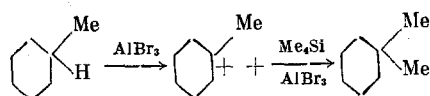


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The reaction of electrophilic substitution of hydrogen on the tertiary C atom in saturated hydrocarbons by a  $\text{CH}_3$  group from TMS under the influence of  $\text{AlBr}_3$  is presented. Thus, methylcyclohexane reacts with TMS under the influence of  $\text{AlBr}_3$  in  $\text{CH}_2\text{Cl}_2$  with the formation of gem-dimethylcyclohexane. The yield of gem-dimethylcyclohexane reaches 50% with the ratio of the reactants hydrocarbon:silane:  $\text{AlBr}_3 = 1:10:3$  ( $50^\circ\text{C}$ , 6 h). Decalin forms 9-methyldecalin under analogous conditions.

Evidently the reaction proceeds according to a scheme including the removal of a hydride ion from the tertiary C atom under the influence of  $\text{AlBr}_3$  with the formation of a carbene ion and the transfer to it of a  $\text{CH}_3$  group from the TMS:



According to [1] the splitting of the Si-C bond does not occur immediately under the influence of the carbene ion. Evidently  $\text{AlBr}_3$  plays the role of the carrier of the  $\text{CH}_3$  group in the reaction. The fact that nonbranched hydrocarbons, for example cyclohexane, not forming a tertiary carbene ion, do not enter into the alkylation reaction speaks in favor of the presented scheme.

The reaction presented here is a new method of synthesis of hydrocarbons with a quaternary C atom. Application of it to saturated bicyclic compounds permits the production of hydrocarbons with an angular  $\text{CH}_3$  group.

## LITERATURE CITED

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