

OCTASUBSTITUTED FLAVONES FROM *AGERATUM HOUSTONIANUM**

LEOVIGILDO QUIJANO, JOSE S. CALDERON, FEDERICO GOMEZ G., EDGAR ESCOBAR and TIRSO RIOS

Instituto de Química, Universidad Nacional Autónoma de México (UNAM), Circuito Exterior, Ciudad Universitaria, Coyoacán 04510, Mexico

(Received 11 September 1984)

Key Word Index—*Ageratum houstonianum*, Asteraceae, Eupatorieae, octasubstituted flavones, 3'-hydroxy-5,6,7,8,2',4',5'-heptamethoxyflavone, 5,3'-dihydroxy-6,7,8,2',4',5'-hexamethoxyflavone

Abstract—Two new highly oxygenated flavones, were isolated from aerial parts of *Ageratum houstonianum*. Their structures were established as 3'-hydroxy-5,6,7,8,2',4',5'-heptamethoxyflavone and 5,3'-dihydroxy-6,7,8,2',4',5'-hexamethoxyflavone on the basis of spectral data and chemical degradation. The structure of the latter compound was confirmed by X-ray analysis.

INTRODUCTION

In a previous paper we described the isolation and structure elucidation of agehoustins A (1c) and B (2), two highly oxygenated flavones from the less polar extract of *Ageratum houstonianum* [1]. Now we have undertaken a study of the chloroform extract of *A. houstonianum* which resulted in isolation of the flavonoids previously isolated from the petrol extract [1] as well as two new members of the rare group of octasubstituted flavones with a 2',3',4',5'-oxygenation pattern, which we have named agehoustins C (1a) and D (1b).

RESULTS AND DISCUSSION

The chloroform extract of the aerial parts of *A. houstonianum* afforded a mixture of several flavones previously isolated from the petrol extract [1], as well as two new octasubstituted flavones, agehoustin C (1a) and D (1b).

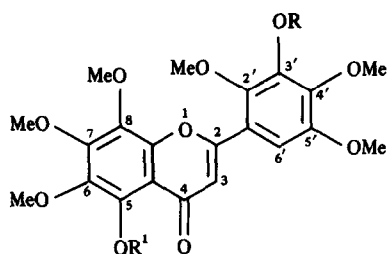
Agehoustin C (1a), $C_{22}H_{24}O_{10}$ ($[M]^+$ at m/z 448), mp 145–146°, was isolated as a pale yellow crystalline compound. Both the UV (268, 318 nm) and the IR (3240, 1635 cm^{-1}) absorptions were typical of flavones [2]. The 1H NMR spectrum (Table 1) was similar to that of agehoustin A (1c), except that one methoxyl signal was missing and instead it showed a non-chelated hydroxyl signal at δ 6.0. Methylation of 1a with diazomethane afforded agehoustin A (1c) indicating that 1a has the same substitution pattern as 1c. The mass spectral peaks at m/z 225 $[A_1 - Me]^+$ and 197 $[A_1 - Me - CO]^+$ indicated that the A-ring was fully methoxylated [1], therefore, the hydroxyl group must be at the B-ring. Consistent with this, alkaline hydrolysis of 1a afforded 2-hydroxy-3,4,5,6-tetramethoxyacetophenone (3a) and 2-hydroxy-3,4,5,6-tetramethoxybenzoic acid (3b) [1]. These fragments resulted from the A-ring. In addition, a monohydroxytri-

methoxybenzoic acid and a monohydroxytrimethoxyacetophenone were isolated. These fragments must have arisen from the B-ring and they are products expected from fission of the β -diketone, 4, which was also isolated, mainly as the keto-enol form, 5. The remaining problem was the relative position of the hydroxyl group in the B-ring. The 1H NMR data showed the absence of a weak hydrogen-bonded hydroxyl group at the 2'-position, since the hydroxyl proton signal in the 1H NMR spectrum of 1a appeared at δ 6.0. In contrast, a hydroxyl group at C-2' would give rise to a proton signal at ca δ 8.0 [3]. The 4'-position was ruled out, since the UV spectrum of 1a on addition of sodium methoxide, indicated the presence of a blocked hydroxyl at C-4' (bathochromic shift of band I of only 14 nm). These facts indicated that the lone hydroxyl was not on C-2' or C-4'. On the other hand, benzene-induced shift data suggested the presence of a methoxy group at C-5', since a large upfield shift was observed for one of the seven methoxyl signals [4] similar to that observed for the methoxyl group at C-5' in 1c and 2 (Table 1). Hence, the hydroxyl group must be attached at C-3' and 1a is 3'-hydroxy-5,6,7,8,2',4',5'-heptamethoxyflavone and, therefore, the fragments from the B-ring in the alkaline hydrolysis must be 3-hydroxy-2,4,5-trimethoxyacetophenone (6a) and 3-hydroxy-2,4,5-trimethoxybenzoic acid (6b).

Agehoustin D (1b), $C_{21}H_{22}O_{10}$ ($[M]^+$ at m/z 434), mp 168–169°, was a bright yellow crystalline dihydroxyhexamethoxyflavone. The 1H NMR data (Table 1), which was very similar to that of agehoustin C (1a), showed the following differences from 1a. One methoxy signal was missing and, instead, an extra hydroxyl proton signal appeared at δ 12.46, indicating that an additional hydroxyl group must be at C-5. Acetylation of 1b afforded the corresponding diacetate, 1e, which showed a C-5 acetyl signal at δ 2.46 [4]. Hence, 1b must be 5-O-demethylagehoustin C. Selective demethylation of 1a with aqueous hydrochloric acid [5], confirmed the above assumption and established the structural relationship between agehoustins C and D. Therefore, agehoustin D must be 3',5-dihydroxy-6,7,8,2',4',5'-hexamethoxyflavone (1b).

Alkaline hydrolysis of 1b furnished the same com-

* Part 4 in the series "Flavonoids from *Ageratum* Species". For Part 3 see ref. [1]. This is contribution No. 713 of the Instituto de Química, UNAM.



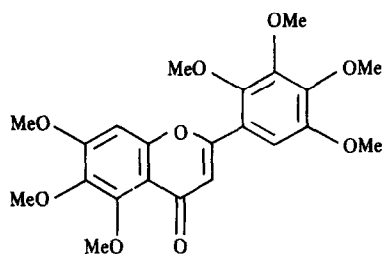
1a R = H, R¹ = Me

1b R = R¹ = H

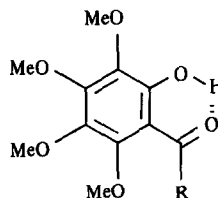
1c R = R¹ = Me

1d R = Ac, R¹ = Me

1e R = R¹ = Ac

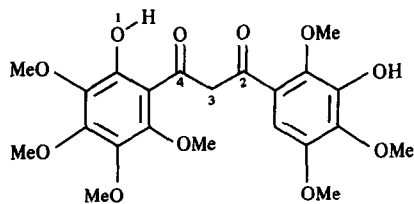


2

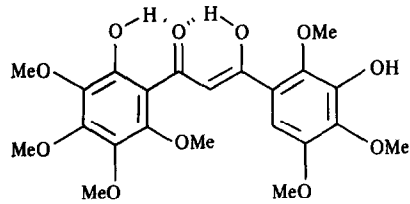


3a R = Me

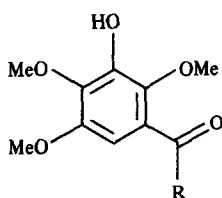
3b R = OH



4

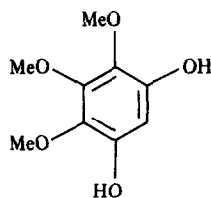


5



6a R = Me

6b R = OH



7

pounds, **6a** and **6b**, from the B-ring as given by **1a**. In addition, 4,5,6-trimethoxyresorcinol (**7**) was obtained. Compound **7** can be derived from the A-ring by decarboxylation of the corresponding acid.

Final confirmation of the structures of **1a** and **1b** was obtained by single crystal X-ray diffraction studies performed on **1b**. The molecular structure of **1b**, as determined by the X-ray data, is illustrated in Fig. 1.

Finally, our results of the alkaline hydrolysis require a comment. Kostanecký pointed out that four possible products are expected from fission of the intermediary β -diketone, which can seldom be isolated in the alkaline hydrolysis of flavones [6]. Nevertheless, the literature reports reveal that the carbonyl group adjacent to the B-

ring (C-2) is, in general, the one chiefly attacked, favouring formation of an *ortho*-hydroacetophenone from the A-ring and a benzoic acid from the B-ring. Now, in contrast with the above generality, we have observed that methoxy substitution at C-2' favours the isolation of the β -diketone (mainly as the keto-enol form) as well as its four possible further degradation products, the two different acetophenones and two different benzoic acids from both rings A and B [1, 7].

EXPERIMENTAL

Aerial parts of *Ageratum houstonianum* Mill. were collected ca 60 km south of Mexico City at the end of the road to Cuernavaca.

Table 1 ^1H NMR data of agehoustins A (1c), B (2), C (1a), D (1b) and their acetates 1d and 1e*

	1a		1b		1c	2	1d		1e	
	CDCl_3	C_6D_6	CDCl_3	C_6D_6	C_6D_6	C_6D_6	CDCl_3	C_6D_6	CDCl_3	C_6D_6
H-3	6.89†	7.20	6.95	7.08	7.13	7.00	6.86	7.07	6.84	6.98
H-8	—	—	—	—	—	6.40	—	—	—	—
H-6'	6.93†	6.84	6.95	6.81	7.06	6.88	7.27	7.07	7.26	6.98
OMe	3.86	3.38	3.85	3.38	3.40	3.17	3.77	3.32	3.75	3.33
	3.90	3.62	3.88	3.60	3.62	3.40	3.91	3.53	3.87	3.46
	3.94	3.65	3.93	3.63	3.65	3.60	3.91	3.66	3.91	3.58
	3.95	3.67	3.93	3.66	3.65	3.67	3.94	3.75	3.91	3.70
	3.98	3.71	3.98	3.82	3.71	3.75	3.94	3.77	3.99	3.75
	3.99	3.74	4.07	3.85	3.72	3.75	3.97	3.82	4.07	3.80
	4.08	4.00	—	—	3.74	4.07	4.07	4.05	—	—
					4.00					
OH	6.05	6.43	6.05	5.97	—	—	—	—	—	—
			12.46	13.24						
OAc	—	—	—	—	—	—	2.40	1.90	2.39	1.90
									2.46	2.37

Values are in δ -values all signals are singlets

*At 80 MHz with TMS as int. standard

†Values may be reversed

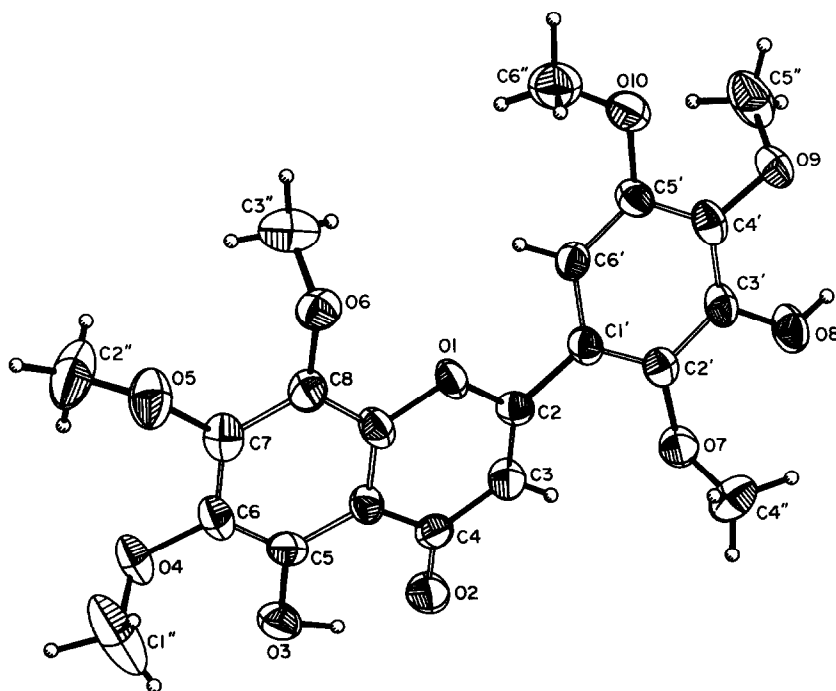


Fig. 1 Molecular structure of agehoustin D (1b)

(Edo Morelos, Mexico) The plant material was extracted with CHCl_3 as described previously [1, 7]. The crude syrup (80 g) obtained was pre-adsorbed on 120 g silica gel (Merck 35–70 mesh) and chromatographed on 500 g silica gel (Merck 70–230 mesh), using petrol and petrol–EtOAc as eluants.

Agehoustin C (1a) Fractions 166–180 eluted with petrol–EtOAc mixtures (7:3 and 3:2) were combined and purified by TLC (CHCl_3 – Me_2CO , 9:1, $\times 3$), yielding 115 mg 1a which was crystallized from CHCl_3 – Et_2O , mp 145°.

UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 268 (25 000), 318 (30 000). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} 3520, 1637, 1590, 1510, 1465. EIMS (probe) 70 eV m/z (rel. int.) 448 $[\text{M}]^+$ (23), 433 $[\text{M} - \text{Me}]^+$ (100), 418 $[\text{M} - \text{CH}_2\text{O}]^+$ (61), 403 $[\text{M} - \text{Me} - \text{CH}_2\text{O}]^+$ (20), 225 $[\text{A}_1 - \text{Me}]^+$ (95), 197 $[\text{A}_1 - \text{Me} - \text{CO}]^+$ (20), 193 $[\text{B}_1 - \text{Me}]^+$ (6).

Agehoustin D (1b) Purification of fractions 101–110 (eluted with petrol–EtOAc, 17:3) by TLC (Et_2O –petrol, 4:1, $\times 3$) afforded 90 mg 1b, which was crystallized from CHCl_3 – Et_2O , mp 168–169° UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 279 (23 600), 324 (19 500).

IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} 3520, 1655, 1610, 1590, 1576, 1465 EIMS (probe) 70 eV m/z (rel int) 434 $[\text{M}]^+$ (52), 419 $[\text{M} - \text{Me}]^+$ (100), 211 $[\text{A}_1 - \text{Me}]^+$ (13), 183 $[\text{A}_1 - \text{Me} - \text{CO}]^+$ (13), 193 $[\text{B}_1 - \text{Me}]^+$ (4)

Agehoustin C acetate (1d) Acetylation of 27 mg **1a**, with Ac_2O -pyridine, as usual, gave the acetate, **1d**, mp 109–111° (CHCl_3 - Et_2O) UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 270 (19 300), 324 (16 850) IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} 1770, 1640, 1582, 1498, 1460 EIMS (probe) 70 eV m/z (rel int) 490 $[\text{M}]^+$ (28), 475 $[\text{M} - \text{Me}]^+$ (100), 433 $[\text{M} - \text{Me} - \text{CH}_2\text{CO}]^+$ (23), 225 $[\text{A}_1 - \text{Me}]^+$ (14), 197 $[\text{A}_1 - \text{Me} - \text{CO}]^+$ (32), 43 $[\text{MeCO}]^+$ (45)

Methylation of agehoustin C (1a) Compound **1a** (23 mg) in dry Me_2CO (25 ml), dry K_2CO_3 (200 mg) and 0.5 ml MeI were refluxed for 4 hr and worked-up as usual. Crystallization from CHCl_3 - Et_2O gave a compound identical in all respects with agehoustin A (**1c**)

Alkaline degradation of agehoustin C (1a) A 55 mg sample of **1a** was refluxed with 50% KOH (15 ml) in EtOH (15 ml) under N_2 for 15 hr. The reaction mixture was worked-up as described before [1]. The neutral fraction after TLC separation (Et_2O -petrol, 7/3, twice) yielded three products. The less polar compound was the only 2-hydroxy-3,4,5,6-tetramethoxyacetophenone (**3a**), identified by NMR and comparison with an authentic sample [1].

The second neutral product crystallized from CHCl_3 -petrol, mp 98–100°. Its spectral data were in accord with 3-hydroxy-2,4,5-trimethoxyacetophenone (**6a**) UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 218 (24 500), 274 (11 300), 318 (4000) IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} 3250, 1655, 1594, 1495, 1466, 1425 EIMS (probe) 70 eV m/z (rel int) 226 $[\text{M}]^+$ (100), 211 $[\text{M} - \text{Me}]^+$ (78), 183 $[\text{M} - \text{MeCO}]^+$ (36), 193 $[\text{M} - \text{Me} - \text{H}_2\text{O}]^+$ (17), 179 $[\text{M} - \text{Me} - \text{MeOH}]^+$ (29), 196 $[\text{M} - 3\text{O}]^+$ (10), 43 $[\text{MeCO}]^+$ (17) ^1H NMR (80 MHz, CDCl_3) δ 2.64 (COMe), 3.86, 3.90, 3.97 (OMe), 5.84 (OH), 6.84 (H-6)

The third neutral product was an oil identified as the tautomeric mixture of **4** and **5** UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 204 (38 000), 285 (9750), 365 (12 850) IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} 3380, 1585, 1560, 1495, 1402 EIMS (probe) 70 eV m/z (rel int) 466 $[\text{M}]^+$ (10), 435 $[\text{M} - \text{OMe}]^+$ (17), 241 $[\text{C}_{11}\text{H}_{13}\text{O}_6]^+$ (12), 240 $[\text{C}_{12}\text{H}_{15}\text{O}_6 - \text{Me}]^+$ (27), 225 $[\text{C}_{11}\text{H}_{13}\text{O}_5]^+$ (18), 211 $[\text{C}_{10}\text{H}_{11}\text{O}_5]^+$ (100) ^1H NMR (80 MHz, CDCl_3) δ 3.80–4.05 (OMe), 5.85 (OH-3'), 7.0 (H-3), 7.65 (H-6'), 12.4 ($\phi_{\text{A}} - \text{OH}$), 15.74 (OH-2). The following signals at δ 4.65 (OC- CH_2 -CO), 7.01 (H-6'), 12.90 ($\phi_{\text{A}} - \text{OH}$), indicated the presence of the β -diketone, **4**.

From the acidic fraction after TLC separation (Et_2O -petrol, 17/3, \times 3), two benzoic acids were isolated. The less polar one was crystallized from CHCl_3 -petrol, mp 125° and identified as 3-hydroxy-2,4,5-trimethoxybenzoic acid (**6b**) UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 208 (34 200), 256 (8250), 300 (3000) IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} 3240, 1730, 1605, 1585, 1492, 1431 EIMS (probe) 70 eV m/z (rel int) 228 $[\text{M}]^+$ (86), 210 $[\text{M} - \text{H}_2\text{O}]^+$ (22), 195 (65), 167 (100), 152 (24), 181 (14) ^1H NMR (80 MHz, CDCl_3) δ 3.88, 4.01, 4.08 (OMe), 5.9 (br, COOH), 7.24 (H-6), (C_6D_6) δ 3.16, 3.43, 3.44 (OMe), 7.35 (H-

6). The more polar acidic product was identified as 2-hydroxy-3,4,5,6-tetramethoxybenzoic acid (**3b**) by ^1H NMR and comparison with an authentic sample [1].

Agehoustin D diacetate (1e) Acetylation of **1b** (29 mg) with Ac_2O and a drop of HClO_4 , followed by usual work-up and TLC purification (CHCl_3 - Me_2CO , 9/1), gave the diacetate **1e**, mp 140–142° UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 265 (38 700), 322 (38 300) IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} 1775, 1645, 1607, 1570, 1503, 1465 EIMS (probe) 70 eV m/z (rel int) 518 $[\text{M}]^+$ (8), 476 $[\text{M} - \text{CH}_2\text{CO}]^+$ (100), 461 $[\text{M} - \text{Me} - \text{CH}_2\text{CO}]^+$ (84), 419 $[\text{M} - \text{Me} - 2\text{CH}_2\text{CO}]^+$ (41), 211 $[\text{A}_1 - \text{Me}]^+$ (16), 183 $[\text{A}_1 - \text{Me} - \text{CO}]^+$ (19), 193 $[\text{B}_1 - \text{Me}]^+$ (7)

Alkaline degradation of agehoustin D (1b) A 50 mg sample of **1b**, treated under the same degradative conditions as described above, afforded the same acetophenone (**6a**) and benzoic acid (**6b**) as obtained from **1a**. In addition, 4,5,6-trimethoxyresorcinol (**7**) was also isolated UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 202 (31 100), 233 (3180) IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} 3400, 1606, 1485, 1425 EIMS (probe) 70 eV m/z (rel int) 200 $[\text{M}]^+$ (63), 185 $[\text{M} - \text{Me}]^+$ (100), 157 $[\text{M} - 43]^+$ (20), 142 $[\text{M} - 68]^+$ (37) ^1H NMR (80 MHz, CDCl_3) δ 3.85 (6H), 3.93 (3H) (OMe), 5.49 (2H) (OH), 6.32 (H-2)

Selective demethylation of agehoustin C (1a) A mixture of **1a** (5 mg), conc HCl (0.5 ml) and conc HOAc (0.5 ml) was heated on a steam bath for 12 hr, diluted with H_2O and extracted with EtOAc. The EtOAc soln was washed with H_2O , dried and evaporated. The residue was purified by TLC (CHCl_3 - Me_2CO , 9/1, twice) giving a crystalline compound identified as agehoustin D (**1b**) by ^1H NMR and comparison with an authentic sample.

Acknowledgements—We thank Mrs T German and Mr F Ramos, Herbarium of the Instituto de Biología, (UNAM), for identification of the plant material, Messrs R Saucedo,* J Cárdenas, H Bojórquez, L Velasco, A Toscano and R Villena for ^1H NMR, IR, UV and mass spectra. Grateful acknowledgement is due to Mr Alfredo Toscano for assistance in performing the X-ray crystallographic analysis.

REFERENCES

- Quijano, L., Calderón, J. S., Gómez G., F. and Ríos, T. (1982) *Phytochemistry* **21**, 2965.
- Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) *The Systematic Identification of Flavonoids*, p. 41. Springer, New York.
- Braz, F. R. and Gottlieb, O. R. (1971) *Phytochemistry* **10**, 2433.
- Harborne, J. B., Mabry, T. J. and Mabry, H. (1975) *The Flavonoids*, p. 45. Chapman & Hall, London.
- Sondheimer, F. and Meisels, A. (1960) *Tetrahedron* **9**, 143.
- Dean, F. M. (1963) *Naturally Occurring Oxygen Ring Compounds*, p. 280. Butterworths, London.
- Quijano, L., Calderón, J. S., Gómez G., F., Soria, I. E. and Ríos, T. (1980) *Phytochemistry* **19**, 2439.