lo[4,5-b]pyridines (2). In the latter, the isoxazole ring is not aromatic but has an o-quinonoid structure and hence it might be isomerizing to the more stable structure 3. The 1,2-dihy-dropyridine structure assigned to the product 3 is supported by I.R. (3400-3300 cm<sup>-1</sup>) and <sup>1</sup>H-N.M.R. (a broad one-proton peak in the downfield area which disappears on shaking with deuterium oxide) spectra which point to the presence of NH. A 1,4-dihydropyridine structure for the product is ruled out as the <sup>1</sup>H-N.M.R. spectrum does not show any methylene protons.



## Chemistry of Heterocycles; 7. Synthesis of Isoxazolopyridines by $6\pi$ -Electrocyclization<sup>1</sup>

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Electrocyclic ring-closures of aza-1,3,5-hexatriene to aza-1,3cyclohexadiene which amount to isomerization, have not been reported sofar. The only known related reactions are the electrocyclic-ring closures accompanied by eliminations, which have been exhaustively reviewed<sup>2</sup>, and transformations of these trienes into five-membered rings<sup>3</sup>. In this communication, we report a thermal electrocyclic reaction of 3-methyl-4benzylideneamino-5-styrylisoxazoles 1, strictly an isomerization, leading to isoxazolopyridines 3. This is a convenient, one-step, general synthesis of compounds 3, which hitherto required a multistep sequence<sup>4-7</sup>.

Diphenyl ether solutions of 3-methyl-4-benzylideneamino-5styrylisoxazoles 1<sup>8</sup> were heated under reflux for 3 h. Removal of the solvent followed by crystallization furnished T.L.C.pure products which have been characterized as 3-methyl-5,6diaryl-4,5-dihydroisoxazolo[4,5-b]pyridines (3, Table) rather than the expected 3-methyl-5,6-diaryl-5,6-dihydroisoxazoWith a view to extend the reaction to aza-1,3,5,7-octatetraenes, the vinylogue of 1, namely, 3-methyl-4-cinnamylideneamino-5-styrylisoxazole (4) was subjected to thermal cyclization in diphenyl ether to get the azocine 5. Instead of the expected 5, the product was again the pyridine 6. I.R. and <sup>1</sup>H-N.M.R. spectral data show the presence of the NH group. The <sup>1</sup>H-N.M.R. spectrum displays a doublet of doublets at  $\delta = 7.1$  ppm (J = 15 Hz) which is diagnostic of a *trans*-1,2-disubstituted ethylene, thus strongly supporting the structure 6. The reason as to why 4 is behaving like a hexatriene rather than a octatetraene is not clear, but cyclooctatrienes are obtained<sup>9</sup> only by the electrocyclic ring-closure of 2,4,6,8-decatetraenes.

## 3-Methyl-5,6-diaryl-4,5-dihydroisoxazolo[4,5-b]pyridines 3; General Procedure:

A solution of 3-methyl-4-benzylideneamino-5-styrylisoxazole 1 (5 mmol) in diphenyl ether (15 ml) is refluxed for 3 h. The dark brown substance obtained after the removal of solvent by steam distillation is triturated with benzene (20 ml). Crystallization from a large excess of benzene gives isoxazolopyridines 3.

## 3-Methyl-5-styryl-6-phenyl-4,5-dihydroisoxazolo[4,5-b]pyridines (6); General Procedure:

3-Methyl-4-cinnamylideneamino-5-styrylisoxazole 4 (4 mmol) is refluxed in diphenyl ether (15 ml) for 3 h, cooled, and the solid formed is filtered and purified by crystallization.

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Table. 4,5-Dihydro-isoxazolo[4,5-b]pyridines 3 and 6 prepared

Produ No.		Ar <sup>2</sup>	Yield [%]	m.p. [°C] (solvent)	Molecular formula <sup>a</sup>	I.R. (KBr) v [cm <sup>-1</sup> ]	<sup>1</sup> H-N.M.R. (CDCl <sub>3</sub> ) $\delta$ [ppm]	M.S. <i>m/e</i> (rel. int. %)
3a	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	25	212° (C <sub>6</sub> H <sub>6</sub> )	C <sub>19</sub> H <sub>16</sub> N <sub>2</sub> O (288.4)	33503400 (NH)	2.6 (s, 3 H); 6.9-8.0 (m, 12 H); 12.0 (br. s, 1 H)	288 (M <sup>+</sup> , 100); 211 (30), 185 (20), 104 (22)
3b	C <sub>6</sub> H <sub>5</sub>	$4-H_3C-C_6H_4$	60	220° (C <sub>6</sub> H <sub>6</sub> )	C <sub>20</sub> H <sub>18</sub> N <sub>2</sub> O (302.4)	3250-3280 (NH)	b	b
3c	C <sub>6</sub> H <sub>5</sub>	4-ClC <sub>6</sub> H <sub>4</sub>	45	257° (CH <sub>3</sub> OH/H <sub>2</sub> O)	C <sub>19</sub> H <sub>15</sub> ClN <sub>2</sub> O (322.8)	3300-3360 (NH)	b	b
3đ	4-H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	35	215° (C <sub>6</sub> H <sub>6</sub> )	C <sub>20</sub> H <sub>18</sub> N <sub>2</sub> O (302.4)	3270-3300 (NH)	2.4 (s, 3H); 2.6 (s, 3H); 7.3-8.2 (m, 11H); 11.5 (br. s, 1H)	b
6a	C <sub>6</sub> H <sub>5</sub>		20	194-196° (C <sub>6</sub> H <sub>6</sub> /PE)	C <sub>21</sub> H <sub>18</sub> N <sub>2</sub> O (314.4)	3300-3370 (NH) 970 ( <i>trans</i> - CH≕CH)	2.5 (s, 3 H); 7.1 (2 d, 2 H, $J = 15$ Hz); 7.0-8.1 (m, 12 H); 9.5 (br. s, 1 H)	314 (M <sup>+</sup> , 100); 285 (24), 209 (40), 103 (17)
6b	4-H <sub>3</sub> CC <sub>6</sub> H <sub>4</sub>		25	226° (C <sub>6</sub> H <sub>6</sub> )	C <sub>22</sub> H <sub>20</sub> N <sub>2</sub> O (328.4)	3320-3380 (NH)	b	b

<sup>a</sup> Satisfactory microanalyses obtained: C  $\pm$  0.50, H  $\pm$  0.20, N  $\pm$  0.06.

<sup>b</sup> Not recorded.

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