

A SIMPLIFIED SYNTHESIS OF (+)-BREFELDIN A

E. J. Corey and Philip Carpino

Department of Chemistry, Harvard University, Cambridge, Massachusetts, 02138

Summary: An effective and enantioselective process for the total synthesis of (+)-brefeldin A (**1**) is described which starts from the dextrorotatory ketone **2**.

Brefeldin A¹ (**1**) has long been known to exhibit an extraordinary range of biological activities including antibiotic, antiviral, cytostatic and antimitotic effects.² Recently, the underlying reason for such biological activities has been clarified by a number of important discoveries including the following: (1) **1** inhibits protein transport in mammalian cells from the endoplasmic reticulum (ER) to the golgi complex (GC)³ and causes disassembly of the GC; (2) **1** inhibits sialation, fucosylation, and sulfation of glycoproteins as well as transport of proteins;⁴ (3) **1** inhibits exocytosis and the presentation of viral proteins to cytotoxic T lymphocytes.⁵ These are remarkable actions for such a small molecule.

Since the first synthesis of (±)-brefeldin A in 1976,⁶ several groups have described syntheses of **1**, both in racemic and chiral form.⁷ Because of a substantial current demand for experimental quantities of **1**, its commercial unavailability, and the impracticality of previously developed syntheses, we were led to develop a simplified and practically useful new synthetic route to **1** with the results which are described herein.

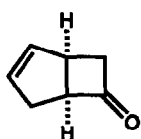
The starting point for synthesis was the readily available dextrorotatory form of bicyclo[3.2.0]heptenone **2**, $[\alpha]_D^{25} +61.4^\circ$ ($c=1$, CHCl_3),⁸ which was converted to bromo lactone **3** by (1) bromohydrin formation with 1 equiv of N-bromosuccinimide in 5:1 acetone-water at 23°C (60% yield)⁹ and (2) Baeyer-Villiger oxidation with 30% H_2O_2 in HOAc at 5 °C for 4 h (67% yield, not optimized). Bromo lactone **3** was transformed into enone **4** by the sequence: (1) debromination with 1.15 equiv of Bu_3SnH and 0.2 equiv of bisazoisobutyronitrile in C_6H_6 at reflux for 24 h (92%), (2) oxidation by treatment with dimethyl sulfoxide (1.1 equiv) and $(\text{CF}_3\text{CO})_2\text{O}$ (1.1 equiv) in CH_2Cl_2 at -65 °C followed by Et_3N (2.15 equiv), warming to 25 °C (1 h at 25 °C) to give **4**, and (3) esterification (CH_3I , K_2CO_3 , DMF, 23 °C for 48 h) to

afford **5** (60% overall), $[\alpha]_D^{23} +120.8^\circ$ ($c=1.1$, CH_2Cl_2). The next step, **5** \rightarrow **7**, required the reagent **6** which was prepared by reaction of the *t*-butyldimethylsilyl (BMS) ether of (*S*)-hept-6-yn-2-ol^{7b} with Cp_2ZrHCl ¹⁰ in THF.¹¹ A mixture of **5** and **6** (1.6 equiv) in THF at 0 °C was treated with a THF solution prepared from 0.4 equiv of $\text{Ni}(\text{acac})_2$ and 0.4 equiv of diisobutylaluminum hydride at -10 °C¹⁰ to give, after a reaction time of 8 h at 0 °C and isolation by silica gel chromatography, keto ester **7** in 72% yield as an oil, $[\alpha]_D^{23} -69.5^\circ$ ($c=2.6$, CH_2Cl_2). Reduction of **7** with 1 equiv of Li-*sec*- Bu_3BH (L-Selectride) in THF at -78 °C for 2 h gave a 4:1 mixture of **8** and the C-7 epimer (brefeldin numbering) in 89% combined yield. Pure **8** was obtained by heating the mixture with 2 mole% of tosic acid in toluene at reflux for 10 h, to lactonize the minor diastereomer, followed by silica gel chromatography; oil, $[\alpha]_D^{23} -14.8^\circ$ ($c=0.3$, CH_2Cl_2). Reaction of **8** with methoxyethoxymethyl chloride (MEM chloride, 3 equiv) and diisopropylethylamine (4.5 equiv) in methylene chloride at 23 °C for 10 h afforded the MEM ether **9** (97%), $[\alpha]_D^{23} -14.7^\circ$ ($c=1.9$, CH_2Cl_2), reduction of which with 1 equiv of *i*- Bu_2AlH in CH_2Cl_2 at -78 °C for 1.5 h produced aldehyde **10** (98%), $[\alpha]_D^{23} -15.1^\circ$ ($c=1.4$, CH_2Cl_2). Reaction of aldehyde **10** with 4 equiv of (*S*)-**11**¹² and 6 equiv of piperidine in CH_3CN at 5 °C for 48 h afforded after silica gel chromatography allylic alcohol **12** and its C(4) diastereomer in a ratio of 7.5:1 (94% total yield). Since the mixture was difficult to separate it was used as such, since the purification of the final product **1** was easily accomplished by chromatography.

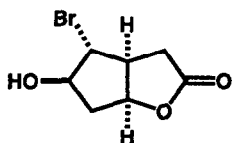
Intermediate **12** was converted to the bis MEM ether as described above for **8** \rightarrow **9** (90% yield) and then converted to hydroxy acid **13** by desilylation (3:1 HOAc- H_2O at 23 °C for 5 h, 87% yield) and saponification (0.15 M LiOH in 6:1 MeOH- H_2O at 23 °C for 24 h, 93% yield). Lactonization of **13** and MEM ether cleavage, as previously described,⁶ provided after silica gel chromatography and recrystallization from ethyl acetate, (+)-brefeldin A (**1**), mp and mixture mp with an authentic sample 201-202 °C, $[\alpha]_D^{23} +90^\circ$ ($c=0.1$, MeOH). Synthetic and authentic samples of brefeldin A were identical by infrared, 500 MHz ^1H NMR, and tlc comparison.¹³

In an effort to achieve and even shorter synthesis of **1**, esters **14** and **15** were synthesized from **4** and the corresponding (*S*)- α -*p*-toluenesulfinylacetate esters, as described above for **10** \rightarrow **12**. Attempts to effect macrolactonization-conjugate addition from **14** using reagents such as $\text{Bu}_3\text{Sn}^\bullet$ radical precursors or sources of Pd(0) were not successful. In addition attempts to hydrozirconate **15** with Cp_2ZrHCl (for Ni-catalyzed internal conjugation analogous to **5** + **6** \rightarrow **7**) were thwarted by the facile reduction of the enone carbonyl by Cp_2ZrHCl .

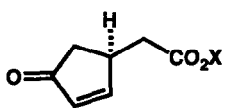
We believe that the synthesis of (+)-**1** reported herein is substantially simpler and more effective than any previously described.¹⁴



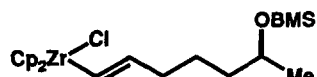
2



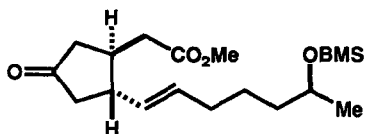
3



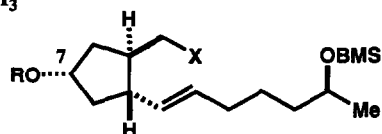
4 X = H

5 X = CH₃

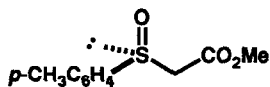
6



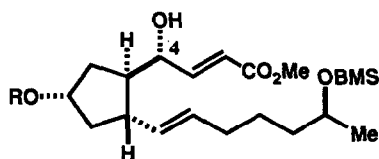
7

8 R = H; X = CO₂Me9 R = MEM; X = CO₂Me

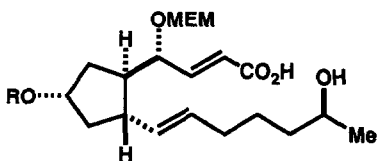
10 R = MEM; X = CHO



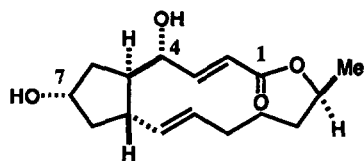
11



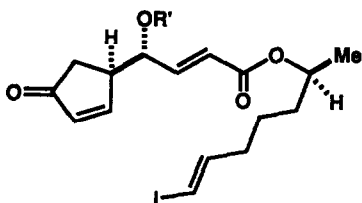
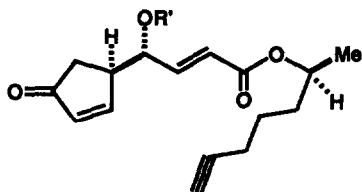
12



13



1

14 R' = *t*-BuPh₂Si

15

BMS - *t*-butyldimethylsilyl

References and Notes

1. (a) H. P. Sigg, *Helv. Chim. Acta*, **47**, 1401 (1964); (b) R. G. Coombe, P. S. Foss, J. J. Jacobs, and T. R. Watson, *Aust. J. Chem.*, **22**, 1943 (1969); (c) H. P. Weber, D. Hauser, and H. P. Sigg, *Helv. Chim. Acta*, **54**, 2763 (1971).
2. (a) V. Betina, *Neoplasma*, **16**, 23 (1969); (b) V. Betina, K. Horáková, and Z. Baráth, *Naturwiss*, **49**, 241 (1962); (c) V. Betina and A. Murin, *Cytologia* (Tokyo) **29**, 370 (1964); (d) D. Baciková, V. Betina, and P. Nemec, *Naturwiss*, **51**, 445 (1964); (e) G. Tamura, K. Ando, S. Suzuki, A. Takatsuki, and K. Arima, *J. Antibiotics*, **21**, 160 (1968); (f) A. Takatsuki, I. Yamaguchi, G. Tamura, T. Misato, K. Arima, *J. Antibiotics*, **22**, 442 (1969).
3. Y. Misumi, Y. Misumi, K. Miki, A. Takatsuki, G. Tamura, and Y. Ikehara, *J. Biol. Chem.*, **261**, 11398 (1986).
4. (a) V. S. Perkel, Y. Miura, and J. A. Magner, *Proc. Soc. Exp. Biol. Med.*, **190**, 286 (1989); (b) J. A. Magner and E. Papogiannes, *Endocrinol.*, **122**, 912 (1988).
5. J. W. Yewdell and J. R. Bennink, *Science*, **244**, 1072 (1989).
6. (a) E. J. Corey and R. H. Wollenberg, *Tetrahedron Letters*, 4705 (1976); (b) E. J. Corey, R. H. Wollenberg and D. R. Williams, *Tetrahedron Letters*, 2243 (1977).
7. (a) T. Kitahara and K. Mori, *Tetrahedron*, **40**, 2935 (1984); (b) C. L. LeDrian and A. E. Greene, *J. Am. Chem. Soc.*, **104**, 5473 (1982); (c) Y. Koeksal, P. Raddatz and E. Winterfeldt, *Angew. Chem. Int. Ed. Engl.*, **19**, 472 (1980); (d) K. Sakai, Y. Ishiguro, K. Funakoshi, K. Ueno, and H. Suemune, *Tetrahedron Letters*, **25**, 961 (1984); (e) B. M. Trost, J. Lynch, P. Renaut, and D. H. Steinman, *J. Am. Chem. Soc.*, **108**, 284 (1986).
8. The dextro ketone **2** is readily available from (\pm)-**2** by resolution of the crystalline bisulfite addition product obtained from (\pm)-**2**, ($-$)- α -phenylethylamine, sulfur dioxide and 1 equiv of water; see E. W. Collington, C. J. Wallis, and I. Waterhouse, *Tetrahedron Letters*, **24**, 3125 (1983). We are grateful to Dr. Wallis for a generous supply of (+)-**2**.
9. Z. Grudzinski and S. M. Roberts, *J. Chem. Soc. Perkin I*, 1767 (1975).
10. J. Schwartz, M. J. Loots, and H. Kosugi, *J. Am. Chem. Soc.*, **102**, 1333 (1980).
11. Satisfactory spectroscopic data were obtained for each synthetic intermediate using chromatographically purified and homogeneous samples.
12. (a) R. Tanikaga, Y. Nozaki, T. Tamura, and A. Kaji, *Synthesis*, 134 (1983); (b) K. Burgess and I. Henderson, *Tetrahedron Letters*, **30**, 3633, 4325 (1989).
13. We are grateful to Dr. H. P. Sigg of Sandoz Ltd. for an authentic sample of brefeldin A.
14. This research was assisted financially by grants from the National Institutes of Health and the National Science Foundation.

(Received in USA 4 October 1990)