

Zinc cyanamide, $\text{Zn}(\text{CN}_2)$

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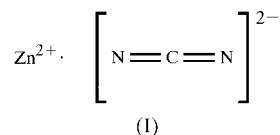
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Single crystals of the title compound have been grown by annealing microcrystalline zinc cyanamide at 843 K in silver crucibles. $\text{Zn}(\text{CN}_2)$ crystallizes as colourless prisms. The crystal structure is composed of corner-linked ZnN_4 tetrahedra. Carbon and nitrogen form $(\text{CN}_2)^{2-}$ dumb-bells with the C atom on a twofold axis. Nitrogen is approximately trigonally planar, coordinated by two Zn atoms and one C atom.

Comment

The structural characterization of M^{II} -cyanamides (M^{II} is a transition metal) has up until now been limited to cadmium cyanamide (Dvoinin *et al.*, 1982). Recently, mercury cyanamide has been prepared and its crystal structure determined (Becker & Jansen, 2000). In the course of further structural investigations of crystalline metal cyanamides, we have grown single crystals of $\text{Zn}(\text{CN}_2)$, which was first described by Grube (Grube & Nitsche, 1914). Nowadays, zinc cyanamide, (I), is a component in non-polluting anticorrosive white pigments (Nagayama *et al.*, 1994).



We present here the first structural investigation on $\text{Zn}(\text{CN}_2)$. In the title compound, zinc is tetrahedrally coordinated by N atoms, with bond lengths of 1.985 (2) and 2.035 (2) Å. Nitrogen is surrounded in an approximately trigonally planar manner by two Zn^{2+} cations and one C atom. The ZnN_4 tetrahedra share corners while the C atoms are each situated between two N atoms to form slightly bent $(\text{CN}_2)^{2-}$ dumb-bells [$\text{N}-\text{C}-\text{N}$ 176.3 (3)°] (Fig. 1). The cyanamide dumb-bells are surrounded by four Zn atoms, which are arranged in a staggered conformation (Fig. 2). Although the corner-linked ZnN_4 tetrahedra are similar to the silicon dioxide polymorphs, there is no topological relationship to any of them; the ZnN_4 tetrahedra form Zn_4N_4 rings which have no precedence in SiO_2 structures. The ZnN_4 tetrahedra form chains along the (111) direction with every second ZnN_4 unit being twisted counterclockwise (rotation angle 81.5°). With

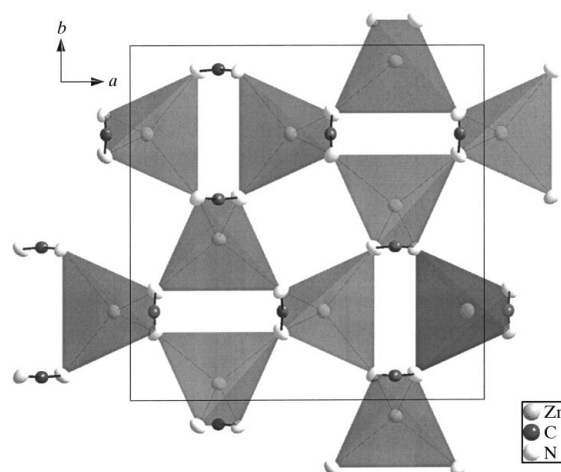


Figure 1

View of the crystal structure of $\text{Zn}(\text{CN}_2)$. The ZnN_4 building units have been shown as polyhedra.

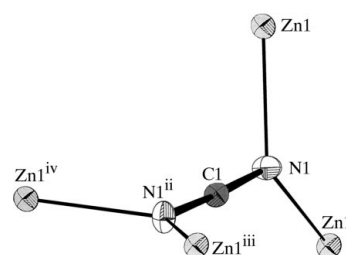


Figure 2

View of the staggered conformation of the anion of $\text{Zn}(\text{CN}_2)$ and the cations. Displacement ellipsoids are shown at the 50% probability level. [Symmetry codes: (i) $1 - y, x - 1, -z$; (ii) $\frac{3}{2} - x, y, \frac{3}{4} - z$; (iii) $1 - y, x - 1, 1 - z$; (iv) $\frac{3}{2} - x, \frac{1}{2} - y, \frac{1}{2} + z$.]

regard to zinc and nitrogen, the structure can be related to the high pressure polymorph of Willemite (Zn_2SiO_4 -II; Marumo & Syono, 1970), with nitrogen being substituted for oxygen. The sites of the tetrahedrally coordinated Si atoms are empty in the structure of $\text{Zn}(\text{CN}_2)$. The position of carbon in zinc cyanamide can be reproduced by shifting silicon in Zn_2SiO_4 -II towards the edges of its coordination polyhedron, thus forming dumb-bells.

Experimental

$\text{Zn}(\text{CN}_2)$ was precipitated from aqueous solutions of ZnSO_4 and $\text{Na}_2(\text{CN}_2)$ (in a 1:1 molar ratio). The precipitate was washed several times with distilled water and dried *in vacuo* at 10^{-3} mbar (1 bar = 10^5 Pa). The resulting powder was compacted. The pellet obtained was placed in a silver crucible, sealed in a glass tube under argon and heated to 843 K over a period of 70 h. It was held at this temperature for 150 h. After slow cooling to room temperature (6 K h^{-1}), crystals suitable for single-crystal investigation were obtained.

Crystal data

$\text{Zn}(\text{CN}_2)$
 $M_r = 105.40$
 Tetragonal, $I\bar{4}2d$
 $a = 8.8047$ (2) Å
 $c = 5.4329$ (2) Å
 $V = 421.17$ (2) Å³
 $Z = 8$
 $D_x = 3.324 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation
 Cell parameters from 557 reflections
 $\theta = 4.41\text{--}37.55^\circ$
 $\mu = 11.227 \text{ mm}^{-1}$
 $T = 293$ (2) K
 Prism, colourless
 $0.06 \times 0.04 \times 0.04 \text{ mm}$

Data collection

Bruker AXS SMART CCD
diffractometer
 ω scans
4078 measured reflections
557 independent reflections
509 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.075$
 $\theta_{\text{max}} = 37.51^\circ$
 $h = -15 \rightarrow 15$
 $k = -14 \rightarrow 15$
 $l = -9 \rightarrow 9$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.023$
 $wR(F^2) = 0.053$
 $S = 1.007$
557 reflections
20 parameters

$w = 1/[\sigma^2(F_o^2) + (0.0262P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.75 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.66 \text{ e } \text{\AA}^{-3}$
Absolute structure: Flack (1983)
Flack parameter = 0.09 (4)

Data collection: *SMART* (Bruker, 1997); cell refinement: *SMART*;
data reduction: *SMART*; program(s) used to solve structure:
SHELXS97 (Sheldrick, 1997); program(s) used to refine structure:

SHELXL97 (Sheldrick, 1997); molecular graphics: *DIAMOND*
(Bergerhoff, 1996).

Supplementary data for this paper are available from the IUCr electronic
archives (Reference: IZ1010). Services for accessing these data are
described at the back of the journal.

References

- Becker, M. & Jansen, M. (2000). *Z. Anorg. Allg. Chem.* **626**, 1639–1641.
Bergerhoff, G. (1996). *DIAMOND*. Gerhard-Domagk-Straße 1, 53121 Bonn,
Germany.
Bruker (1997). *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.
Dvonin, V. I., Skornyyakov, L. G., Yatlova, L. E., Degtyarev, M. V. & Kitaev,
G. A. (1982). *Zh. Prikl. Khim. (Leningrad)*, **55**, 213–215.
Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
Grube, H. H. & Nitsche, R. (1914). *Z. Anorg. Chem.* **27**, 374.
Marumo, F. & Syono, Y. (1970). *Acta Cryst.* **B27**, 1868.
Nagayama, H. (1994). Patent No. JP 08041374.
Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Gött-
ingen, Germany.