CHLORONITROACETYL CHLORIDE IN THE SYNTHESIS OF 3-CHLOROISOXAZOLES

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The cycloaddition of chloronitrile oxide generated $in \ situ$ from dichloroformaldoxime to acetylenes leads to the formation of 3-chloroisoxazoles, which had been difficult to prepare [1].

We have shown that chloronitroacetyl chloride (I), which is readily obtained by the nitration of trichloroethylene using $HNO_3-H_2SO_4$ as a nitrating mixture [2], may be a source of chloronitrile oxide. Carrying out the reaction of (I) with a twofold excess of monosubstituted acetylenes (II) in methylene chloride in the presence of an equimolar amount of sodium bicarbonate gave the corresponding 5-substituted 3-chloroisoxazoles (III) in 30-45% yield.



R = Ph (a). CH_2Br (b), OEt (c).

A solution of 19.6 ml (0.02 mole) (I) in 40 ml CH_2Cl_2 was added dropwise with stirring over 4 h to a mixture of 0.04 mole (IIa)-(IIc) and 1.7 g (0.02 mole) NaHCO₃ in 10 ml CH_2Cl_2 . The reaction mixture was washed with water and dried over magnesium sulfate. The solvent was distilled off. The residue was distilled in vacuum. The yield of isoxazole (IIIa) was 1.1 g (30%), bp 100-102°C (1 mm). PMR spectrum in $CDCl_3$ (δ , ppm): 7.8-7.6 m (2H), 7.5-7.3 m (3H), 6.5 s (1H).

The yield of isoxazole (IIIb) was 1.4 g (35%), bp 90-91°C (10 mm). PMR spectrum in $CDCl_3$ (δ , ppm): 6.41 t (1H, J = 0.6 Hz), 4.46 d (2H, J = 0.6 Hz).

The yield of isoxazole (IIIc) was 1.3 g (45%), bp 60-61°C (1 mm). PMR spectrum in $CDCl_3$ (δ , ppm): 5.32 s (1H), 4.28 q (2H, J = 8 Hz), 1.48 t (3H, J = 8 Hz).

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