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Synthetic Applications of 2-(1,3-Dithian-2-yl)indoles V.¹ Asymmetric Synthesis of Dasycarpidone-Type Indole Alkaloids

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Abstract- Chiral tetracyclic dasycarpidone-type compounds (αR , 1*S*, 5*R*)-4 and (αR , 1*R*, 5*S*)-16 have been synthesized from 2-(1,3-dithian-2-yl)indole 6 and (αR)-*N*-(β -hydroxy- α -phenylethyl)- Δ^3 -piperidein-2-one 8 in three steps, following the methodology that we described recently.

Continuing our studies on the synthesis of indole alkaloids by the use of 2-(1,3-dithian-2-yl)indoles (6) as umpoled synthons, we described recently the synthesis of dasycarpidone type compound 1 (Scheme 1).^{1,2} Thus, the conjugate addition of 11, the dianion of 6, on Δ^3 -piperidein-2-one 7, followed by reductive cyanation of the resulting adduct and aqueous AcOH cyclization was shown to be a straightforward method for the preparation of the tetracyclic ABED systems, key intermediates in the synthesis of *Strychnos* alkaloids.³⁻⁶ We describe now the application of the methodology to the preparation of chiral dasycarpidone system 4 and its diastereomer 16 by the use of Δ^3 -piperidein-2-one 8, an asymmetric equivalent of 7.



Scheme 1

The synthesis of compound 8 has been developed in the context of studies on the reactivity of its precursor, lactam 9.7 Phenylselenylation of lactam 9 followed by MCPBA oxidation of the selenyl derivative 10, using slight

modifications of the method applied for the preparation of 7,1 yielded 8 (Scheme 2), which was identified from its spectral data.⁸



Reagents and conditions; i) 1. sec-BuLi (2.5 equivalents), THF, -78°C, 20 min; 2. C₆H₅SeBr (1.1 equivalents), -78°C, 1 h (64%, **10a:10b** = 7:3); ii) MCPBA (1.1 equivalents), CH₂Cl₂, 0°C to room temperature, 30 min (95%).

Scheme 2

Condensation of 2 equivalents of dithianylindole dianion 11 with 1 equivalent of piperideinone 8 in THF at -78°C yielded an equimolecular mixture of diastereomeric lactams 12a and 12b, epimers on C-4 (88%) which were easily separated by flash column chromatography (Scheme 3). As expected, both diastereomers showed very similar NMR data. The main spectral differences observed between lactams 12 were the ¹³C NMR signals corresponding to C-6 and C- α , which were 1.3 ppm shielded in 12b.⁹ The partial reduction of the carbonyl group was carried out with an excess of Red-Al[®] in THF.¹⁰



<u>Reagents and conditions</u>: i) *n*-BuLi (2.1 equivalents), THF, -78°C; ii) **8** (0.5 equivalents), -78°C, 5 h (88% based on **8**); iii) RedAl[®] (excess), THF, -78°C to r.t., 8 h (45% on transformed product); iv) 50% aq. AcOH, CH₂Cl₂, r.t., 8 h (55%); v) Red-Al[®] (excess), THF, -45°C, 30 min, 2 h (56%); vi) 50% aq. AcOH, CH₂Cl₂, 2 h (84%).

In the case of **12a** the formation of aminal **14** was observed together with the expected oxazolidines **13**, obtained as a mixture of epimers on C-2. The reduction of compound **12b** occurred much faster to give exclusively the corresponding oxazolidines **15**. The most characteristic NMR data for all oxazolidines were the signals corresponding to 2-position: δ_H -3.7, δ_C -90. Tetracyclic compound **4**¹¹ was obtained by treating the mixture of oxazolidines **13** and aminal **14** with 50% aqueous AcOH at room temperature.¹² Compound **16**¹³ was similarly prepared from oxazolidines **15**, and once again the reaction proved to be much easier when the starting compounds were (αR , 4S). Such different behavior, and the splitting of the aromatic signals in the ¹H NMR spectrum of **4**, can be attributed to the steric hindrance between the α -phenyl ring and the piperidine C-2 substituent in the (αR , 4R) series.



Scheme 4. Double arrows indicate NOESY correlations.

In addition, the transformation of racemic tetracyclic compound 17¹ to compounds 2 and 3⁸ has been carried out by Raney-Nickel reduction of the dithiane ring and by (CF₃COO)₂IPh deprotection of the C-6 carbonyl group respectively, followed by *O*-debenzylation. Since the closure of ring C from compounds type 2 has already been described,¹⁴ its preparation constitutes a new formal synthesis of the *Strychnos* pentacyclic framework. By extrapolation to the chiral series this work opens a new perspective on the preparation of *Strychnos* alkaloids.



<u>Reagents and conditions:</u> i) W-2 Raney-Ni (excess), EtOH, reflux, 4 h (70%); ii) Me₂S (excess), BF₃.Et₂O (10.6 equivalents), CH₂Cl₂ (~90%); iii) (CF₃COO)₂IC₆H₅ (1.4 equivalents), CH₃CN, r.t., 45 min (83%).

Scheme 5

By applying our methodology, we have achieved the synthesis of chiral dasycarpidone type tetracyclic systems 4 and 16 in three steps from indolyldithiane 6 and asymmetric Δ^3 -piperidein-2-one 8. Further work will aim at the chiral synthesis of *Strychnos* pentacyclic framework.

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References and notes

1H, In-NH).

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- All compounds have been fully characterized by their spectral data and elemental analysis or HRMS. 8.
- The absolute configuration of C-4 in compounds 12a and 12b was determined on the basis of the 9 stereochemistry of the corresponding cyclization products 16 and 4.
- Reaction with Red-Al[®] avoided the further reduction to the corresponding piperidine, observed when using 10. LiAIH₄.
- 11. Compound 4: [a]D = -17° (c = 1.1, EtOH); ¹H NMR (300 MHz, CDCl₃) 1.90-2.45 (m, 7H), 2.70-2.95 (m, 4H), 3.05 and 3.20 (2 td, J = 12 and 2 Hz, 2H, SCHa), 3.45 (dd, J = 9 and 2 Hz, 1H, CHAOH), 3.62 (dd, J = 4 and 2 Hz, 1H, CH α), 3.80 (dd, J = 9 and 4Hz, 1H, CH_BOH), 4.15 (br s, 1H, 1-H), 7.05 (t, J = 7 Hz, 1H, 10-H), 7.15 (t, J = 7 Hz, 1H, 9-H), 7.27-7.55 (m, 3H, Ph-p, 8-H and 11-H), 7.49 (t, J = 7 Hz, 2H, Ph-m), 7,64 (d, J = 7 Hz, 2H, Ph-o), 8.60 (br s, 1H, NH).

12. Treatment of oxazolidines 13 in refluxing aqueous AcOH yielded compounds 18, as the result of a fragmentation of intermediate 4. 18 (Diastereomeric mixture): ¹H NMR (300 MHz, CDCl₃) 2.10 (br t, J = 7 Hz, 2H, CH₂), 2.80-2.95 (m, 7H, SCH₂CH₂, and CH), 3.25 (t, J = 7 Hz, 2H, NCH₂), 3.50 (dd, J = 9 and 7 Hz, 1H, CHAOH), 3.69 (dd, J = 9 and 4 Hz, 1H, CHBOH), 3.72 (dd, J = 7 and 4 Hz, 1H, CH α), 7.11 (d, J = 7 Hz, 1H, CH=), 7.15-7.60 (m,

8H, Ar-H), 7.95 (d, J = 7 Hz, 1H, CH=), 8.03 (d, J = 7 Hz, 1H, 5-H), 8.60 (br s,



- 13. Compound 16: [a]D = -37.5° (c = 1.23, EtOH); ¹H NMR (500 MHz, CDCl₃) 1.87-1.92 (m, 1H, 4-Ha), 1.95 (qt, J = 14 and 2 Hz, 1H, SCH₂CH_a), 2.04 (td, J = 12 and 2.5 Hz, 1H, 3-H_a), 2.11 (dm, J = 14 Hz, 1H, SCH₂CH_e), 2.14 (dm, J = 12 Hz, 1H, 12-He), 2.22 (dm, J = 11 Hz, 1H, 4-He), 2.44 (ddd, J = 12, 6 and 3 Hz, 1H, 12-Ha), 2.68 (dt, J = 14 and 2 Hz, 1H, SCH_e), 2.71 (dt, J = 14 and 2 Hz, SCH_e), 2.72-2.76 (m, 1H, 5-He), 2.92 (ddd, J = 14, 12 and 2 Hz, 1H, SCH_a), 3.13 (td, J = 14, 12 and 2 Hz, 1H, SCH_a'), 3.70 (dd, J = 4 and 2 Hz, 1H, CHa), 4.10 (dd, J = 10 and 2 Hz, 1H, CHAOH), 4.41 (dd, J = 10 and 4 Hz, 1H, CHBOH), 4.50 (br s, 1H, 1-H), 7.12 (t, J = 7 Hz, 1H, 10-H), 7.14-7.22 (m, 6H, Ph-H and 9-H), 7.36 (d, J = 7 Hz, 1H, 8-H), 7.59 (d, J = 7 Hz, 1H, 11-H), 8.73 (br s, 1H, NH); ¹³C NMR 24.5 (SCH₂CH₂), 26.6 (C-4), 28.9 and 29.1 (SCH₂), 31.4 (C-12), 33.3 (C-5), 42.1 (C-3), 51.9 (C-1), 53.4 (SCS), 63.4 (C-β), 66.5 (C-α), 111.3 (C-8), 119.2 (C-11), 120.4 (C-10), 122.6 (C-9), 128.3 (Ph-o and Ph-m), 128.7 (Ph-p).
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