

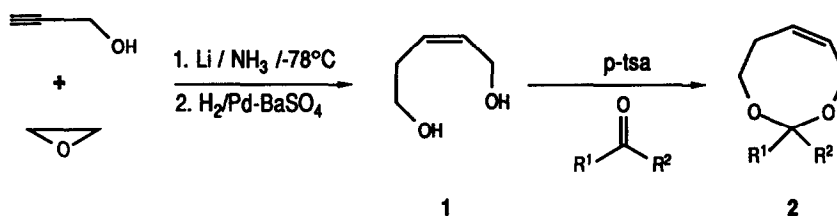
SYNTHESIS OF 2,3-SUBSTITUTED TETRAHYDROPYRANS BY RE-ARRANGEMENT OF 5,6-DIHYDRO-4H-1,3-DIOXOCINS

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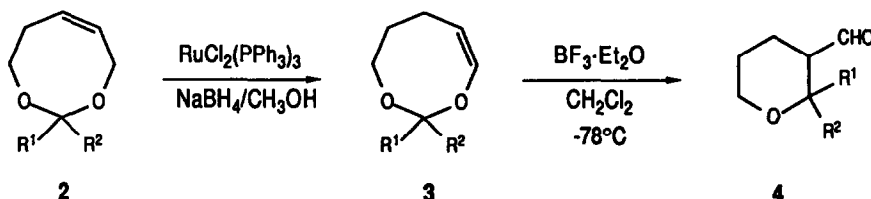
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Abstract - The synthesis of 5,8-dihydro-4H-1,3-dioxocins **2** and 5,6-dihydro-4H-1,3-dioxocins **3** is described. 2-Substituted tetrahydropyran-3-carbaldehydes are obtained by acid-catalyzed rearrangement of **3**.

Tetrahydrofurans and tetrahydropyrans are often found as structural subunits in naturally occurring polyether antibiotics.^{1,2} One strategy for the stereoselective synthesis of 2,3- and 2,3,5-substituted tetrahydrofurans is the rearrangement of 4,5-dihydro-1,3-dioxepins.^{3,4} This procedure generally involves acetalization, double-bond isomerization and acid-catalyzed rearrangement. Analogous to this reaction sequence we have now investigated the synthesis of tetrahydropyrans by rearrangement of 5,6-dihydro-4H-1,3-dioxocins **3**.



Even though eight-membered cyclic acetals of type **2** and **3** are not well established in the literature, 5,8-dihydro-4H-1,3-dioxocins **2** are easily obtained in high yields ($\text{R}^1 = \text{H}$, $\text{R}^2 = \text{CH}_3$, C_6H_5 , $n\text{-C}_3\text{H}_7$; $\text{R}^1 = \text{CH}_3$, $\text{R}^2 = \text{CH}_3$; 68 - 80%) by acetalization of aldehydes or ketones with **1**⁵ by usual procedures.



The double-bond isomerization of the allylic precursors **2** is achieved with hydridic transition metal catalysts, whereas the isomerization with KO-*tert*-Bu only gives sluggish results. When

the eight-membered cyclic vinyl acetals⁴ are reacted with Lewis acids at -78°C , tetrahydropyran-3-carbaldehydes **4** with substituents in the 2-position are formed. Surprisingly, this rearrangement proceeds with high stereoselectivity, although no preferential conformations could be detected by first inspections of the NMR spectra of **3**. Probably the trans isomers are formed predominantly. Detailed stereochemical investigations are now in progress.

As a typical example, the synthesis and the transformation of 5,8-dihydro-2-propyl-4H-1,3-dioxocin are described in the following experimental details.

5,8-Dihydro-2-propyl-4H-1,3-dioxocin (**2**, $\text{R}^1 = \text{H}$, $\text{R}^2 = n\text{-C}_3\text{H}_7$): Obtained from *n*-butyraldehyde and **1** by azeotropic removal of water in the presence of *p*-tsa (71%); bp. $81^{\circ}\text{C}/12\text{ Torr}$; $^1\text{H NMR}$ (300 MHz, CDCl_3): $\delta = 5.81$ (dt, $J = 11, 7.7, 1.2\text{ Hz}$, 1 H, $\text{CH-CH}_2\text{-CH}_2\text{-O}$), 5.63 (dt, $J = 11, 5\text{ Hz}$, 1 H, $\text{CH-CH}_2\text{-O}$), 4.63 (t, $J = 5.7\text{ Hz}$, 1 H, O-CH-O), 4.32 (dd, $J = 15.2, 5.5\text{ Hz}$, 1 H, $\text{CH-CH}_2\text{-O}$), 4.06 (dd, $J = 15.2, 4.2\text{ Hz}$, 1 H, $\text{CH-CH}_2\text{-O}$), 3.88 (ddd, $J = 11.8, 6.4, 4.7\text{ Hz}$, 1 H, $\text{CH}_2\text{-CH}_2\text{-O}$), 3.46 (ddd, $J = 11.8, 8.2, 4.0\text{ Hz}$, 1 H, $\text{CH}_2\text{-CH}_2\text{-O}$), 2.59–2.33 (m, 2 H, $\text{O-CH}_2\text{-CH}_2\text{-CH}$), 1.62 (td, $J = 7.8, 5.7\text{ Hz}$, 2 H, $\text{CH}_2\text{-CH}_2\text{-CH}_3$), 1.38 (sext, $J = 7.6\text{ Hz}$, 2 H, $\text{CH}_2\text{-CH}_3$), 0.93 (t, $J = 7.4\text{ Hz}$, 3 H, CH_3); $^{13}\text{C NMR}$ (300 MHz, CDCl_3): $\delta = 129.79, 127.46$ ($\text{O-CH}_2\text{-CH}$, $\text{OCH}_2\text{-CH}_2\text{-CH}$), 103.24 (O-CH-O), 66.27, 63.40 ($\text{O-CH}_2\text{-CH}$, $\text{O-CH}_2\text{-CH}_2$), 36.13 ($\text{CH}_2\text{-CH}_2\text{-CH}_3$), 28.49 ($\text{CH}_2\text{-CH}_2\text{O}$), 18.13 ($\text{CH}_2\text{-CH}_3$), 13.98 (CH_3).

5,6-Dihydro-2-propyl-4H-1,3-dioxocin (**3**, $\text{R}^1 = \text{H}$, $\text{R}^2 = n\text{-C}_3\text{H}_7$): Obtained by isomerization of **2** with $\text{RuCl}_2(\text{PPh}_3)_3$ (0.1 mol%) and NaBH_4 (0.2 mol%) in CH_3OH (83%); $48^{\circ}\text{C}/3.5\text{ Torr}$; $^1\text{H NMR}$ (300 MHz, CDCl_3): $\delta = 6.16$ (d, $J = 5.7\text{ Hz}$, 1 H, O-CH=CH), 5.27 (td, $J = 7, 5.7\text{ Hz}$, 1 H, O-CH=CH), 4.68 (t, $J = 5.9\text{ Hz}$, O-CH-O), 4.02 (dt, $J = 12.1, 4.7\text{ Hz}$, 1 H, $\text{CH}_2\text{-O}$), 3.64 (ddd, $J = 12.1, 8.7, 3.8\text{ Hz}$, 1 H, $\text{CH}_2\text{-O}$), 2.29 (dddd, $J = 14, 10, 8, 3.5\text{ Hz}$, 1 H, CH=CH-CH_2), 2.11 (m, 1 H, CH=CH-CH_2), 1.9–1.5 (m, 4 H, $\text{CH}_2\text{-CH}_2\text{-CH}_3$, $\text{O-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}$), 1.42 (sext, $J = 7.1\text{ Hz}$, 2 H, $\text{CH}_2\text{-CH}_3$), 0.93 (t, $J = 7\text{ Hz}$, 3 H, CH_3); $^{13}\text{C NMR}$ (300 MHz, CDCl_3): $\delta = 141.37$ (CH=CHO), 120.25 (CH=CHO), 105.16 (OCHO), 69.46 (CH_2O), 35.90 ($\text{CH}_2\text{-CH}_2\text{-CH}_3$), 29.86 ($\text{CH}_2\text{-CH=CH}$), 22.82 ($\text{O-CH}_2\text{-CH}_2\text{-CH}_2$), 18.22 (CH_2CH_3), 13.84 (CH_3).

2-Propyl-tetrahydrofuran-3-carbaldehyde (**4**, $\text{R}^1 = \text{H}$, $\text{R}^2 = n\text{-C}_3\text{H}_7$): Obtained by rearrangement of **3** with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in CH_2Cl_2 at -78°C and aqueous workup (74%); $43\text{--}45^{\circ}\text{C}/1\text{ Torr}$; $^1\text{H NMR}$ (300 MHz, CDCl_3): $\delta = 9.63$ (d, $J = 2.9\text{ Hz}$, 1 H, CHO), 3.98 (dq, $J = 11, 2.5\text{ Hz}$, 1 H, O-CH_2), 3.49 (m, O-CH-CH), 3.38 (m, 1 H, O-CH_2), 2.33 (m, 1 H, CH-CHO), 1.96 (m, 1 H, $\text{O-CH}_2\text{-CH}_2\text{-CH}_2$), 1.66–1.59 (m, 3 H, $\text{O-CH}_2\text{-CH}_2\text{-CH}_2$, $\text{O-CH}_2\text{-CH}_2$), 1.55–1.47 (m, 4 H, $\text{CH}_2\text{-CH}_2\text{-CH}_3$, $\text{CH}_2\text{-CH}_3$), 0.91 (t, $J = 7\text{ Hz}$, 3 H, CH_3); $^{13}\text{C NMR}$ (300 MHz, CDCl_3): $\delta = 202.74$ (CHO), 76.35 (O-CH), 67.53 (O-CH_2), 54.00 (O-CH-CH), 36.71 ($\text{CH}_2\text{-CH}_2\text{-CH}_3$), 24.67 ($\text{O-CH}_2\text{-CH}_2$), 23.97 ($\text{O-CH}_2\text{-CH}_2\text{-CH}_2$), 18.56 ($\text{CH}_2\text{-CH}_3$), 14.02 (CH_3).

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5. **1** was prepared by a procedure analogous to that described for hex-3-in-1,6-diol.⁶
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