Synthesis and antitumour activity of certain pyrido[2,3-*d*] pyrimidine and 1,8-naphthyridine derivatives

Afaf K. Elansary, Ashraf A. Moneer, Hanan H. Kadry* and Ehab M. Gedawy

Pharmaceutical Organic Chemistry Department, Faculty of Pharmacy, Cairo University, Cairo, 11562, Egypt

In an effort to establish new candidates with improved anticancer activity, we report here the synthesis of various series of 2,4,5,7-tetrasubstituted pyrido[2,3-*d*]pyrimidines and their related isosteres substituted 1,8-naphthyridines. The cytotoxic activity of the newly synthesised compounds against human breast cancer cell line, MCF7 was investigated. Most of the tested compounds exploited potent to moderate growth inhibitory activity, in particular 7-(4-chlorophenyl)-5-(3-nitrophenyl)pyrido[2,3-*d*]pyrimidin-4-amine exhibited superior potency to the reference drug doxorubicin (IC_{50} =7.54 and 8.48 µM respectively).

Keywords: synthesis, pyrido[2,3-d]pyrimidine, naphthyridine, inhibitor, substituent effect, antitumour

The pyrido[2,3-d]pyrimidine ring system are an interesting class of heterocycles because of several biological activities associated with this scaffold such as antibacterial,1 antiinflammatory,² antihypertensive,³ diuretic,⁴ antihistaminic ⁵ and more specifically as cytotoxic agents.⁶ Their cytotoxic activities might be attributed to inhibition of several enzymes such as tyrosine kinases,7-9 cyclin-dependant kinase (CDK)10,11 and dihydrofolate reductase (DHFR).¹² In a clinical study, The second generation dihydrofolate reductase inhibitor piritrexim (PTX), 2,4-diamino-6-(2,5-dimethoxybenzvl)-5methylpyrido[2,3-d]pyrimidine, has been developed as a potent small-molecule inhibitor that is effective against cancer cells resistant to methotrexate.13 Piritrexim was active against a broad range of tumours, with evidence of high tissue penetration due to its lipophilicity.¹⁴ Also, lometrexol (5,10-dideazatetrahydrofolic acid) was a new antifolate. One of the most promising preclinical features of lometrexol in animal models was its significant activity against a broad panel of solid tumours.¹⁵ (Fig. 1).

On the other hand, naphthyridine derivatives continued to be of great interest due to the wide spectrum of their physiology such as anti-inflammatory,16 anti-aggressive,17 and antitumour activities.^{18,19} Several synthetic approaches have been developed to synthesise the 1,8-naphthyridine derivatives,²⁰ but due to their great importance, the development of new synthetic methods remain an active research area.Based on the above finding, coupled with our efforts to synthesise novel fused pyridines acting as antitumour agents,²¹ we decided to prepare and evaluate certain new pyrido[2,3-d]pyrimidines and their bioisosteric analogue 1,8-naphthyridine derivatives, structurally-related to the aforementioned anticancer agents (Fig. 1). Also the effect of different substituent at positions 2,4,5, and 7 to substantiate the possible influence of such substitution on the in vitro antitumour activities against human breast adenocarcinoma MCF-7 cell line.

Results and discussion

The synthetic approaches adopted to obtain the target compounds are outlined in Schemes 1-3. Compounds 2a-d were synthesised by one-pot reaction of the corresponding cycloalkanones 1a-d, appropriate aromatic aldehydes, ethyl cyanoacetate and ammonium acetate following a procedure previously reported by Fathalla et al.22 in 55-68% yield. The formation of compounds 2a-d was confirmed by IR spectra that showed C≡N band at v 2200 cm⁻¹ as well as NH bands at v 3300-3150 cm⁻¹, in addition to C=O absorption at v 1660-1650 cm⁻¹. The¹H NMR spectra of compounds **2a-d** showed an exchangeable singlet signal at δ 12.504-12.740 ppm corresponding to the NH protons. Chlorination of 3-cyanopyridones 2a-d with phosphorus oxychloride/N,Ndimethylaniline mixture gave the corresponding 2-chloro derivatives, 3a-d, which were confirmed by IR spectra that showed the disappearance of C=O absorption. The latter compounds were reacted with guanidine base, after liberation from its hydrochloride salt, in pyridine to afford the substituted 2,4-diaminopyrido[2,3-d]pyrimidines 4a-d. The structures of compounds 4a-d were established on the basis of elemental analyses and spectral data. The IR spectra of compounds 4a-d showed no absorption band for C=N group, meanwhile revealed characteristic absorption bands at 3400-3180 cm⁻¹ indicating the presence of NH₂ functionalities. Furthermore, their ¹H NMR spectra revealed the appearance of singlet signals at δ 6.418–7.158 and 11.890–12.026 ppm corresponding to two NH₂ groups. Scheme 2 illustrates the synthesis of several substituted pyrido[2,3-d]pyrimidines 6-12 starting from the 2-amino-4,6-disubstituted pyridine-3-carbonitriles 5a,b. Compounds 5a,b were synthesised by applying a one-pot reaction of the respective substituted acetophenone with appropriate aromatic aldehydes and malononitrile in the presence of ammonium

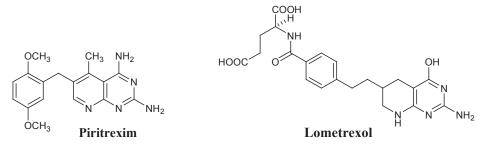
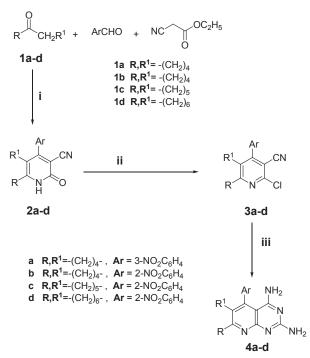
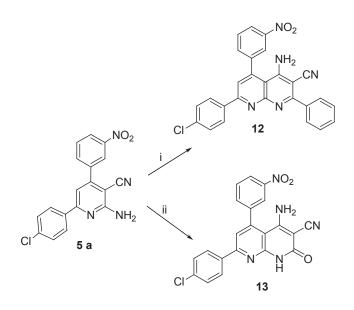


Fig. 1 Some drugs containing pyrido[2,3-d]pyrimidine currently in clinical development.

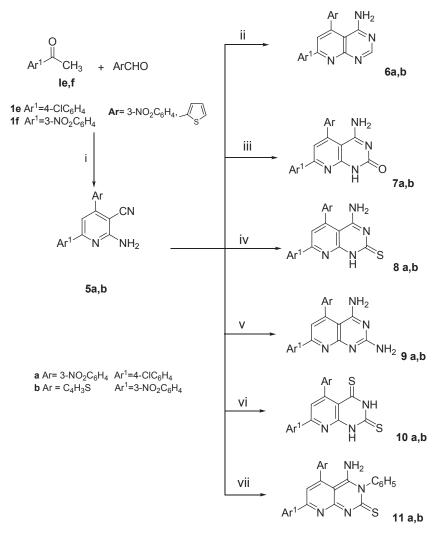
^{*} Correspondent. E-mail: hanankadry2005@yahoo.com





 $\begin{array}{l} \textbf{Scheme 1} \quad \text{Reagent and conditions: (i) } \text{CH}_3\text{COONH}_4, \text{ n-butanol, reflux 8 h; (ii) } \text{POCL}_3 \text{ dimethylaniline, reflux 10 h; (iii) guanidine HCL, } \text{NaOC}_2\text{H}_5, \\ \text{pyridine, reflux 5 h.} \end{array}$

Scheme 3 Reagent and condition: (i) benzylidenemalononitrile, piperidine, dioxane, reflux 3 h; (ii) ethyl cyanoacetate, triethylamine, ethanol, reflux 3 h.



Scheme 2 Reagent and condition: (i) malononitrile, $CH_{3}COONH_{4}$, n-butanol, reflux 8 h; (ii) formamide, oil bath, reflux 15 h; (iii) urea, oil bath, reflux 4 h; (iv) thiourea, oil bath, reflux 4 h; (v) guanidine HCL methanol, reflux 6 h; (vi) CS_{2} , pyridine, reflux 15 h; (vii) PhCNS, DMF, TEA, reflux 10 h.

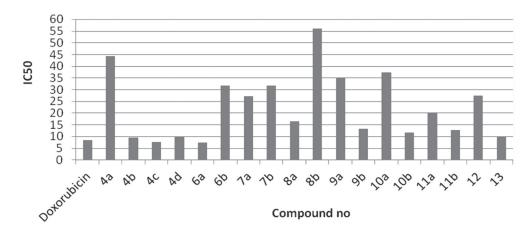


Fig. 2 IC50 in µM of the synthesised compounds and doxorubicin against human breast adenocarcinoma cell line (MCF7).

acetate. The IR spectra of **5a**,**b** revealed the appearance of absorption bands at 3362–3140 cm⁻¹ corresponding to NH₂ and absorption bands at 2203 cm⁻¹ attributed to CN group. The ¹H NMR spectra was consistent with the proposed structure.

Accordingly, treatment of 5a,b with excess formamide produced 4-amino-5,7-disubstitutedpyrido[2,3-d]pyrimidines 6a,b following reported reaction conditions.²³ The IR spectra proved as useful in tracing the disappearance of the C=N stretching absorption of the parent compound 5a,b. Whereas the ¹H NMR spectra of **6a** and **6b** displayed signals at δ 10.02 and 10.442 ppm respectively corresponding to NH₂ group. Furthermore, the mass spectrum of 6a displayed molecular ion peaks at m/z 377 and 379 corresponding to (M⁺) and (M+2) in a ratio 3:1. On the other hand, reacting compounds 5a,b with urea or thiourea afforded 7a,b and 8a,b respectively, following the reaction conditions reported for the preparation of related compounds.²³ The structures of compounds 7a,b and 8a,b were confirmed by IR and ¹H NMR spectra. The IR spectra clearly showed the disappearance of the C≡N stretching absorption of the parent compound 5a,b and the appearance of C=O (in compounds 7a,b) or C=S (in compounds 8a,b) stretching signals together with NH, absorption band. The ¹H NMR spectra of compounds 7a,b and 8a,b revealed two D₂O exchangeable singlet signals at δ 6.51–7.04 and δ 11.18–11.32 ppm corresponding to NH, and NH, respectively. Whereas, 2,4-diamino-5,7-disubstituted pyrido[2,3-d] pyrimidines 9a,b were obtained by reacting 5a,b with guanidine hydrochloride in refluxing methanol in the presence of sodium hydroxide. ¹H NMR spectra of compounds 9a,b revealed the appearance of two exchangeable singlet signal at δ 5.02,6.95 and δ 7.23–8.5 ppm corresponding to NH₂ protons. Reacting 5a,b with carbon disulfide in refluxing pyridine afforded the corresponding pyrido[2,3-d]pyrimidine-2,4(1H,3H)-dithione derivatives 10a,b, which were confirmed by spectral data. The IR spectra proved as useful in tracing the disappearance of the $C \equiv N$ stretching absorption of the parent compound 5a,b and the appearance of a C=S absorption. Finally, reaction of 5a,b with phenyl isothiocyanate in refluxing dimethyl formamide in the presence of triethylamine yielded the corresponding 4-amino-3-phenyl-5,7-disubstituted pyrido[2,3-d]pyrimidine-2(3H)-thiones (11a,b) which were confirmed by spectral data. One of the principal objectives of the present work was the synthesis of substituted 1,8-naphthyridine derivatives12 and 13 (Scheme 3), thereby, reacting 5b with either benzylidene malononitrile in refluxing dioxane in the presence of piperidine or ethyl cyanoacetate in refluxing ethanol in the presence of triethylamine afforded compounds 12 and 13 respectively.

The ¹H NMR spectra of compounds **12** and **13** displayed a singlet signal at δ 6.95 ppm corresponding to C-6 proton of naphthyridine. Both the analytical and spectral data (IR, ¹H NMR ¹³C NMR and MS) of the newly synthesised compounds were in full agreement with the proposed structures.

In vitro anticancer screening

The newly synthesised compounds were evaluated for their in vitro cytotoxic activity against human breast cancer cell line, MCF-7. Doxorubicin, which is one of the most effective anticancer agents, was used as the reference drug in this study. The IC_{50} (dose of the compound which caused a 50% reduction of survival values) are shown in Table 1 and represented graphically in Fig. 2. From the results in Table 1, it is evident that all the tested compounds displayed potent to moderate growth inhibitory activity, in particular compounds 4c and 6a $(IC_{50} = 7.70 \text{ and } 7.54 \,\mu\text{M}$ respectively) were found to be more potent and efficacious than doxorubicin (IC₅₀ values 8.48 μ M). Further studies are recommended to explore the mechanism of action as well as the effect of substitution on the anticancer effect of these compounds. We hope that the synthesised compounds serve as lead chemical entities for further modification to render them clinically useful drug agents.

Experimental

Melting points are uncorrected and determined in one-end open capillary tubes using Gallenkamp melting point apparatus MFB-595-010M (Gallenkamp, London, UK). Microanalysis was carried out at the Micro-analytical Unit, Faculty of Science, Cairo University and the Regional Centre for Microbiology and Biotechnology, Al-Azhar University. Analyses indicated were within $\pm 0.4\%$ of the theoretical values. Infrared Spectra were recorded on Schimadzu FT-IR 8400S spectrophotometer (Shimadzu, Kyoto, Japan), and expressed in wavenumber (cm-1), using potassium bromide discs. The NMR spectra were recorded on a Varian Gemini 200 MHz and Varian Mercury VX-300 or Jeol-ECA500, 500 MHz Japan NMR spectrometer. ¹H NMR spectra were run at 300 MHz and ¹³C NMR spectra were run at 125 MHz in dimethylsulfoxide (DMSO-d_s). Chemical shifts are quoted in δ and were related to that of the solvents. Mass spectra were recorded using Hewlett Packard Varian (Varian, Polo, USA) and Shimadzu gas chromatograph mass spectrometer-QP 1000 EX (Shimadzu, Kyoto, Japan). TLC was carried out using Art. DC-Plastikfolien, Kieselgel 60 F254 sheets (Merck, Darmstadt, Germany), the developing solvents were chloroform: benzene: methanol [6.5: 3.5: 0.5] and the spots were visualised at 366, 254 nm by UV Vilber Lourmat 77202 (Vilber, Marne La Vallee, France). Compound 5a,²⁴ benzylidenemalononitrile,²⁵ were obtained according to reported procedure.

	R^2 R^1 N R^4 N N R^4		R^{2} R^{1} $R^{-C_{6}H_{5}}$ R^{4} N N R		R^2 R^1 CN R^4 N N R	
	4–10		11		12 and 13	
Compound	R	R ¹	R ²	R ³	R ⁴	IC50/µMª
Doxorubicin ^b	_	_	-	_	_	8.48
4a	NH ₂	NH_2	3-NO ₂ C ₆ H ₄		$-(CH_2)_4$	44.34
4b	NH ₂	NH_2	$2-NO_2C_6H_4$		$-(CH_2)_4$	9.63
4c	NH ₂	NH_2	2-NO ₂ C ₆ H ₄		$-(CH_2)_5$	7.70
4d	NH ₂	NH_2	$2-NO_2C_6H_4$		$-(CH_2)_6$	9.69
6a	Н	NH_2	3-NO ₂ C ₆ H ₄	Н	$4-CIC_6H_4$	7.54
6b	Н	NH_2	C_4H_3S	Н	$3-NO_2C_6H_4$	31.77
7a	=0	NH_2	3-NO ₂ C ₆ H ₄	Н	4-CIC ₆ H ₄	27.17
7b	=0	NH_2	C_4H_3S	Н	$3-NO_2C_6H_4$	31.75
8a	=S	NH_2	3-NO ₂ C ₆ H ₄	Н	$4-CIC_6H_4$	16.47
8b	=S	NH_2	C_4H_3S	Н	$3-NO_2C_6H_4$	56.15
9a	NH ₂	NH_2	3-NO ₂ C ₆ H ₄	Н	4-CIC ₆ H ₄	34.87
9b	NH ₂	NH_2	C_4H_3S	Н	$3-NO_2C_6H_4$	13.17
10a	=S	=S	$3-NO_2C_6H_4$	Н	$4-\text{CIC}_6\text{H}_4$	37.24
10b	=S	=S	C_4H_3S	Н	$3-NO_2C_6H_4$	11.66
11a	=S	NH_2	$3-NO_2C_6H_4$	Н	$4-CIC_6H_4$	20.32
11b	=S	NH_2	C_4H_3S	Н	$3-NO_2C_6H_4$	12.89
12	C_6H_5	NH_2	$3-NO_2C_6H_4$	Н	4-CIC ₆ H ₄	27.41
13	=0	NH_2	$3-NO_2C_6H_4$	Н	4-CIC ₆ H ₄	10.05

Table 1 In vitro cytotoxic activities of the synthesised compounds against MCF-7 cancer cell line

^aThe values given are means of three experiments.

^bDoxorubicin was used as reference drug.

Synthesis of 2-oxo-4,5,6-trisubstituted-1,2-dihydropyridine-3carbonitriles 2a-d; general procedure

A mixture of appropriate cycloalkanones 1a-d (0.05 mol), appropriate aromatic aldehyde (0.05 mol), ethyl cyanoacetate (5.65 mL, 0.05 mol) and ammonium acetate (30.8 g, 0.4 mol) in butan-1-ol (150 mL) was heated under reflux for 8 h. The separated solid was filtered, washed with ethanol, dried and crystallised from appropriate solvent.

4-(3-Nitrophenyl)-2-oxo-1,2,5,6,7,8-hexahydroquinoline-3carbonitrile (2a): Crystallised from butan-1-ol. Yield 63%; m.p. 291–292 °C; IR v_{max}/cm⁻¹: 3280, 3150 (NH), 3080 (CH aromatic), 2949, 2910, 2866 (CH aliphatic), 2225 (C=N), 1647(C=O), 1533, 1348 (NO₂); ¹H NMR (CDCl₃) δ ppm: 1.59–1.72 (m, 4H, 2CH₂), 2.04–2.12 (m, 2H, CH₂), 2.62–2.82 (m, 2H, CH₂), 7.88–8.39 (m, 4H, ArH), 12.56 (s, 1H, NH, exch. D₂O). Anal. calcd for C₁₆H₁₃N₃O₃ (295.28): C, 65.07; H, 4.43; N, 14.23; found: C, 64.99; H, 4.49; N, 14.13%.

4-(2-Nitrophenyl)-2-oxo-1,2,5,6,7,8-hexahydroquinoline-3carbonitrile (2b): Crystallised from ethanol. Yield 55%; m.p. 276-278°C; IR v_{max}/cm⁻¹: 3300 (NH),3050 (CH aromatic), 2950, 2850 (CH aliphatic), 2200 (C=N), 1640(C=O), 1520, 1340 (NO₂); ¹H NMR (DMSO-d₄) δ ppm: 1.51–1.66 (m, 4H, 2CH₂), 1.81–1.99 (m, 2H, CH₂), 2.63–2.69 (m, 2H, CH₂), 7.52 (d, 1H, J=7.5 Hz, H6 of ArH), 7.84 (t, 1H, J=7.5 Hz, H5 of ArH), 7.96 (t, 1H, J=7.8 Hz, H4 of ArH), 8.31 (d, 1H, J=7.8 Hz, H3 of ArH), 12.51 (s, 1H, NH, exch. D₂O). Anal. calcd for C₁₆H₁₃N₃O₃ (295.28): C, 65.07; H, 4.43; N, 14.23; found: C, 64.90; H, 4.38; N, 14.45%.

4-(2-Nitrophenyl)-2-oxo-2,5,6,7,8,9-hexahydro-1H-cyclohepta[b] pyridine-3-carbonitrile (2c): Crystallised from ethanol. Yield 68%: m.p. 273–275 °C; IR v_{max} /cm⁻¹: 3300 (NH), 3050 (CH aromatic), 2920, 2820 (CH aliphatic), 2210 (C=N), 1640 (C=O), 1540, 1350 (NO₂); ¹H NMR (DMSO-*d_e*) δ ppm: 1.26–1.46 (m, 4H, 2C*H_a*), 1.62–1.84 (m, 2H, CH,), 2.08-2.26 (m, 4H, 2CH,), 2.82-2.98 (m, 2H, CH,), 7.54-8.32 (m, 4H, ArH), 12.74 (br s, 1H, NH, exch. D₂O). Anal. calcd for C₁₇H₁₅N₂O₂ (309.30): C, 66.00; H, 4.88; N, 13.58; found: C, 66.14; H, 4.92; N, 13.74%.

4-(2-Nitrophenyl)-2-oxo-1,2,5,6,7,8,9,10-octahydrocycloocta[b] pyridine-3-carbonitrile (2d): Crystallised from ethanol. Yield 68%; m.p.>300 °C; IR v_{max}/cm⁻¹: 3300 (NH), 3050 (CH aromatic), 2920, 2820 (CH aliphatic), 2210 (C=N), 1640 (C=O), 1540, 1340 (NO₂); ¹H NMR (DMSO-*d_s*) δ ppm: 1.15–1.20 (m, 4H, 2C*H*₂), 1.37–1.41 (m, 4H, 2C*H*₂), 1.65-1.70 (m, 2H, CH₂), 2.16-2.24 (m, 4H, 2CH₂), 2.76-2.80 (m, 2H, CH₂), 7.56 (d, 1H, J=7.8 Hz, H6 of ArH), 7.84 (t, 1H, J=8.1 Hz, H5 of ArH), 7.96 (t, 1H, J=8.4 Hz, H4 of ArH), 8.33 (d, 1H, J=8.1 Hz, H3 of ArH), 12.66 (s, 1H, NH, exch. D₂O). Anal. calcd for C₁₈H₁₇N₃O₃ (323.32): C, 66.86; H, 5.29; N, 12.99; found: C, 66.79; H, 5.10; N, 13.41%

Synthesis of 2-chloro-4,5,6-trisubstituted pyridine-3-carbonitriles (3a-d); general procedure

A mixture of the corresponding pyridones 2a-d (0.0075 mol), N, N-dimethylaniline (10 mL), and phosphorous oxychloride (10 mL) was heated under reflux for 10 h. The reaction mixture was cooled and poured onto crushed ice. The formed solid was filtered, dried and crystallised from ethanol.

2-Chloro-4-(3-nitrophenyl)-5,6,7,8-tetrahydroquinoline-3carbonitrile (3a): Yield 52%; m.p. 200–202 °C; IR v_{mv}/cm⁻¹: 3090 (CH aromatic), 2960, 2935 (CH aliphatic), 2229 (C=N), 1530, 1340 (NO₂); ¹H NMR (CDCl₃) δ ppm: 1.76–1.91 (m, 4H, 2CH₂), 2.33–2.62 (m, 2H, CH₂), 2.98-3.04 (m, 2H, CH₂), 7.63-8.38 (m, 4H, ArH). Anal. calcd for C₁₆H₁₂ClN₂O₂ (313.73): C, 61.25; H, 3.85; N, 13.39; found: C, 61.45; H, 3.80; N, 13.65%.

2-Chloro-4-(2-nitrophenyl)-5,6,7,8-tetrahydroquinoline-3*carbonitrile* (3b): Yield 55%; m.p. 154–156°C; IR v_{max}/cm⁻¹: 3050(CH aromatic), 2920 (CH aliphatic), 2200 (C≡N), 1520, 1340 (NO₂); ¹H NMR (DMSO-d₆) δ ppm: 1.64–1.66 (m, 2H, CH₂), 1.78–1.80 (m, 2H, CH₂), 2.10–2.35 (m, 2H, CH₂), 2.96–2.98 (m, 2H, CH₂), 7.63 (d, 1H, J=7.5 Hz, H6 of ArH), 7.88 (t, 1H, J=7.5 Hz, H5 of ArH), 7.98 (t, 1H, J=7.5 Hz, H4 of ArH), 8.37 (d, 1H, J=8.1 Hz, H3 of ArH); MS m/z(%): 316.00 (M+3,12.45), 315(M+2, 5.95), 314(M+1, 25.71) 313(M+18.67), (C₅H₃,100). Anal. calcd for C₁₆H₁₂ClN₃O₂ (313.73): C, 61.25; H, 3.85; N, 13.39; found: C, 60.88; H, 3.80; N, 13.42%.

2-Chloro-4-(2-nitrophenyl)-6,7,8,9-tetrahydro-5H-cyclohepta[b] pyridine-3-carbonitrile (3c): Yield 60%; m.p. 165–166 °C; IR v_{max} / cm⁻¹: 3060 (CH aromatic), 2920 (CH aliphatic), 2200 (C=N), 1520, 1340 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 1.33–1.46 (m, 2H, CH₂), 1.66–1.76 (m, 4H, 2CH₂), 2.40–2.43 (m, 2H, CH₂), 3.10–3.12 (m, 2H, CH₂), 7.60 (d, 1H, J=7.5 Hz, H6 of ArH), 7.87 (t, 1H, J=7.5 Hz, H5 of ArH), 7.96 (t, 1H, J=7.8 Hz, H4 of ArH), 8.36 (d, 1H, J=7.8 Hz, H3 of ArH). Anal. calcd for C₁₇H₁₄CIN₃O₂ (327.75): C, 62.29; H, 4.30; N, 12.82; found: C, 62.47; H, 4.28; N, 13.01%.

2-*Chloro-4-(2-nitrophenyl)-5,6,7,8,9,10-hexahydrocycloocta[b]* pyridine-3-carbonitrile **(3d):** Yield 54%; m.p. 157–158 °C; IR ν_{max} /cm⁻¹: 3064 (CH aromatic), 2937, 2926, 2856 (CH aliphatic), 2210 (C≡N), 1520, 1320 (NO₂); ¹H NMR (DMSO– d_6) δ ppm: 1.18–1.51 (m, 6H, 3C H_2), 1.71–1.82 (m, 2H, C H_2), 2.12–2.14 (m, 2H, C H_2), 3.05–3.12 (m, 2H, C H_2), 7.71–8.43 (m, 4H, ArH). Anal. calcd for C₁₈H₁₆CIN₃O₂ (341.78): C, 63.25; H, 4.71; N, 12.29; found: C, 63.41; H, 4.76; N, 12.42%.

Synthesis of 2,4-diamino-5,6,7-trisubstituted pyrido[2,3-d]pyrimidines **(4a–d)***: general procedure*

Guanidine hydrochloride (1.84 g, 0.022 mol) was added to a cold solution of sodium (0.506 g, 0.022 mol) in absolute ethanol (50 mL). The separated sodium chloride was filtered off and washed with ethanol, the filtrate was evaporated under reduced pressure. The formed oily mass was diluted with pyridine (5 mL) and finally the corresponding 2-chloro-4,5,6-trisubstituted pyridine-3-carbonitrile **3a–d** (0.0033 mol) was added. The reaction mixture was heated under reflux for 5 h, cooled and finally poured onto water (25 mL). The formed precipitate was filtered, dried and crystallised from DMF/H₂O.

5-(3-Nitrophenyl)-6,7,8,9-tetrahydropyrimido[4,5-b]quinoline-2,4diamine (4a): Yield 72%; m.p.>300 °C; IR v_{max} /cm⁻¹: 3283, 3180 (NH₂), 3050(CH aromatic), 2950, 2850 (CH aliphatic), 1580, 1380 (NO₂); ¹H NMR (DMSO-*d*₆) δ ppm: 1.22–1.25 (m, 2H, CH₂), 1.82–1.90(m, 2H, CH₂), 2.73–2.91(m, 4H, 2CH₂), 6.58 (br s, 2H, NH₂, exch. D₂O), 7.19–8.37 (m, 4H, ArH), 11.89 (s, 2H, NH₂, exch. D₂O). Anal. calcd for C₁₇H₁₆N₆O₂ (336.35): C, 60.71; H, 4.79; N, 24.99; found: C, 60.73; H, 4.79; N, 25.00%.

5-(2-Nitrophenyl)-6,7,8,9-tetrahydropyrimido[4,5-b]quinoline-2,4-diamine (4b): Yield 68%; m.p.>300 °C; IR v_{max} cm⁻¹: 3400–3200 (NH₂), 3100 (CH aromatic), 2920, 2850 (CH aliphatic), 1580, 1330 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 1.17–1.23 (m, 2H, CH₂), 1.71–1.98 (m, 2H, CH₂), 2.74–2.89(m, 4H, 2CH₂), 7.15 (s, 2H, NH₂, exch. D₂O), 7.48–7.96 (m, 4H, ArH), 12.02 (s, 2H, NH₂, exch. D₂O); ¹³C NMR (DMSO) δ ppm: 22.98, 24.69, 46.19, 52.15.23, 93.91, 110.03, 115.54, 116.28, 130.82, 134.69, 136.13, 138.22, 141.21, 148.49, 155.58, 158.68, 162.23; Anal. calcd for C₁₇H₁₆N₆O₂(336.35): C, 60.71; H, 4.79; N, 24.99; found: C, 60.76; H, 4.69; N, 25.20%.

5-(2-Nitrophenyl)-6,7,8,9-tetrahydro-5H-cyclohepta[b] pyrido[2,3-d]pyrimidine-2,4-diamine (4c): Yield 84%; m.p. 255– 258 °C; IR v_{max} /cm⁻¹: 3480, 3310 (NH₂), 3090 (CH aromatic), 2920 (CH aliphatic) 1580, 1340 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 1.76–1.91 (m, 6H, 3CH₂), 3.01–3.21 (m, 4H, 2CH₂), 6.46 (s, 2H, NH₂, exch. D₂O), 7.25–8.35 (m, 4H, ArH), 12.02 (s, 2H, NH₂, exch. D₂O); ¹³C NMR (DMSO) δ ppm: 26.09, 29.34, 29.77, 31.23, 38.15, 112.12, 114.41, 118.42, 119.0, 127.89, 132.36, 138.33, 149.74, 155.70, 158.58, 158.88, 167.2, 169.87. Anal. calcd for C₁₈H₁₈N₆O₂ (350.37): C, 61.70; H, 5.18; N, 23.99; found: C, 61.80; H, 5.00; N, 24.06%.

5-(2-Nitrophenyl)-5,6,7,8,9,10-hexahydrocycloocta[b]pyrido[2,3-d] pyrimidine-2,4-diamine (4d):Yield 75%; m.p. > 300 °C; IR v_{max} /cm⁻¹: 3400–3200 (NH₂), 3050 (CH aromatic), 2920, 2820 (CH aliphatic), 1520, 1340 (NO₂); ¹H NMR (DMSO– d_{6}) δ ppm: 1.36–1.50 (m, 4H, 2CH₂), 1.76–1.80 (m, 2H, CH₂), 1.85–1.92 (m, 2H, CH₂), 2.86–2.88 (m, 2H, CH₂), 3.06–3.12 (m, 2H, CH₂), 6.41 (s, 2H, NH₂, exch. D₂O), 7.23 (d, 1H, *J*=6.9 Hz, H6 of ArH), 7.47 (t, 1H, *J*=8.7 Hz, H3 of ArH), 12.01 (s, 2H, NH₂, exch. D₂O); ¹³C NMR (DMSO) δ ppm: 26.41, 26.61, 30.12, 30.57, 35.50, 42.55, 96.91, 112.17, 125.72, 128.49, 131.59, 131.89, 135.32, 146.8, 149.9, 156.28, 156.19, 161.69, 165.37. Anal. calcd for C₁₉H₂₀N₆O₂ (364.40): C, 62.62; H, 5.53; N, 23.06; found: C, 62.47; H, 5.51; N, 23.01%.

Synthesis of 2-amino-4,6-diarylpyridine-3-carbonitriles **5a,b**; general procedure

A mixture of substituted acetophenone either Ie or 1f (0.01 mol), appropriate aromatic aldehydes (0.01 mol), malononitrile (5.3 g, 0.08 mol) and ammonium acetate (6.16 g, 0.08 mol) in butan-1-ol (10 mL) was heated under reflux for 8 h then cooled. The obtained precipitate was filtered, washed with ethanol, dried and crystallised from suitable solvent to afford 5a,b.

2-Amino-6-(4-chlorophenyl)-4-(3-nitrophenyl)nicotinonitrile (5a): Crystallised from butan-1-ol, yield 73%; m.p.>300 °C; IR v_{max} /cm⁻¹: 3300–3100 (NH₂), 3020(CH aromatic), 2200 (CN), 1550, 1350 (NO₂); ¹H NMR (DMSO- d_0) δ ppm: 6.61 (s, 2H, NH₂, exch. D₂O), 7.11–8.86 (m, 9H, 8ArH and H5). Anal. calcd for C₁₈H₁₁ClN₄O₂ (350.76): C, 61.64; H, 3.16; N, 15.97; found: C, 61.47; H, 3.13; N, 15.64%.

2-*Amino-6-(3-nitrophenyl)-4-(thiophen-2-yl)nicotinonitrile* (5b): Crystallised from ethanol. Yield 80%; m.p. > 300 °C; IR v_{max} /cm⁻¹: 3362–3140 (NH₂), 3080(CH aromatic), 2203 (CN), 1537, 1330 (NO₂); ¹H NMR (DMSO-*d*₀) δ ppm: 6.86 (s, 2H, N*H*₂, exch. D₂O), 7.11–7.86 (m, 8H, 7ArH and *H5*). Anal. calcd for C₁₆H₁₀N₄O₂S (322.34): C, 59.62; H, 3.13; N, 17.38; found: C, 59.52; H, 3.12; N, 17.35%.

Synthesis of 4-amino-5,7-disubstituted pyrido[2,3-d]pyrimidines (6a,b); general procedure

A mixture of the appropriate 2-aminopyridine-3-carbonitriles **5a,b** (0.01 mol) and excess formamide (15 mL) was refluxed in an oil-bath for 15 h at 210 °C. The mixture was cooled. The separated solid was filtered, dried and crystallised from ethanol.

7-(4-Chlorophenyl)-5-(3-nitrophenyl)pyrido[2,3-d]pyrimidin-4amine (6a): Yield 58%; m.p. 255–257 °C; IR v_{max} cm⁻¹: 3293, 3168 (NH₂), 3050 (CH aromatic), 1559, 1337 (NO₂); ¹H NMR (DMSO– d_6) & ppm: 7.39–8.91 (m, 8H, ArH), 8.02 (s, 1H, *H*2), 8.53 (s, 1H, *H*6), 10.44 (s, 2*H*, NH, exch.D₂O₅, ¹³C NMR (DMSO) & ppm: 106.88, 121.19, 122.08, 123.99, 129.26, 129.77, 130.62, 130.89, 131.08, 133.24, 135.06, 135.85, 136.56, 139.90, 147.94, 148.60, 155.01, 156.05, 159.66; MS *m/z* (%): 379.22 (M+2, 6.52), 377.70 (M⁺, 22.23). Anal. calcd for C₁₉H₁₂ClN₅O₂ (377.78): C, 60.41; H, 3.20; N, 18.54; found: C, 60.98; H, 3.45; N, 18.88%.

7-(3-Nitrophenyl)-5-(thiophen-2-yl)pyrido[2,3-d]pyrimidin-4-amine (6b): Yield 83%; m.p. > 300 °C; IR v_{max} /cm⁻¹: 3327–3193 (NH₂), 3050 (CH aromatic), 1561, 1338 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 6.85– 8.95 (m, 7H, ArH and thienyl H), 7.99 (s, 1H, H2), 8.35 (s,1H, H6), 10.01 (s, 2H, NH₂ exch.D₂O). Anal. calcd for C₁₇H₁₁N₅O₂S (349.37): C, 58.44; H, 3.17; N, 20.05; found: C, 58.89; H, 3.09; N, 20.43%.

Synthesis of 4-amino-5,7-disubstituted pyrido[2,3-d]pyrimidin-2(1H)ones (7a,b); general procedure

A mixture of the appropriate 2-aminopyridine-3-carbonitrile (5a,b) (0.01 mol) and urea (1.2 g, 0.02 mol) was heated in an oil bath at 120–130 °C for 2 h. The temperature was then raised to 180 °C for another 2 h. The residue was washed with water then ethanol and finally crystallised from appropriate solvent.

4-*Amino*-7-(4-chlorophenyl)-5-(3-nitrophenyl)pyrido[2, 3-d] pyrimidin-2(1H)-one (7a): Crystallised from butan-1-ol. Yield 55%; m.p. 275–277 °C; IR v_{max} (cm⁻¹: 3330–3200 (NH, NH₂), 3050 (CH aromatic), 1700 (C=O),1520, 1340 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 6.93 (s, 1H, H6), 7.01 (s, H, NH, exch.D₂O), 7.58–8.13 (m, 3H, H4, H5 and H6 of 3-NO₂C₆H₄), 8.85 (d, 2H, J=8 Hz, H3, H5 of 4-ClC₆H₄), 8.36 (d, 2H, J=7.8 Hz, H2, H6 of 4-ClC₆H₄), 8.47 (s, 1H, H2 of 3-NO₂C₆H₄), 11.18 (s, 2H, NH₂, exch.D₂O); ¹³C NMR (DMSO) δ ppm: 94.87, 95.32, 116.19, 119.08, 123.99, 124.71, 129.26, 130.87, 131.08, 135.06, 135.85, 136.56, 139.25, 147.96, 148.32, 149.37, 150.46, 154.54; MS *m/z* (%): 393.15 (M⁺, 41.18), 104.20 (100). Anal. calcd for C₁₉H₁₂ClN₅O₃ (393.78): C, 57.95; H, 3.07; N, 17.78; found: C, 57.69; H, 3.30; N, 17.20%.

4-Amino-7-(3-nitrophenyl)-5-(thiophen-2-yl)pyrido[2,3-d] pyrimidin-2(1H)-one (7b): Crystallised from ethanol. Yield 80%; m.p.>300 °C; IR v_{max} /cm⁻¹: 3419, 3350, 3197 (NH, NH₂), 3100 (CH aromatic), 1676 (C=O),1540, 1368 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 6.88 (s, 1H, NH, exch. D₂O), 7.13–7.92 (m, 8H, 7ArH+H6), 11.22 (s, 2H, NH₂, exch.D₂O); MS m/z (%): 365.24 (M⁺, 58.59), 359 (100). Anal. calcd for C₁₇H₁₁N₅O₃S (365.37): C, 55.88; H, 3.03; N, 19.17; found: C, 55.83; H, 3.40; N, 19.30%.

Synthesis of 4-amino-5,7-disubstituted pyrido[2,3-d]pyrimidine-2(1H)-thiones (8a,b); general procedure

A mixture of thiourea (1.52 g, 0.02 mol) and the appropriate 2-aminopyridine-3-carbonitrile **5a,b** (0.01 mol) was heated in an oilbath at 120–130 °C for 2 h then the temperature was raised to 180 °C for additional 2 h. The residue was washed with water then ethanol and finally crystallised from appropriate solvent.

4-*Amino*-7-(4-chlorophenyl)-5-(3-nitrophenyl)pyrido[2,3-d] pyrimidine-2(*IH*)-thione **(8a):** Crystallised from n-butanol. Yield 62%; m.p. 262–264 °C; IR v_{max} /cm⁻¹: 3450–3300 (NH, NH₂), 3050 (CH aromatic), 1090 (C=S), 1560, 1330 (NO₂); ¹H NMR (DMSO– d_6) & ppm: 6.97 (s, 1H, *H6*), 7.04 (s, 1H, N*H*₂, exch. D₂O), 7.62–7.90 (m, 3H, *H4*, *H5* and *H6* of 3-NO₂C₆H₄), 7.88 (d, 2H, *J*=7.6 Hz, *H3*, *H5* of 4-ClC₆H₄), 8.15 (d, 2H, *J*=7.6 Hz, *H2*, *H6* of 4-ClC₆H₄), 8.51(s, 2H, N*H*₂, exch. D₂O); MS *m/z* (%): 412 (M+2, 0.6), 410 (M⁺, 1.9), 68 (100). Anal. calcd for C₁₉H₁₂ClN₅O₂S (409.85): C, 55.68; H, 2.95; N, 17.09; found: C, 55.85; H, 2.68; N, 17.10%.

4-*Amino*-7-(3-*nitrophenyl*)-5-(*thiophen*-2-*yl*)*pyrido*[2,3-d] *pyrimidine*-2(*1H*)-*thione* (8b): Crystallised from ethanol. Yield 55%; m.p. 178–180 °C; IR v_{max} /cm⁻¹: 3377, 3277, 3176 (NH, NH₂), 3096 (CH aromatic), 1520, 1340 (NO₂), 1087 (C=S); ¹H NMR (DMSO-*d*₆) & ppm: 6.93 (s, 1H, N*H*, exch.D₂O), 7.42–8.07 (m, 8H, 7ArH+*H*6), 11, 32 (s, 2H, N*H*, exch.D₂O); MS *m/z* (%): 381.762 (M⁺, 100). Anal. calcd for C₁₇H₁₁N₅O₂S₂ (381.42): C, 53.53; H, 2.91; N, 18.36; found: C, 53.44; H, 3.26; N, 18.08%.

Synthesis of 2,4-diamino-5,7-disubstituted pyrido[2,3-d]pyrimidines (9a,b); general procedure

Guanidine hydrochloride (2.1 g, 0.025 mol) was added to a solution of sodium (2.6 g, 0.113 mol) in methanol (95 mL), the precipitated sodium chloride was filtered off. The *o*-amino cyano derivative **5a**,**b** (0.015 mol) was added to the filtrate. The reaction mixture was heated under reflux for 6 h. The mixture was cooled and the separated solid was filtered, washed well with ethanol and crystallised from butan-1-ol.

7-(4-Chlorophenyl)-5-(3-nitrophenyl)pyrido[2,3-d]pyrimidine-2,4-diamine (9a): Yield 65%; m.p.>300 °C; IR v_{max}/cm^{-1} : 3410– 3200 (NH₂), 3050 (CH aromatic), 1540, 1350 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 6.95 (s, 2H, NH₂, exch.D₂O), 7.34–8.51 (m, 11H, 8ArH+H6+NH₂, exch. D₂O); MS m/z (%): 395 (M+3, 45.16), 393 (M+1, 67.74), 258 (M-Cl+1, 61.29), 213.65 (100). Anal. calcd for C₁₉H₁₃ClN₆O₂ (392.80): C, 58.10; H, 3.34; N, 21.40; found: C, 58.10; H, 3.39; N, 21.73%.

7-(3-Nitrophenyl)-5-(thiophen-2-yl)pyrido[2,3-d]pyrimidine-2,4-diamine (9b): Yield 68%; m.p.>300 °C; IR v_{max} /cm⁻¹: 3463, 3349, 3245 (NH₂), 3019 (CH aromatic), 1566, 1380 (NO₂); ¹H NMR (DMSO- d_6) δ ppm: 5.02 (s, 2H, NH₂, exch.D₂O), 7.23–8.28 (m, 10H, 7ArH+H6+NH₂, exch.D₂O); ¹³C NMR (DMSO) δ ppm: 94.01, 112.12, 114.17, 114.70, 117.0, 119.0, 120.28, 129.44, 130.07, 133.71, 135.26, 154.02, 154.13, 159.39, 159.70, 160.43. Anal. calcd for C₁₇H₁₂N₆O₂S (364.37): C, 56.03; H, 3.32; N, 23.06; found: C, 56.30; H, 3.29; N, 22.97%.

Synthesis of 5,7-disubstituted pyrido[2,3-d]pyrimidine-2,4(1H,3H)-dithiones (10a,b); *general procedure*

Carbon disulfide (3.04 g, 0.04 mol) was added to a solution of 2-aminopyridine-3-carbonitriles **5a,b** (0.01 mol) in pyridine (15 mL), the mixture was heated under reflux for 15 h. After cooling excess pyridine was removed by distillation, the precipitated solid was filtered, washed with ethanol and finally crystallised from appropriate solvent.

7-(4-Chlorophenyl)-5-(3-nitrophenyl)pyrido[2,3-d]pyrimidine-2,4(1H,3H)-dithione (10a): Crystallise from DMF/H₂O. Yield 62%; m.p. 266–268 °C; IR v_{max} /cm⁻¹: 3360, 3200 (NH), 3035 (CH aromatic), 1527, 1346 (NO₂), 1100 (C=S); ¹H NMR (DMSO–d₆) δ ppm: 6.95–8.5 (m, 9H, 8ArH+H6), 7.03 (s, 1H, N1H, exch.D₂O), 8.5 (s, 1H, N3H, exch. D₂O); MS *m/z* (%): 428 (M+2, 10.4), 426 (M⁺,3.7), 374 (100). Anal. calcd for C₁₉H₁₁ClN₄O₂S₂ (426.90): C, 53.46; H, 2.60; N, 13.12; found: C, 53.35; H, 2.29; N, 13.28%. 7-(3-Nitrophenyl)-5-(thiophen-2-yl)pyrido[2,3-d]pyrimidine-2,4(1H,3H)-dithione (10b): Crystallise from ethanol, Yield 58%; m.p. 258–260 °C; IR v_{max} /cm⁻¹: 3377, 3277 (NH), 3095 (CH aromatic), 1521, 1340 (NO₂), 1085 (C=S); ¹H NMR (DMSO-*d*₆) δ ppm: 6.89 (s, 1H, N1*H* exch. D₂O), 7.06 (s, 1H, N3*H*, exch D₂O), 7.17–8.6 (m, 8H, 7ArH+*H*6); MS *m*/z (%): 398.30 (M⁺, 100). Anal. calcd for C₁₇H₁₀N₄O₂S₃ (398.48): C, 51.24; H, 2.53; N, 14.06; found: C, 51.60; H, 2.92; N, 14.28%.

Synthesis of 4-amino-3-phenyl-5,7-disubstitutedpyrido[2,3-d] pyrimidine-2(3H)-thiones (11a,b); general procedure

A few drops of triethylamine was added to a mixture of the appropriate 2-aminopyridine-3-carbonitrile 5a,b (0.005 mol) and phenyl isothiocyanate (0.67 g, 0.005 mol) in DMF (30 mL). The mixture was heated under reflux for 10 h, then left to cool and finally poured onto cold water (25 mL). The formed solid was filtered, washed with ethanol and crystallised from butan-1-ol.

 $\begin{array}{l} 4-A\min o-7-(4-ch \log n \log n \log n)-5-(3-nitroph en yl)-3-\\ phenylpyrido[2,3-d]pyrimidine-2(3H)-thione (11a): Yield 62%; m.p. 262-263 °C; IR v_{max} cm^{-1}: 3377, 3277 (NH_2), 3095 (CH aromatic), 1521, 1340 (NO_2), 1083 (C=S); ¹H NMR (DMSO-d_6) \delta ppm: 7.02 (s, 2H, NH_2, exch.D_2O), 7.02-8.52 (m, 14H, 13ArH+H6); MS m/z (%): 487.2(M+2, 1.2), 485.1(M⁺, 3.7), 85.9 (100) Anal. calcd for C_{23}H_{16}ClN_5O_2S (485.94): C, 61.79; H, 3.32; N, 14.41; found: C, 61.75; H, 3.24; N, 14.45%. \end{array}$

4-Amino-7-(3-nitrophenyl)-3-phenyl-5-(thiophen-2-yl)pyrido[2,3-d] pyrimidine-2(3H)-thione (11b): Yield 58%; m.p. 188–190 °C; IR $v_{max}/$ cm⁻¹: 3350, 3250 (NH₂), 3050 (CH aromatic),1520, 1370 (NO₂) 1040 (C=S); ¹H NMR (DMSO- d_6) δ ppm: 6.98 (s, 2H, NH₂, exch.D₂O), 7.2–7.84 (m, 12H, ArH), 7.99 (s, 1H, H6). Anal. calcd for C₂₃H₁₅N₅O₂S₂ (457.51): C, 60.37; H, 3.30; N, 15.30; found: C, 60.69; H, 3.05; N, 15.00%.

4-Amino-7-(4-chlorophenyl)-5-(3-nitrophenyl)-2-phenyl-1,8-naphthyridine-3-carbonitrile (12): A mixture of the benzylidenemalononitriles (0.15 g; 0.001 mol), and corresponding 2-aminopyridine-3-carbonitrile derivatives 5a (0.35 g; 0.001 mol) and piperidine (1 mL) was dissolved in dioxane (10 mL). The mixture was heated under reflux for 3 h. The solvent was removed under reduced pressure. The formed solid was filtered, washed and crystallised from DMF/H₂O. Yield 74%; m.p. 273–274 °C; IR v_{max}/cm⁻¹: 3300 (NH₂), 3050 (CH aromatic), 2200 (CN), 1520, 1340 (NO₂); ¹H NMR $(DMSO-d_s) \delta ppm: 6.94 (s, 1H, H-6 of naphthridine), 6.96 (s, 2H, NH_2)$ exch.D₂O), 7.59–8.47 (m, 13H, ArH).; MS m/z (%): 479.75 (M+3, 9.3), 478 (M+1, 9.3), 374 ($M-C_6H_4CN$, 100). Anal. calcd for $C_{22}H_{16}CIN_5O_2$ (477.88): C, 67.85; H, 3.37; N, 14.65; found: C, 68.05; H, 3.70; N, 14.74%. 4-Amino-7-(4-chlorophenyl)-5-(3-nitrophenyl)-2-oxo-1,2-dihydro-1,8-naphthyridine-3-carbonitrile (13): A mixture of 2-amino-6-(4chlorophenyl)-4-(3-nitrophenyl)nicotinonitrile (5a) (3.50 g, 0.01 mol), ethyl cyanoacetate (1.13 g, 0.01 mol) and triethylamine (2 mL) in ethanol 95% (30 mL) was heated under reflux for 3 h. The formed solid was filtered, washed with ethanol and crystallised from ethanol. Yield 58%; m.p. 270–272 °C; IR v_{max} /cm⁻¹: 3480, 3360, 3200 (NH₂, NH), 3050 (CH aromatic), 1640 (C=O), 1520, 1350 (NO₂); ¹H NMR (DMSO-d_c) & ppm: 7.99 (s, 2H, NH, exch.D₂O), 6.99-8.4 (m, 9H, 8ArH+H6), 8.51(s, 1H, NH, exch.D₂O); MS m/z (%): 419.75 (M+3, 16.67), 418.25 (M+1, 18.33), 55 (100). Anal. calcd for C₂₁H₁₂ClN₅O₃ (417.79): C, 60.36; H, 2.89; N, 16.76; found: C, 60.69; H, 3.19; N, 16.36%.

Cytotoxic activity studies

Anticancer activity studies were done at Cairo University, National Cancer Institute, Cancer Biology Department, Pharmacology Unit.

Measurement of potential cytotoxicity: The cytotoxic activity of the newly synthesised compounds was measured *in vitro* on human breast adenocarcinoma cell line (MCF-7) using Sulforhodamine-B stain (SRB) assay applying the method of Skehan *et al.* ²⁶ Cells were plated in 96-multiwell plate (104 cells/well) for 24 h before treatment with the test compounds to allow attachment of the cells to the wall of the plate. Test compounds were dissolved in DMSO and diluted with saline to the appropriate volume. Different concentrations of the test compounds (0, 1, 2.5, 5 and 10 mg mL⁻¹) were added to the cell monolayer. Triplicate

wells were prepared for each individual dose. Monolayer cells were incubated with the test compound for 48 h at 37 °C in atmosphere of 5% CO_2 . After 48 h, cells were fixed with trichloroacetic acid, washed with water and stained for 30 min with 0.4% (wt/vol). Sulforhodamine-B stain dissolved with 1% acetic acid. Excess stain was removed by four washes with 1% acetic acid and attached stain was recovered with Tris EDTA buffer. Colour intensity was measured in an ELISA reader. The relation between surviving fraction and compound concentration was plotted and IC₅₀ [the concentration required for 50% inhibition of cell viability] was calculated for each compound and results are given in Table 1.

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