This article was downloaded by: [The University of Manchester Library] On: 09 December 2014, At: 14:26 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

The Still-Wittig Rearrangement in the Preparation of a Synthetically Useful 4C-Hydroxymethyl-hex-2-enopyranoside

Andrew G. Wee ^a & Lin Zhang ^a

^a Department of Chemistry , University of Regina , Regina, Saskatchewan, S4S 0A2, Canada Published online: 23 Sep 2006.

To cite this article: Andrew G. Wee & Lin Zhang (1993) The Still-Wittig Rearrangement in the Preparation of a Synthetically Useful 4C-Hydroxymethylhex-2-eno-pyranoside, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 23:3, 325-334, DOI: 10.1080/00397919308009784

To link to this article: http://dx.doi.org/10.1080/00397919308009784

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and

are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <u>http://www.tandfonline.com/page/terms-and-conditions</u>

THE STILL-WITTIG REARRANGEMENT IN THE PREPARATION OF A SYNTHETICALLY USEFUL 4C-HYDROXYMETHYL-HEX-2-ENO-PYRANOSIDE

Andrew G. Wee* and Lin Zhang

Department of Chemistry, University of Regina, Regina, Saskatchewan, S4S 0A2, Canada

ABSTRACT: The Still-Wittig rearrangement of the stannyl-ether 9 is successfully carried out under optimal reaction conditions to give the 4C-hydroxymethy-hex-2-enopyranoside 10 in good yields (70-74%).

Unsaturated sugars typified by the 1,2-dideoxy-hex-1-enopyranosides and the 2,3dideoxy-hex-2-enopyranosides undergo a wide variety of chemical reactions¹ and consequently, they are commonly employed in organic synthesis.² The reactions that are commonly carried out on these unsaturated sugars are the Claisen rearrangement and its modified versions.³ These reactions are especially useful for the introduction of functionalized two-carbon units into the molecule and are effective for stereospecific preparation of C-branched unsaturated sugars. However, the use of the corresponding [2,3]-Wittig rearrangement⁴ for the placement of functionalized one-carbon units, such as the hydroxymethyl group, into the unsaturated sugar moiety has been limited only to the reaction $1 \rightarrow 2 + 3$.

^{*} To whom correspondence should be addressed

Copyright @ 1993 by Marcel Dekker, Inc.

Tulshian and Fraser-Reid reported^{3b} that the Still-Wittig⁴ rearrangement of the stannyl-ether 1 proceeded to give only a very low yield (25%) of the desired C-glycoside 3; the majority of the product was the methyl ether 2 (75%) resulting from proton abstraction by the incipient carbanion.



We have been interested in the preparation of the 4C-hydroxymethyl-hex-2enopyranoside 10 from the stanny-ether 9, which is readily obtained from methyl α -D-glucoside. The application of the Still-Wittig rearrangement reaction to 3,4dideoxy-hex-3-enopyranosides have not been reported. We report, here, that the Still-Wittig rearrangement protocol is successfully employed for the conversion, $9\rightarrow 10$. Compound 10 is a useful chiral intermediate that is rich in functional groups; for example, it already possesses two of the three stereogenic centers present in 4 and can be readily converted to 4. Compound 4 is the key intermediate⁵ used for the synthesis of the medicinally important antibiotic, thienamycin.



a, R=t-BuPh₂Si. i) (n-Bu)₂SnO, MeOH; PhC(O)Cl, Et₃N, 1,4dioxane. ii) Ph₃P, CHI₃, imidazole, PhMe. iii) NaOMe, MeOH-CH₂Cl₂. iv) KH, THF; (n-Bu)₃SnCH₂I, Bu₄NI, THF. v) n-BuLi, THF, -84°C; NH₄Cl.

Scheme Ia

The synthesis of the Still-Wittig precursor 9 is shown in Scheme I. Selective benzoylation of the known triol 5^6 using the procedure of Munavu and Szmant⁷ provided an excellent yield (92%) of the 2-O-benzoate 6. Subsequent treatment of 6 with iodoform-triphenylphosphine⁸ gave the unsaturated benzoate 7 in high yield (94%). It is noteworthy that 6-O-desilylation did not occur, suggesting that the reductive-elimination reaction conditions used here are mild.⁹

Removal of the benzoyl group in 7 using NaOMe in dry MeOH proceeded uneventfully to yield the allylic alcohol 8^{3a} ([α]_D^{23.5} -7.09; c 1.07, CHCl₃) in 99% yield. Treatment of 8 with KH in dry THF generated the corresponding alkoxide which was alkylated with n-Bu₃SnCH₂I,¹⁰ in the presence of a catalytic amount of Bu₄NI, to give the stannyl-ether 9 in 93% yield.

Compound 9 was readily characterised by NMR; in the ¹H NMR, the stannylmethylene protons appear as a pair of AB doublets (J= 10.0 Hz), one centered at δ 3.76 and the other at δ 3.82, and in its ¹³C NMR, the C-6 and the stannylmethylene carbon resonated at δ 59.91 and δ 75.77, respectively.

Having the stannyl-ether 9 in hand, the key rearrangement reaction was examined. Treatment of a 0.05M solution of compound 9 in dry THF with 1.5 equivalent of n-BuLi at -78°C for 2h and, followed by NH₄Cl quench, afforded a 55–60% yield of the homoallylic alcohol 10. The methyl ether arising from protonation of the carbanion intermediate was not detected. Further experiments (varying reaction time, reaction temperature, and the amount of n-BuLi) were conducted in order to optimize the yield of 10. When the reaction was conducted at -40°C (1.5 eq. n-BuLi) for 30 min and then quenched, the yield of 10 was 58%. Conducting the reaction at -78°C (3h), but employing 10 equivalent of n-BuLi¹¹ resulted in a lower yield of 10 (33–55%). The optimal yield of 10 (70–74%) was obtained when the rearrangement was carried out at -84°C and using 1.5 equivalent of n-BuLi. It was also noted that for larger scale reactions ($\approx 0.8-1.0$ g) a lower yield (50–63%) of 10 was obtained.

The salient features of the ¹H NMR of **10** are the multiplet at δ 3.48–3.70 attributed to the C-7 methylene protons and a broad hump ranging from δ 2.60–2.96 due to the hydroxyl group. In its ¹³C NMR the C-6 signal resonated at δ 63.43 and the C-7 signal was observed at δ 66.19.

In summary, we have shown that the Still-Wittig rearrangement of compound 9 is successfully effected at -84° C using 1.5 eq. of n-BuLi and at a reaction scale of $\approx 0.50-0.60$ g of 9. This procedure is suited for the preparation of the synthetically useful compound 10. Also, it is found that the iodoform-triphenylphosphine method is a mild method for effecting reductive-elimination in the presence of a 6-O-(tert-butyldiphenylsilyl) group.

Experimental.

For general procedures, see reference 12. Optical rotations were measured using an Optical Activity AA-5 polarimeter at the sodium-D line (589.6 nm). Tetrahydrofuran (THF) was distilled from sodium-benzophenone ketyl, dichloromethane and toluene were dried by distillation from calcium hydride, and methanol was dried by distillation from magnesium methoxide.

Methyl 2-O-benzoyl-6-O-(tert-butyldiphenylsilyl)-α-D-glucopyranoside (6).

Dibutyltin oxide (11.9g, 47.9 mmol) was added to the triol⁶ 5 (20.6 g, 47.6 mmol) in dry methanol (190 mL) and the resulting milky mixture was refluxed until it became homogenous and clear. After an additional 30 min., the mixture was evaporated to give a thick oil, which was dried under high vacuum overnight. The 2,3-O-dibutylstannylene, obtained as a white foam, was dissolved in dry 1,4-dioxane (324 mL) and dry triethylamine (7.3 mL, 52.4 mmol) was added. Then benzoyl chloride (6.3 mL, 54.3 mmol) was added dropwise at R.T. and, after a brief period, some inorganic salts were precipitated. After 15 h, the salts were filtered off, the residue washed with 1,4-dioxane (20 mL) and the combined filtrates were concentrated. The resulting syrup was chromatographed (EtOAc-

petroleum ether, 1:2) to provide 23.7 g (93%) of **6**. $[\alpha]_D^{23.5}$ + 58.5° (c. 1.20, CHCl₃). IR v_{max} (CH₂Cl₂, film): 3430, 1720, 1600, 1580, 1500 cm⁻¹. ¹H NMR, δ : 1.12 (s, 9H, t-Bu), 2.90-3.30 (br hump, 2H, OH), 3.33 9s, 3H, OMe), 3.71 (dd, 2H, H-6, J= 5.1, 1.5 Hz), 3.91 (m, 2H, H-4, H-5), 4.06-4.19 (m, 1H, H-3), 4.91 (dd, 1H, H-2, J= 9.6, 3.6 Hz), 4.98 (d, 1H, H-1, J= 3.6 Hz), 7.35-7.60 (m, 10H, Ph-H), 7.65-7.75 (m, 3H, Ph-H), 8.13 (d, 2H, Ph-H, J= 7.6 Hz). ¹³C NMR, δ : 19.14, 26.74, 55.02, 64.32(-), 70.38, 71.70, 72.36, 73.78, 96.93, 127.70, 128.32, 129.51, 129.78, 129.85, 132.97, 133.23, 135.57, 166.39. Anal. Calcd for $C_{30}H_{30}O_7Sii$: C, 67.12; H, 6.76. Found: C, 66.91; H, 6.76.

Methyl 2-O-benzoyl-6-O-(tert-butyldiphenylsilyl)-3,4-dideoxy- α -D-hex-3enopyranoside (7).

Ph₃P (3.03 g, 11.6 mmol), imidazole (0.393 g, 5.774 mmol) and recrystallized iodoform (2.27 g, 5.77 mmol) were added to a solution of the diol 6 (1.52 g, 2.83 mmol) in dry toluene (80 mL). The reaction mixture was refluxed for 40 min., cooled to R.T. and then transferred to a separatory funnel. The remaining brown gummy residue was washed several times with ethyl acetate (15 mL). The combined organic phases were washed with satd. NaHCO₃ (2 x 50 mL), dried, filtered and concentrated. The residue was chromatographed (Et₂O-petroleum ether, 1:8) to yield 7 as a colorless oil (1.34 g, 94%). $[\alpha]_D^{23.5}$ –34.3° (c 1.24, CHCl₃). IR v_{max}(film): 1720, 1613, 1600, 1581, 1500 cm⁻¹. ¹H NMR, δ: 1.10 (s, 9H, t-Bu), 3.46 (s, 3H, OMe), 3.78 (dd, 1H, H-6, J= 10.0, 6.0 Hz), 3.83 (dd, 1H, H-6', J= 10.0, 5.5 Hz), 4.23-4.35 (m, 1H, H-5), 5.22 (d, 1H, H-1, J= 4.5 Hz), 5.49-5.59 (m, 1H, H-2), 5.83 (br d, 1H, H-3, J= 10.5 Hz), 6.06 ("dt", 1H, H-4, J=

10.5, 1.9, 1.9 Hz), 7.30-7.65 (m, 10H, Ph-H), 7.62-7.75 (m, 3H, Ph-H), 8.10 (d, 2H, Ph-H, J= 7.6 Hz). ¹³C NMR, δ : 19.34, 26.90, 53.47, 56.00, 65.97, 67.39, 69.07, 96.05, 122.95, 127.77, 128.41, 129.82, 129.90, 133.21, 133.47, 135.67, 135.71, 166.14. Anal. Calcd for C₃₀H₃₄O₅Si.¹/₄H₂O: C, 71.05; H, 6.86. Found: C, 70.92; H, 6.88.

Methyl 6-O-(tert-butyldiphenylsilyl)-2-O(tributylstannylmethyl)-3,4-dideoxy-α-D-erythro-hex-3-enopyranoside (9).

KH (35% dispersion in mineral oil, 0.208 g, 5.19 mmol) was washed with dry THF (2x 4 mL), under N₂ and then dry THF (40 mL) was added. The suspension was cooled to 0°C and a solution of 8 (1.01 g, 2.54 mmol) in dry THF (5 mL) was added, via cannula, to the suspension. After addition was complete, the ice bath was removed and the mixture warmed slowly to R.T. During this time there was vigorous evolution of H₂ gas. The mixture was recooled to 0°C after 1 h, Bu₄NI (0.982 g, 2.66 mmol) was added and a solution of Bu₃SnCH₂I (1.49 g, 3.44 mmol) in dry THF (5 mL) was slowly added via cannula. Then the mixture was warmed slowly to R.T., stirred at R.T. for 1 h and t-BuOH (1 mL) was added. MeOH (4 mL), satd. NH_4Cl solution (8 mL) and water (5 mL) were added, successively, followed by EtOAc (15 mL). The aqueous layer was removed and the organic phase was washed with brine (2x 25 mL), dried, filtered and concentrated. Chromatographic separation of the residue (ether-petroleum ether, 1:16) provided 1.69 g (95%) of the stannyl ether 9 as a colorless oil. $[\alpha]_{D}^{23.5}$ -16.7° (c, 1.20, CHCl₃). IR v_{max} (neat): 1600, 1590, 1550, 1500 cm⁻¹. ¹H NMR, δ : 0.66-0.96 (m, 15H, 3[SnCH₂], 3Me), 1.03 (s, 9H, t-Bu), 1.14-1.68 (m, 12H, 3[(CH₂)₂], 3.44 (s, 3H, OMe), 3.62 (dd, 1H, H-6, J= 10.0, 6.0 Hz), 3.72 (dd, 1H, H-6', J= 10.0, 5.0 Hz), 3.76 (d, 1H, OCHSn, J= 10.0 Hz), 3.82 (d, 1H, OCHSn, J= 10.0 Hz), 4.10-4.20 (m, 1H, H-2), 4.98 (d, 1H, H-1, J= 4.0 Hz), 5.66-5.86 (m, 2H, H-3, H-4), 7.26-7.50 (m, 6H, Ph-H), 7.57-7.78 (m, 4H, Ph-H). ¹³C NMR, δ : 9.02, 13.73, 19.27, 27.30, 29.11, 29.31, 55.80, 59.91, 66.17, 69.36, 75.77, 96.83, 124.91, 127.42, 127.64, 129.65, 133.54, 135.60, 135.66. Anal. Calcd for C₃₆H₅₈O₄SiSn: C, 61.61; H, 8.34. Found: C, 61.53; H, 8.52.

Methyl6-O-(tert-butyldiphenylsilyl)-4-C-(hydroxymethyl)-2,3,4-trideoxy-α-Derythro-hex-2-enopyranoside (10).

A solution of compound 9 (0.550 g, 0.790 mmol) in dry THF (15 mL) was cooled to -84°C (CO₂ slurry in acetone), under argon, and followed by dropwise addition of n-BuLi (0.90 mL, 1.19 mmol, 1.32M in hexane). After 3 h, the reaction was judged complete by t.l.c. and satd. NH₄Cl (3 mL) was added to the mixture. The reaction mixture was allowed to warm slowly to R.T., and EtOAc (10 mL) followed by brine (10 mL) were added. The aqueous layer was removed and reextracted with more EtOAc (3 x 15 mL). The combined organic layers were dried, filtered and evaporated *in vacuo* to give the crude product 6. Purification by chromatography (EtOAc-petroleum ether, 1:4) gave pure 10 (0.239 g, 74%). $[\alpha]_D^{23.5}$ +31.5° (c 1.11, CHCl₃). IR v_{max} (film, CH₂Cl₂): 3575–3275, 1600, 1589, 1500 cm⁻¹. ¹H NMR, δ : 1.04 (s, 9H, t-Bu), 2.36-2.58 (m, 1H, H-4), 2.60-2.96 (br hump, 1H, OH), 3.32 (s, 3H, OMe), 3.48-3.70 (m, 2H, H-7), 3.70-4.00 (m, 3H, H-5, H-6), 4.83 (d, 1H, H-1, J= 2.0 Hz), 5.84 (br s, 2H, H-2,H-3), 7.31-7.60 (m, 6H, Ph-H), 7.68-7.88 (m, 4H, Ph-H). ¹³C NMR, δ : 19.23, 26.84, 40.58, 55.12, 63.43(-), 66.19(-), 70.40, 95.08, 126.40, 127.83, 129.90, 130.90, 133.04, 135.68. Anal. Calcd for C₂₄H₃₂O₄Si: C, 69.85; H, 7.82. Found: C, 70.14; H, 7.84.

Compound **10** was also characterised as its 7-O-acetate: $[\alpha]_D^{23.5} + 39.7$ (c 0.817, CHCl₃). IR υ_{max} (CH₂Cl₂, film): 1743, 1600, 1589, 1500 cm⁻¹. ¹H NMR, δ : 1.10 (s, 9H, t-Bu), 1.96 (s, 3H, C(O)Me), 2.50-2.62 (m, 1H, H-4), 3.45 (s, 3H, OMe), 3.75-3.92 (m, 3H, H-5, H-6), 3.90 (dd, 1H, H-7, J= 10.2, 6.2 Hz); 4.05 (dd, 1H, H-7', J=10.2, 5.7 Hz), 4.88 ("t", 1H, H-1, J=2.0 Hz), 5.81 (dd, 1H, H-2, J= 10.3, 2.0 Hz), 5.89 (br d, 1H, H-3, J= 10.3 Hz), 7.30-7.50 (m, 6H, Ph-H), 7.57-7.77 (m, 4H, Ph-H). ¹³C NMR, δ : 19.24, 20.77, 26.78, 35.72, 55.23, 64.29(-), 65.10(-), 69.71, 95.00, 126.52, 127.64, 127.69, 129.69, 133.36, 133.50, 135.61, 135.72, 171.07. Anal. Calcd for C₂₅H₃₀O₅Si: C, 68.69; H, 7.54. Found: C, 68.74; H, 7.67.

Acknowledgements. We thank the University of Regina for financial support. References.

- a) Fraser-Reid, B. Acc. Chem. Res. 1985, 18, 347. b) Holder, N. Chem. Rev. 1982, 82, 287. c) Ferrier, R. J. Adv. Carbohydr. Chem. Biochem. 1969, 24, 199.
- a) See ref. 1. b) Hanessian, S. Total Synthesis of Natural Products. The Chiron Approach., Pergamon Press, Oxford, 1983. c) Inch, T. D. Tetrahedron, 1984, 40, 3194.

- 3. For leading references: a) Pelyvas, I.; Lindhorst, T.; Thiem, J. Liebigs Ann. Chem. 1990, 761 and refs. cited. b) Tulshian, D. B.; Fraser-Reid, B. J. Org. Chem. 1984, 49, 518 and refs. cited.
- 4. Recent reviews: a) Marshall, J. A. The Wittig Rearrangement, Trost, B. M.; Fleming, I. eds, Pergamon Press, 1991, vol. 3, 975. b) Bruckner, R. (2,3)-Sigmatropic Rearrangement, Trost, B. M.; Fleming, I. eds, Pergamon Press, 1991, vol. 6, 873. c) Nakai, T.; Mikami, T. Chem. Rev. 1986, 86, 885.
- Udodong, U. K.; Fraser-Reid, B. J. Org. Chem. 1989, 54, 2103 and refs. 5. cited.
- Hanessian, S.; Lavallee, P. Can. J. Chem. 1975, 53, 2975. 6.
- Munavu, R. M.; Szmant, H. H. J. Org. Chem. 1976, 41, 1832. 7.
- Bessodes, M.; Abushanab, E.; Panzica, R. P. J. Chem. Soc., Chem. 8. Commun. 1981, 26.
- Cf. Thiem and coworkers^{3a} have observed the lost of the 6-O-(t-9. butyldiphenylsily) protecting group in the Tipson-Cohen reductiveelimination of the corresponding 3.4-di-O-mesvlate.
- 10. Still, W. C. J. Am. Chem. Soc. 1978, 100, 1481.
- The use of excess n-BuLi was based on the premise that coordination of the 11. lithium cation to the oxygens of the sugar moiety (cf. Jarosz, S.; Fraser-Reid, B. Tetrahedron Lett. 1989, 30, 2359) would result in the decrease in the "effective" concentration of n-BuLi if 1.5 equivalent were used.
- 12. Wee, A. G. H. Tetrahedron 1990, 46, 5065 (Received in USA 12 August, 1992)

334