## **Collective Synthesis of Cladiellins Based on the Gold-Catalyzed Cascade Reaction of 1,7-Diynes**\*\*

Guozong Yue, Yun Zhang, Lichao Fang, Chuang-chuang Li, Tuoping Luo,\* and Zhen Yang\*

Abstract: The cladiellin family of natural products, which includes molecules with various biological activities, continues to invite new synthetic studies. A gold-catalyzed tandem reaction of 1,7-diynes to construct the 6-5-bicyclic ring systems that are present in a number of natural products was developed. This reaction was applied as the key step to realize the formal and total syntheses of nine members of the cladiellin family in an enantio- and diastereoselective manner. This modular and efficient approach could also be used for the construction of other cladiellins, as well as their analogues, for follow-up studies.

The cladiellins (compounds 1–9) are a group of structurally complex natural products that have attracted considerable attention from the organic synthesis community.<sup>[1]</sup> Among them, (-)-sclerophytin A (1) is of particular interest because it has been reported to be toxic to mouse lymphocytic leukemia L1210 cells at concentrations as low as 3 nm.<sup>[2]</sup> Several other cladiellins have also been reported to possess potential antitumor activities, including (+)-vigulariol (6), which exhibited in vitro cytotoxicity towards human lung adenocarcinoma A-549 cells, and (-)-pachycladin D (9), which showed anti-invasive activity.<sup>[3,4]</sup> Another cladiellin, (+)-polyanthellin A (4), has been reported to be an antimalarial reagent based on its inhibitory activity towards Plasmodium falciparum.<sup>[5]</sup> Despite these reports, the antitumor and antimicrobial activities of the many other compounds belonging to the cladiellin family have not been extensively evaluated. Comparison studies with cladiellins and their unnatural structural analogues through the use of biological assays could provide a detailed understanding of their structure-activity relationships and could effectively highlight the potential of cladiellins as lead compounds for the

[\*] Prof. Dr. T. Luo, Prof. Dr. Z. Yang Key Laboratory of Bioorganic Chemistry and Molecular Engineering of Ministry of Education and Beijing National Laboratory for Molecular Science (BNLMS) Peking-Tsinghua Center for Life Sciences, Peking University Beijing 100871 (China)
E-mail: tuopingluo@pku.edu.cn zyang@pku.edu.cn
G. Yue, Y. Zhang, L. Fang, Prof. Dr. C.-c. Li, Prof. Dr. Z. Yang

Laboratory of Chemical Genomics, School of Chemical Biology and Biotechnology, Peking University Shenzhen Graduate School Shenzhen 518055 (China) development of small-molecule probes and drugs.<sup>[6]</sup> Several members of the cladiellin family have been successfully synthesized by a number of research groups.<sup>[7-15]</sup> In fact, research into the total synthesis of cladiellins has played an important role in determining their structures.<sup>[7,8,9e]</sup> The tricyclic compound **10** has been used as a common late-stage intermediate in a number synthetic strategies for accessing various cladiellins, including **1**, **2**, **5**, and **6**.<sup>[9f,12a,c,15]</sup> Given that compounds **1–9** differ only in terms of their substituents at C6, C7, and C11 (sclerophytin A numbering used throughout), the development of a method for the conversion of **10** into **3**, **4**, **7**, **8**, and **9** would represent a significant addition to the techniques currently available for the synthesis of cladiellins (Figure 1).<sup>[12c]</sup>

Among the different strategies that have been developed for the construction of enantiopure **10**,<sup>[9f,12c,15]</sup> the preferential ring-closing olefin metathesis of bicyclic compound 11 is relatively straightforward but only provides the desired product in 40–50% yield under the reported conditions.<sup>[15]</sup> Operating under the assumption that the yield for this ringclosing olefin metathesis step could be improved, it was envisaged that the 6-5-bicyclic ring system of 11 could be accessed through the gold-catalyzed cascade reaction of 1,7diynes that we recently developed during our synthesis of compounds belonging to the drimane family of natural products.<sup>[16]</sup> The precursor for 11, compound 12, could be produced from 1,7-diyne 14 in one step via intermediate 13. Enantiopure 1,7-diyne 14, which has three contiguous stereogenic centers, could itself be prepared by a Marshall coupling reaction between propargyl mesylate 15 and ethyl glyoxalate (16).<sup>[17]</sup> Herein, we disclose our contribution towards the development of an efficient and modular strategy for the synthesis of cladiellins that complements other approaches in terms of providing a platform for the synthesis of related analogues.

Our modular and scalable synthesis of chiral 1,7-diyne 14 is shown in Scheme 1. The diastereoselective  $\alpha$ -alkylation of amide 18, which was prepared from commercially available hex-5-ynoic acid (17), afforded the tertiary stereogenic center at C14 of the cladiellins.<sup>[18]</sup> The superiority of pseudoephenamine as an auxiliary in asymmetric alkylation reactions involving isopropyl iodide as the electrophile allowed 19 to be prepared on a multigram scale.<sup>[19]</sup> Successive hydrolysis and amide coupling reactions yielded a Weinreb amide, which was subsequently allowed to react with an acetylenyl Grignard reagent to give ketone 21. The asymmetric reduction of ketone 21 with (*S*)-alpine-borane<sup>[20]</sup> followed by mesylation of the resulting alcohol afforded mesylate 15 as a single diastereomer in 56% yield. Marshall coupling of 15 and aldehyde 16 under optimized conditions afforded a separable

 <sup>[\*\*]</sup> We thank the National Basic Research Program of China (973 Program, Grant No. 2009CB940904) and the National 863 Program (Grant No. 2013AA092903) for financial support.

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/anie.201309449.





*Figure 1.* Retrosynthetic analysis of cladiellins via key intermediate **10**: a strategy enabled by the gold-catalyzed cascade reaction of 1,7-diyne **14**.

mixture of diynes **14** and **22** in 90% overall yield with a moderate level of stereoselectivity (d.r. = 2.5:1; Table S1, see the Supporting Information for details). Pleasingly, the undesired diastereomer **22** could be readily converted into **14** in two steps through the inversion of one stereogenic center.

With abundant 1,7-diyne **14** in hand, we proceeded to investigate the feasibility of our key cascade transformation for the construction of the 6-5-bicyclic skeleton. Subjecting **14** to the gold-catalyzed cascade reaction in the presence of excess methylallyl trimethylsilane, however, led only to the *O*-TMS protection product **23**,<sup>[21]</sup> and decomposition was observed when phenol was added to scavenge the TMS group.<sup>[22]</sup> By contrast, the use of *p*-nitrobenzylic alcohol as a nucleophile allowed the desired 6-5-bicyclic skeleton to be formed successfully, albeit as a pair of diastereomers **24a** and **24b**. Following a period of optimization (Table S2 in the Supporting Information), **14** could be converted into **24a** (98% *ee*) and **24b** in 65% yield with 3:1 diastereoselectivity on a gram scale (Scheme 2). Given that both **24a** and **24b** 



Scheme 1. Scalable enantioselective synthesis of 1,7-diyne 14. Yields of isolated products (%) are indicated. Reagents and conditions: a) EDCI (1.1 equiv), HOBt (1.1 equiv), (-)-(15,25)-pseudoephenamine (1.1 equiv), DIPEA (3.0 equiv), DMF, RT. b) LiCl (6.0 equiv), LDA (4.0 equiv), isopropyl iodide (6.0 equiv), THF. c) nBu₄NOH (5.0 equiv), tert-butyl alcohol/water = 1:1, reflux. d) EDCI (3.0 equiv), HOBt (3.0 equiv), DIPEA (6.0 equiv), N,O-dimethylhydroxylamine hydrochloride (3.0 equiv), DMF. e) ethynylmagnesium chloride (2.0 equiv), THF. f) (S)-alpine borane (2.0 equiv). g) MsCl (2.0 equiv), Et<sub>3</sub>N (4.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>. h) 5% [Pd(dppf)Cl<sub>2</sub>]·CH<sub>2</sub>Cl<sub>2</sub>, InI (1.1 equiv), 16 (2.0 equiv), THF/HMPA = 3:1. i) PPh<sub>3</sub> (2.5 equiv), DIAD (3.0 equiv), p-NO<sub>2</sub>PhCO<sub>2</sub>H (2.5 equiv), THF. j) K<sub>2</sub>CO<sub>3</sub> (2.0 equiv), EtOH. EDCI = 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide, HOBt = 1-hydroxybenzotriazole, LDA = lithium diisopropylamide, DIPEA = diisopropylethylamine, Ms = methanesulfonyl, TEA = triethylamine, dppf = 1,1'-bis(diphenylphosphanyl)ferrocene, HMPA = (hexamethylphosphoramide, DIAD = diisopropyl azodicarboxylate.

could be used in the next step, the mixture of 24a and 24b was treated with excess (methylallyl)trimethylsilane and  $\text{TMSOTf}^{[23]}$  to give 12 and its C9 diastereomer 25 in yields of 60 and 20%, respectively. Ester 12 was then converted into 11 through sequential Weinreb amide formation and Grignard reactions. Morken and co-workers. reported the ring-closing metathesis of 11 to afford 10 in 55% yield after resubjection of the recovered starting material and side product to the reaction conditions,<sup>[15]</sup> and this result was successfully reproduced in our hands. Pleasingly, however, the use of the Zhan 1B catalyst 26 led to a significant increase in the efficiency of the desired transformation, presumably because of the enhanced stability of the catalyst.<sup>[24]</sup> This modification afforded the key intermediate 10 in 70% yield in a single operation. The analytical data for 10 were identical to those reported<sup>[9f,12a,c,15]</sup> and confirmed that we had produced the material required to explore the synthesis of compounds 1-9. Our initial efforts focused on the synthesis of compounds



(-)-Sclerophytin B (2)

Scheme 2. Synthesis of (-)-sclerophytin A (1), (-)-sclerophytin B (2), (+)-cladiella-6Z,11(17)-dien-3-ol (5), and (+)-vigulariol (6). Yields of isolated products (%) are indicated. Reagent and conditions: a) [(IPr)AuCl] (5 mol%), AgSbF<sub>6</sub> (5 mol%), *p*-NO<sub>2</sub>PhCH<sub>2</sub>OH (1.5 equiv). b) TMSOTf (1.6 equiv), methylallyl-trimethylsilane (3.2 equiv), CH<sub>3</sub>CN. c) N,O-Dimethylhydroxylamine hydrochloride (3.0 equiv), isopropylmagnesium chloride (6.0 equiv), THF. d) 3-Butenyl magnesium bromide (2.0 equiv), THF. e) Grubbs 2nd generation catalyst (10 mol%), PhH, reflux. f) Zhan 1B catalyst (10 mol%), toluene, reflux. g) mCPBA (1.4 equiv), CHCl<sub>3</sub>, -12°C. h) LiOH (18.0 equiv), dioxane, H<sub>2</sub>O; then KHSO<sub>4</sub> (18.0 equiv), Sc(OTf)<sub>3</sub> (1.5 equiv), CH<sub>3</sub>CN, H<sub>2</sub>O, RT. i) MeMgCl (120.0 equiv), THF, 52 °C, 24 h. j) Ac<sub>2</sub>O (3.0 equiv), Et<sub>3</sub>N (7.5 equiv), DMAP (1.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0°C. k) MeMgCl (20.0 equiv), THF, 0°C. l) mCPBA (2.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0°C. TMS = trimethylsilyl, Tf = trifluoromethanesulfonyl, mCPBA = mchloroperbenzoic acid, DMAP = 4-dimethylaminopyridine.

**1**, **2**, **5**, and **6** by using the published procedures.<sup>[8c,9f,12,15]</sup> The spectroscopic data (<sup>1</sup>H and <sup>13</sup>C NMR spectra, HRMS analyses, and optical rotation) of the synthesized samples of **1**, **2**, **5**, and **6** were in good agreement with those reported for the corresponding natural materials.<sup>[9f,12,13,15]</sup>

Given that (+)-polyanthellin A (4) can be prepared by the acetylation of (+)-deacetylpolyanthellin A (3),<sup>[11,12c,14]</sup> we



**Scheme 3.** Synthesis of (+)-deacetylpolyanthellin A (**3**) and (+)-polyanthellin (**4**). Yields of isolated products (%) are indicated. Reagents and conditions: a) Hg(OAc)<sub>2</sub> (2.0 equiv), THF, RT, 2 h; then H<sub>2</sub>O, Hg-(OAc)<sub>2</sub> (2.0 equiv), 2 h; then Et<sub>3</sub>B (10.0 equiv), NaBH<sub>4</sub> (10.0 equiv), -20 °C to RT, overnight. b) Ac<sub>2</sub>O (5.0 equiv), Et<sub>3</sub>N (5.0 equiv), DMAP (5.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, RT, 24 h.

aimed to prepare **3** from **5** by installing additional cyclic ether and tertiary alcohol moieties (Scheme 3). A regio- and stereoselective double oxymercuration followed by a reduction, the combination of which is an efficient one-pot protocol that has successfully been used for the synthesis of **3** from the other two advanced intermediates,<sup>[11,14]</sup> accomplished this task in moderate yield. Thus, the total synthesis of the natural product (+)-polyanthellin A (**4**) was achieved in 62 % yield by the reaction of **3** with acetic anhydride in the presence of Et<sub>3</sub>N and DMAP.<sup>[11,12c,14]</sup>

To access (-)-cladiellinsin (7), (-)-pachycladin C (8), and (-)-pachycladin D (9), we first converted 10 into hemiketal 27 by following known procedures (Scheme 4).<sup>[9f,15]</sup> Selective protection of the hemiketal hydroxy group in 27 afforded tertiary alcohol 28, which underwent regioselective dehydration followed by deprotection to furnish hemiketal 29 in 72 % yield over three steps.<sup>[25,26]</sup> The addition of methylmagnesium chloride at an elevated temperature afforded 7 in 93 % yield as a single diastereoisomer. Finally, 7 was converted into 8 and 9 by acetylation and oxidation of the secondary alcohol, respectively. The analytical data for 7, 8, and 9 corresponded well with those reported from the isolated natural materials.<sup>[4]</sup>

In summary, we have developed a concise and general strategy to access a number of cladiellins through the gold-catalyzed cascade reaction of 1,7-diynes. The modularity of our approach will enable the efficient preparation of a variety of natural product analogues. The biological activities of our synthesized cladiellins, especially their cytotoxicities against various cancer cells, are under evaluation.

Received: October 30, 2013

## Angewandte Communications



**Scheme 4.** Total synthesis of (-)-cladiellinsin (7), (-)-pachycladin C (8) and (-)-pachycladin D (9). Yields of isolated products (%) are indicated. Reagents and conditions: a) *m*CPBA (1.4 equiv), CHCl<sub>3</sub>, -12 °C. b) LiOH (18.0 equiv), dioxane, H<sub>2</sub>O; then KHSO<sub>4</sub> (18.0 equiv), Sc(OTf)<sub>3</sub> (1.5 equiv), CH<sub>3</sub>CN, H<sub>2</sub>O, RT. c) *N*-(Trimethylsilyl)imidazole (10.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>. d) Burgess reagent (4.0 equiv), toluene, 70 °C. e) TBAF (8.0 equiv), THF. f) MeMgCl (90.0 equiv), toluene, 100 °C. g) Ac<sub>2</sub>O (3.0 equiv), Et<sub>3</sub>N (6.0 equiv), DMAP (2.0 equiv). h) MnO<sub>2</sub> (60.0 equiv). TBAF = tetra-*n*-butylammonium fluoride.

**Keywords:** cascade reaction · cyclization · gold · natural products · total synthesis

- Reviews: a) P. Bernardeli, L. A. Paquette, *Heterocycles* 1998, 49, 531–556; b) J. M. Ellis, M. T. Crimmins, *Chem. Rev.* 2008, 108, 5278–5298; c) A. J. Welford, I. Collins, *J. Nat. Prod.* 2011, 74, 2318–2328.
- [2] a) P. Sharma, M. Alam, J. Chem. Soc. Perkin Trans. 1 1988, 2537 2540; b) M. Alam, P. Sharma, A. S. Zektzer, G. E. Martin, X. Ji, D. J. van der Helm, J. Org. Chem. 1989, 54, 1896 1900.
- [3] J.-H. Su, H.-C. Huang, C.-H. Chao, L.-Y. Yan, Y.-C. Wu, J.-H. Sheu, Bull. Chem. Soc. Jpn. 2005, 78, 877–879.
- [4] H. M. Hassan, M. A. Khanfar, A. Y. Elnagar, R. Mohammed, L. A. Shaala, D. T. A. Youssef, M. S. Hifnawy, K. A. El Sayad, J. *Nat. Prod.* 2010, *73*, 848–853.
- [5] C. A. Ospina, A. D. Rodríguez, E. Ortega-Barria, T. L. Capson, J. Nat. Prod. 2003, 66, 357–363.
- [6] a) D. J. Newman, J. Med. Chem. 2008, 51, 2589–2599; b) S. J. Miller, J. Clardy, Nat. Chem. 2009, 1, 261–263; c) E. E. Carlson, ACS Chem. Biol. 2010, 5, 639–653.
- [7] a) D. W. C. MacMillan, L. E. Overman, J. Am. Chem. Soc. 1995, 117, 10391-10392; b) L. E. Overman, L. D. Pennington, Org. Lett. 2000, 2, 2683-2686; c) D. W. C. MacMillan, L. E. Overman, L. D. Pennington, J. Am. Chem. Soc. 2001, 123, 9033-9044; d) F. Gallou, D. W. C. MacMillan, L. E. Overman, L. A. Paquette, L. D. Pennington, J. Yang, Org. Lett. 2001, 3, 135-137.

- [8] a) L. A. Paquette, O. M. Moradei, P. Bernardelli, T. Lange, Org. Lett. 2000, 2, 1875–1878; b) D. Friedrich, R. W. Doskotch, L. A. Paquette, Org. Lett. 2000, 2, 1879–1882; c) P. Bernardelli, O. M. Moradei, D. Friedrich, J. Yang, F. Gallou, B. P. Dyck, R. W. Doskotch, T. Lange, L. A. Paquette, J. Am. Chem. Soc. 2001, 123, 9021–9032.
- [9] a) M. T. Crimmins, B. H. Brown, J. Am. Chem. Soc. 2004, 126, 10264–10266; b) M. T. Crimmins, J. M. Ellis, J. Am. Chem. Soc. 2005, 127, 17200–17201; c) M. T. Crimmins, B. H. Brown, H. R. Plake, J. Am. Chem. Soc. 2006, 128, 1371–1378; d) M. T. Crimmins, M. Ellis, J. Org. Chem. 2008, 73, 1649–1660; e) M. T. Crimmins, M. C. Mans, A. D. Rodríguez, Org. Lett. 2010, 12, 5028–5031; f) M. T. Crimmins, C. S. Stauffer, M. C. Mans, Org. Lett. 2011, 13, 4890–4893.
- [10] a) G. A. Molander, D. J. St. Jean, Jr., J. Haas, J. Am. Chem. Soc.
   2004, 126, 1642-1643; b) G. A. Molander, B. Czakó, D. J. St. Jean, Jr., J. Org. Chem. 2006, 71, 1172-1180.
- [11] H. Kim, H. Lee, J. Kim, S. Kim, D. Kim, J. Am. Chem. Soc. 2006, 128, 15851–15855.
- [12] a) J. S. Clark, S. T. Hayes, C. Wilson, L. Gobbi, Angew. Chem. 2007, 119, 441-444; Angew. Chem. Int. Ed. 2007, 46, 437-440;
  b) J. S. Clark, R. Berger, S. T. Hayes, L. H. Thomas, A. J. Morrison, L. Gobbi, Angew. Chem. 2007, 119, 10063-10066; Angew. Chem. Int. Ed. 2010, 49, 9867-9870; c) J. S. Clark, R. Berger, S. T. Hayes, H. M. Senn, L. J. Farrugia, L. H. Thomas, A. J. Morrison, L. Gobbi, J. Org. Chem. 2013, 78, 673-696.
- [13] J. Becker, K. Bergander, R. Fröhlich, D. Hoppe, Angew. Chem. 2008, 120, 1678–1681; Angew. Chem. Int. Ed. 2008, 47, 1654– 1657.
- [14] a) M. J. Campbell, J. S. Johnson, J. Am. Chem. Soc. 2009, 131, 10370–10371; b) M. J. Campbell, J. S. Johnson, Synthesis 2010, 2841–2852.
- [15] B. Wang, A. P. Ramirez, J. J. Slade, J. P. Morken, J. Am. Chem. Soc. 2010, 132, 16380-16382.
- [16] H. Shi, L. Fang, C. Tan, L. Shi, W. Zhang, C.-c. Li, T. Luo, Z. Yang, J. Am. Chem. Soc. 2011, 133, 14944–14947.
- [17] J. A. Marshall, C. M. Grant, J. Org. Chem. 1999, 64, 696-697.
- [18] M. R. Morales, K. T. Mellem, A. G. Myers, Angew. Chem. 2012, 124, 4646–4649; Angew. Chem. Int. Ed. 2012, 51, 4568–4571.
- [19] D. A. Sandham, R. J. Taylor, J. S. Carey, A. Fassler, *Tetrahedron Lett.* 2000, 41, 10091–10094.
- [20] M. M. Midland, A. Tramontano, A. Kazubski, R. S. Graham, D. J. S. Tsai, D. B. Cardin, *Tetrahedron* **1984**, *40*, 1371–1380.
- [21]



- [22] a) T. Morita, Y. Okamoto, H. Sakurai, *Tetrahedron Lett.* 1980, 21, 835–838; b) T. Suzuki, T. Watahiki, T. Oriyama, *Tetrahedron Lett.* 2000, 41, 8903–8906.
- [23] X.-F. Ma, Q. Tian, Q. Tang, J.-Z. Zhao, H.-W. Shao, Org. Lett. 2011, 13, 4276–4279.
- [24] a) Z.-Y. Zhan, US Patent, 20070043180A1, 2007; b) Z.-Y. Zhan, WO Patent, 2007003135A1, 2007.
- [25] Y. Li, Z.-X. Chen, Q. Xiao, Q.-D. Ye, T.-W. Sun, F.-K. Meng, W.-W. Ren, L. You, L.-M. Xu, Y.-F. Wang, J.-H. Chen, Z. Yang, *Chem. Asian J.* **2012**, *7*, 2321–2333.
- [26] E. M. Burgess, H. R. Penton, Jr., E. A. Taylor, J. Org. Chem. 1973, 38, 26–31.