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One-Step Synthesis of 5-Mercaptopyrazoles from 2-Methylene-1,3-dithietanes

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2-Methylene-1,3-dithietanes bearing electron-withdrawing groups on the sp² carbon afford 5-mercaptopyrazole derivatives by the reaction with methylhydrazine.

The 5-mercaptopyrazoles 3 are of value as a key intermediate of potent herbicide. The hitherto reported synthetic methods of them can be classified in two groups. One involves the transformation of 5-hydroxypyrazoles into the objective compounds via chlorination and subsequent thiolation, and the other consists of their preparation from 5-aminopyrazoles by the Sandmeyer reaction. However, both methods are not satisfactory with respect to the total yield and versatility.

In a series of studies on four-membered heterocycles, the authors have found that 2-(acetylethoxycarbonyl)methylene-1,3-dithietane (1a) on treatment with amines underwent facile ring-opening to yield thioamides, and the latter on treatment with bromine were cyclized to 3-isothiazolones.⁴

The authors report here a new one-pot preparation of 3,4-disubstituted 5-mercapto-1-methylpyrazoles from the dithietanes 1a, b by treating with methylhydrazine. On refluxing with excess methylhydrazine in acetonitrile, 2-methylene-1,3-dithietanes 1a, b were directly converted into the 5-mercaptopyrazoles 3a, b in almost quantitative yield. However, attempts to condense the dithietanes 1a, b with excess phenylhydrazine were unsuccessful. On methylation, 5-mercaptopyrazoles 3 were converted to 5-methylthiopyrazoles⁵ 4.

$$\begin{array}{c} R \longrightarrow S \\ EtO_2C \longrightarrow S \end{array} + \begin{array}{c} H \longrightarrow NH_2 \end{array} \begin{array}{c} CH_3CN \\ rt, 0.5h \\ Me \end{array} \begin{array}{c} O \longrightarrow CO_2Et \\ S \longrightarrow NH_2 \\ Me \end{array}$$

1-4	R
a h	CH ₃ Ph
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The compounds prepared were characterized by IR- and 1 H-NMR spectra which were consistent with the proposed structure. 2-Methylene-1,3-dithietanes **1a**, **b** were prepared by the reaction of ethyl α -acylacetates with carbon disulfide in the presence of a base followed by cyclization with dibromomethane in 52–74% overall yields. $^{6.7}$

The melting points were determined on Yanagimoto micro melting point apparatus and are uncorrected. IR spectra were recorded with a Shimadzu IR-420 spectrophotometer. ¹H-NMR spectra were determined with a Varian EM-390 (90 MHz) spectrometer with TMS as internal standard.

4-Ethoxycarbonyl-5-mercapto-1,3-dimethylpyrazole (3a):

To a solution of 2-methylene-1,3-dithietane (1a, 8.0 g, 36 mmol) in CH₃CN (50 mL) is added dropwise methylhydrazine (3.5 g, 71 mmol) at r.t. over 30 min. The reaction mixture is refluxed for 3 h, and the contents are concentrated under reduced pressure. After addition of H₂O (50 mL) to the residual mixture the contents are acidified with 1 N HCl to pH 2 and extracted with EtOAc (50 mL). The organic layer is dried over anhydrous Na₂SO₄, evaporated to dryness, and the residue is chromatographed on a silica gel column with hexane/EtOAc (2:1) as eluent to give 3a; yield: 3.7 g (99%); mp 64-66°C.

IR (Nuiol): $v = 1670 \text{ cm}^{-1}$.

¹H-NMR (CDCl₃): δ = 1.35 (t, 3 H, J = 7 Hz); 2.37 (s, 3 H); 3.67 (s, 3 H); 4.30 (q, 2 H, J = 7 Hz); 6.55 (br s, 1 H).

4-Ethoxycarbonyl-5-mercapto-1-methyl-3-phenylpyrazole (3 b):

Yield: 2.3 g (98%) from 1a (2.0 g, 9 mmol), brown oil.

IR (liquid): $v = 1670 \, \text{cm}^{-1}$.

¹H-NMR (CDCl₃): δ = 1.15 (t, 3 H, J = 7 Hz); 3.85 (s, 3 H); 4.20 (q, 2 H, J = 7 Hz); 7.20 - 7.60 (m, 5 H).

4-Ethoxycarbonyl-1,3-dimethyl-5-methylthiopyrazole (4a):

To a solution of 5-mercaptopyrazole (3a, 0.5 g, 2.5 mmol) in MeOH (50 mL) is added dropwise MeONa (2.5 mmol) and then MeI (0.4 g, 2.5 mmol) at r.t.. The reaction mixture is stirred at r.t. for 1 h, and concentrated under reduced pressure. After addition of $\rm H_2O$ (30 mL), the solution is acidified with 1 N HCl to pH2 and extracted with $\rm CH_2Cl_2$ (50 mL). The organic layer is dried over anhydrous Na₂SO₄ and evaporated to dryness to give 4a; yield: 0.5 g (92%); mp 33–34°C.

C₉H₁₄N₂O₂S calc. C 50.45 H 6.59 N 13.07 (214.3) found 50.44 6.70 13.11

IR (Nujol): v = 1710, 1510, 1250 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.40 (t, 3 H, J = 7 Hz); 2.45 (s, 6 H); 3.92 (s. 3 II); 4.35 (q, 2 H, J = 7 Hz).

4-Ethoxycarbonyl-1-methyl-5-methylthio-3-phenylpyrazole (4b):

Yield: 1.9 g (79%) from 3b (2.3 g, 8.8 mmol), yellow oil.

C₁₄H₁₆N₂O₂S calc. C 60.85 H 5.84 N 10.14 (276.4) found 60.88 5.84 10.15

IR (liquid): $v = 1710 \text{ cm}^{-1}$.

¹H-NMR (CDCl₃): δ = 1.18 (t, 3 H, J = 7 Hz); 2.50 (s, 3 H); 4.00 (s, 3 H); 4.24 (q, 2 H, J = 7 Hz); 7.20–7.60 (m, 5 H).

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- Aoki, I., Kuragano, T., Okajima, N., Okada, Y. Eur. Patent. Appl. 220695 (1987), Takeda Chemical Industries, Ltd.; C. A. 1987, 107, 176057.
- (2) Suzuki, F., Iwasawa, Y., Sato, T., Ikai, T., Oguchi, T. Jpn. Patent 84219281 (1984), Nissan Chemical Industries, Ltd.; C. A. 1985, 102, 220905.
- (3) Jacob, L., in: *Heterocyclic Compounds*, Elderfield, R.C. (ed.), Vol. 5, John Wiley and Sons. London, 1957, p. 141.
- (4) Okajima, N., Okada, Y. *Synthesis*, submitted for preparation.
- (5) The reason why the formation of the 3-mercapto isomer was excluded is as follows. 4-Ethoxycarbonyl-1,3-dimethyl-5-methyl-thiopyrazole which was synthesized from 5-amino-4-ethoxycarbonyl-1,3-dimethylpyrazole (4a) was prepared alternatively by methylthioation of 5-chloro-4-ethoxycarbonyl-1,3-dimethylpyrazole (5) by the Sandmeyer reaction. 1,3 5-Aminopyrazole 5 was prepared from methylhydrazine and ethyl 2-cyano-3-ethoxycrotonate. 8 On this reaction the formation of 3-aminopyrazole was excluded. 9,10
- (6) Taninaka, K. Dissertation, University of Kyoto, Kyoto, Japan, 1981.
- (7) See: Borrmann, D., in: Houben-Weyl, 4th ed., Vol. 7/4, Georg Thieme Verlag, Stuttgart, 1968, p. 404.
- (8) US Patent 2824121 (1958); C. A. 1958, 52, 11909.
- (9) Schmidt, P. Helv. Chim. Acta 1959, 42, 349.
- (10) Elnagoli, M. Z. Naturforsch. Teil B 1976, 31, 795.