Facile Synthesis of Trifluoromethyl-substituted Enynes: Remarkable Reactivity and Stereoselectivity of Tributyl(3,3,3-trifluoropropynyl)stannane in Carbostannylation of Alkynes

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Carbostannylation of alkynes with tributyl(3,3,3-trifluoropropynyl)stannane is found to proceed at *room temperature* in a syn-manner, giving rise to the corresponding CF₃-substituted enynes as *a single stereoisomer* in good yields. Both terminal and, CF₃- or RO₂C-substituted internal alkynes are applicable to the addition reaction. Synthetic applications of the adduct are also demonstrated.

Since carbon-carbon triple bonds can undergo various kinds of transformations, 3,3,3-trifluoropropynyl-containing compounds¹ serve as versatile building blocks for the preparation of trifluoromethylated molecules, to which much attention has been paid in the fields of pharmaceuticals, agrochemicals, and organic materials.² Hence, generation and reactions of 3,3,3-trifluoropropynyllithium, -magnesium, -silyl, and -zinc reagents have been studied well.³ Meanwhile, transition metal-catalyzed carbostannylation of alkynes with alkynylstannanes has emerged as a powerful synthetic tool because an alkynyl and stannyl groups are simultaneously incorporated into alkynes in a synmanner to afford stereodefined alkenyltins, which can further be transformed into a variety of alkenes via stereospecific carbon-carbon and carbon-heteroatom bond formations with the aid of the tin functionality.⁴ Therefore, carbostannylation of alkynes with 3,3,3-trifluoropropynylstannanes is highly attractive for the preparation of diverse trifluoromethylated enynes. We report here palladium-catalyzed carbostannylation of alkynes with tributyl(3,3,3-trifluoropropynyl)stannane (1),⁵ which proceeds smoothly at room temperature to give CF₃-substituted envnes **3** as a single stereoisomer in good yields (Eq 1).



In the course of our synthetic study utilizing 1, we attempted cross-coupling reaction of 1 with iodobenzene to prepare 3,3,3-trifluoro-1-phenylpropyne. Thus, a toluene solution of 1 and io-dobenzene in the presence of Pd₂(dba)₃ (1 mol %) and *t*-Bu₃P (2 mol %) was stirred at room temperature.⁶ The isolated product unexpectedly turned out to be alkenylstannane 3 (R¹ = CF₃, R² = Ph) as a single stereoisomer in 40% yield, which was considered to form via carbostannylation of the cross-coupled product with 1. This result prompted us to investigate generality of the carbostannylation with 1. The results are summarized in Table 1. Under the same conditions, aryl acetylenes 2a–2d were carbostannylated with 1 at room temperature to give 3a–3d as a

stereochemically pure form in good yields (Entries 1-4),⁷ respectively. Z-Stereochemistry of 3b determined by NOE data of the vinyl hydrogen and protonolysis of the C-Sn bond⁸ indicated that the reaction proceeded via syn-addition and the Bu₃Sn group that was bulkier than a CF₃CC group added to the less hindered sp carbon. Methyl propiolate (2e) and N.N-dimethyl propiolamide (2f) also reacted with 1 to afford stereochemically pure 3e and 3f, whose stereochemistries were deduced by protonolysis, with the opposite regioselectivity (Entries 5 and 6).⁸ Since these stereochemical outcome is consistent with typical carbostannylation, the present reaction is considered to proceed via the well-accepted reaction mechanism,⁴ which involves oxidative addition of 1 to the Pd(0) complex and successive insertion of an alkyne to the Pd-C bond followed by reductive elimination, resulting in production of 3 and regeneration of the Pd(0) complex.

Furthermore, the reaction of internal alkynes, which was usually difficult to achieve in the typical alkynylstannylation chemistry,⁴ was scrutinized. The results are shown in entries 7–17 in Table 1. To our delight, the addition to CF₃-substituted aryl acetylenes 2g-2n occurred also at room temperature under the same conditions to give 3g-3n as a sole product (Entries 7–14). Various functional groups were tolerant under the conditions. Phenyl- and methyl-substituted propiolate derivatives 2o and 2p as well as dimethyl acetylenic dicarboxylate (2q) reacted

Table 1. Carbostannylation of alkynes 2 with 1^a

Entry	2	\mathbb{R}^1	\mathbb{R}^2	3	Yield/%
1	2a	Н	C_6H_5	3a	79
2	2b	Н	<i>p</i> -MeO-C ₆ H ₄	3b	73
3	2c	Н	$p-C_6H_5-C_6H_4$	3c	82
4	2d	Н	<i>p</i> -CF ₃ -C ₆ H ₄	3d	79
5	2e	CO_2Me	Н	3e	78
6	2f	CONMe ₂	Н	3f	58
7	2g	CF ₃	Ph	3g	76
8	2h	CF ₃	p-Me-C ₆ H ₄	3h	52
9	2i	CF ₃	p-MeO-C ₆ H ₄	3i	67
10	2ј	CF ₃	p-Cl-C ₆ H ₄	3j	62
11	2k	CF ₃	p-Ac-C ₆ H ₄	3k	68
12	21	CF ₃	p-EtO ₂ C-C ₆ H ₄	31	57
13	2m	CF ₃	$p-O_2N-C_6H_4$	3m	58
14	2n	CF ₃	p-CF ₃ -C ₆ H ₄	3n	67
15	20	CO_2Et	Ph	30	87
16	2p	CO_2Et	Me	3p	77
17	2q	CO_2Me	CO ₂ Me	3q	71

^aReagents and conditions: **1** (0.6 mmol), **2** (0.9 mmol), Pd₂-(dba)₃ ($6.0 \,\mu$ mol), *t*-Bu₃P (12 μ mol), toluene (1.6 mL), and rt.



Scheme 1. Synthetic application of 3a.

in a Michael fashion to afford **30–3q** as a single stereoisomer in good yields, respectively (Entries 15-17). Since diphenylacetylene and 4-octyne did not react with 1 at all, the presence of such an electron-withdrawing group as CF3 and CO2R appears to be essential for the realization.

The fact that all the reactions took place at room temperature definitely shows remarkably higher reactivity of 1 than those of common alkynyltins which require heating at 50 or 90 °C to effect the carbostannylation reaction.4 Strong electron-withdrawing effect by a CF3 group may induce acceleration of the oxidative addition step to undergo the reaction at room temperature, which lead to perfect stereoselectivity.⁹

Alkenyltin functionality of 3 can be readily utilized for further transformation.¹⁰ Representative examples with 3a are demonstrated in Scheme 1. Pd-catalyzed cross-coupling reaction with aryl and alkenyl iodides gave CF₃-substituted envne 4 and dienyne 5, while iodinated enyne 6 was prepared in good yield by treatment with I₂ in THF.

In summary, we have demonstrated that carbostannylation of alkynes with tributyl(3,3,3-trifluoropropynyl)stannane constitutes facile and stereoselective synthesis of 1-tributylstannyl-5,5,5-trifluoropent-1-en-3-ynes. Both terminal and internal alkynes are applicable to the reaction. Synthetic application of the CF₃-substituted envnes is in progress in our laboratory.

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References and Notes

a) M. G. Barlow, S. Tajammal, and A. E. Tipping, Chem. Commun., 1989, 1637. b) N. Yoneda, S. Matsuoka, N. Miyaura, T. Fukuhara, and A. Suzuki, Bull. Chem. Soc. Jpn., 63, 2124 (1990). c) M. G. Barlow, S. Tajammal, and A. E. Tipping, J. Fluorine Chem., 63, 139 (1993). d) M. N. Bobrovnikov, E. S. Turbanova, and A. A. Petrov, Russ. J. Org. Chem., 29, 1445 (1993). e) M. G. Barlow, N. N. E. Suliman, and A. E. Tipping, J. Fluorine Chem., 70, 109 (1995). f) T. Yamazaki, K. Mizutani, and T. Kitazume, J. Org. Chem., 60, 6046 (1995). g) A. K. Brisdon, I. R. Crossley, K. R. Flower, R. G. Pritchard, and J. E. Warren, Angew. Chem., Int. Ed., 42, 2399 (2003). h) A. K. Brisdon, I. R. Crossley, R. G. Pritchard, G. Sadiq, and J. E. Warren, Organometallics, 22, 5534 (2003). i) I. H. Jeong, S. L. Jeon, and B. T. Kim, Tetrahedron Lett., 44, 7213 (2003). j) J. Chae, T. Konno, T. Ishihara, and H. Yamanaka, Chem. Lett., 33, 314 (2004). k) T. Konno, J. Chae, T. Ishihara, and H. Yamanaka, Tetrahedron, 60, 11695 (2004). 1) T. Konno, J. Chae, T. Tanaka, T. Ishihara, and H. Yamanaka, Chem. Commun., 2004, 690, m) T. Konno, T. Daitoh, A. Noiri, J. Chae, T. Ishihara, and H. Yamanaka, Org. Lett., 6, 933 (2004). n) T. Konno, T. Takehana, J. Chae, T. Ishihara, and H. Yamanaka, J. Org. Chem., 69, 2188 (2004).

- a) J. T. Welch, Tetrahedron, 43, 3123 (1987). b) J. F. Liebman, 2 A. Greenberg, and J. W. R. Dolbier, "Fluorine-containing Molecules: Structure, Reactivity, Synthesis, and Applications," VCH, New York (1988). c) J.-P. Begue and D. Bonnet-Delpon, Tetrahedron, 47, 3207 (1991). d) M. A. McClinton and D. A. McClinton, Tetrahedron, 48, 6555 (1992). e) R. E. Banks, B. E. Smart, and J. C. Tatlow, "Organofluorine Chemistry. Principles and Commercial Applications," Plenum Press, New York (1994). f) M. Hudlicky and A. E. Pavlath, "Chemistry of Organic Fluorine Compounds II A Critical Review," American Chemical Society-ACS Monograph 187, Washington, DC (1995). g) T. Hiyama, "Organofluorine Compounds-Chemistry and Applications," Springer, Berlin (2000). h) M. Shimizu and T. Hiyama, Angew. Chem., Int. Ed., 44, 214 (2005).
- 3 a) A. L. Henne and M. Nager, J. Am. Chem. Soc., 74, 650 (1952). b) F. G. Drakesmith, O. J. Stewart, and P. Tarrant, J. Org. Chem., 33, 280 (1968). c) J. E. Bunch and C. L. Bumgardner, J. Fluorine Chem., 36, 313 (1987). d) A. R. Katritzky, M. Qi, and A. P. Wells, J. Fluorine Chem., 80, 145 (1996). e) A. K. Brisdon and I. R. Crossley, Chem. Commun., 2002, 2420. f) T. Konno, J. Chae, M. Kanda, G. Nagai, K. Tamura, T. Ishihara, and H. Yamanaka, Tetrahedron, 59, 7571 (2003).
- a) E. Shirakawa, H. Yoshida, T. Kurahashi, Y. Nakao, and T. Hiyama, J. Am. Chem. Soc., 120, 2975 (1998). b) E. Shirakawa and T. Hiyama, J. Organomet. Chem., 576, 169 (1999). c) E. Shirakawa, H. Yoshida, Y. Nakao, and T. Hiyama, J. Am. Chem. Soc., 121, 4290 (1999). d) E. Shirakawa, K. Yamasaki, H. Yoshida, and T. Hiyama, J. Am. Chem. Soc., 121, 10221 (1999). e) E. Shirakawa, H. Yoshida, Y. Nakao, and T. Hiyama, Org. Lett., 2, 2209 (2000). f) H. Yoshida, E. Shirakawa, T. Kurahashi, Y. Nakao, and T. Hiyama, Organometallics, 19, 5671 (2000). g) E. Shirakawa, Y. Nakao, H. Yoshida, and T. Hiyama, J. Am. Chem. Soc., 122, 9030 (2000). h) E. Shirakawa, Y. Nakao, and T. Hiyama, Chem. Commun., 2001, 263. i) H. Yoshida, E. Shirakawa, Y. Nakao, Y. Honda, and T. Hiyama, Bull. Chem. Soc. Jpn., 74, 637 (2001). j) H. Yoshida, Y. Honda, E. Shirakawa, and T. Hiyama, Chem. Commun., 2001, 1880. k) E. Shirakawa and T. Hiyama, J. Organomet. Chem., 653, 114 (2002). 1) E. Shirakawa and T. Hiyama, Bull. Chem. Soc. Jpn., 75, 1435 (2002). m) E. Shirakawa, Y. Yamamoto, Y. Nakao, S. Oda, T. Tsuchimoto, and T. Hiyama, Angew. Chem., Int. Ed., 43, 3448 (2004).
- 5 T. Hanamoto, Y. Hakoshima, and M. Egashira, Tetrahedron Lett., 45, 7573 (2004).
- 6 No carbostannylated product formed when N-[2-(diphenylphosphino)benzylidene]cyclohexylamine, which was essential for conventional carbostannylation of alkynes, or PPh3 was employed as a phosphine ligand with $Pd_2(dba)_3$ complex.
- 7 Reaction of alkyl acetylenes with 1 failed to give the carbostannylation products under the conditions.
- 8 See Supporting Information.
- 9 In marked contrast, no reaction took place with tributylpropynylstannane under the same conditions, suggesting that the fluorine atoms played a crucial role in the reaction of 1.
- a) M. Pereyre, J.-P. Quintard, and A. Rahm, "Tin in Organic 10 Synthesis," Butterworth, London (1987). b) A. G. Davies, "Organotin Chemistry," VCH, Weinheim (1997).