## Formation of N-Tributylstannyl Heterocycle from Bis(tributyltin) Oxide and ω-Haloalkyl Isocyanate. One-Pot Convenient Synthesis of 2-Oxazolidinones and Tetrahydro-2H-1,3-oxazin-2-one

Ikuya Shibata,\* Kenji Nakamura, Akio Baba, and Haruo Matsuda Department of Applied Chemistry, Faculty of Engineering, Osaka University, 2-1 Yamadaoka, Suita, Osaka 565 (Received September 30, 1988)

Novel types of compounds, N-tributylstannyl-2-oxazolidinone (4a) and tetrahydro-2H-1,3-oxazin-2-one (4b), are formed from the adduct of  $(n-Bu_3Sn)_2O$  (1) with  $\omega$ -haloalkyl isocyanate (2), and the subsequent coupling reaction with alkyl halides gives a variety of N-substituted 2-oxazolidinones and tetrahydro-2oxazinones in a one-pot procedure. Both the cyclization and the coupling reaction proceed quantitatively in the presence of HMPA which enhances the reactivity of the Sn-heteroatom bond by coordination.

Many organotin reagents have become widely used in organic synthesis.1) Some important features of tin compounds are facile insertion of a heterocumulene such as an isocyanate toward a Sn-O bond,2 and a great affinity toward a sulfur<sup>3)</sup> or a halogen atom.<sup>4)</sup>

We have reported the preparation of five- and sixmembered heterocyclic compounds from the adduct of tributyltin  $\omega$ -haloalkoxides  $(n-Bu_3SnO(CH_2)_nX)$  with a heterocumulene<sup>5)</sup> (Scheme 1), in which a Lewis base plays an important role in the cyclization.

The coordination of a base toward the Sn atom increases the basicity of the adjacent heteroatom. which induces facile cyclization.

In connection with our interest in the use of this type of cyclization, we wish to report here a convenient method for the preparations of new compounds, N-

tributylstannyl-2-oxazolidinone (4a) and tetrahydro-2H-1.3-oxazin-2-one (4b), by the cyclication of the adduct (3), generated from  $(n-Bu_3Sn)_2O$  (1) and  $\omega$ haloalkyl isocyanate (2). Furthermore, the preparation of a variety of N-substituted 2-oxazolidinones and tetrahydro-2-oxazinones by the subsequent coupling reaction could be achieved under mild and neutral conditions (Scheme 2).6) 2-Oxazolidinones are an important class of heterocyclic compounds which have many biological uses. 7 In particular, the use of Nstannyl-2-oxazolidinones and tetrahydro-2-oxazinones in organic synthesis has never been reported, although the silvlated analogue is widely known as a silvlating agent.8)

$$Bu_3^nSnO(CH_2)_nX \xrightarrow{Y=C=Z} Bu_3^nSnY-C-O(CH_2)_nX \xrightarrow{Base} O \bigvee_{Z}^{(CH_2)_n} X$$

n: 2,3 X; halogen

Y=C=Z; heterocumulene

Scheme 1.

$$(Bu_{3}^{n}Sn)_{2}O + CI(CH_{2})_{n}NCO \longrightarrow \begin{pmatrix} CH_{2}\end{pmatrix}_{n}CI \\ 1 \qquad 2 \qquad O-SnBu_{3}^{n} \\ 3; b \qquad 3a,b \qquad 3a,b \qquad ABU_{3}^{n} + Bu_{3}SnCI \longrightarrow \begin{pmatrix} CH_{2}\end{pmatrix}_{n} \\ -Bu_{3}SnCI \longrightarrow \begin{pmatrix} CH_{2}\end{pmatrix}_{n} \\ -Bu_{3}SnCI \longrightarrow \begin{pmatrix} CH_{2}\end{pmatrix}_{n} \\ -Bu_{3}SnX \longrightarrow \begin{pmatrix} CH_{2}\\ -Bu_{3}\\ -Bu_{3} \longrightarrow \begin{pmatrix} CH_{2}\\ -Bu_{3}\\ -Bu_{3} \longrightarrow \begin{pmatrix} CH_{2}\\ -Bu_{3} \longrightarrow \begin{pmatrix} CH_{2}\\ -Bu_{3} \longrightarrow \begin{pmatrix} CH_{2}\\ -Bu_{3} \longrightarrow \begin{pmatrix} CH_{2}\\ -Bu_{3} \longrightarrow \begin{pmatrix}$$

Scheme 2.

## **Results and Discussion**

As shown in Scheme 2, N-tributylstannyl-2-oxazolidinone (4a) could be readily formed from the reaction of  $(n\text{-Bu}_3\text{Sn})_2\text{O}$  (1) and 2-chloroethyl isocyanate (2a). At first, the isocyanate inserts into the Sn-O bond spontaneously to give N-stannylcarbamate (3a). This intermediate was indicated by the appearance of an IR absorption band at  $1600~\text{cm}^{-1}$  in place of the NCO group (2280 cm<sup>-1</sup>). <sup>1</sup>H NMR showed the multiplet peaks between  $\delta$  3.30 and 3.70 due to methylene groups adjacent the nitrogen and chloro atoms. To trap the intermediate 3a, we tried the reaction with benzoyl chloride (Eq. 1). The N-acylation proceeded smoothly, and subsequent work-up gave the amide (5) in 98% yield, accompanied by decarboxylation.

$$3a \xrightarrow{PhCOCl} PhC-N-C-OSnBu_3 \xrightarrow{H^+} -CO_2$$

$$PhC-NH \sim Cl \qquad (1)$$

In the next stage, heating of 3a induces the cyclization to give 4a. IR band of the C=O group shifted to  $1680 \, \text{cm}^{-1}$ , and  $^1\text{H}$  NMR indicated the peaks of methylene groups adjacent to the nitrogen and chloro groups at  $\delta$  3.40—3.80 and 4.25—4.65, respectively. Moreover, GLC analysis showed the quantitative formation of the eliminated by-product, n-Bu<sub>3</sub>SnCl.

The compound 4a is so reactive that it was readily protonated to give 2-oxazolidinone (6) during work-up. Interestingly, the cyclization step is affected by additives (Table 1). Without any additives, the reaction hardly took place under the mild conditions (Entry 1), and elevated temperature (80 °C) was necessary to complete the cyclization (Entry 1, paren-

Table 1. Effect of Additive<sup>a)</sup>

$$(Bu_3Sn)_2O + C1 \sim NCO \xrightarrow{additive} O N-SnBu_3 \xrightarrow{MeOH} O NH$$
1 2a 4a 6

| Entry | Additive           | Yield of 6/%b) |
|-------|--------------------|----------------|
| 1     | _                  | tr (98)°)      |
| 2     | Et <sub>3</sub> N  | tr             |
| 3     | Bu <sub>3</sub> P  | tr             |
| 4     | DMF                | 73             |
| 5     | Bu <sub>3</sub> PO | 62             |
| 6     | HMPA               | 85             |

a) 1; 5 mmol, 2a; 5 mmol, additive 10 mmol. b) GLC yield. c) 80 °C, 1 h.

theses). Among the additives, the ones bearing a high coordinating ability toward a Sn atom are effective for the cyclization, and particularly, hexamethylphosphoric triamide (HMPA) was the most suitable one. This coordination increases the basicity of the heteroatom adjacent to the Sn atom. Thus, the cyclization to form 4a proceeded in high yield at room temperature in the presence of HMPA (Entry 6).

We next examined the coupling reactions of 4a with organic halides. It is possible to carry out a one pot cyclization-coupling reaction by simply adding an acid chloride to the cyclization mixture. In this way, 3-benzoyl-2-oxazolidinone (7) was formed directly in 86% yield (Table 2, Entry 1). The acylation proceeded cleanly, affording the corresponding 3-acyl-2-oxazolidinones<sup>10</sup> (7—13). The yields are good or excellent, no side reactions were observed. Moreover, purification of the products was simple, and the by-product, n-Bu<sub>3</sub>SnCl, could be removed completely by column chromatography using hexane as an eluent. With a half molar equivalent of oxalyl dichloride and adipoyl dichloride, bis(2-oxazolidinones) (14, 15) were obtained, respectively (Entries 8 and 9).

Other organic halides also gave good yields. Similar to acid chlorides, the reactions with phosphoryl chlorides and a sulfonyl chloride proceeded spontaneously, and gave 3-phosphoryl- and 3-sulfonyl-2-oxazolidinones (16—18), respectively (Entries 10—12). With activated organic halides such as  $\alpha$ -bromo carbonyl compounds, allyl and benzyl bromides, the coupling reactions proceeded at 80 °C to form 19—25 in high yields (Entries 13—19).

As in the cyclization reactions, HMPA was found to facilitate these coupling reactions, in contrast to the severe conditions previously reported for the alkylation of N-unsubstituted 2-oxazolidinone (6).<sup>7a)</sup> For example, reaction of 4a with cinnamyl bromide in the absence of HMPA at 80 °C for 1 h afforded only 31% of compound 22; in the presence of HMPA, 73% yield was obtained (Entry 16). Apparently, HMPA enhances the reactivity of the Sn-N bond by coordination.

We next tried to prepare six-membered compounds, tetrahydro-2*H*-1,3-oxazin-2-one derivatives, by the analogous reactions using 3-chloropropyl isocyanate (**2b**) (Eq. 2).

1 
$$\frac{1) \text{ CI} \sim \text{NCO (2b)}}{2) \text{ HMPA}} \circ N \text{SnBu}_3^n \xrightarrow{RX} \circ N \text{R}$$
4b (2)

These results are shown in Table 3. Compared to the case of five-membered ring formation, the cyclization to form **4b** required severe conditions (80 °C, 1 h) for completion. Subsequent reaction with organic halides gave a variety of tetrahydro-2-oxazinone derivatives (**26—30**) in good yields.

Table 2. Synthesis of 2-Oxazolidinones

|        |  | 4ā                       |   |           |
|--------|--|--------------------------|---|-----------|
| Entry  | RX   | Conditions <sup>a)</sup> | Product                                       | Yield/%ª) |
|        | RCOCl  |                          | 0_N_R   |           |
|        | RCOCI  |                          | JL.   |           |
| l      | R = Ph   | rt, 10 min               | 7   | 86        |
| 9      | Me   | rt, 10 min               | 8   | 100       |
| 2<br>3 | i-Pr   | rt, 10 min               | 9   | 100       |
| 4      | PhCH <sub>2</sub>  | rt, 10 min               | 10  | 94        |
| 7      |  |                          | 10  | 94        |
| 5      | PhOCH <sub>2</sub>                                       | rt, 10 min               | 11  | 99        |
| 6      | PhCH=CH  | rt, 10 min               | 12  | 88        |
| 7      | PhCH <sub>2</sub> O                                      | rt, 10 min               | 13  | 100       |
|        |  |                          | 0 N (CH <sub>2</sub> ) <sub>n</sub> N 0       |           |
|        | C1 (CH <sub>2</sub> ) (C1) (C1) (C1) (C1) (C1) (C1) (C1) |                          | Ö Ö Ö   |           |
| 8      | Ö Ö<br>n= 0  | rt, 10 min               | 14  | 78°)      |
| 9      | 4  | rt, 10 min               | 15  | 84°)      |
|        |  |                          | 0N-SO <sub>2</sub> Ph                         |           |
| 10     | DI CO CI   | 10                       | 16  | 00        |
| 10     | PhSO <sub>2</sub> Cl                                     | rt, 10 min               |   | 80        |
|        |  |                          | Ó√N-P(OPh) <sub>2</sub>                       |           |
| 11     | (PhO)2PC1<br>  <br> 0                                    | rt, 10 min               |   | 93        |
|        | Ö  |                          | OPh C   |           |
| 12     | PhopC12  | rt, 10 min               | 0_N-P-N_0                                     | 100°)     |
|        | PhOPC1 <sub>2</sub><br>II<br>O                           |                          | , , <u>, , , , , , , , , , , , , , , , , </u> |           |
|        |  |                          | 18  |           |
|        | R√Br   |                          | 0 N R   |           |
|        | R Br   |                          | <b>"</b> "                                    |           |
| 13     | R = Ph   | 80°C, 15 h               | 19  | 76        |
| 14     | EtO  | 80°C, 15 h               | <b></b>                                       | 78        |
|        | R <b>◇</b> ◇Br   |                          | O√N∕≪R  |           |
| 15     | R = H  | 80°C, 15 h               | 0 <b>21</b>                                   | 81        |
| 16     | Ph   | 80°C, 15 h               | 22  | 73        |
|        |  | 00 0, 10                 |   | 73        |
|        |  |                          | 0 N Ph  |           |
| 17     | Ph∕Br  | 80°C, 15 h               | 23  | 60        |
|        |  |                          |   |           |
|        |  |                          | 0 N O Ph                                      |           |
| 18     | Ph 0 0 0 1   | 80°C, 15 h               | <b>" 24</b>                                   | 90        |
|        | Ph 0 C1  |                          |   |           |
| 19     | ~~~i   | 80°C, 15 h               | 0 N 25  | 91        |
| 13     | · · · · · · · · · · · · · · · · · · ·                    | 00 G, 13 H               | 0 43  | 31        |
|        |  |                          |   |           |

a) At the coupling reaction. b) Isolated yield. c) 0.5 equiv of RX was used.

Next, the preparation of 4,5-disubstituted 2-oxazolidinones was accomplished as follows. It is reported that iodine isocyanate (INCO) derived from silver cyanate and iodine, adds to an olefin in a stereospecific manner to produce 2-iodoethyl isocyanates<sup>11)</sup> (Eq. 3).

$$\mathsf{AgCNO} \xrightarrow{\mathsf{I}_2} \mathsf{INCO} \xrightarrow{\mathsf{R-CH=CH-R}} \xrightarrow{\mathsf{R}} \overset{\mathsf{R}}{\mathsf{NCO}} ^{(3)}$$

Employing these substrates, the preparation of 4,5-dialkyl-2-oxazolidinones could be achieved by the

Table 3. Synthesis of Tetrahydro-2-oxazinones

1 + 2b 
$$\xrightarrow{\text{HMPA}} 0 \xrightarrow{\text{N-SnBu}_3} \xrightarrow{\text{RX}} 0 \xrightarrow{\text{N-R}} 0$$

| Entry | RX          | Condition <sup>a)</sup> | Product   | Yield/%b) |
|-------|-------------|-------------------------|-----------|-----------|
| 1     | Ph C1       | rt, 10 min              | 0 N Ph    | 100       |
| 2     | Ph Br       | 80°C, 15 h              | 0 Ph 0 27 | 72        |
| 3     | Ph 🌭 Br     | 80°C, 15 h              | 0 Ph 28   | 100       |
| 4     | Ph^Br       | 80°C, 15 h              | O Ph 29   | 87        |
| 5     | <b>~</b> ~₁ | 80°C, 15 h              | 0 N N 30  | 80        |

a) At the coupling reaction. b) Isolated yield.

combination with our method (Table 4). For example, as shown in Eq. 4, *cis*-2-butene afforded *threo*-iodo isocyanate which reacted with 1 to give the adduct. Next, the cyclization to form *N*-stannyl-2-oxazolidinone proceeded via backside attack of the oxygen on the terminal alkyl iodide. As a result, *cis*-4,5-dimethyl-2-oxazolidinone (31) was prepared in 88% overall yield (Entry 1).

Similarly, the diastereoisomer, trans-dimethyl-2-oxazolidinone (32), was obtained from trans-2-butene (Entry 2).<sup>13)</sup> Thus, 4,5-disubstituted 2-oxazolidinones could be easily prepared from olefins in a stereospecific manner by a one-pot reaction. The additional results obtained using other olefins are shown in Table 4.

In summary, we have demonstrated an effective and direct method for the preparation of 2-oxazolidinones and tetrahydro-2*H*-1,3-oxazin-2-ones by using organotin reagents. Our method provides several advantages in terms of mild and neutral conditions, high yields of products and operational convenience.

Table 4. Synthesis of 2-Oxazolidi-

| nones from Olefins |            |                             |           |  |  |
|--------------------|------------|-----------------------------|-----------|--|--|
| Entry              | Olefin     | Product                     | Yield/%ª) |  |  |
| 1                  | $\bigvee$  | 0 N Ph<br>0 0 31            | 88        |  |  |
| 2                  | <b>—</b>   | )                           | 98        |  |  |
| 3                  |            | 0 N Ph<br>0 0 33            | 66        |  |  |
| 4                  | $\bigcirc$ | 0<br>0<br>0<br>0<br>0<br>34 | 50        |  |  |
| 5                  | Ph         | 0 N Ph<br>0 0 35            | 52        |  |  |
| a) Isolated yield  |            |                             |           |  |  |

a) Isolated yield.

## Experimental

Melting points were obtained by using a Yanaco Micromelting point apparatus and are uncorrected. IR spectra were recorded on a Hitachi 260-30 spectrometer using KBr pelletes or KRS-5 cells. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were performed on Hitachi R-90HS spectrometer. Mass spectra were recorded on a Hitachi RMU-6E instrument. Analytical GLC was performed on Shimadzu GC-3B with TCD using a 2 m×3 mm glass column packed with Silicone OV-1 on uniport HP (5%, 60—80 mesh). Silica-gel column chromatography was used (Wakogel C-200).

**Benzoyl-2-chloroethylamide** (5). Under dry nitrogen, 0.53 g (5 mmol) of 2-chloroethyl isocyanate (2a) was added to 2.98 g (5 mmol) of bis(tributyltin) oxide (1). This process to form 3a proceeded spontaneously. Next, to this solution was added 1.40 g (10 mmol) of PhCOCl at room temperature. After stirring for 10 min, the reaction mixture was chromatographed by silica-gel column, where decarboxylation occurred. Elusion with benzene gave the compound 5 which was purified by recrystallization from benzene–hexane (1:1): Mp 86–87 °C; MS m/z 183.5 (M+); IR 1630 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ=3.50–4.00 (m, 4H, CH<sub>2</sub>), 6.35–6.85 (br, 1H, NH<sub>1</sub>), 7.20–7.90 (m, 5H, phenyl). Calcd for C<sub>9</sub>H<sub>10</sub>NOCl: C, 58.86; H, 5.45; H, 7.63%. Found: C, 58.55; H, 5.41; N, 7.97%.

3-Benzoyl-2-oxazolidinone (7) (General Procedure for the Preparation of 2-Oxazolidinones): To the liquid of 3a formed in situ was added 1.80 g (10 mmol) of HMPA. This mixture was stirred at 40 °C for 1 h, giving N-stannyl-2-oxazolidinone (4a) which was converted to 6 upon MeOH. Without HMPA, the cyclization required severe conditions (80 °C, 1 h). To this mixture containing 4a was added 0.70 g (5 mmol) of benzoyl chloride at room temperature. After

10 min, the reaction mixture was chromatographed, and purified by recrystallization from benzene-hexane (1:1): mp 166-168 °C (lit, <sup>14</sup>) mp 168 °C); IR (KBr) 1670, 1770 cm<sup>-1</sup>; MS m/z 191 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =3.80-4.35 (m, 2H, CH<sub>2</sub>O), 4.35-4.90 (m, 2H, CH<sub>2</sub>N), 7.10-8.00 (m, 5H, phenyl). The following compounds **8–18** were obtained in a similar manner by using suitable organic halides.

**3-Acetyl-2-oxazolidinone (8):** Mp 85 °C (lit, <sup>7a)</sup> mp 87—87.5 °C); IR (KBr) 1690, 1780 cm<sup>-1</sup>; MS m/z 129 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =2.55 (s, 3H, C $\underline{\text{H}}_3$ ), 4.05 (t, J=8 Hz, 2H, C $\underline{\text{H}}_2$ N), 4.45 (t, J=8 Hz, 2H, C $\underline{\text{H}}_2$ O).

**3-Isopropionyl-2-oxazolidinone** (9): Wax, purified by Kugelrohr distillation at  $100\,^{\circ}$  C/ $10^{-3}$  mmHg (1 mmHg= 133.322 Pa); IR (neat) 1700, 1780 cm<sup>-1</sup>; MS m/z 157 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =1.20 (d, J=7 Hz, 6H, C $\underline{\text{H}}_3$ ), 3.60—3.90 (m, 1H, Me<sub>2</sub>C $\underline{\text{H}}$ ), 4.05 (dd, J=7 and 9 Hz, 2H, C $\underline{\text{H}}_2$ N), 4.45 (dd, J=7 and 9 Hz, 2H, C $\underline{\text{H}}_2$ O); Calcd for C<sub>7</sub>H<sub>11</sub>NO<sub>3</sub>: C, 53.50; H, 7.01; N, 8.92%. Found: C, 53.38; H, 6.98; N, 9.04%.

**3-Phenylacetyl-2-oxazolidinone** (10): Mp 55—56 °C; IR (KBr) 1690, 1760 cm<sup>-1</sup>; MS m/z 205 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =3.95 (t, J=8 Hz, 2H, C $\underline{H}_2$ N), 4.10—4.60 (m, 4H, C $\underline{H}_2$ O and PhC $\underline{H}_2$ ), 6.90—7.60 (m, 5H, phenyl); Calcd for C<sub>11</sub>H<sub>11</sub>NO<sub>3</sub>: C, 64.39; H, 5.37; N, 6.83%. Found: C, 64.18; H, 5.07; N, 6.88%.

**3-Phenoxyacetyl-2-oxazolidinone** (11): Mp 93—95 °C; IR (KBr) 1720, 1790 cm<sup>-1</sup>; MS m/z 221 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =4.05 (t, J=8 Hz, 2H, C $\underline{H}_2$ N), 4.50 (t, J=8 Hz, 2H, C $\underline{H}_2$ O), 5.25 (s, 2H, PhOCH<sub>2</sub>), 6.80—7.50 (m, 5H, phenyl); Calcd for C<sub>11</sub>H<sub>11</sub>O<sub>4</sub>N: C, 59.72; H, 4.98; N, 6.33%. Found: C, 59.66; H, 4.93; N, 6.32%.

**3-Cinnamoyl-2-oxazolidinone** (12): Mp 148—149 °C; IR (KBr) 1620, 1770 cm<sup>-1</sup>; MS m/z 217 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =4.00—4.30 (m, 2H, CH<sub>2</sub>N), 4.30—4.60 (m, 2H, CH<sub>2</sub>O), 7.20—8.00 (m, 7H, phenyl and CH<sub>2</sub>=CH-Ph); Calcd for C<sub>12</sub>H<sub>11</sub>NO<sub>3</sub>: C, 66.36; H, 5.07; N, 6.45%. Found: C, 66.24; H, 4.95; N, 6.55%.

**3-Benzyloxycarbonyl-2-oxazolidinone (13):** Mp 98—100 °C (lit, <sup>7a)</sup> mp 101—102 °C); IR (KBr) 1800 cm<sup>-1</sup>; MS m/z 221 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =3.80—4.20 (m, 2H, C $\underline{\text{H}}_2$ N), 4.30—4.60 (m, 2H, C $\underline{\text{H}}_2$ O), 5.30 (s, 2H, PhC $\underline{\text{H}}_2$ ), 7.20—7.70 (m, 5H, phenyl).

**3,3'-Oxalylbis(2-oxazolidinone)** (14). The compounds **14** and **15** were prepared by using 0.5 equiv of acyl chlorides: mp 234—236 °C (lit, <sup>7a)</sup> mp 238 °C); IR (KBr) 1710, 1790 cm<sup>-1</sup>; MS m/z 228 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =4.00—4.50 (m, 4H, C $\underline{H}_2$ N), 4.50—4.90 (m, 4H, C $\underline{H}_2$ O).

3,3'-Adipoylbis(2-oxazolidinone) (15): Mp 138 °C (lit, $^{7a}$ ) mp 138 °C); IR (KBr) 1710, 1780 cm $^{-1}$ ; MS m/z 284 (M+);  $^{1}$ H NMR (CDCl<sub>3</sub>)  $\delta$ =1.70—2.00 (m, 4H, COCH<sub>2</sub>CH<sub>2</sub>), 2.96 (t, J=7 Hz, 4H, COCH<sub>2</sub>), 4.01 (t, J=7 Hz, 4H, CH<sub>2</sub>N), 4.32 (t, J=7 Hz, 4H, CH<sub>2</sub>O).

**3-Phenylsulfonyl-2-oxazolidinone** (16): Mp 136—138 °C (lit, <sup>15)</sup> mp 130—135 °C); IR (KBr) 1770 cm<sup>-1</sup>; MS m/z 227 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =3.90—4.30 (m, 2H, C $\underline{\text{H}}_2$ N), 4.30—4.60 (m, 2H, C $\underline{\text{H}}_2$ O), 7.40—8.20 (m, 5H, phenyl).

**3-Diphenyloxyphosphinyl-2-oxazolidinone** (17): Mp 94—95 °C; IR (KBr) 1770 cm<sup>-1</sup>; MS m/z 319 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =3.84 (t, J=8 Hz, 2H, CH<sub>2</sub>N), 4.33 (t, J=8 Hz, 2H, CH<sub>2</sub>O), 7.20—7.40 (m, 10H, phenyl); Calcd for C<sub>15</sub>H<sub>14</sub>NO<sub>5</sub>P: C, 56.43; H, 4.39; N, 4.39%. Found: C, 56.57; H, 4.44; N, 4.49%.

3,3'-Phenoxyphosphinylidenebis(2-oxazolidinone) (18). This compound was prepared by using 0.5 equiv of phenyl

phosphorodichloridate: mp 158 °C; IR (KBr) 1770 cm<sup>-1</sup>; MS m/z 312 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ=4.00—4.30 (m, 4H, CH<sub>2</sub>N), 4.40—4.70 (m, 4H, CH<sub>2</sub>O), 7.20—7.50 (m, 5H, phenyl); Calcd for C<sub>12</sub>H<sub>13</sub>N<sub>2</sub>O<sub>6</sub>P: C, 46.15; H, 4.17; N, 8.97%. Found: C, 45.99; H, 4.20; N, 8.89%.

**3-Phenacyl-2-oxazolidinone (19).** The coupling reaction of **4a** formed in situ with phenacyl bromide was performed at 80 °C for 15 h. The following compounds **20—25** were obtained in a similar manner using suitable organic halides: mp 105—108 °C; IR (KBr) 1700, 1750 cm<sup>-1</sup>; MS m/z 205 (M+): <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ=3.73 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>N), 4.45 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>O), 4.70 (s, 2H, PhCOCH<sub>2</sub>), 7.20—8.30 (m, 5H, phenyl); Calcd for C<sub>11</sub>H<sub>11</sub>-NO<sub>3</sub>; C, 64.38; H, 5.40; N, 6.83%. Found: C, 64.54; H, 5.29; N, 6.66%.

**3-Ethoxycarbonylmethyl-2-oxazolidinone (20):** Wax, purified by Kugelrohr distillation at  $100\,^{\circ}\text{C}/10^{-3}\,\text{mmHg}$ ; IR (neat) 1740, 1760 cm<sup>-1</sup>; MS m/z 173 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =1.29 (t, J=7 Hz, 3H, C $\underline{\text{H}}_3$ ), 3.70 (dd, J=7 and 9 Hz, 2H, C $\underline{\text{H}}_2$ N), 4.02 (s, 2H, C $\underline{\text{H}}_2$ COOEt), 4.10—4.50 (m, 4H, C $\underline{\text{H}}_2$ Me and C $\underline{\text{H}}_2$ O); Calcd for C<sub>7</sub>H<sub>11</sub>NO<sub>4</sub>: C, 48.55; H, 6.36; N, 8.09%. Found: C, 48.75; H, 6.47; N, 7.99%.

**3-Allyl-2-oxazolidinone (21):** Wax, purified by Kugelrohr distillation at  $100 \,^{\circ}\text{C}/10^{-3} \,\text{mmHg}$  (lit,  $^{16)}$  bp  $123-125 \,^{\circ}\text{C}/0.7 \,\text{mmHg}$ ); IR (neat)  $1750 \,\text{cm}^{-1}$ ; MS  $m/z \, 127 \,(\text{M}^{+})$ ;  $^{1}\text{H} \,\text{NMR}$  (CDCl<sub>3</sub>)  $\delta$ =3.51 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>N), 3.90 (d, J=6 Hz, 2H, NCH<sub>2</sub>CH=CH<sub>2</sub>), 4.34 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>O), 5.10—6.10 (m, 3H, CH=CH<sub>2</sub>).

**3-Cinnamyl-2-oxazolidinone** (22): Mp 69 °C; IR (KB<sub>I</sub>) 1740 cm<sup>-1</sup>; MS m/z 203 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ=3.62 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>N), 4.08 (d, J=7 Hz, 2H, CH<sub>2</sub>CH= CHPh), 4.38 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>O), 6.00—6.70 (m, 2H, CH=CHPh), 7.20—7.60 (m, 5H, phenyl); Calcd for C<sub>12</sub>H<sub>13</sub>NO<sub>2</sub>: C, 70.94; H, 6.40; N, 6.90%. Found: C, 70.78; H, 6.46; N, 6.66%.

**3-Benzyl-2-oxazolidinone (23):** Mp 77—78 °C (lit,<sup>7a)</sup> mp 79—80 °C); IR (KBr) 1750 cm<sup>-1</sup>; MS m/z 177 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =3.40 (dd, J=7 and 9 Hz, 2H, C $\underline{\text{H}}_2$ N), 4.28 (dd, J=7 and 9 Hz, 2H, C $\underline{\text{H}}_2$ O), 4.41 (s, 2H, PhC $\underline{\text{H}}_2$ ), 7.40—7.60 (m, 5H, phenyl).

**3-Benzyloxymethyl-2-oxazolidinone** (24): Wax, purified by Kugelrohr distillation at  $150 \,^{\circ}$  C/ $10^{-3}$  mmHg; IR (neat)  $1760 \,^{\circ}$  cm<sup>-1</sup>; MS m/z 207 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =3.62 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>N), 4.25 (dd, J=7 and 9 Hz, CH<sub>2</sub>O), 4.56 (s, 2H, CH<sub>2</sub>Ph), 4.83 (s, 2H, NCH<sub>2</sub>O), 7.30—7.50 (m, 5H, phenyl); Calcd for C<sub>11</sub>H<sub>13</sub>NO<sub>3</sub>:C, 63.77; H, 6.28; N, 6.76%. Found: C, 63.37; H, 6.41; N, 6.86%.

3-Hexyl-2-oxazolidinone (25): Wax, purified by Kugelrohr distillation at  $100 \,^{\circ}\text{C}/10^{-3}$  mmHg (lit, <sup>7a)</sup> bp  $176 \,^{\circ}\text{C}/1$  mmHg); IR (neat)  $1740 \,^{\circ}\text{cm}^{-1}$ ; MS m/z  $171 \,^{\circ}\text{M}^{+}$ ); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =0.80—1.90 (m, 11H, NCH<sub>2</sub>C<sub>5</sub>H<sub>11</sub>), 3.28 (t, J=7 Hz, 2H, CH<sub>2</sub>C<sub>5</sub>H<sub>11</sub>), 3.58 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>N), 4.33 (dd, J=7 and 9 Hz, 2H, CH<sub>2</sub>O).

3-Benzoyltetrahydro-2*H*-1,3-oxazin-2-one (26) (General Procedure for the Preparation of Tetrahydro-2*H*-1,3-oxazin-2-ones). To a neat liquid of 2.98 g (5 mmol) of 1 was added 0.60 g (5 mmol) of 3-chloropropyl isocyanate (2b). After 10 min, 1.80 g (10 mmol) of HMPA was added to this reaction mixture, and heated at 80 °C for 1 h, where *N*-stannyltetrahydro-2*H*-1,3-oxazin-2-one (4b) was formed. Next, 0.70 g (5 mmol) of benzoyl chloride was added to this solution, and stirred at room temperature for 10 min. The product 26 was obtained by column chromatography: mp

87 °C; IR (KBr) 1680, 1720 cm<sup>-1</sup>; MS m/z 205 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =2.00—2.40 (m, 2H, C $\underline{\text{H}}_2$ ), 3.91 (t, J=6 Hz, 2H, C $\underline{\text{H}}_2$ N), 4.43 (t, J=5 Hz, 2H, C $\underline{\text{H}}_2$ O), 7.20—8.20 (m, 5H, phenyl); Calcd for C<sub>11</sub>H<sub>11</sub>NO<sub>3</sub>: C, 64.39; H, 5.37; N, 6.83%. Found: C, 64.11; H, 5.43; N, 6.84%.

**3-Phenacyltetrahydro-2***H***-1,3-oxazin-2-one (27).** The coupling reaction of **4b** with organic halides to form **27—30** was performed at 80 °C for 15 h: mp 93 °C; IR (KBr) 1680, 1700 cm<sup>-1</sup>; MS m/z 219 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =2.16 (m, 2H, CH<sub>2</sub>), 3.43 (t, J=6 Hz, 2H, CH<sub>2</sub>N), 4.41 (t, J=5 Hz, CH<sub>2</sub>O), 4.82 (s, 2H, PhCOCH<sub>2</sub>), 7.20—8.00 (m, 5H, phenyl); Calcd for C<sub>12</sub>H<sub>13</sub>NO<sub>3</sub>: C, 65.75; H, 5.94; N, 6.39%. Found: C, 65.55; H, 5.91; N, 6.45%.

**3-Cinnamyltetrahydro-2***H***-1,3-oxazin-2-one** (28): Wax, purified by Kugelrohr distillation at 150 °C/10<sup>-3</sup> mmHg; IR (neat) 1690 cm<sup>-1</sup>; MS m/z 217 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =1.90—2.00 (m, 2H, CH<sub>2</sub>), 3.32 (t, J=6 Hz, 2H, CH<sub>2</sub>N), 4.11 (d, J=6 Hz, 2H, CH<sub>2</sub>CH=CHPh), 4.26 (t, J=5 Hz, 2H, CH<sub>2</sub>O), 6.00—6.70 (m, 2H, CH=CHPh), 7.10—7.60 (m, 5H, phenyl); Calcd for C<sub>13</sub>H<sub>15</sub>NO<sub>2</sub>: C, 71.89; H, 6.91; N, 6.45%. Found: C, 71.65; H, 7.08; N, 6.34%.

**3-Benzyltetrahydro-2***H***-1,3-oxazin-2-one (29):** Mp 35 °C; IR (KBr) 1690 cm<sup>-1</sup>; MS m/z 191 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =2.03 (m, 2H, C $\underline{\text{H}}_2$ ), 3.21 (t, J=6 Hz, 2H, C $\underline{\text{H}}_2$ N), 4.25 (t, J=5 Hz, 2H, C $\underline{\text{H}}_2$ O), 4.55 (s, 2H, PhC $\underline{\text{H}}_2$ ), 7.20—7.30 (m, 5H, phenyl); Calcd for C<sub>11</sub>H<sub>13</sub>NO<sub>2</sub>: C, 69.11; H, 6.81; N, 7.33%. Found: C, 68.93; H, 6.83; N, 7.34%.

**3-Hexyltetrahydro-2***H***-1,3-oxazin-2-one (30):** Wax, purified by Kugelrohr distillation at  $100 \,^{\circ}\text{C}/10^{-3}$  mmHg; IR (neat)  $1690 \,^{\circ}\text{cm}^{-1}$ ; MS m/z 185 (M+);  $^{1}\text{H}$  NMR (CDCl<sub>3</sub>)  $\delta$ =0.80—1.90 (m, 11H, NCH<sub>2</sub>C<sub>5</sub>H<sub>11</sub>), 1.90—2.20 (m, 2H, CH<sub>2</sub>), 3.30 (m, 4H, CH<sub>2</sub>N and NCH<sub>2</sub>C<sub>5</sub>H<sub>11</sub>), 4.23 (t, *J*=5 Hz, 2H, CH<sub>2</sub>O); Calcd for C<sub>10</sub>H<sub>19</sub>NO<sub>2</sub>: C, 64.86; H, 10.27; N, 7.57%. Found: C, 64.97; H, 9.96; N, 7.68%.

3-Benzoyl-cis-4,5-dimethyl-2-oxazolidinone (31). cis-2-Butene (0.56 g, 10 mmol) was added to AgNCO (1.50 g, 10 mmol) in 5 ml of ether at -10 °C. The mixture was stirred vigorously while I2 (2.54 g, 10 mmol) was added. Stirring was continued for 3 h. The slurry was filtered, and concentration of the solution gave a mobile brown liquid. To this liquid was added 1 (5.96 g, 10 mmol) at room temperature. After 10 min, HMPA (3.60 g, 20 mmol) was added to this reaction mixture, and heated at 40 °C for 1 h. Moreover, the addition of benzoyl chloride (1.40 g, 10 mmol) at room temperature and stirring for 10 min gave 31. Mp 78 °C; IR (KBr) 1670, 1780 cm<sup>-1</sup>; MS m/z 219 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =1.39 (d, J=6.5 Hz, 3H, NCHCH<sub>3</sub>), 1.46 (d, J=6.5 Hz, 3H, OCHC $\underline{H}_3$ ), 4.50-5.00 (m, homodecoupling J=8 Hz, 2H, CH), 7.20-7.90 (m, 5H, phenyl); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ =13.2, 14.8, 54.9, 74.7, 127.9, 128.8, 132.2, 133.6, 153.0, 169.9; Calcd for C<sub>12</sub>H<sub>13</sub>NO<sub>3</sub>: C, 65.75; H, 5.94; N, 6.39%. Found: C, 65.46; H, 5.94; N, 6.42%.

**3-Benzoyl-***trans***-4,5-dimethyl-2-oxazolidinone** (32): Mp 47 °C; IR (KBr) 1680, 1780 cm<sup>-1</sup>; MS m/z 219 (M+); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =1.46 (d, J=6 Hz, 3H, NCHC $\underline{H}$ <sub>3</sub>), 1.53 (d, J=6 Hz, 3H, OCHC $\underline{H}$ <sub>3</sub>), 4.00—4.50 (m, homodecoupling J=6 Hz, 2H, C $\underline{H}$ ), 7.20—7.90 (m, 5H, phenyl); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ =17.8, 19.3, 58.2, 77.9, 128.0, 128.4, 129.2, 132.5, 162.7, 186.1; Calcd for C<sub>12</sub>H<sub>13</sub>NO<sub>3</sub>: C, 65.75; H, 5.94; N, 6.39%. Found: C, 65.91; H, 6.03; N, 6.23%.

**3-Benzoylhexahydro-2***H***-cyclopentoxazol-2-one (33):** Mp 127 °C; IR (KBr) 1670, 1760 cm<sup>-1</sup>; MS m/z 231 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =1.80—2.30 (m, 6H, (C<u>H</u><sub>2</sub>)<sub>3</sub>), 4.80—5.20

(m, 2H, C<u>H</u>), 7.20—7.80 (m, 5H, phenyl); Calcd for  $C_{13}H_{13}NO_3$ : C, 67.53; H, 5.63; N, 6.06%. Found: C, 67.20; H, 5.55; N, 6.08%.

3-Benzoylhexahydrobenzoxazol-2(3*H*)-one (34): Mp 112 °C (lit,<sup>17)</sup> mp 114—115 °C); IR (KBr) 1670, 1690 cm<sup>-1</sup>; MS m/z 245 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =1.20—2.50 (m, 8H, (C $\underline{\text{H}}_2$ )<sub>4</sub>), 4.20—4.50 (m, 1H, C $\underline{\text{H}}$ N), 4.50—4.80 (m, 1H, C $\underline{\text{H}}$ O), 7.20—7.80 (m, 5H, phenyl).

**3-Benzoyl-4-phenyl-2-oxazolidinone** (35): Mp 167 °C; IR (KBr) 1680, 1790 cm<sup>-1</sup>; MS m/z 267 (M<sup>+</sup>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ =4.32 (dd, J=7 and 9 Hz, 1H, C $\underline{\text{H}}_2\text{O}$ ), 4.78 (t, 1H, J=9 Hz, C $\underline{\text{H}}_2\text{O}$ ), 5.64 (dd, J=7 and 9 Hz, 1H, C $\underline{\text{H}}\text{N}$ ), 7.20—7.80 (m, 10H, phenyl); Calcd for C<sub>16</sub>H<sub>13</sub>NO<sub>3</sub>: C, 71.91; H, 4.87; N, 5.24%. Found: C, 71.80; H, 4.92; N, 5.19%.

This work was supported by the Grant-in-Aid for Scientific Research No. 62550613 from the Ministry of Education, Science and Culture.

## References

- 1) For reviews see: A. K. Sawyer, "Organotin Compounds," Marcel Dekker New York (1971); E. Negishi "Organometallics in Organic Synthesis," Wiley, New York (1980), Vol. 1, p. 394; A. G. Davies and P. J. Smith, "Comprehensive Organometallic Chemistry," Pergamon, Oxford (1982), Vol. 2, p. 519; M. Pereyre, P. J. Quintard, and A. Rahm, "Tin in Organic Synthesis," Butterworth, London (1987).
- 2) A. J. Bloodworth and A. G. Davies, *J. Chem. Soc.*, **1965**, 5238; A. J. Bloodworth, A. G. Davies, and S. C. Vasishtha, *J. Chem. Soc. C*, **1967**, 1309.
- 3) S. Sakai, Y. Asai, Y. Kiyohara, K. Itoh, and Y. Ishii, Organomet. Chem. Synth., 1, 45 (1970); S. Sakai, Y. Fujimura, and Y. Ishii, J. Organomet. Chem., 50, 113 (1973); S. Sakai, H. Niimi, Y. Kobayashi, and Y. Ishii, Bull. Chem. Soc. Jpn., 50, 3271 (1977).
- 4) B. Delmond, J. C. Pommier, and J. Valade, J. Organomet. Chem., 35, 91 (1972); B. Delmond, J. C. Pommier, and J. Valade, J. Organomet. Chem., 47, 337 (1973).
- 5) A. Baba, H. Kishiki, I. Shibata, and H. Matsuda, Organometallics, 4, 1329 (1985); A. Baba, I. Shibata, H. Kashiwagi, and H. Matsuda, Bull. Chem. Soc. Jpn., 59, 341 (1986); I. Shibata, A. Baba, and H. Matsuda, Bull. Chem. Soc. Jpn., 59, 4000 (1986).
- 6) A part of study on this cyclization-coupling reaction has been published as a communication: I. Shibata, A. Baba, and H. Matsuda, J. Chem. Soc., Chem. Commun., 1986, 1703.
- 7) For reviews see: a) M. E. Dyen, and D. Swern, *Chem. Rev.*, **67**, 197 (1967); b) V. A. Pankrotov, T. M. Frenkel, and A. M. Fainleib, *Usp. Khim. Zh. (Russ. Ed)*, **52**, 1018 (1983).
- 8) For example: C. Palomo, Synthesis, 1981, 809; A. Arrieta and C. Palomo, *ibid.*, 1982, 1050; J. M. Aizpurua, C. Palomo, and A. L. Palomo, Can. J. Chem., 62, 336 (1984). 336 (1984).
- 9) HMPA acts as an efficient ligand to organotin compounds bearing an electron withdrawing group. T. F. Bolles and R. S. Drago, J. Am. Chem. Soc., 88, 5730 (1966).
- 10) N-Acyl-2-oxazolidinones are known versatile key reagents in organic synthesis. For example; a) D. A. Evans, Aldrichimica Acta, 15, 23 (1982). b) K. Narasaka, M. Inoue, and N. Okada, Chem. Lett., 1986, 1109.
- 11) A. Hassner, R. P. Hoblitt, C. Heathcock, J. E. Kropp,

- and M. Lorber, J. Am. Chem. Soc., 92, 1326 (1970).
- 12) Sn-O bond has been reported to cause the backside attack toward the alkyl halides.
- 13) The diastereoisomers, 31 and 32, were confirmed by the coupling constant between the ring protons obtained by decoupling of the methyl protons. It has been reported that the value of cis isomer is larger than that of trans isomer: B. D. Harris, K. L. Bhat, and M. Joullie, *Tetrahedron Lett.*, 28,
- 2837 (1987); D. J. Kempf, J. Org. Chem., 51, 3921, (1986).
- 14) O. Tsuge, T. Itoh, and K. Sakai, *Nippon Kagaku Zasshi*, **90**, 1031 (1969).
- 15) J. W. McFarland and R. W. Houser, *J. Org. Chem.*, **33**, 340 (1968).
- 16) J. S. Pierce, J. Am. Chem. Soc., 50, 241 (1928).
- 17) D. Ben-Ishai, J. Am. Chem. Soc., 78, 4962 (1956).