STEREOSPECIFIC SYNTHESIS AND STEREOCHEMICAL

STRUCTURE CONFIRMATION OF DUMETORINE¹.

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Abstract. A short stereospecific route to the alkaloid dumetorine <u>1</u> and its 5-epimer <u>2</u> is presented. This provides the first reported synthesis of this new alkaloid as well as stereochemical confirmation of its structure.

(+)Dumetorine, an alkaloid accompanying dioscorine, was recently isolated from tubers of Dioscorea dumetorum Pax, a plant used by natives in Africa for medicinal purposes². Based on extensive NMR studies, dumetorine was identified as <u>1</u> and the indicated syn (erythro) configuration was proposed on the basis of optical rotation additivities². We now confirm structure <u>1</u> for dumetorine by unequivocal stereospecific synthesis of (±)<u>1</u> and its epimer <u>2</u>.



Our interest in stereospecific 1,3-dipolar cycloadditions^{1,3} led us to consider a nitrone-olefin cycloaddition as an approach to the regiospecific and stereospecific introduction of the two stereochemical centers in 1. It had been shown⁴ that dipolar cycloaddition of cyclic nitrone 3 with α , β -unsaturated carbonyl compounds leads to a mixture of 4 and 5 (e.g. R = CO₂Me) in which the syn adduct 4 predominates. The latter could be a useful building block of dumetorine. Nevertheless, we decided in favor of a more convergent approach (see Scheme 1). This has the advantage of utilizing the completely stereospecific cycloaddition^{4,5} of non-conjugated terminal olefins to nitrone 3 leading to 5 with the opposite stereochemistry (anti) to that found in dumetorine, even though it requires an epimerization step at the CH-O center. At the same time 5-epidumetorine 2 became accessible for comparison.



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Reaction of allyl bromide with Zn in the presence of t-butyl acetoacetate <u>6</u> at room temperature provided in 70% yield the olefin <u>7</u>, ⁶ which underwent regio- and stereospecific cycloaddition to <u>3</u> to give exclusively the anti adduct <u>5a</u> as two diastereomers⁷ in 72% yield.



Methylation with methyl iodide followed by zinc, reduction in 50% aqueous acetic acid led to amino diol <u>8</u> in 78% as a mixture of diastereomers due to the tertiary hydroxyl center but containing carbons 1 and 5 in the stereospecific array indicated .² Attempts to remove the t-butyl protecting group under standard conditions (HCO₂H at room temperature) led to formation of oxetane <u>9</u> as a single isomer. Apparently <u>9</u> resulted from internal trapping of a tertiary carbocation by the secondary alcohol. Further heating of oxetane <u>9</u> in formic acid proceeded via hydrolysis of the t-butyl ester, ring opening with elimination and lactonization to afford 5-epidumetorine 2 in 65% yield.



(±)Dumetorine 1 was obtained from diol 8 by an inversion of C-5 according to the Mitsunobu procedure⁸. The resulting benzoate 10 (57%) which already contained the desired stereochemistry, was saponified (82%) and converted to dumetorine 1 in 50% yield by heating with formic acid. The product was identical by ¹H and ¹³C-NMR to natural dumetorine⁹. 5-Epidumetorine 2 exhibited almost identical spectra except for a few characteristic differences⁷. For instance, C-5 absorbs at 73.3 ppm in dumetorine 1 and at 75.96 in the 5-epimer 2. In the ¹H-NMR the C-5 proton appears at 4.53 as an 8 line signal in 1, whereas it is found at 4.50 as a 6-line multiplet in 2. Our short and stereospecific approach to 1 and 2 provides the first reported synthesis of dumetorine, as well as confirmation of its stereochemical assignment.



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REFERENCES

- Cycloadditions 33. For paper 32 see A. Hassner and K.K. Murthy, Tetrahedron Lett. 28, 98 (1987).
- D.G. Corley, M.S. Tempesta and M.M. Iwu, Tetrahedron Lett. <u>26</u>, 1615 (1985). The numbering in dumetorine is based on its relationship to dioscorine.
- 3. A. Hassner and K.K. Murthy, Tetrahedron Lett. 27, 1407 (1986).
- 4. J.J. Tufariello and S.A. Ali, Tetrahedron Lett. 4647 (1978).
- 5. E. Gössinger, Tetrahedron Lett. 2229 (1980).
- 6. G.C. Fischer, R.H. Turakhia and C.J. Morrow, J. Org. Chem. 50, 2011 (1985).
- Characteristic NMR data (in CDCl₃, ¹H-300 MHz, ¹³C-75 MHz): <u>5a</u> (1:1 mixture) ¹H-1.28 and 1.31 (2s,Me), 1.46 (brs, ^tBu) 3.43 (brd, J=10 Hz, C-6, 2H), 3.87 (m,C-1,1H), 4.28 (m,C-5,1H). ¹³C-23.84, 24.75, 27.07, 27.38, 28.10, 29.28, 41.38, 46.18, 46.41, 46.88, 47.30, 55.22, 66.18, 70.64, 70.79, 72.78, 73.05, 80.98, 81.12, 172.01. <u>8</u> (1:1 mixture) ¹H-1.29 (s,Me), 1.45 and 1.46 (2s,^tBu) 2.34 (s,Me), 2.90 (brd, J=12 Hz, C-1,1H), 4.48 (m,C-5,1H). ¹³C-24.27, 25.47, 28.08, 28.28, 29.56, 37.39, 46.57, 46.83, 57.09, 62.54, 67.08, 71.68, 80.39, 171.86. <u>10</u> ¹H (1:1 mixture) ¹H-1.28 and 1.29 (2s,Me), 2.61 (s,Me) 3.13 (brd,J=12 Hz, C-1;1H), 5.42 (m,C-5,1H), 7.45 and 8.05 (m,Ph). <u>2</u> ¹H-1.98 (s,C-13 Me), 2.28 (s,N-Me), 2.86 (t,d,J=4,12 Hz, C-1H) 4.50 (6 line multiplet, C-5H), 5.80 (brs,C-11,1H), 1.26-2.30 (m, remaining H's). ¹³C-22.96(q), 23.84(t), 24.92(t), 31.51(t) 35.27(t), 38.54(t), 42.58(q), 56.23(t), 60.08(d), 75.96(d), 116.53(d), 156.82(s), 165.14(s).
 Mitsunobu and M. Eguchi, Bull Chem. Soc. Jpn. <u>44</u>, 3427 (1971); 0. Mitsunobu
- and M. Yamada, Bull. Chem. Soc. Jpn. <u>40</u>, 2380 (1967).
- 9. We thank Dr. M.S. Tempesta for detailed spectra of dumetorine used in the comparison.

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