- 18 W. H. Dyson, R. H. Hall, C. I. Hong, S. P. Dutta und G. B. Chheda, Can. J. Biochem. 50, 237 (1972).
- 19 C. I. Hong, G. B. Chheda, S. P. Dutta, A.O'Grady-Curtis und G. L. Tritsch, J. Med. Chem. 16, 139 (1973).
- 20 C. I. Hong und g. B. Chheda, J. Med. Chem. 18, 75 (1975).
- 21 C. I. Hong, G. L. Tritsch, A. Mittelman, P. Hebborn und G. B. Chheda, J. Med. Chem. 18, 465 (1975).
- 22 S. P. Dutta, C. I. Hong, G. L. Tritsch, C. Cox, R. Parthasarthy und G. B. Chheda, J. Med. Chem. 20, 1598 (1977).
- 23 C. I. Hong, A. Mittelman und G. B. Chheda, J. Pharm. Sci. 67, 569 (1978).

[Ph 720]

Arch. Pharm. (Weinheim) 317, 198-202 (1984)

Activated Nitriles in Heterocyclic Chemistry

A Novel Synthesis of Pyridones, Pyrazolo[1,5-a]pyrimidines and Pyrazolo[1,5-b]oxazines

Mohamed A.E. Khalifa*, Gamal H. Tammam and Hadeer M. Bakeer

Department of Chemistry, Faculty of Science, Cairo University, Giza, A.R. Egypt. Eingegangen am 3. Januar 1983

Several new pyridones, pyrazolo[1,5-a] pyrimidines and pyrazolo[1,5-b] oxazines were prepared from the cyanoacetohydrazide derivatives 1, 11 and the cinnamonitrile derivatives 2a-c.

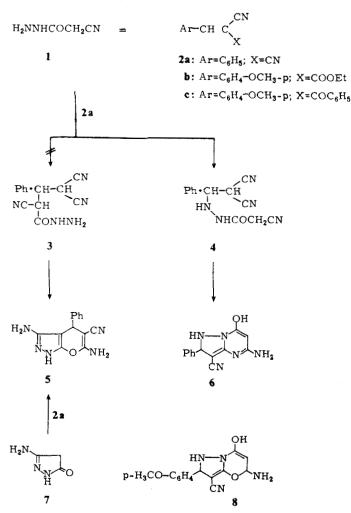
Aktivierte Nitrile in der heterocyclischen Chemie: Eine neue Synthese von Pyridonen, Pyrazolo[1,5-a]pyrimidinen und Pyrazolo[1,5-b]oxazinen

Einige neue Pyridone, Pyrazolo[1,5-a]pyrimidine and Pyrazolo[1,5-b]oxazine werden hergestellt aus den Cyanoacethydrazid-Derivaten 1 und 11 und den Cinnamonitril-Derivaten 2a-c.

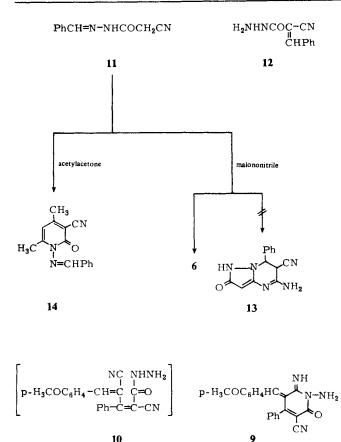
The considerable biological activities of both pyridones and pyrazoles have stimulated enormous interest in the synthesis and chemistry of this class of compounds^{1,2)}. As a part of a medicinal chemistry program^{3,5)} in our laboratories, the syntheses of several pyridones, pyrazolo[1,5-a]pyrimidines and pyrazolo[1,5-b]oxazines were required.

In conjugation of this work, we report here on the reaction of cyanoacethydrazide (1) with the cinnamonitrile derivatives 2a-c. Thus, it has been found that 1 reacts with benzylidenemalononitrile (2a) to yield an 1:1 adduct. Several isomeric cyclic and acyclic structures seemed possible for the reaction product. Acyclic structures 3 and 4 were readily eliminated based on the ¹H-NMR spectrum of the product which revealed only one singlet proton at $\delta = 5.6$ ppm. Acyclic structures 3 and 4 should have revealed multiplets for more

than one proton linked to sp^3 carbons. Thus, structures 5 and 6 were considered and spectroscopy seemed to be of little help for discrimination of both structures. Structure 5 could be eliminated based on the non-identity of the reaction product with the product of the reaction of 2a with 3-amino-2-pyrazolin-5-one (7). Thus, structure 6 was established for the reaction product. Compound 6 is assumed to be formed via initial addition of the hydrazide NH to the activated double bond in 2a and subsequent cyclisation of the formed 4 into 6 (cf. chart. 1).



Similar to the behaviour of 2a, compound 2b reacted with 1 to yield the pyrazolo[1,5-b]oxazine derivative 8. In contrast to the behaviour of 2a, b, compound 2c reacted with 1 to yield the condensation product 9. The formation of 9 in this reaction is assumed to proceed via condensation of 1 with 2c to yield the intermediate 10 which then cyclises via addition to the cyano group to yield the final isolable product 9.



The behaviour of 2a-c towards 1 is different than their behaviour towards other active methylene reagents. This may indicate that the hydrazino moiety in 1 is the most active nucleophilic center in the molecule. This assumption was supported by our observation that 1 reacts with benzaldehyde to yield the Schiff's base 11 rather than the ylidene derivative 12.

The behaviour of 11 towards a variety of active methylene nitriles has been investigated. It has been found that 11 reacts with malononitrile to yield a product of molecular formula $C_{13}H_{11}N_5O$. Two isomeric structures seemed possible (cf. structures 13 and 6). Structure 6 was readily established based on identity of the reaction product with a sample of 6 prepared from the reaction of 1 with 2a. Similarly the pyridine derivative 14 was formed from the reaction of 11 with acetylacetone. Now the behaviour of 11 towards a variety of activated double bond systems is under investigation.

Experimental Part

MP: uncorr. *IR Spectra* (KBr): Unicam Sp-1000. ¹*H*-*NMR* (DMSO-D₆), TMS int. stand., chemical shifts: δ (ppm). *Analytical data:* microanalytical center at Cairo University.

Pro- duct	Colour	Yield (%)	M.P.[°C] (Solv.)		Molecular formula	% Analysis		Calc.
			(30	JIV.)	Tormula	С	Н	Found N
5	brown	50	>	300	C ₁₃ H ₁₁ N ₅ O (253)	61.7 61.5	4.3 4.1	27.7 27.5
6	pale yellow	70	>	280 (acetic acid)	C ₁₃ H ₁₁ N ₅ O (253	61.7 61.3	4.3 4.1	27.7 27.1
8	Colourless	73		200 (ethanol)	C ₁₄ H ₁₄ N ₄ O ₃ (286)	58.7 58.5	4.8 4.6	19.6 19.4
9	colourless	68	>	300 (ethanol)	C ₂₀ H ₁₆ N ₄ O ₂ (344)	69.8 69.5	4.6 4.4	16.3 16.0
11	colourless	78		173 (ethanol)	C ₁₀ H ₉ N ₃ O (187)	64.2 63.9	4.8 4.6	22.5 22.1
14	Colourless	65		198 (ethanol)	C ₁₅ H ₁₃ N ₃ O (251)	71.7 71.4	5.2 5.0	16.7 16.5

Table 1: Characterisation data of compounds 5, 6, 8, 9, 11 and 14

Table 2: I.R. and ¹H-NMR data of compounds 5, 6, 8, 9, 11 and 14

Pro- duct	IR (KBr) [cm ⁻¹]	¹ H-NMR (DMSO-d ₆ /TMS _{int}) δ [ppm]		
5	3300, 3200, 3400 (NH ₂); 2940 (CH) and 2210 (CN).	5.66 (s, 4H, NH ₂); 7.4 (m, 5H _{arom.}); 8.3 (s, br, 1H, NH), 7.9 (s, 1H, CH).		
6	3400, 3320, 3200 (NH ₂); 2940 (CH) and 2200 (CN).	5.66 (s, 2H, NH ₂); 7.5 (m, 5H _{arom.}); 8.34 (s, br, 2H, CH and NH) and 11.9 (s, 1H, OH).		
8	3450, 3300, 3200 (NH ₂); 2970, 2950 (CH and CH ₃) and 2225 (CN).			
9	3460, 3300, 3210 (NH and NH ₂); 2950 (CH and CH ₃); 2220 (CN) and 1750 (ring CO).	3.3 (s, 3H, NH and NH_2); 4.0 (s, 3H, OCH ₃) and 6.9 ~8.0 (m, $9H_{arom}$)		
11	3200 (NH); 2940 (CH ₂); 2220 (CN) and 1700 (CO).	4.2 (s, 2H, CH ₂); 7.3 ~7.8 (m, 5H _{arom}); 7.9 (s, 1H, CH) and 12.3 (s, br, 1H, NH).		
14	3025, 2900 (CH and CH ₃) 2200 (CN) and 1680 (CO).	3.3 (s, 6H, two CH ₃); 6.38 (s, 1H, pyridine H-3) and 9.8 (s, 1H, azomethine CH).		

5-Cyano-3,6-diamino-4-phenyl-pyrazolo[1,5-a]-4H-pyran (5)

A solution of 0.99 g 3-amino-5-pyrazolone (7) in 100 ml ethanol was heated under reflux with 1.5 g benzalmalononitrile (2a) and 1 ml of triethylamine for 5 h, the solvent was removed, the residue was triturated with ethanol and the resulting solid product was crystallized to give 5 (cf. Table 1 and 2).

5-Amino-3-cyano-7-hydroxy-2-phenylpyrazolo[1,5-a]pyrimidine (6)

Method A: A solution of 0.99 g cyanoacethydrazide in 100 ml ethanol was treated with 1.54 g 2a and 1 ml of triethylamine. The reaction mixture was heated under reflux for 5 h and then the solvent was removed. The residue was triturated with ethanol and the resulting solid product was crystallised to give 6 (cf. Tables 1 and 2).

Method B: Compound 6 could also be prepared by the reaction of 1.87 g 11 and 0.66 g malononitrile applying the same procedure described in method A.

5-Amino-3-cyano-7-hydroxy-2-(p-methoxyphenyl)-pyrazolo[1,5-b]oxazine (8)

A solution of 0.99 g 1 in 100 ml ethanol was treated with 2.31 g anisalethyl cyanoacetate and 2 ml triethylamine. The reaction mixture was heated under reflux for 7 h, then the solvent was evaporated i.vac. The residue was triturated with ethanol and the resulting solid product was crystallised to give 8 (cf. Tables 1 and 2).

1-Amino-5-anisylidene-3-cyano-6-imino-4-phenylpyridin-2-one (9)

9 was prepared by the reaction of 0.99 g 1 with 2.63 g 2c applying the same procedure as for synthesis of compound 8 (cf. Tables 1 and 2).

Benzal cyanoacethydrazide (11)

A mixture of 0.99 g 1 and 1.06 g benzaldehyde was heated on a boiling water bath for 20 min. The oily layer that formed readily solidified on cooling and was crystallised to give 11 (cf. Tables 1 and 2).

1-N-(Benzalamino)-3-cyano-4,6-dimethylpyridin-2-one (14)

A solution of 1.87 g **11**, 1 g acetylacetone and 2 ml triethylamine in 100 ml ethanol was heated under reflux for 7 h. The solvent was removed i.vac. and the residue crystallised to give **14** (cf. Tables 1 and 2).

References

- 1 M. A. E. Khalifa and M. H. Elnagdi, Indian J. Chem. 12, 46 (1974); C. A. 81, 91310n (1974).
- 2 M. Sammour, Egypt. J. Chem. 14, 213 (1971).
- 3 M.H. Elnagdi, M.A.E. Khalifa, M.K.A. Ibraheim and M.R.H. Elmoghayar, J. Heterocycl. Chem. 18, 877 (1981).
- 4 M.H. Elnagdi and H. Wamhoff, Chem. Lett. 1981, 419.
- 5 M.A.E. Khalifa, G.H. Tammam and A.A.A. Elbanany, Arch. Pharm. (Weinheim) 315, (1982).

202