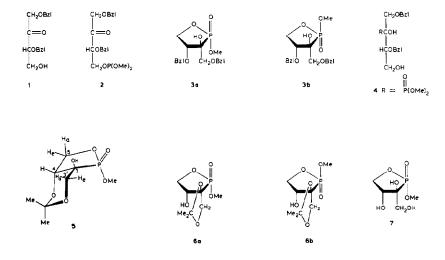
Synthesis of carbohydrates having phosphorus in the anomeric position

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Analogues of carbohydrates with phosphorus as the ring heteroatom have been described^{1,2} and we now report on analogues in which phosphorus replaces the anomeric carbon. Our interest in this area was stimulated by the transformation³ of a model γ -ketophosphite into diastereomeric 1,2-oxaphospholan-3-ols and the corresponding α , γ -dihydroxyphosphonate. We anticipated that this approach could be applied to the synthesis of C-2 branched-chain sugars.

1,3-Di-O-benzyl-D-glycero-tetrulose⁴ (1) was transformed quantitatively under standard conditions [(MeO)₂PCl/NEt₃] into the mixed phosphite 2, which was immediately hydrolysed³. ³¹P-N.m.r. spectroscopy of the crude reaction mixture revealed (2R,3R,4R)- (3a) and (2S,3R,4R)-4-O-benzyl-3-C-(benzyloxymethyl)-2-methoxy-2-oxo-1,2-oxaphospholane-3,4-diol (3b), and 1,3-di-O-benzyl-2-C-(dimethoxyphosphinyl)-Derythritol (4) together with a trace of 1,3-di-O-benzyl-2-C-(dimethoxyphosphinyl)-Dthreitol. The C-3 epimers of 3a and 3b were not detected and the cyclisation of 2 appears to be stereospecific. Column chromatography of the crude product gave 23% of a 2:1 mixture of 3a and 3b, and 22% of 4 {oil, $[\alpha]_D^{20} - 13^\circ$ (c 2.4, chloroform)}.



Treatment of 4 with 10% triethylamine in benzene at 20° afforded 73% of a 2:1 mixture of 3a and 3b, and attempts to separate these products failed. Hydrogenolysis of the 2:1 mixture of 3a and 3b gave the corresponding triols, which were immediately treated with 2,2-dimethoxypropane-toluene-*p*-sulfonic acid monohydrate⁵. From the resulting mixture, (2R, 3R, 4R)-3-C-(hydroxymethyl)-4,3¹ O-isopropylidene-2-methoxy-2-oxo-1,2-oxaphospholane-3,4-diol (5, 8.1%) and (2R, 3R, 4R)- (6a, 22.6%) and (2S, 3R, 4R)-3-C-(hydroxymethyl)-3,3¹-O-isopropylidene-2-methoxy-2-oxo-1,2-oxaphospholane-3,4-diol (6b, 10.2%) were isolated.

The ¹³C resonances at 99.15 (CMe₂) and 19.97 and 28.17 p.p.m. (CMe₂) for 5 strongly support the formation of the 1,3-dioxane ring⁶. On the other hand, the ¹³C resonances at 112.43 and 111.91 p.p.m. for CMe₂ in 6a and 6b, respectively, together with those for Me₂C at 25.48 and 26.29 p.p.m. for 6a, and 25.38 and 26.29 p.p.m. for 6b, are characteristic for non-fused 1,3-dioxolane rings⁶. Furthermore, both spiro derivatives had ³J_{P,CMe}, values of 4.4 Hz. Detailed configurational and conformational relationships for 5 were assigned on the basis of ¹H-n.m.r. data (Table I). Each ring hydrogen in 5 is three bonds away from phosphorus, and the J_{HP} values reflect the stereochemistry⁷. The large ${}^{3}J_{4,P}$ value of 24.5 Hz is observed only for *cis*-fused 1,2-oxaphospholane and 1,3-dioxane rings. From the values of ${}^{3}J_{4,5a}$ and ${}^{3}J_{4,5e}$, and ${}^{3}J_{5aP}$ and ${}^{3}J_{5eP}$ it is evident that the 1,2-oxaphospholane ring in 5 exists in the E_4 conformation. Furthermore, the ${}^{3}J_{P,3^{1}a}$ and ${}^{3}J_{P,3^{1}e}$ values clearly show that the 1,3-dioxane ring is in the chair conformation. In the alternative 4,3¹-O-isopropylidene derivative, having the opposite configuration at position 3, no coupling between H-4 and phosphorus is expected. The above conclusion reflects the ${}^{1}T_{3}$ conformation that the 1,2-oxaphospholane ring is forced to adopt in the trans-fused system with the 1,3-dioxane ring. Isopropylidenation studies proved the 2:1 mixture of 3a and 3b obtained by hydrolysis of 2 to be identical with that prepared by intramolecular cyclisation of 4.

The relative configurations of HO-3 and P=O in 3a and 3b were deduced from the chemical shifts of the ³¹P resonances, namely, 41.7 p.p.m. for the isomer (3a) with *cis* groups and 39.5 p.p.m. for the isomer (3b) with *trans* groups.

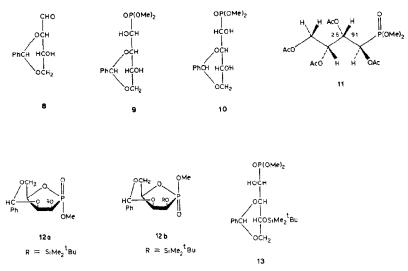
Treatment of **6a** in chloroform with toluene-*p*-sulfonic acid monohydrate at 40° for 1 h and then with methanol gave (2R, 3R, 4R)-3-*C*-(hydroxymethyl)-2-methoxy-2-oxo-1,2-oxaphospholane-3,4-diol (7, 72%), m.p. 145–146°, $[\alpha]_D^{22}$ –14° (c 1, methanol). ¹³C-N.m.r. data (CD₃OD): δ 54.78 (d, *J* 6.6 Hz, Me), 63.20 (d, *J* 8.1 Hz, C-3¹), 73.99 (d, *J* 8.1 Hz, C-5), 76.17 (d, *J* 136.8 Hz, C-3), and 76.41 (d, *J* 14.7 Hz, C-4).

Me ₂ C	MeOP	H-3 ¹ a	<i>H-3</i> ¹ e	H-4	H-5a	H-5e
		4.00 dd J _{31a,31e} 13.5 J _{31a,P} 23.7	J ₃₁ _{e,P} 12.9	4.25 dd J _{4,5a} 3.2 J _{4,P} 24.5	4.55 ddd J _{5a,5e} 10.3 J _{5a,P} 3.2	4.08 dd J5e,P 18.2

TABLE I

^a Chemical shifts on the δ scale, J in Hz.

Triethylamine-catalysed addition of dimethyl phosphite to 2,4-O-benzylidene-D-threose⁸ (8) afforded (1*R*)-2,4-O-benzylidene-1-C-(dimethoxyphosphinyl)-D-threitol (9) (³¹P 27.1 p.p.m.) and (1*S*)-2,4-O-benzylidene-1-C-(dimethoxyphosphinyl)-D-threitol (10) (³¹P 27.5 p.p.m.) in the ratio 9:1. Column chromatography of the mixture on silica gel gave 9 (14%), m.p. 121–122°, $[\alpha]_D^{18}$ +13° (c 2.1, chloroform). Hydrogenolysis of 9 followed by acetylation afforded the tetra-acetate 11, m.p. 54–55°, $[\alpha]_D^{18}$ +22° (c 1.1, chloroform). N.m.r. data: $J_{2,3}$ 2.6 and $J_{1,2}$ 9.1 Hz. The former J value accords with the *threo* configuration around the C-2–C-3 bond, and the latter shows⁹ that the configuration at C-1 is *R*.



Treatment of 9 with 2.2 equiv. of tert-butyldimethylsilyl chloride in N,N-dimethylformamide in the presence of 4.4 equiv. of imidazole¹⁰ afforded (2R,3R,4S,5R)-(12a) and (2S,3R,4S,5R)-4,5¹-O-benzylidene-3-O-(tert-butyldimethylsilyl)-5-C-(hydroxymethyl)-2-methoxy-2-oxo-1,2-oxaphospholane-3,4-diol (12b), and (1R)-2,4-Obenzylidene-3-O-(tert-butyldimethylsilyl)-1-C-(dimethoxyphosphinyl)-D-threitol (13). On the basis of ³¹P-n.m.r. data, the ratios of 12a (³¹P 35.27 p.p.m.), 12b (³¹P 34.00 p.p.m.), and 13 (³¹P 27.36 p.p.m.) were 1:1:1. Column chromatography of the mixture on silica gel gave 13 as a colorless syrup, $[\alpha]_D^{20} - 5^\circ$ (c 1.3, chloroform). For the p-nitrobenzoate of 13 {m.p. 106-107°, $[\alpha]_D^{20} - 25^\circ$ (c 1.5, chloroform)}, H-1 resonated at δ 5.72 and a ²J₁p value of 4.8 Hz was observed. These data proved that 13 was formed by preferential silylation of the ring hydroxyl group in 9. After the removal of 13, fractional crystallisation gave 12a {m.p. 159-160°, $[\alpha]_D^{20} + 49.5^\circ$ (c 1.6, chloroform)} and 12b {m.p. 189-190°, $[\alpha]_D^{20} + 30^\circ$ (c 1.5, chloroform)}. The assignment of configuration at phosphorus was based on the deshielding effect of the phosphoryl group⁷, which causes the resonance of H-3 in 12b to fall 0.16 p.p.m. below that in 12a. The ²J₃p values for 12a and 12b were similar (-8.6 and -9.9 Hz, respectively) and could not be used for configurational assignments⁷.

The method for cyclisation of 9 should be useful for the preparation of other

analogues of pentofuranosides having phosphorus in the anomeric position. The intramolecular cyclisation of 4 corresponds to the hemiacetal formation—glycosidation sequence of simple monosaccharides.

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REFERENCES

- 1 R. L. Whistler and W. Chi-Cheng, J. Org. Chem., 33 (1968) 4455-4458.
- 2 H. Yamamoto, Y. Nakamura, H. Kawamoto, S. Inokawa, M. Yamashita, M.-A. Armour, and T. T. Nakashima, *Carbohydr. Res.*, 102 (1982) 185-196, H. Yamamoto, K. Yamamoto, S. Inokawa, M. Yamashita, M.-A. Armour, and T. T. Nakashima, *J. Org. Chem.*, 48 (1983) 435-440, and references therein.
- 3 A. E. Wróblewski, Tetrahedron, 39 (1983) 1809-1816.
- 4 A. E. Wróblewski, unpublished data.
- 5 A. Lipták, J. Imre, and P. Nánási, Carbohydr. Res., 92 (1981) 154-156.
- 6 J. G. Buchanan, M. E. Chacón-Fuertes, A. R. Edgar, S. J. Moorhouse, D. I. Rawson, and R. H. Wightman, *Tetrahedron Lett.*, (1980) 1793-1796.
- 7 L. D. Quin, The Heterocyclic Chemistry of Phosphorus, Wiley-Interscience, New York, 1981, pp. 319-359.
- 8 A. S. Perlin, Methods Carbohydr. Chem., 1 (1962) 68-70.
- 9 R. E. Moore, G. Bartolini, J. Barchi, A. A. Bothner-By, J. Dadok, and J. Ford, J. Am. Chem. Soc., 104 (1982) 3776-3779.
- 10 E. J. Corey and A. Venkateswarlu, J. Am. Chem. Soc., 94 (1972) 6190-6191.