

SIMPLE SYNTHESIS OF SUBSTITUTED 3-OXABICYCLO[3.1.0]HEXAN-2-ONES AND THEIR REARRANGEMENT IN THE PRESENCE OF LITHIUM IODIDE

A. N. Kasatkin, A. N. Kulak, R. Kh. Biktimirov,
G. A. Tolstikov, O. V. Shitikova, and V. R. Sultanmuratova

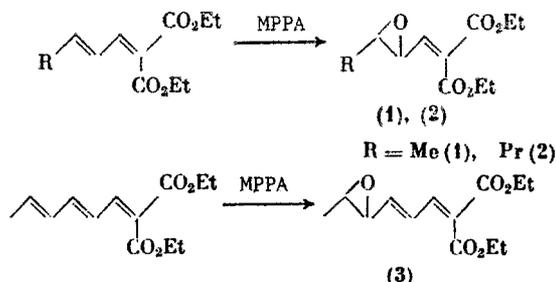
UDC 547.512:547.514.71:547.473.2

2-Alkenylidenemalonic ester monoepoxides react with organolithium and organomagnesium compounds to give 1-ethoxycarbonyl-4-alkyl-6-alkenyl(alkyl, aryl, alkynyl)-3-oxabicyclo[3.1.0]hexan-2-ones in high yields. The latter, when an alkenyl fragment is present, undergo rearrangement in the presence of LiI to substituted 1-ethoxycarbonyl-3-oxabicyclo[3.3.0]hex-6-ene-2-ones or 3-(1,3-butadienyl)-4-butanolides.

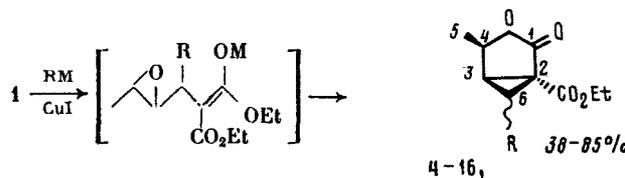
Keywords: 2-alkenylidenemalonic ester monoepoxides, 1-ethoxycarbonyl-4-alkyl-6-alkenyl-3-oxabicyclo[3.1.0]hexan-2-ones, 1-ethoxycarbonyl-3-oxabicyclo[3.3.0]hex-6-ene-2-ones, 3-(1,3-butadienyl)-4-butanolides, organolithium compounds, organomagnesium compounds, lithium iodide, synthesis, rearrangement.

The conjugate addition of nucleophilic reagents to γ -halogen-containing α,β -unsaturated carbonyl substrates and subsequent cyclization of the resulting enolates is an effective method for the synthesis of various cyclopropanes containing functional groups. Halo derivatives of alkyl crotonates [1-3], diethyl alkylidenemalonates [4-6], and unsaturated acylphosphoranes [7] can be used as substrates in reactions of this type. In the present research we studied the possibility of the utilization of 2-alkenylidenemalonic ester monoepoxides for the synthesis of cyclopropanes.

Oxiranes **1-3** can be readily obtained by treatment of diethyl *trans*-2-butenylidenemalonate, diethyl *trans*-2-hexenylidenemalonate, and diethyl *trans,trans*-2,4-hexadienylidenemalonate with monopero-phthalic acid (MPPA) in Et₂O at 20°C [8].



We have found that the reaction of epoxide **1** with various organolithium and organomagnesium reagents (Et₂O, -78°C → 20°C, 3 h) in the presence of 5 mole % CuI leads to lactones **4-16** containing a cyclopropane fragment (Table 1).



R=Me (**4**), Pr (**5**), CH₂=CH (**6**), 1,3-Butadien-2-yl (**7**), CH₂=C(Bu) (**8**),
Z-*i*-BuCH=CH (**9**), E-BuCH=CH (**10**), Pr₂C=CH (**11**), 1-Cyclohexenyl
(**12**), E-PhCH=CH (**13**), Ph (**14**), PrC≡C (**15**), PhC≡C (**16**);
M=Li, MgCl, MgBr, MgI.

Institute of Chemistry, Bashkir Science Center, Ural Branch, Russian Academy of Sciences, Ufa 450054. Translated from *Izvestiya Akademii Nauk, Seriya Khimicheskaya*, No. 4, pp. 946-954, April, 1992. Original article submitted February 27, 1991.

TABLE 1. Reactions of Epoxide **1** with Organolithium and Organomagnesium Compounds (RM) (Et₂O, -78°C → 20°C, 3 h, [RM]₀:**1**₀ = 1.05:1)

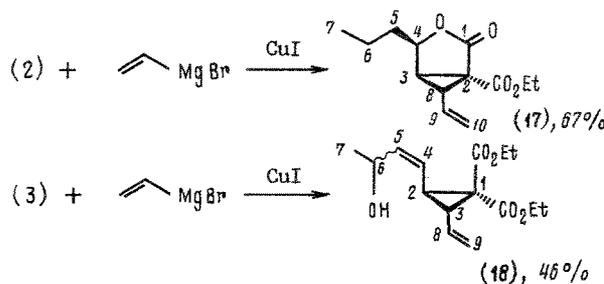
RM	Reaction product	Yield,* %	trans:cis †
MeMgI	4	65	>95 : 5
PrMgBr	5	85	60 : 40
CH ₂ =CHMgBr	6	70	76 : 24
CH ₂ =CHC(MgCl)=CH ₂	7	72	>95 : 5
CH ₂ =C(Bu)Li	8	54	>95 : 5
Z- <i>i</i> -BuCH=CHLi	9	75	>95 : 5
E-BuCH=CHLi	10	84	>95 : 5
Pr ₂ C=CHLi	11	71	>95 : 5
(1-Cyclohexenyl)Li	12	66	92 : 8
E-PhCH=CHMgBr	13	64	90 : 10
PhMgBr	14	69	>95 : 5
PrC≡CLi	15	38	75 : 25
PhC≡CLi	16	66	82 : 18

*The yields of the isolated products are presented.

†According to NMR spectroscopic data.

According to the PMR spectroscopic data, substituent R in the lactone ring of products **4-16** has primarily a *trans* orientation [9]. The constants of spin-spin coupling of the H³ and H⁶ protons in isomerically pure cyclopropanes **4**, **7-11**, and **14** range from 4.5 to 5.6 Hz; in the case of **6** and **16** J_{H³-H⁶} = 4.5 and 8 Hz, respectively, for the major and minor components. Let us note that in the spectrum of product **14** the signals of the protons of the CO₂Et group [δ (ppm) 3.93 m and 0.8 t] are located at anomalously strong field, evidently as a consequence of the shielding effect of the phenyl substituent. It follows from the data in Table 1 that the maximum stereoselectivity of the formation of the *trans*-cyclopropanes is observed when alkenyl derivatives of lithium, as well as MeMgI, PhMgBr, and 1,3-butadien-2-yl magnesium chloride, are used (the corresponding *cis* isomers are not recorded in the ¹H and ¹³C NMR spectra).

As compared with substrate **1**, epoxide **2** reacts more selectively. Thus its reaction with vinylmagnesium bromide leads to lactone **17** in the form of virtually one stereoisomer. The reaction of CH₂=CHMgBr with epoxide **3** is accompanied by allylic rearrangement in the step involving cyclization of the magnesium enolate and gives *trans*-dialkenylcyclopropane **18** (a mixture of *E* and *Z* isomers with respect to the disubstituted double bond, *E/Z* = 20:80).^{*} When other organomagnesium or organolithium compounds (MeMgI, Z-*iso*-BuCH=CHLi, PhMgBr, PhC≡CLi) are used, cyclopropanes of the **18** type cannot be synthesized in greater than 10-15% yields because of the formation of a complex mixture of products.



We found that bicyclic lactones **6-12**, which contain an alkenyl fragment, undergo rearrangement with opening of the cyclopropane ring when they are refluxed in DMF in the presence of three equivalents of LiI (0.5-2 h) [10] (Table 2).

Cyclopropanes **6-8**, which have a terminal double bond, undergo rearrangement to give cyclopentenes **19-21** [11, 12]. Compounds **20** and **21** are formed in a mixture with decarboxylation products **22** and **23**. Replacement of LiI by NaI and Bu₄NI in the reaction with the participation of substrate **6** leads to a decrease in the yield of cyclopentene **19** from 61% to 39% and 23%, respectively. A still more significant decrease (to 10-15%) in the yield of product **19** is observed on passing from DMF to hexamethylphosphoric triamide (HMPT) and DMSO.

Cyclopropanes **9-12**, which contain an internal double bond, react to give conjugated dienes **24-27** [11]. According to the ¹H and ¹³C NMR spectroscopic data [9], the disubstituted C⁶-C⁷ bonds in products **24-26** and the C⁸=C⁹ bonds in products

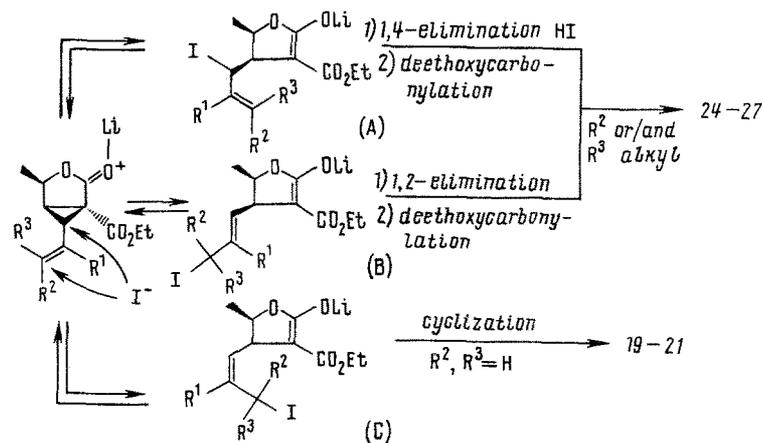
^{*}In the ¹³C NMR spectrum of **18** a single signal corresponds to both carbonyl C atoms; the pair of signals (Δδ = 1.03 ppm) corresponds to the allyl C⁶ atom.

24 and **25** have a *trans* configuration (J_{H^6, H^7} and $J_{H^8, H^9} = 14-16$ Hz). Compounds **26** and **27** are mixtures of *E* and *Z* isomers with respect to the trisubstituted $C^8=C^9$ and $C^6=C^7$ bonds, respectively.

It should be noted that complete retention of the relative configuration of the C^3 and C^4 atoms is observed during the reaction.*

β -Styryl derivative **13** and 1,1-diethoxycarbonyl-2-vinylcyclopropane do not undergo rearrangement under the investigated conditions.

The key step in the rearrangement of alkenylcyclopropanes **6-12** to cyclopentenes **19-21** or 1,3-dienes **24-27** is evidently orbital nucleophilic opening of the cyclopropane ring by the I^- anion to give enolates A (1,5-addition) or B and C (1,7-addition), which is facilitated by coordination of the Li^+ cation at the carbonyl oxygen atom [14-16].



Upon cyclization, derivatives A and B give only the starting cyclopropane,† while enolate C is capable of irreversible cyclization to a cyclopentene. The 1,4- and/or 1,2-elimination of HI in enolates A and B when the corresponding CH fragments are present leads to a diene system. Let us note that a necessary condition for the occurrence of the reaction is a rigid bicyclic structure of compounds **6-12**.

Thus the reactions described in this paper open up a simple and effective pathway to the synthesis of various γ -lactones containing a cyclopropane, cyclopentene, or 1,3-diene fragment.

EXPERIMENTAL

The 1H and ^{13}C NMR spectra of solutions of the compounds in $CDCl_3$ were recorded with a Bruker AM-300 spectrometer with tetramethylsilane (TMS) as the internal standard. The IR spectra of thin layers of the substances were obtained with a UR-20 spectrometer. The mass spectra were recorded with an MKh-1300 spectrometer at an input-cylinder temperature of $100^\circ C$ and electron-ionization energies of 70 and 12 eV.

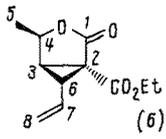
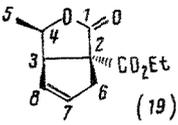
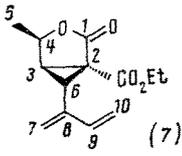
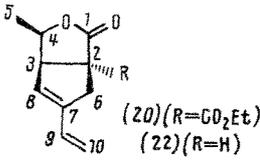
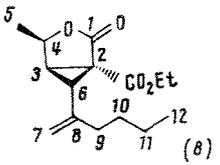
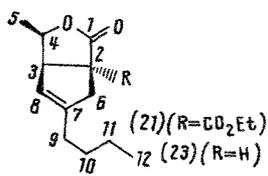
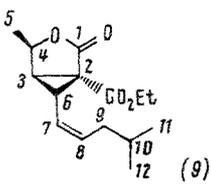
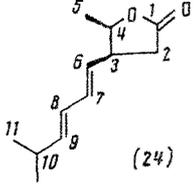
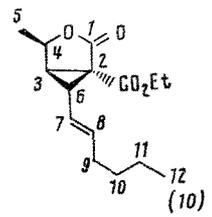
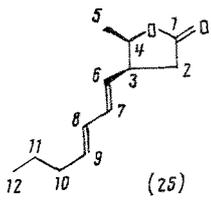
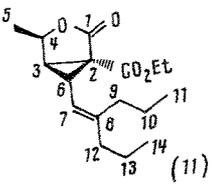
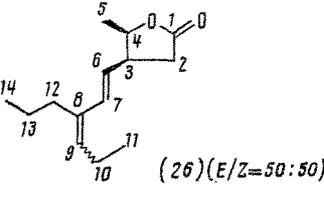
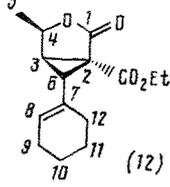
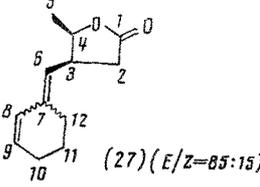
The starting organolithium and organomagnesium compounds were obtained by known methods. All of the experiments were carried out in a dry Ar atmosphere using absolute solvents.

Reactions of Monoepoxides 1-3 with Organolithium and Organomagnesium Compounds. A 1-ml (2.1 mmoles) sample of a 2.1 M ether solution of $MeMgI$ was added with stirring and cooling to $-78^\circ C$ to a solution of 0.456 g (2 mmoles) of epoxide **1** in 15 ml of Et_2O containing 0.019 g (0.1 mmole) of CuI , after which the mixture was stirred for 15 min at $-78^\circ C$, heated for 3 h to $20^\circ C$, and hydrolyzed by the addition of 10 ml of 8% HCl solution. The ether layer was separated, and the aqueous layer was extracted with Et_2O (3×5 ml). The combined ether extracts were dried with $MgSO_4$ and concentrated. Column chromatography [silica gel L 40/100, hexane: Et_2O (2:1)] gave 0.257 g (65%) of 1-ethoxycarbonyl-4,6-dimethyl-3-oxabicyclo[3.1.0]-hexan-2-one (**4**) in the form of a colorless oil. PMR spectrum (δ , ppm; J, Hz): 1.35 t (3H, Me, $J = 7.2$), 1.35 d (3H, Me, $J = 5$), 1.38 d (3H, Me, $J = 5$), 1.8 dq (1H, CH, $J_1 = J_2 = 5$), 2.55 dd (1H, CH, $J_1 = J_2 = 5$), 4.32 q (2H, CH_2O , $J =$

*An analysis of molecular models of compounds **19-21** and **24-27** by the MM2 method [13] gives the following dihedral angles $H^3-C^3-C^4-H^4$ and constants of spin-spin coupling of the H^3 and H^4 protons for syn and anti orientations of the substituents attached to the C^3 and C^4 atoms relative to the lactone ring: $37-38^\circ$, 7.4-7.6 Hz (syn); $85-86^\circ$, 1.9-2.0 Hz (anti). The observed J_{H^3, H^4} values correspond to a syn configuration.

†The cyclization of enolate A to a cyclopentene is an unfavorable 5-endo-trigonal process [17].

TABLE 2. Rearrangement of Alkenylcyclopropanes 6-12 in the Presence of Li (3 equivalents) in DMF at 150°C

Substrate	Reaction time	Rearrangement product	Yield, % ^a
 (6)	40 min	 (19)	61
 (7)	2 h	 (20) (R=CO ₂ Et) (22) (R=H)	65 ^b
 (8)	2 h	 (21) (R=CO ₂ Et) (23) (R=H)	84 ^c
 (9)	1.5 h	 (24)	62
 (10)	1.5 h	 (25)	57
 (11)	30 min	 (26) (E/Z=50:50)	71
 (12)	1 h	 (27) (E/Z=85:15)	57

^aThe yields of the isolated products are presented.

^bOverall yield of 20 and 22: 20:22 = 74:26.

^cOverall yield of 21 and 23: 21:23 = 63:37.

7.2), 4.8 dq (1H, CHO, $J_1 = J_2 = 5$). ^{13}C NMR spectrum (δ , ppm): 170.67 (C^1), 36.46 s (C^2), 35.98 d (C^3), 73.96 (C^4), 17.50 q (C^5), 26.33 d (C^6), 11.92 q (Me), 14.30 q, 61.82 t, 165.67 s (CO_2Et). IR spectrum (ν , cm^{-1}): 1780, 1720. Mass spectrum (m/z): 198 M^+ .

1-Ethoxycarbonyl-4-methyl-6-propyl-3-oxabicyclo[3.1.0]hexan-2-one (5). PMR spectrum (δ , ppm; J , Hz): 0.88 t (3H, Me, $J = 7$); 1.06-1.63 m (10H, Me + CH_2); 2.21-3.08 m (2H, CH); 4.03 q, 4.1 q (2H, CH_2O , $J = 7$); 4.63 dq, 4.71 dq (1H, CHO, $J_1 = J_2 = 4$). ^{13}C NMR spectrum (δ , ppm): 171.64 s, 174.21 s (C^1); 36.81 s, 37.10 s (C^2); 36.03 d, 36.23 d (C^3); 73.89 d, 74.28 d (C^4); 16.71 q, 17.49 q (C^5); 31.70 d, 32.90 d (C^6); 13.64 q, 14.17 q, 61.82 t, 161.50 s, 167.63 s (CO_2Et); 21.67 t, 22.58 t, 27.41 t, 28.53 t (CH_2); 11.13 q, 22.58 q (Me). IR spectrum (ν , cm^{-1}): 1770, 1710. Mass spectrum (m/z): 226 M^+ .

1-Ethoxycarbonyl-4-methyl-6-vinyl-3-oxabicyclo[3.1.0]hexan-2-one (6). PMR spectrum (δ , ppm; J , Hz): 0.88 t (3H, Me, $J = 7$), 1.06-1.63 m (10H, Me + CH_2), 2.21-3.08 m ($J = 7$ and 4.5), 2.63 dd (1H, CH, $J_1 = J_2 = 4.5$), 4.12 q (2H, CH_2O , $J = 7$), 4.66 dq (1H, CHO, $J = 4.5$ and 6), 4.96-6.0 m (3H, $\text{CH}=\text{CH}_2$) (trans isomer). ^{13}C NMR spectrum (δ , ppm): 168.92 s, 169.59 s (C^1); 37.53 s, 37.65 s (C^2); 35.12 d, 36.88 d (C^3); 73.83 d, 74.48 d (C^4); 16.64 q, 17.30 q (C^5); 34.20 d (C^6); 129.83 d, 130.36 d (C^7); 120.11 t, 121.15 t (C^8); 14.23 q, 61.88 t, 164.69 s, 166.65 s (CO_2Et). IR spectrum (ν , cm^{-1}): 3080, 1770, 1710, 1640, 980, 910. Mass spectrum (m/z): 210 M^+ .

1-Ethoxycarbonyl-4-methyl-6-(1,3-butadien-2-yl)-3-oxabicyclo[3.1.0]hexan-2-one (7). PMR spectrum (δ , ppm; J , Hz): 1.21 t (3H, Me, $J = 7.2$), 1.42 d (3H, Me, $J = 6.1$), 2.44 d (1H, $\text{CHC}=\text{C}$, $J = 5.1$), 3.06 dd (1H, CH, $J_1 = J_2 = 5.1$), 4.14 q (2H, CH_2O , $J = 7.2$), 4.87 dq (1H, CHO, $J = 5.1$ and 6.1), 5.1-5.6 m (4H, $\text{CH}_2=\text{C}$), 6.4 dd (1H, $\text{CH}=\text{C}$, $J = 18$ and 11). ^{13}C NMR spectrum (δ , ppm): 169.48 s (C^1); 36.87 s (C^2); 32.00 d (C^3); 73.68 d (C^4); 17.41 q (C^5); 32.00 d (C^6); 119.53 t (C^7); 137.80 s (C^8); 136.56 d (C^9); 115.64 t (C^{10}); 14.11 q, 61.41 t, 163.53 s (CO_2Et). IR spectrum (ν , cm^{-1}): 3100, 1770, 1725, 1605, 940, 930, 905. Mass spectrum (m/z): 136 M^+ .

1-Ethoxycarbonyl-4-methyl-6-(1-hexen-2-yl)-3-oxabicyclo[3.1.0]hexan-2-one (8). PMR spectrum (δ , ppm; J , Hz): 0.91 t (3H, Me, $J = 7.1$); 1.25 t (3H, Me, $J = 7.1$); 1.38 d (3H, Me, $J = 6.2$); 1.2-1.5 m (4H, CH_2); 2.09 t (2H, $\text{CH}_2\text{C}=\text{C}$, $J = 7.4$); 2.26 d (1H, $\text{CHC}=\text{C}$, $J = 5.6$); 2.95 dd (1H, CH, $J = 5.6$ and 4.6); 4.2 q (2H, CH_2O , $J = 7.1$), 4.82 dq (1H, CHO, $J = 4.6$ and 6.2); 4.9 s, 5.02 s (2H, $\text{CH}_2=\text{C}$). ^{13}C NMR spectrum (δ , ppm): 169.99 (C^1); 38.39 s (C^2); 36.22 d (C^3); 73.95 d (C^4); 17.36 q (C^5); 34.89 d (C^6); 113.94 t (C^7); 140.87 s (C^8); 31.98 t (C^9); 29.83 t (C^{10}); 22.39 t (C^{11}); 13.84 q (C^{12}); 14.16 q, 61.69 t, 163.95 s (CO_2Et). IR spectrum (ν , cm^{-1}): 3100, 1780, 1730, 1650, 920. Mass spectrum (m/z): 266 M^+ .

1-Ethoxycarbonyl-4-methyl-6-(Z-4-methyl-1-pentenyl)-3-oxabicyclo[3.1.0]hexan-2-one (9). PMR spectrum (δ , ppm; J , Hz): 0.87 d (6H, Me, $J = 6.3$), 1.26 t (3H, Me, $J = 7.3$), 1.35 d (3H, Me, $J = 6.1$), 1.62 m (1H, CH), 2.02 dd (2H, $\text{CH}_2\text{C}=\text{C}$, $J_1 = J_2 = 7.2$), 2.47 dd (1H, $\text{CHC}=\text{C}$, $J = 8.9$ and 5.1), 2.73 dd (1H, CH, $J_1 = J_2 = 5.1$), 4.22 q (2H, CH_2O , $J = 7.3$), 4.74 dq (1H, CHO, $J = 5.1$ and 6.1), 5.24 dd (1H, $\text{CH}=\text{C}$, $J = 10.8$ and 8.9), 5.64 dt (1H, $\text{CH}=\text{C}$, $J = 10.8$ and 7.5). ^{13}C NMR spectrum (δ , ppm): 169.75 s (C^1); 37.49 s (C^2); 35.60 d (C^3); 73.74 d (C^4); 17.45 q (C^5); 29.53 d (C^6); 121.96 d, 134.85 d (C^7 , C^8); 36.79 t (C^9); 28.34 d (C^{10}); 22.11 q, 22.21 q (C^{11} , C^{12}). IR spectrum (ν , cm^{-1}): 3080, 1770, 1720, 1650. Mass spectrum (m/z): 266 M^+ .

1-Ethoxycarbonyl-4-methyl-6-(E-1-hexenyl)-3-oxabicyclo[3.1.0]hexan-2-one (10). PMR spectrum (δ , ppm; J , Hz): 0.88 t (3H, Me, $J = 6.1$), 1.31 t (3H, Me, $J = 7.3$), 1.38 d (3H, Me, $J = 6.3$), 1.3-1.7 m (4H, CH_2), 2.0 m (2H, $\text{CH}_2\text{C}=\text{C}$), 2.35 dd (1H, $\text{CHC}=\text{C}$, $J = 8.7$ and 5.2), 2.75 dd (1H, CH, $J_1 = J_2 = 5.2$), 4.26 q (2H, CH_2O , $J = 7.3$), 4.75 dq (1H, CHO, $J = 5.2$ and 6.3), 5.33 dd (1H, $\text{CH}=\text{C}$, $J = 15.5$ and 8.7), 5.84 dt (1H, $\text{CH}=\text{C}$, $J = 15.5$ and 6.6). ^{13}C NMR spectrum (δ , ppm): 169.85 s (C^1); 37.53 s (C^2); 34.99 d (C^3); 73.83 d (C^4); 17.36 q (C^5); 34.14 d (C^6); 121.81 d, 137.28 (C^7 , C^8); 32.18 t (C^9); 31.07 t (C^{10}); 22.06 t (C^{11}); 13.84 q (C^{12}); 14.23 q, 61.69 t, 164.76 s (CO_2Et). IR spectrum (ν , cm^{-1}): 1780, 1720, 1640, 970. Mass spectrum (m/z): 266 M^+ .

1-Ethoxycarbonyl-4-methyl-6-(2-propyl-1-pentenyl)-3-oxabicyclo[3.1.0]hexan-2-one (11). PMR spectrum (δ , ppm; J , Hz): 0.87 t, 0.93 t (6H, Me, $J = 7.2$), 1.3 t (3H, Me, $J = 7.1$), 1.38 d (3H, Me, $J = 6.3$), 1.3-1.5 m (4H, CH_2), 1.9-2.2 m (4H, $\text{CH}_2\text{C}=\text{C}$), 2.51 dd (1H, $\text{CHC}=\text{C}$, $J = 8.9$ and 5.2), 2.76 dd (1H, CH, $J_1 = J_2 = 5.2$), 4.25 q (2H, CH_2O , $J = 7.1$), 4.78 dq (1H, CHO, $J = 5.2$ and 6.3), 5.0 d (1H, $\text{CH}=\text{C}$, $J = 8.9$). ^{13}C NMR spectrum (δ , ppm): 170.01 s (C^1); 37.81 s (C^2); 35.58 d (C^3); 73.87 d (C^4); 17.47 q (C^5); 30.57 d (C^6); 116.60 d (C^7); 147.29 s (C^8); 38.98 t (C^9); 20.98 t, 21.42 t (C^{10} , C^{13}); 13.69 q, 14.01 q (C^{11} , C^{14}); 32.83 t (C^{12}); 14.24 q, 61.64 t, 164.76 s (CO_2Et). IR spectrum (ν , cm^{-1}): 1780, 1730, 1650, 860. Mass spectrum (m/z): 294 M^+ .

1-Ethoxycarbonyl-4-methyl-6-(1-cyclohexenyl)-3-oxabicyclo[3.1.0]hexan-2-one (12). PMR spectrum (δ , ppm; J , Hz): 1.3 t (3H, Me, $J = 7.1$); 1.36 d (3H, Me, $J = 6.2$); 1.55 m (4H, ring CH_2); 1.8-2.3 m (5H, $\text{CHC}=\text{C} + \text{CH}_2\text{C}=\text{C}$); 2.67 m, 2.95 m (1H, CH); 4.17 q (2H, CH_2O , $J = 7.1$); 4.8 m (1H, CHO); 5.63 m, 5.83 m (1H, $\text{CH}=\text{C}$). ^{13}C NMR spectrum (δ ,

ppm): 170.30 s (C¹); 37.66 s (C²); 36.56 d (C³); 73.89 d (C⁴); 17.36 q (C⁵); 31.20 d (C⁶); 128.92 s (C⁷); 127.29 d (C⁸); 25.13 t (C⁹); 21.93 t, 22.52 t (C¹⁰, C¹¹); 28.79 t (C¹²); 14.23 q, 61.49 t, 164.04 s (CO₂Et) (trans isomer). IR spectrum (ν , cm⁻¹): 3100, 1760, 1725, 880. Mass spectrum (m/z): 264 M⁺.

1-Ethoxycarbonyl-4-methyl-6-(E- β -styryl)-3-oxabicyclo[3.1.0]hexan-2-one (13). PMR spectrum (δ , ppm; J , Hz): 1.29 t (3H, Me, $J = 7.3$), 1.4 d (3H, Me, $J = 6.4$), 2.53 dd (1H, CHC=, $J = 8.8$ and 5), 2.9 dd (1H, CH, $J_1 = J_2 = 5$), 4.25 q (2H, CH₂O, $J = 7.3$), 4.8 dq (1H, CHO, $J = 5$ and 6.4), 6.05 dd (1H, CH=, $J = 16$ and 8.8), 6.65 d (1H, CH=, $J = 16$), 7.31 m (5H, arom.) (trans isomer). IR spectrum (ν , cm⁻¹): 3090, 3070, 3040, 1775, 1720, 1605, 1500, 980, 770, 750, 715. Mass spectrum (m/z): 286 M⁺.

1-Ethoxycarbonyl-4-methyl-6-phenyl-3-oxabicyclo[3.1.0]hexan-2-one (14). PMR spectrum (δ , ppm; J , Hz): 0.8 t (3H, Me, $J = 7.2$), 1.45 d (3H, Me, $J = 6.3$), 2.95 d (1H, CHPh, $J = 5.5$), 3.3 dd (1H, CH, $J_1 = J_2 = 5.5$), 3.93 m (2H, CH₂O), 4.94 dq (1H, CHO, $J = 5.5$ and 6.3), 7.3 m (5H, arom.). ¹³C NMR spectrum (δ , ppm): 169.85 s (C¹); 39.17 s (C²); 32.12 d, 34.30 d (C³, C⁶); 74.06 d (C⁴); 17.45 q (C⁵); 128.08 d, 128.35 d, 128.79 d, 132.09 s (Ph), 13.71 q, 61.50 t, 163.47 s (CO₂Et). IR spectrum (ν , cm⁻¹): 3115, 3075, 3045, 1780, 1725, 1610, 1510, 790, 755, 715. Mass spectrum (m/z): 260 M⁺.

1-Ethoxycarbonyl-4-methyl-6-(1-pentynyl)-3-oxabicyclo[3.1.0]hexan-2-one (15). PMR spectrum (δ , ppm; J , Hz): 0.9 t (3H, Me, $J = 6.3$), 1.26 t (3H, Me, $J = 7.2$), 1.33 d (3H, Me, $J = 6.2$), 1.1-1.6 m (2H, CH₂), 2.0 m (3H, CHC= + CH₂C=), 2.6 m (1H, CH), 4.16 q (2H, CH₂O, $J = 7.2$), 4.63 m (1H, CHO). ¹³C NMR spectrum (δ , ppm): 168.77 s (C¹); 37.82 s, 38.30 s (C²); 35.49 d, 35.92 d (C³); 73.96 d, 74.12 d (C⁴); 15.12 q, 17.45 q (C⁵); 19.88 d (C⁶); 72.65 s, 84.20 s (C≡C); 20.70 t, 21.78 t, 21.94 t, 13.38 q (Ph); 14.14 q, 14.30 q, 62.04 t, 163.46 s (CO₂Et). IR spectrum (ν , cm⁻¹): 2300, 1780, 1720. Mass spectrum (m/z): 250 M⁺.

1-Ethoxycarbonyl-4-methyl-6-phenylethynyl-3-oxabicyclo[3.1.0]hexan-2-one (16). PMR spectrum (δ , ppm; J , Hz): 1.29 t, 1.33 t (3H, Me, $J = 7$); 1.44 d, 1.67 d (3H, Me, $J = 6.5$); 2.42 d, 2.45 d (1H, CHC=, $J = 8$ and 4.9); 2.75 dd, 3.04 dd (1H, CH, $J = 8$ and 4.9, $J_1 = J_2 = 4.9$); 4.29 q, 4.31 q (2H, CH₂O, $J = 7$); 4.81 dq, 4.88 dq (1H, CHO, $J = 4.9$ and 6.5); 7.33 m (5H, arom.). ¹³C NMR spectrum (δ , ppm): 168.45 s (C¹); 37.93 s, 38.74 s (C²); 35.87 d, 36.30 d (C³); 73.96 d, 74.12 d (C⁴); 15.28 q, 17.45 q (C⁵); 19.94 d, 20.64 d (C⁶); 81.87 s, 82.03 s, 83.38 s (C≡C); 122.07 s, 128.30 d, 128.68 d, 131.66 d (Ph); 14.09 q, 14.25 q, 62.15 t, 62.36 t, 163.25 s, 165.73 s (CO₂Et). IR spectrum (ν , cm⁻¹): 3090, 3075, 2270, 1780, 1725, 1600, 1580, 1500, 775, 750, 710. Mass spectrum (m/z): 284 M⁺.

1-Ethoxycarbonyl-4-propyl-6-vinyl-3-oxabicyclo[3.1.0]hexan-2-one (17). PMR spectrum (δ , ppm; J , Hz): 0.97 t (3H, Me, $J = 7.3$), 1.31 t (3H, Me, $J = 7.1$), 1.53 m (4H, CH₂), 2.37 dd (1H, CHC=, $J = 8.2$ and 4.8), 2.79 dd (1H, CH, $J_1 = J_2 = 4.8$), 4.26 q (2H, CH₂O, $J = 7.1$), 4.62 dt (1H, CHO, $J = 4.8$ and 7.5), 5.0-5.8 m (3H, CH= + CH₂=). IR spectrum (ν , cm⁻¹): 3095, 1780, 1725, 1640. Mass spectrum (m/z): 238 M⁺.

1,1-Diethoxycarbonyl-2-(3-hydroxy-1-butenyl)-3-vinylcyclopropane (18). PMR spectrum (δ , ppm; J , Hz): 1.22 t (6H, Me, $J = 7$), 1.28 d (3H, Me, $J = 6.3$), 2.55 m (2H, CHC=), 3.05 broad s (1H, OH), 4.1 q (4H, CH₂O, $J = 7$), 4.0-4.3 m (1H, CHOH), 4.9-6.0 m (5H, CH= + CH₂=). ¹³C NMR spectrum (δ , ppm): 42.97 s (C¹); 34.03 d, 35.49 d (C², C³); 123.68 d (C⁴); 138.70 d (C⁵); 68.24 d, 69.27 d (C⁶); 21.56, 23.30 q (C⁷); 132.09 d (C⁸); 118.66 t (C⁹); 14.20 q, 61.66 t, 167.15 s (CO₂Et). IR spectrum (ν , cm⁻¹): 3500, 3095, 1730, 1645, 985, 930. Mass spectrum (m/z): 282 M⁺.

Rearrangement of Alkenylcyclopropanes 6-12 in the Presence of LiI. A solution of 0.42 g (2 mmoles) of cyclopropane 6 and 0.804 g (6 mmoles) of LiI in 10 ml of DMF was stirred at 150°C, during which the progress of the reaction was monitored by TLC. After 40 min, the reaction mixture was cooled to 20°C, 20 ml of H₂O was added, and the aqueous mixture was extracted with Et₂O (3 × 10 ml). The ether layer was dried with MgSO₄, concentrated, and subjected to column chromatography [silica gel L 40-100, hexane—Et₂O (1:1)] to give 0.256 g (61%) of 1-ethoxycarbonyl-4-methyl-3-oxabicyclo[3.3.0]-6-octen-2-one (19) in the form of a colorless oil. PMR spectrum (δ , ppm; J , Hz): 1.3 t (3H, Me, $J = 7.2$), 1.43 d (3H, Me, $J = 6.5$), 3.05 ddd (1H, ring CH₂, $J = 17.4$, 4.3, and 2.5), 3.14 ddd (1H, ring CH₂, $J = 17.4$, 4.6, and 2.5), 3.73 m (1H, CH), 4.25 q (2H, CH₂O, $J = 7.2$), 4.94 dq (1H, CHO, $J_1 = J_2 = 6.5$), 5.6 m (1H, CH=), 5.88 m (1H, CH=). ¹³C NMR spectrum (δ , ppm): 175.85 s (C¹); 61.75 s (C²); 56.86 d (C³); 78.46 d (C⁴); 16.78 q (C⁵); 40.41 t (C⁶); 125.26 d (C⁷); 132.84 d (C⁸); 14.01 q, 62.21 t, 169.00 s (CO₂Et). IR spectrum (ν , cm⁻¹): 3080, 1770, 1735, 1620. Mass spectrum (m/z): 210 M⁺.

1-Ethoxycarbonyl-4-methyl-7-vinyl-3-oxabicyclo[3.3.0]-6-octen-2-one (20). PMR spectrum (δ , ppm; J , Hz): 1.3 t (3H, Me, $J = 6.9$), 1.45 d (3H, Me, $J = 6.4$), 3.15 d (1H, ring CH₂, $J = 16.7$), 3.25 d (1H, ring CH₂, $J = 16.7$), 3.77 (1H, CH), 4.26 q (2H, CH₂O, $J = 6.9$), 4.93 dq (1H, CHO, $J_1 = J_2 = 6.4$), 5.25 m (2H, CH₂=), 5.51 broad s (1H, CH=), 6.48 dd (1H, CH=, $J = 17.4$ and 10.7). ¹³C NMR spectrum (δ , ppm): 175.64 s (C¹); 61.47 s (C²); 56.63 d (C³); 78.17 d (C⁴); 16.79 q (C⁵); 38.61 t (C⁶); 144.29 s (C⁷); 131.69 d (C⁸); 123.30 d (C⁹); 117.76 t (C¹⁰); 14.09 q, 62.29 t, 168.95 s (CO₂Et). IR spectrum (ν , cm⁻¹): 3085, 3040, 1770, 1740, 1650, 990, 910. Mass spectrum (m/z): 236 M⁺.

1-Ethoxycarbonyl-4-methyl-7-butyl-3-oxabicyclo[3.3.0]-6-octen-2-one (21). PMR spectrum (δ , ppm; J , Hz): 0.9 t (3H, Me, $J = 7$), 1.30 t (3H, Me, $J = 7.1$), 1.42 d (3H, Me, $J = 6.3$), 1.3-1.5 m (4H, CH₂), 2.1 t (2H, CH₂C=, $J = 7.2$), 2.88 d (1H, ring CH₂, $J = 16.9$), 3.09 d (1H, ring CH₂, $J = 16.9$), 3.67 m (1H, CH), 4.25 q (2H, CH₂O, $J = 7.1$), 4.88 dq (1H, CHO, $J_1 = J_2 = 6.3$), 5.19 broad s (1H, CH=). ¹³C NMR spectrum (δ , ppm): 176.02 s (C¹); 65.54 s (C²); 56.62 d (C³); 78.24 d (C⁴); 16.70 q (C⁵); 42.48 t (C⁶); 147.77 s (C⁷); 117.70 d (C⁸); 30.69 t (C⁹); 29.63 t (C¹⁰); 22.38 t (C¹¹); 13.78 q (C¹²); 14.09 q, 62.05 t, 169.24 s (CO₂Et). IR spectrum (ν , cm⁻¹): 1770, 1740, 1640. Mass spectrum (m/z): 266 M⁺.

4-Methyl-7-vinyl-3-oxabicyclo[3.3.0]-6-octen-2-one (22). PMR spectrum (δ , ppm; J , Hz): 1.41 d (3H, Me, $J = 6.4$), 2.76 dd (1H, ring CH₂, $J = 16.2$ and 8.7), 2.97 d (1H, ring CH₂, $J = 16.2$), 3.29 dd (1H, CH, $J_1 = J_2 = 8.7$), 3.62 m (1H, CHC=), 4.74 dq (1H, CHO, $J_1 = J_2 = 6.4$), 5.21 m (2H, CH₂=), 5.60 broad s (1H, CH=), 6.54 dd (1H, CH=, $J = 17.3$ and 10.7). IR spectrum (ν , cm⁻¹): 3085, 3040, 1770, 1640, 1595, 990, 910. Mass spectrum (m/z): 164 M⁺.

3-(5-Methyl-1,3-hexadienyl)-4-pentanolide (24). PMR spectrum (δ , ppm; J , Hz): 1.05 d (6H, Me, $J = 6.8$), 1.27 d (3H, Me, $J = 6.6$), 2.35 m (1H, CH), 2.47 dd (1H, ring CH₂, $J = 17.4$ and 7.4), 2.67 dd (1H, ring CH₂, $J = 17.4$ and 8.1), 3.19 m (1H, CH), 4.7 dq (1H, CHO, $J = 7.2$ and 6.6), 5.45 dd (1H, CH=, $J = 14.6$ and 8.8), 5.68 dd (1H, CH=, $J = 15$ and 6.7), 6.0 dd (1H, CH=, $J = 15.0$ and 10.2), 6.1 dd (1H, CH=, $J = 14.6$ and 10.2). ¹³C NMR spectrum (δ , ppm): 176.20 s (C¹); 34.57 t (C²); 42.69 d (C³); 79.62 d (C⁴); 16.53 q (C⁵); 134.02 d (C⁶); 126.27 d, 126.29 d (C⁷, C⁸); 142.82 d (C⁹); 31.12 d (C¹⁰); 22.24 q (C¹¹). IR spectrum (ν , cm⁻¹): 3040, 1775, 1655, 960. Mass spectrum (m/z): 194 M⁺.

3-(1,3-Heptadienyl)-4-pentanolide (25). PMR spectrum (δ , ppm; J , Hz): 0.91 t (3H, Me, $J = 7.4$), 1.27 d (3H, Me, $J = 6.6$), 1.42 tq (2H, CH₂, $J_1 = J_2 = 7.4$), 2.06 dt (2H, CH₂C=, $J = 7.4$ and 7.2), 2.45 dd (1H, ring CH₂, $J = 17.4$ and 7.4), 2.67 dd (1H, ring CH₂, $J = 17.4$ and 8.1), 3.2 m (1H, CH), 4.7 dq (1H, CHO, $J = 7.2$ and 6.6), 5.44 dd (1H, CH=, $J = 14.1$ and 8.9), 5.7 dt (1H, CH=, $J = 14.5$ and 7.2), 6.02 dd (1H, CH=, $J = 14.5$ and 10.2), 6.1 dd (1H, CH=, $J = 14.1$ and 10.2). ¹³C NMR spectrum (δ , ppm): 176.18 s (C¹); 34.53 t (C²); 42.63 d (C³); 79.57 d (C⁴); 16.54 q (C⁵); 135.71 d (C⁶); 125.99 d (C⁷); 129.25 d (C⁸); 133.82 (C⁹); 34.53 t (C¹⁰); 22.39 t (C¹¹); 13.71 q (C¹²). IR spectrum (ν , cm⁻¹): 1775, 1660, 960. Mass spectrum (m/z): 194 M⁺.

3-(3-Propyl-1,3-hexadienyl)-4-pentanolide (26). PMR spectrum (δ , ppm; J , Hz): 0.9 t, 0.93 t, 0.99 t, 1.0 t (6H, Me, $J = 7.4$); 1.26 d, 1.28 d (3H, Me, $J = 6.6$); 1.43 m (2H, CH₂); 2.08-2.24 m (4H, CH₂C=); 2.4 dd, 2.45 dd (1H, ring CH₂, $J = 17.4$ and 6.9); 2.7 dd, 2.72 dd (1H, ring CH₂, $J = 17.4$ and 8.1); 3.22 m (1H, CH); 4.71 dq, 4.73 dq (1H, CHO, $J_1 = J_2 = 6.6$); 5.37 t, 5.44 t (1H, CH=, $J = 7.3, 7.2$); 5.42 dd, 5.54 dd (1H, CH=, $J = 15.7$ and 8.8, 15.7 and 9); 6.01 d, 6.4 d (1H, CH=, $J = 15.7$). ¹³C NMR spectrum (δ , ppm): 176.24 s, 176.36 s (C¹); 34.85 t (C²); 43.05 d, 43.46 d (C³); 79.12 d, 79.78 d (C⁴); 16.58 q (C⁵); 135.61 d, 137.20 d (C⁶); 129.60 d, 132.85 d (C⁷); 134.47 s, 136.69 s (C⁸); 121.50 d, 124.28 d (C⁹); 21.94 t, 22.23 t (C¹⁰); 13.97 q, 14.13 q, 14.23 q, 14.44 q (C¹¹, C¹⁴); 28.78 t, 36.09 t (C¹²); 20.78 t, 21.51 t (C¹³). IR spectrum (ν , cm⁻¹): 1775, 1650, 960, 860. Mass spectrum (m/z): 222 M⁺.

3-[(2-Cyclohexenylidene)methyl]-4-pentanolide (27). PMR spectrum (δ , ppm; J , Hz): 1.25 d (3H, Me, $J = 6.6$); 1.7 m (2H, ring CH₂); 2.15 m, 2.3 m (4H, CH₂C=), 2.38 dd (1H, ring CH₂, $J = 17.3$ and 7.2); 2.7 dd (1H, ring CH₂, $J = 17.3$ and 8.1); 3.49 m, 3.57 m (1H, CH); 4.72 dq (1H, CHO, $J_1 = J_2 = 6.6$); 4.99 d, 5.12 d (1H, CH=, $J = 9.4, 9.9$); 5.83 dt, 5.95 dt (1H, ring CH=, $J = 8.5$ and 4.1, 10.2 and 4.1); 6.06 d, 6.3 d (1H, ring CH=, $J = 8.5, 10.2$). ¹³C NMR spectrum (δ , ppm): 176.37 s, 176.45 s (C¹); 35.36 t, 35.46 t (C²); 36.76 d, 37.42 d (C³); 79.50 d, 79.71 d (C⁴); 16.33 q, 16.34 q (C⁵); 120.02 d, 121.88 d (C⁶); 137.43 s, 138.34 s (C⁷); 123.38 d, 129.89 d, 130.21 d, 132.60 d (C⁸, C⁹); 25.49 t, 26.14 t (C¹⁰); 22.29 t, 23.00 t (C¹¹); 25.62 t, 32.32 t (C¹²). Mass spectrum (m/z): 192 M⁺.

LITERATURE CITED

1. E. Ghera and Y. Ben-David, *Tetrahedron Lett.*, 4603 (1979).
2. P. Premprec, S. Radviroongit, and Y. Thebtarenonth, *J. Org. Chem.*, **48**, 3553 (1983).
3. M. Joucla, M. F. Goumzili, and B. Fouchet, *Tetrahedron Lett.*, **27**, 1677 (1986).
4. P. Kolsaker and H. J. Storesund, *J. Chem. Soc., Chem. Commun.*, 375 (1972).
5. S. Torii, H. Tanaka, and Y. Nagai, *Bull. Chem. Soc. Jap.*, **50**, 2825 (1977).
6. M. J. De Vos and A. Krief, *Tetrahedron Lett.*, 1891 (1979).
7. M. P. Cook and J. Van Law, Jr., *J. Org. Chem.*, **51**, 758 (1986).
8. A. N. Kasatkin, R. Kh. Biktimirov, A. N. Kulak, and G. A. Tolstikov, *Zh. Org. Khim.* (in press).
9. A. J. Gordon and R. A. Ford, *A Chemist's Guide*, Wiley-Interscience, New York (1973).
10. S. Hashimoto, T. Shinoda, and S. Ikegami, *Tetrahedron Lett.*, **27**, 2885 (1986).
11. Z. Goldschmidt and B. Crommer, *Chem. Soc. Rev.*, **17**, 229 (1988).
12. K. Miura, K. Fugami, K. Oshima, and K. Utimoto, *Tetrahedron Lett.*, **29**, 1543 (1988).

13. N. L. Allinger, *J. Am. Chem. Soc.*, **99**, 8127 (1977).
14. S. Danishefsky, *Acc. Chem. Res.*, **12**, 66 (1979).
15. T. Hiyama, Y. Morizawa, H. Yamamoto, and H. Nozaki, *Bull. Chem. Soc. Japan*, **54**, 2151 (1981).
16. K. Burgess, *J. Org. Chem.*, **52**, 2046 (1987).
17. J. E. Baldwin, *J. Chem. Soc., Chem. Commun.*, 734 (1976).