

### FORMAL SYNTHESIS OF (±)-PERHYDROHISTRIONICOTOXIN

Andrew B. Holmes,<sup>\*a</sup> Keith Russell,<sup>a</sup> Edward S. Stern,<sup>†b</sup> Michael E. Stubbs,<sup>c</sup>  
and Nicholas K. Wellard<sup>a</sup>

<sup>a</sup> University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, U.K.

<sup>b</sup> Gallaher Limited, 65 Kingsway, London WC2B 6TG, U.K.

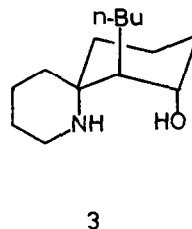
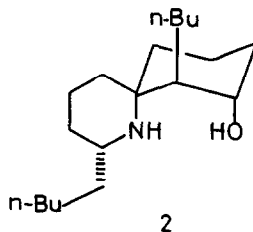
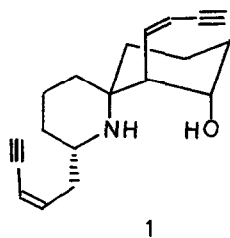
<sup>c</sup> Gallaher Limited, R. and D. Division, Virginia House, Henry Street, Belfast BT15 1JE Northern Ireland, U.K.

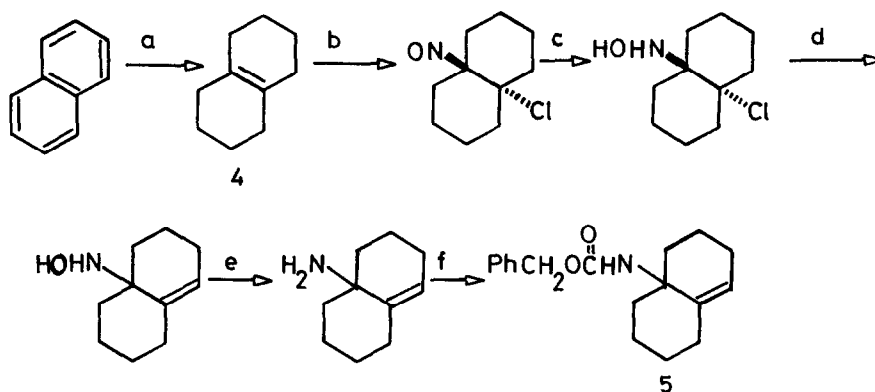
<sup>†</sup> Deceased 7 October, 1982

**Summary:** The syntheses of 1-benzyl-7-butyl-1-azaspirocyclo[5.5]undec-7-ene (**14**) [a formal precursor of perhydrohistrionicotoxin (**2**)] and the thermodynamically preferred, exocyclic isomer (**13**) in six steps from the readily available *N*-benzyloxycarbonyl-10-amino-Δ<sup>1,9</sup>-octalin (**5**) are reported.

The alkaloids of neotropical poison frogs (Dendrobatidae) have provided a rich source of structural variation and biological activity.<sup>1-6,13</sup> In particular, histrionicotoxin (**1**) and a number of reduced derivatives, including the non naturally occurring perhydrohistrionicotoxin (**2**), have been the subject of numerous synthetic investigations.<sup>1,3,5,7-12</sup> Both (**1**) and (**2**) block the passage of potassium and sodium ions in a number of different systems. The enantiomer of (**2**) shows similar activity, and the desamyl-derivative (**3**) exhibits parallel but less potent properties.

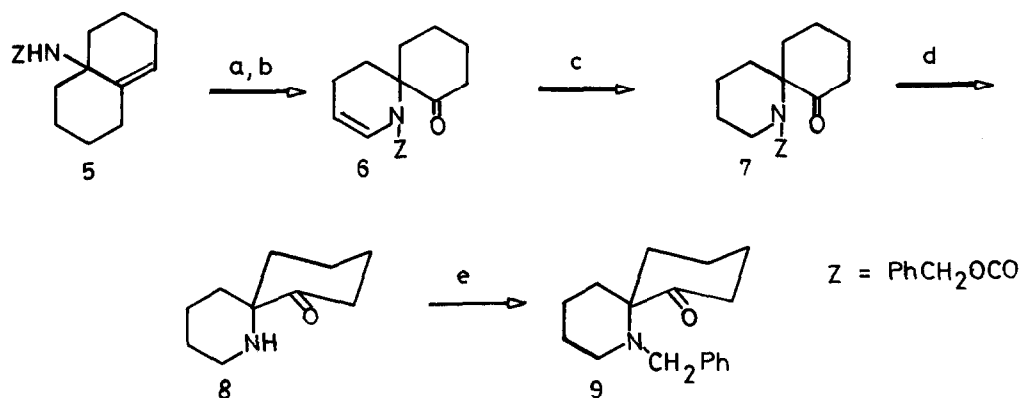
We now report a short synthesis of the azaspirocycle (**14**) which represents a formal synthesis of (**3**) and (**2**). The nitrosyl chloride adduct of octalin (**4**) serves as the starting material for the azaspirocycloundecane system. This is converted by a modification of known procedures into the urethane (**5**) (Scheme 1). Ozonolysis<sup>18</sup> of the urethane (**5**) gave under the work-up conditions the azaspirocyclic enamide (**6**) which upon catalytic reduction yielded the saturated derivative (**7**) (Scheme 2). Removal of the benzyloxycarbonyl group with iodotrimethylsilane gave the amino-ketone (**8**) which could be converted into the *N*-benzyl derivative (**9**) [54% from (**6**)]. The carbonyl groups of both compounds (**7**) and (**9**) were severely hindered to attack by most carbon nucleophiles, and were either rapidly enolised or reduced even in the presence of the recently described organotitanium<sup>19</sup> and organozirconium<sup>20</sup> reagents which have been reported suitable for nucleophilic attack on readily enolisable carbonyl groups. Eventually, a modest yield of the isomeric alcohols (**10**) was obtained by repetitive addition of *n*-butyl-lithium to the amino-ketone (**8**), followed by quenching with methanol.<sup>21</sup>





Reagents: (a) Li, EtNH<sub>2</sub>-Me<sub>2</sub>NH<sup>14</sup> (63%), (b) *iso*-amyl nitrite-HCl<sup>15</sup> (70%), (c) H<sub>2</sub>-Pt (98%)-see Ref 16, (d) NaOMe-MeOH, RT (90%), (e) Al-Hg, THF-H<sub>2</sub>O, reflux (100%)-see Ref 17, (f) PhCH<sub>2</sub>OCOC<sub>2</sub>H<sub>5</sub> (99%).

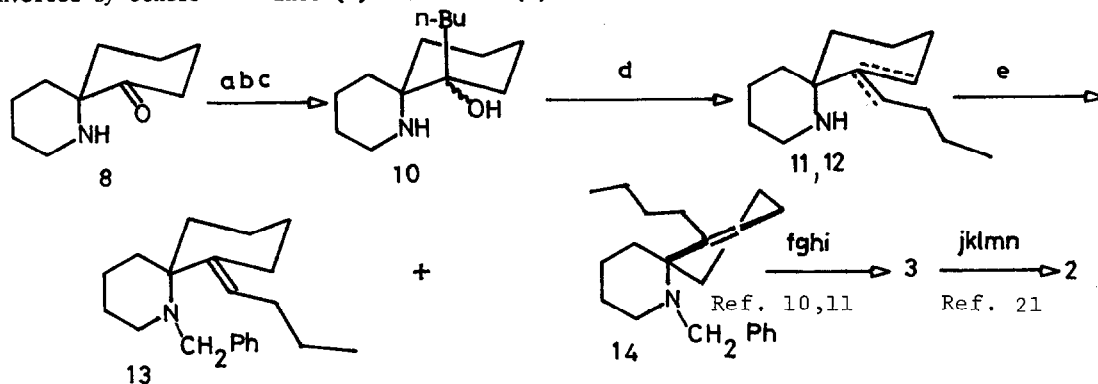
### SCHEME 1



Reagents: (a) O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C, (b) Me<sub>2</sub>S (43-72%), (c) H<sub>2</sub>-PtO<sub>2</sub>, EtOH (95%), (d) Me<sub>3</sub>SiI, CH<sub>3</sub>CN, 16h, RT (85%), (e) PhCH<sub>2</sub>Br, K<sub>2</sub>CO<sub>3</sub>, THF, 4-5d, RT (55%)

### SCHEME 2

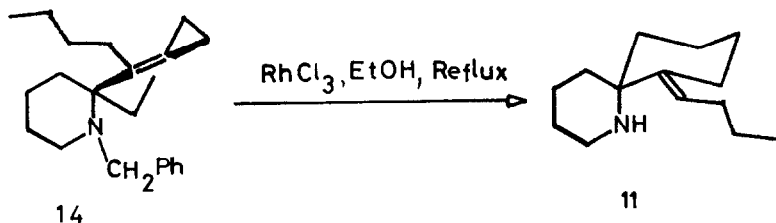
Dehydration of the alcohols (10) gave a 4:1 mixture of exocyclic (11) and endocyclic (12) azaspirocyclic alkenes, identified as the corresponding N-benzyl derivatives (13) and (14). These compounds could be obtained from (7) in a five step sequence in 26% yield. The endocyclic alkene (14) was identical in all respects to an authentic sample which has been converted by others<sup>10,11</sup> into (3) and thence (2).<sup>21,22</sup>



Reagents: (a)  $n\text{-BuLi}$ , (b)  $\text{MeOH}$ , (c) Repeat five times (25%), (d)  $\text{KHSO}_4$ ,  $170^\circ\text{C}$  (53%), (e)  $\text{PhCH}_2\text{Br}$ ,  $\text{KI}$ ,  $(i\text{-Pr})_2\text{NEt}$ ,  $\text{CH}_3\text{CN}$  (35%), (f)  $\text{BH}_3$ , (g)  $\text{Me}_3\text{N}^+\text{-O}^-$ , (h) Acetylation, (i)  $\text{H}_2$ ,  $\text{Pd-C}$ , (j)  $t\text{-BuSiMe}_2\text{Cl}$ , (k)  $\text{NBS}$ , (l)  $\text{KOt-Am}$ , (m)  $\text{C}_5\text{H}_{11}\text{Li}$ , (n)  $n\text{-Bu}_4\text{NF}$

The well known thermodynamic preference for endocyclic double bonds in six-membered rings<sup>21,23</sup> would indicate that the mixture of (13) and (14) could be equilibrated in favour of (14). However, all attempts at base-catalysed isomerisation were unsuccessful. Ultimately it was found that stoichiometric quantities of rhodium trichloride in refluxing ethanol converted the endocyclic N-benzyl alkene (14) to the exocyclic debenzylated<sup>24</sup> amino-alkene (11). We conclude that the usual thermodynamic preference for endocyclic double bonds is overcome in this special case by the  $\text{A}_{1,2}$ -strain<sup>25</sup> present in (14).

We thank the S.E.R.C. for studentships (to N.K.W. and K.R.) and Gallaher Limited (CASE) and Churchill College Cambridge for generous financial support (to K.R.).<sup>26</sup>



## REFERENCES

1. J.W. Daly in "Progress in the Chemistry of Natural Products," W. Herz, H. Grisebach, and G.W. Kirby, Eds. Springer Verlag, Wien, New York, Vol. 41, 1982, pp.206-340.
2. C.W. Myers and J.W. Daly, Scientific American, 1983, **February issue**, pp.96-105.
3. B. Witkop and E. Gössinger in "The Alkaloids - Chemistry and Pharmacology," A. Brossi, Ed., Academic Press, New York and London, 1983, Ch.5, pp. 139-251.
4. B. Witkop, Heterocycles, 1982, **17**, 431.
5. Y. Inubushi and T. Ibuka, Heterocycles, 1982, **17**, 507.
6. T. Tokuyama, J. Yamamoto, J.W. Daly, and R.J. Highet, Tetrahedron, 1983, **39**, 49.
7. G.E. Keck and J.B. Yates, J. Org. Chem., 1982, **47**, 3590.
8. T. Ibuka, H. Minakata, Y. Mitsui, K. Hayashi, T. Taga, and Y. Inubushi, Chem. Pharm. Bull. (Japan), 1982, **30**, 2840.
9. M. Glanzmann, C. Karalai, B. Ostersehl, U. Schön, C. Frese, and E. Winterfeldt, Tetrahedron, 1982, **38**, 2805.
10. (a) S.A. Godleski and D.J. Heacock, J. Org. Chem., 1982, **47**, 4820; S.A. Godleski, D.J. Heacock, J.D. Meinhardt, and S. van Wallendaal, ibid, 1983, **48**, 2101; (b) W. Carruthers and S.A. Cumming, J. Chem. Soc., Chem. Commun., 1983, 360;
11. A.J. Pearson and P. Ham, J. Chem. Soc., Perkin Trans. 1, 1983, 1421.
12. E.R. Koft and A.B. Smith, III, J. Org. Chem., 1984, **49**, 832; T. Ibuka, H. Minakata, M. Hashimoto, L.E. Overman, and R.L. Freerks, Heterocycles, 1984, **22**, 485.
13. K. Takahashi, B. Witkop, A. Brossi, M.A. Maleque, and E.X. Albuquerque, Helv. Chim. Acta, 1982, **65**, 252.
14. E.M. Kaiser and R.A. Benkeser, Org. Synth., 1970, **50**, 88.
15. M. Tuot, Compt. rend., 1937, **204**, 697.
16. J. Meinwald, Y.C. Meinwald, and T.N. Baker, J. Am. Chem. Soc., 1964, **86**, 4074.
17. E.J. Corey and M.J. Chaykovsky, J. Am. Chem. Soc., 1965, **87**, 1345.
18. The ozonolysis could be carried out in yields ranging from 43-72%. Best yields were obtained by adding relatively small quantities of (5) (71 mg) to a solution containing the theoretical quantity of O<sub>3</sub> in CH<sub>2</sub>Cl<sub>2</sub> at -78° C.
19. M.T. Reetz, R. Steinback, J. Wenderoth, and J. Westerman, Chem. and Ind. (London), 1981, 54; M.T. Reetz in "Topics in Current Chemistry," Springer Verlag, Berlin and New York, Vol. 106, 1982, pp.1-54.
20. B. Weidmann, C.D. Maycock, and D. Seebach, Helv. Chim. Acta, 1981, **64** 1552.
21. E.J. Corey, J.F. Arnett, and G.N. Widiger, J. Am. Chem. Soc., 1975, **97**, 430.
22. We thank Professor A.J. Pearson for supplying us with material for this comparison.
23. R.B. Turner and R.H. Garner, J. Am. Chem. Soc., 1958, **80**, 1424.
24. This method of deprotection of N-benzyl amines appears not to have been previously reported, although it has been observed for N-allyl amines (B. Moreau, S. Lavielle, and A. Marquet, Tetrahedron Lett., 1977, 2591). We are investigating the generality of the method.
25. F. Johnson, Chem. Rev., 1968, **68**, 375.
26. All new compounds exhibited spectroscopic and analytical data consistent with the proposed structure.

(Received in UK 12 June 1984)