J</i> =1.5 Hz); 4.48 (br. s, 1 H); 6.03 (br. s, 1 H); 6.26 (d, 1 H, <i>J</i> =10 Hz); 7.33 (d, 1 H, <i>J</i> =10 Hz)	464 (M ⁺ +2); 463, 462 (M ⁺)
3c	63	240–242 ^{od}	—	225–227 ^{o5}	3360 (OH); 1725, 1715, 1645 (C=O)	1.00 (s, 3 H); 1.14 (t, 3 H, <i>J</i> =8 Hz); 1.46 (s, 3 H); 4.27 (d, 2 H, <i>J</i> =1 Hz); 4.55 (br. s, 1 H); 5.71 (s, 1 H)	438 (M ⁺ +2); 437, 436 (M ⁺)
3d	71	223–225°	—	225–227 ^{o5}	3360 (OH); 1725, 1715, 1645 (C=O)	—	436 (M ⁺ +2); 435, 434 (M ⁺)
3e	50	157–159°	+17°	C ₂₉ H ₃₃ ClO ₅ ·0.5H ₂ O (506.0)	3400 (OH); 1715, 1700, 1650 (C=O)	1.08 (s, 3 H); 1.2 (m, 3 H); 1.53 (s, 3 H); 4.18 (s, 2 H); 4.6 (br. s, 1 H); 6.1 (br. s, 1 H); 6.29 (d, 1 H, <i>J</i> =10 Hz); 7.3– 8.1 (m, 6 H)	498 (M ⁺ +2); 497, 496 (M ⁺)
3f	58	241–243°	—	241–243 ^{o5}	3270 (OH); 1730, 1720, 1655 (C=O)	—	468 (M ⁺ +2); 467, 466 (M ⁺)
3g	41	228–230 ^{od}	—	202–205 ^{o5}	3400 (OH); 1738, 1730, 1655 (C=O)	1.01 (s, 3 H); 1.15 (t, 3 H, <i>J</i> =8 Hz); 1.37 (d, 3 H, <i>J</i> =8 Hz); 1.66 (s, 3 H); 3.99 (s, 2 H); 4.56 (br. s, 1 H); 6.06 (br. s, 1 H); 6.28 (d, 1 H, <i>J</i> =10 Hz); 7.23 (d, 1 H, 10 Hz)	486 (M ⁺ +4); 484 (M ⁺ +2); 482 (M ⁺)
3h	68	177–179°	+63°	C ₂₄ H ₃₃ ClO ₅ (421.0)	1725, 1715, 1660 (C=O)	0.73 (s, 3 H); 1.13 (t, 3 H, <i>J</i> =8 Hz); 1.18 (s, 3 H); 4.12 (d, 2 H, <i>J</i> =1.5 Hz); 5.73 (s, 1 H)	422 (M ⁺ +2); 421, 420 (M ⁺)
3i	77	amorphous ^d	—	97–100 ^{o4}	1730, 1725, 1700, 1660 (C=O)	0.77 (s, 3 H); 1.11 (t, 3 H, <i>J</i> =8 Hz); 1.42 (s, 3 H); 4.11 (s, 2 H); 6.12 (s, 1 H); 6.23 (d, 1 H, <i>J</i> =10 Hz); 7.66 (d, 1 H, <i>J</i> =10 Hz)	434 (M ⁺ +2); 433, 432 (M ⁺)

^a Non-optimized yield of isolated product.^b Satisfactory microanalyses obtained: C ±0.26, H ±0.19, Cl ±0.14.^c In 1,2-dichloroethane 49% yield, in benzene 59% yield, in acetonitrile 60% yield.^d These products are homogeneous by T.L.C. and satisfactory microanalyses were obtained (C ±0.26, H ±0.16, Cl ±0.12).

Other aryl chlorides proved to be less suitable than **2**. Acid chloride **2** can be recovered easily as its ethyl or methyl ester from the reaction mixture and recycled if desired. The starting materials **1** were obtained in 84–93% yield from the corresponding corticosteroids and ethyl or methyl ortho esters.

Corticosteroid 17 α ,21-Cyclic Ortho Esters 1a-i; General Procedure:

To a stirred solution of the corticosteroid (2 mmol) in dimethylformamide (4 ml) are added the ethyl or methyl ortho ester (4 mmol) and anhydrous *p*-toluenesulfonic acid (0.1 mmol) under argon. The mixture is heated at 85 °C for 1.5 h, 10% sodium carbonate solution (0.5 ml) and ethyl acetate (50 ml) are added at room temperature. The resultant mixture is washed with water (3 × 30 ml), dried with sodium sulfate, and filtered. The solvents are evaporated and the residual product isolated by recrystallization from ether/hexane to afford the product **1**; yield: 84–93% (Table 1).

Xanthene-9-carbonyl Chloride (2)⁶:

Prepared from commercially available xanthene-9-carboxylic acid (2.26 g, 10 mmol) and thionyl chloride (1.58 g, 13.3 mmol) in dry dichloromethane (12 ml) under reflux for 7 h; yield: 92%; m.p. 86–88 °C; colorless needles from hexane.

I.R. (KBr): ν =1775 cm⁻¹ (CO).

17 α -Acyloxy-21-chloro-corticosteroids 3a-i; General Procedure:

To a solution of the corticosteroid 17 α ,21-ethyl or methyl ortho ester **1** (1 mmol) in dimethylformamide (3 ml) heated at 80 °C under argon is added xanthene-9-carbonyl chloride (**2**; 0.37 g, 1.5 mmol). The mixture is stirred for 30 min, cooled to room temperature and ethyl acetate (50 ml) and 10% sodium carbonate solution (1 ml) are added. The resultant mixture is washed with water (3 × 30 ml), dried with anhydrous sodium sulfate, and evaporated. The residual product is isolated by recrystallization from ether/hexane or by preparative thin layer chromatography on silica gel; yield: 41–88% (Table 2).

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¹ For example, 21-chloro-9 α -fluoro-16 β -methyl-17 α -(1-oxopropoxy)-pregna-1,4-diene-3,20-dione has been used clinically: C. G. Sparkes, L. Wilson, *Br. J. Dermatol.* **90**, 197 (1974).

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⁵ Y. Kamano et al., *German Patent* 2613875 (1978); *C. A.* **86**, 90137 (1977).

⁶ J. W. Cusic, *U.S. Patent* 2650230 (1953); *C. A.* **48**, 11500 (1954).

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