

## Note

### Direct fluorination at positions 3', 4', and 6' of $\beta$ -D-glucopyranosyltheophylline

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Considerable effort has been expended in the synthesis of deoxyfluoro sugars<sup>1-6</sup> as a result of interest in such analogs for use as hexokinase inhibitors<sup>7</sup>, as carriers of <sup>18</sup>F in positron emission tomography<sup>8</sup> (PET), and as possible anticancer agents<sup>9</sup>. Also, fluorinated pentofuranose nucleosides have been investigated for the treatment of infections by *Herpes simplex*<sup>10</sup> (HSV) and human immunodeficiency<sup>11</sup> (HIV) viruses.

Little has been reported on fluorinated hexopyranose nucleosides<sup>12</sup> and, to our knowledge, the use of diethylaminosulfur trifluoride<sup>14</sup> (DAST) has not been reported hitherto for their synthesis.

We now describe the synthesis of 3'-, 4'-, and 6'-fluoro derivatives of  $\beta$ -D-glucopyranosyltheophylline.

DAST can effect<sup>2,13,14</sup> the replacement of OH by F at primary, secondary, and tertiary positions (the last two with inversion of configuration) in high yields.

7-(2,3,4-Tri-*O*-benzoyl-6-deoxy-6-fluoro- $\beta$ -D-glucopyranosyl)theophylline (**3**) was prepared in dry diglyme<sup>3</sup> at 60° from the tribenzoate **2**. The key precursor of 7-(2,4,6-tri-*O*-benzoyl-3-deoxy-3-fluoro- $\beta$ -D-allopyranosyl)theophylline (**13**) was 7-(4,6-*O*-isopropylidene- $\beta$ -D-glucopyranosyl)theophylline (**5**), which reacted with DAST in CH<sub>2</sub>Cl<sub>2</sub> containing 4-dimethylaminopyridine to afford 7-(3-deoxy-3-fluoro-4,6-*O*-isopropylidene- $\beta$ -D-allopyranosyl)theophylline (**11**). 7-(2,3,6-Tri-*O*-benzoyl-4-deoxy-4-fluoro- $\beta$ -D-galactopyranosyl)theophylline (**9**) was obtained by the same procedure from 7-(2,3,6-tri-*O*-benzoyl- $\beta$ -D-glucopyranosyl)theophylline (**8**). If the 4-dimethylaminopyridine was omitted, then 7-(2,3,6-tri-*O*-benzoyl-4-deoxy- $\beta$ -D-erythro-hex-4-enopyranosyl)theophylline was obtained instead of **9**. Attempts to fluorinate 7-(3,4,6-tri-*O*-benzoyl- $\beta$ -D-glucopyranosyl)theophylline were unsuccessful even in hot diglyme<sup>15</sup>, probably because of steric hindrance in the formation of the reaction intermediate.

Debenzoylation (Zemplén) of **3** and **9** gave **4** and **10**, respectively, and deacetalation of **11** gave **12**.

All the compounds synthesised were characterised by their proton-coupled <sup>19</sup>F- and proton-decoupled <sup>13</sup>C-n.m.r. spectra. The n.m.r. data given in Tables I-III

TABLE I

<sup>1</sup>H-N.M.R. DATA ( $\delta$  IN P.P.M.,  $J$  IN HZ)

Compound	H-1'	H-2'	H-3'	H-4'	H-5'	H-6'a	H-6'b	J <sub>1,2</sub>	J <sub>2,3</sub>	J <sub>3,4'</sub>	J <sub>4,5'</sub>	J <sub>5,6'a</sub>	J <sub>5,6'b</sub>	J <sub>6'a6'b</sub>
2 <sup>a</sup>	6.48	6.17	6.04	5.74	3.80	4.10	3.91	9.3	9.5	9.5	9.1	3.5	2.8	10.0
3 <sup>a</sup>	6.56	6.06	6.12	5.77	4.33	4.71	4.55	8.8	9.4	9.7	9.3	2.5	2.7	7.3
4 <sup>b</sup>	5.83	4.10	3.71	—	3.82	4.74	4.74	9.1	9.0	10.0	8.6	—	—	—
8 <sup>a</sup>	6.21	5.98	5.68	4.32	←3.85-4.05→			9.3	9.5	9.4	9.1	—	—	—
9 <sup>a</sup>	6.55	6.18	5.65	5.29	4.63	4.52	4.43	9.2	9.5	2.2	0	4.0	—	6.3
10 <sup>b</sup>	5.85	4.44	3.99	4.99	4.10	←3.95-3.88→			8.2	0	0	2.3	7.4	—
5 <sup>a</sup>	5.78	4.08	3.79	3.92	←3.85-3.69→			5.2	7.0	5.4	10.7	—	—	—
11 <sup>a</sup>	5.92	4.20	5.08	4.00	4.00	←3.89-3.78→			9.2	0.5	—	—	—	—
12 <sup>b</sup>	6.05	4.49	5.20	4.00	←4.14-3.78→			9.4	0.5	0	—	—	—	—
13 <sup>a</sup>	5.64	4.80	5.50	4.80	4.49	3.92	3.53	6.9	0.5	0.5	9.2	3.8	—	12.1

<sup>a</sup>In CDCl<sub>3</sub> (internal Me<sub>4</sub>Si). <sup>b</sup>In D<sub>2</sub>O [internal 3-(trimethylsilyl)-1-propanesulfonic acid, sodium salt].

TABLE Ia

	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	X
1	H	OBz	OBz	H	H	OBz	OTr
2	H	OBz	OBz	H	H	OBz	OH
3	H	OBz	OBz	H	H	OBz	F
4	H	OH	OH	H	H	OH	F
5	H	OH	OH	H	H	O-CMe <sub>2</sub> -O	
6	H	OBz	OBz	H	H	O-CMe <sub>2</sub> -O	
7	H	OBz	OBz	H	H	OH	OH
8	H	OBz	OBz	H	H	OH	OBz
9	H	OBz	OBz	H	F	H	OBz
10	H	OH	OH	H	F	H	OBz
11	H	OH	H	F	H	O-CMe <sub>2</sub> -O	
12	H	OH	H	F	H	OH	OH
13	H	OBz	H	F	H	OBz	OBz

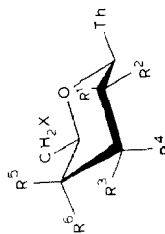


TABLE II

<sup>19</sup>F-N.M.R. DATA ( $\delta$  IN P.P.M.,  $J$  IN HZ)

Compound	F	$J_{F-6',H-6'}$	$J_{F-6',H-5'}$	$J_{F-4',H-4'}$	$J_{F-4',H-3'}$	$J_{F-4',H-5'}$	$J_{F-3',H-3'}$	$J_{F-3',H-4'}$	$J_{F-3',H-2'}$
<b>3<sup>a</sup></b>	-56.3	46.6	21.2						
<b>4<sup>b</sup></b>	-56.2	47.3	25.4						
<b>9<sup>a</sup></b>	-48.2			52.1	25.6	25.6			
<b>10<sup>b</sup></b>	-46.8			51.4	30.9	30.9			
<b>11<sup>a</sup></b>	-50.6						54.0	27.0	27.0
<b>12<sup>b</sup></b>	-48.2						53.6	28.9	28.9
<b>13<sup>a</sup></b>	-48.1						53.7	28.9	28.9

<sup>a</sup>In CDCl<sub>3</sub> (internal C<sub>6</sub>F<sub>6</sub>). <sup>b</sup>In D<sub>2</sub>O (external C<sub>6</sub>F<sub>6</sub>).

establish the location of the fluorine substituents and the configuration of the products.

The <sup>13</sup>C-signals (Table III) were assigned by comparing the spectra with those of the parent compounds. Replacement of OH by F causes a downfield shift of the resonance of the attached carbon by 15–20 p.p.m., upfield shifts of the signals of the adjacent carbons by 2–5 p.p.m., and <sup>1</sup>J<sub>F,C</sub> values of 180–200 Hz (Table III). Thus, the locations of the fluorine substituents were shown to be at position 6 in **3** and **4**, at position 4 in **9** and **10**, and at position 3 in **11–13**.

For **9–13**, the <sup>1</sup>J<sub>F,C</sub> values (Table III) are in accordance with the results of Wray<sup>16</sup> (175.9–183.1 Hz), but those for **3** and **4** (177.5 and 169.5) are higher than the previous values (~167.5)<sup>16</sup>. The  $J_{F-6,C-5}$  value of 18.5 Hz for **3** and **4** is identical to that for 6-deoxy-6-fluoro- $\beta$ -D-glucopyranose<sup>16</sup>. This author recorded <sup>2</sup>J<sub>F,C</sub> values of ~17.5 when the oxygen function on the coupled carbon was *gauche* to the coupled fluorine, but smaller values were found for **9** and **11–13** (Table III).

In the <sup>1</sup>H- and <sup>19</sup>F-n.m.r. spectra (Tables I and II, respectively), <sup>2</sup>J<sub>F,H</sub> was 46.6–47.3 for **3** and **4**, 51.4–52.1 for **9** and **10**, and 53.6–54.0 for **11–13**. These values agree with other data<sup>2,17</sup> on deoxyfluorosugars. The  $J_{F-6,H-5}$  values of 21.2–25.4 for **3** and **4** accord with other observations<sup>3</sup> and indicate the preponderance of the rotamer with F-6 antiperiplanar to H-5. The values of <sup>3</sup>J<sub>F,H</sub> for vicinal *trans*-diaxial nuclei in pyranose structures are generally<sup>18</sup> in the range of 21–30 Hz. The observed <sup>3</sup>J<sub>F,H</sub> values 25.6–30.9 for **9–13** confirm the vicinal *trans*-diaxial H/F relationships. Therefore, the fluorine substituent is axial and the compounds have the <sup>4</sup>C<sub>1</sub>(D) conformation, which shows that replacement of OH by F does not affect the conformation<sup>6</sup>.

## EXPERIMENTAL

*General methods.* — Melting points are uncorrected. T.l.c. was performed on Silica Gel 60-F<sub>254</sub> (Merck) and flash chromatography on Silica Gel 60 (240–400

TABLE III

<sup>13</sup>C-N.M.R. DATA ( $\delta$  IN P.P.M.,  $J$  IN Hz)

Compound	C-1'	C-2'	C-3'	C-4'	C-5'	C-6'	$J_{F6',C6'}$	$J_{F6',C5'}$	$J_{F6',C4'}$	$J_{F4',C4'}$	$J_{F4',C5'}$	$J_{F4',C3'}$	$J_{F3',C3'}$	$J_{F3',C2'}$	$J_{F3',C4'}$	$J_{F3',C1'}$
2 <sup>a</sup>	82.7	71.3	72.8	68.7	77.4	60.9										
3 <sup>a</sup>	82.5	71.9	73.8	68.3	75.8	81.0	177.5	18.5	5.4							
4 <sup>b</sup>	80.0	71.8	76.1	70.7	74.5	80.2	169.5	18.5								
8 <sup>a</sup>	83.2	71.1	79.2	66.6	—	61.7										
9 <sup>a</sup>	82.3	68.9	72.0	86.3	74.3	61.9			186.7	17.6	16.0					
10 <sup>b</sup>	84.1	71.7	72.9	82.9	75.4	62.4			183.5	13.9	14.8					
5 <sup>a</sup>	85.7	72.9	74.6	75.0	62.0	70.8							181.2	14.8	14.8	3.7
11 <sup>a</sup>	84.3	69.9	90.1	70.7	61.9	66.9							175.4	14.8	14.8	
12 <sup>b</sup>	85.3	68.0	95.8	71.3	78.7	63.3							177.6	9.1	14.8	7.4
13 <sup>a</sup>	80.7	69.7	87.8	67.2	72.2	62.4										

<sup>a</sup>In CDCl<sub>3</sub> (internal Me<sub>4</sub>Si). <sup>b</sup>In D<sub>2</sub>O [internal 3-(trimethylsilyl)-1-propanesulfonic acid, sodium salt].

mesh, Merck). N.m.r. spectra were recorded at room temperature with a Bruker 300 MSL spectrometer.

**7-(2,3,4-Tri-O-benzoyl- $\beta$ -D-glucopyranosyl)theophylline (2).** — A solution of  $\beta$ -D-glucopyranosyltheophylline<sup>19</sup> (2 g) and trityl chloride (1.7 g) in anhydrous pyridine (100 mL) was boiled under reflux overnight. Benzoyl chloride (2.2 mL) was then added and, after 2 h, the mixture was cooled, diluted with dichloromethane, washed twice with ice-water, and concentrated. Crystallisation of the residue from ethanol gave 7-(2,3,4-O-benzoyl-6-trityl- $\beta$ -D-glucopyranosyl)theophylline (**1**; 4 g, 77%), m.p. 227–228°,  $[\alpha]_D^{20}$   $-22^\circ$  (c 0.1, methanol).

*Anal.* Calc. for  $C_{53}H_{44}N_4O_9$ : C, 72.27; H, 5.00; N, 6.36. Found: C, 72.76; H, 4.96; N, 6.47.

A solution of **1** (5.6 g, 6 mmol) in 3:2 HCOOH-ether<sup>20</sup> (20 mL) was kept at room temperature for 30 min, then diluted with ether (150 mL), washed successively with brine and with saturated aq. sodium hydrogencarbonate, dried, and concentrated. Flash chromatography (EtOAc-hexane, 7:3) of the syrupy residue gave **2** (3.3 g, 80%), m.p. 146–147° (from methanol),  $[\alpha]_D^{20}$   $-5^\circ$  (c 0.1, methanol).

*Anal.* Calc. for  $C_{34}H_{30}N_4O_{10} \cdot H_2O$ : C, 60.71; H, 4.76; N, 8.33. Found: C, 60.62; H, 4.89; N, 8.17.

**7-(2,3,4-Tri-O-benzoyl-6-deoxy-6-fluoro- $\beta$ -D-glucopyranosyl)theophylline (3).** — DAST (0.585 mL, 6 mmol) was added dropwise to a stirred solution of **2** (980 mg, 1.5 mmol) in dry diglyme (6 mL) at  $-20^\circ$ . The temperature was raised slowly to  $60^\circ$  and maintained for 1 h. The mixture was poured into ice, the precipitate was collected, and a solution in  $CH_2Cl_2$  was washed once with M  $NaHCO_3$  and twice with water, then concentrated. Flash chromatography (dichloromethane-ethyl acetate, 4:1) of the resulting oil gave **3** (615 mg, 68%), m.p. 179–180° (from MeOH),  $[\alpha]_D^{20}$   $-4^\circ$  (c 0.1, methanol).

*Anal.* Calc. for  $C_{34}H_{29}FN_4O_9 \cdot 0.5CH_3OH$ : C, 61.62; H, 4.58; F, 2.83; N, 8.32. Found: C, 61.68; H, 4.47; F, 2.80; N, 8.30.

**7-(6-Deoxy-6-fluoro- $\beta$ -D-glucopyranosyl)theophylline (4).** — To a solution of **3** (665 mg, 1 mmol) in anhydrous methanol (10 mL) was added methanolic M sodium methoxide (1 mL). The mixture was stirred for 2 h at room temperature, neutralised with Amberlite IR-120 ( $H^+$ ) resin, and concentrated. Recrystallisation of the residue from methanol gave **4** (310 mg, 90%), m.p. 218–220°,  $[\alpha]_D^{20}$   $-12^\circ$  (c 0.12, methanol).

*Anal.* Calc. for  $C_{13}H_{17}FN_4O_6 \cdot CH_3OH$ : C, 44.68; H, 5.58; F, 5.05; N, 14.89. Found: C, 44.89; H, 5.53; F, 4.92; N, 15.0.

**7-(4,6-O-Isopropylidene- $\beta$ -D-glucopyranosyl)theophylline (5).** — A mixture of 7- $\beta$ -D-glucopyranosyltheophylline<sup>19</sup> (1.8 g), 2,2-dimethoxypropane (5 mL), and dry *p*-toluenesulfonic acid (15 mg) in dry *N,N*-dimethylformamide (15 mL) was stirred at room temperature for 20 h. M  $NaHCO_3$  (1 mL) was added and the mixture concentrated under vacuum. A solution of the residue in dichloromethane was filtered and concentrated, and the residue was crystallised from ethanol to give **5** (1.39 g, 70%), m.p. 208–210°,  $[\alpha]_D^{20}$   $-33^\circ$  (c 0.13, methanol).

*Anal.* Calc. for  $C_{16}H_{22}N_4O_7$ : C, 50.26; H, 5.76; N, 14.66. Found: C, 50.06; H, 5.80; N, 14.65.

*7-(2,3-Di-O-benzoyl- $\beta$ -D-glucopyranosyl)theophylline (7).* — Conventional treatment of **5** with pyridine and benzoyl chloride and crystallisation of the product from ethanol gave *7-(2,3-di-O-benzoyl-4,6-O-isopropylidene- $\beta$ -D-glucopyranosyl)-theophylline (6, 90%),* m.p. 136–139°,  $[\alpha]_D^{20} -33^\circ$  (c 0.13, methanol).

Amberlite IR-120 ( $H^+$ ) resin was added to a methanolic solution of **6**, which was boiled under reflux for 1 h, then filtered, and concentrated. The syrupy residue was crystallised from ethanol to give **7** (90%), m.p. 225–226°,  $[\alpha]_D^{20} -12^\circ$  (c 0.1, methanol).

*Anal.* Calc. for  $C_{27}H_{26}N_4O_8 \cdot H_2O$ : C, 57.04; H, 4.93; N, 9.86. Found: C, 57.23; H, 4.81; N, 10.08.

*7-(2,3,6-Tri-O-benzoyl- $\beta$ -D-glucopyranosyl)theophylline (8).* — To a stirred solution of **7** (500 mg, 1 mmol) in pyridine (50 mL) at  $-30^\circ$  was added, dropwise, 1 equiv. (0.116 mL) of benzoyl chloride. The mixture was stored at  $0^\circ$  for 72 h and then worked-up in the usual manner. Flash chromatography (ethyl acetate–hexane, 70:30) of the product gave **8** (460 mg, 70%), m.p. 226° (from EtOH),  $[\alpha]_D^{20} -12^\circ$  (c 0.13, methanol).

*Anal.* Calc. for  $C_{34}H_{30}N_4O_{10}$ : C, 62.38; H, 4.59; N, 8.56. Found: C, 62.30; H, 4.63; N, 8.56.

*7-(2,3,6-Tri-O-benzoyl-4-deoxy-4-fluoro- $\beta$ -D-galactopyranosyl)theophylline (9).* — To a stirred solution of **8** (654 mg, 1 mmol) and 4-dimethylaminopyridine (244 mg, 2 mmol) in dry dichloromethane (20 mL) at  $-30^\circ$  was added DAST (0.195 mL, 2 mmol) during 15 min under nitrogen. The mixture was then allowed to attain room temperature. After 24 h, the mixture was cooled to  $0^\circ$ , methanol was added, solvents were removed under vacuum, and the resulting oil was poured into ice–water. The precipitate was collected, and a solution in  $CH_2Cl_2$  was washed with M  $NaHCO_3$  and water to pH 7, dried, and concentrated. Flash chromatography (dichloromethane–ethyl acetate, 9:1) of the residue gave **9** (446 mg, 68%), m.p. 125–126° (from MeOH),  $[\alpha]_D^{20} +25^\circ$  (c 0.1, ethyl acetate).

*Anal.* Calc. for  $C_{34}H_{29}FN_4O_9$ : C, 62.20; H, 4.42; F, 2.90; N, 8.54. Found: C, 62.05; H, 4.38; F, 2.29; N, 8.29.

*7-(4-Deoxy-4-fluoro- $\beta$ -D-galactopyranosyl)theophylline (10).* — Debenzoylation of **9**, as for **3**, gave **10** (90%), m.p. 215–216° (from MeOH),  $[\alpha]_D^{20} +32^\circ$  (c 0.1 methanol).

*Anal.* Calc. for  $C_{13}H_{17}FN_4O_6 \cdot 0.5CH_3OH$ : C, 45.0; H, 5.27; F, 5.27; N, 15.55. Found: C, 44.84; H, 5.22; F, 5.23; N, 15.52.

*7-(3-Deoxy-3-fluoro-4,6-O-isopropylidene- $\beta$ -D-allopyranosyl)theophylline (11).* — A solution of **5** (1.53 g, 4 mmol) and 4-dimethylaminopyridine (1 g, 8 mmol) in dry  $CH_2Cl_2$  at  $-20^\circ$  was treated with DAST (0.8 mL, 8 mmol) as described for **9**. Column chromatography (ethyl acetate–hexane, 3:1) of the product gave **11** (740 mg, 43%), m.p. 235–236° (from EtOH),  $[\alpha]_D^{20} -13^\circ$  (c 0.1, ethyl acetate).

*Anal.* Calc. for  $C_{16}H_{21}FN_4O_6 \cdot 0.5H_2O$ : C, 48.85; H, 5.60; F, 4.83; N, 14.24. Found: C, 49.01; H, 5.59; F, 4.90; N, 14.21.

7-(3-Deoxy-3-fluoro- $\beta$ -D-allopyranosyl)theophylline (**12**). — To a solution of **11** in methanol was added Amberlite IR-120 ( $H^+$ ) resin, and the mixture was boiled under reflux for 90 min, then filtered, and concentrated to give **12** (90%), m.p.  $246^\circ$  (from EtOH),  $[\alpha]_D^{20} -22^\circ$  (c 0.1, methanol).

*Anal.* Calc. for  $C_{13}H_{17}FN_4O_6 \cdot 0.25C_2H_5OH$ : C, 45.56; H, 5.20; F, 5.34; N, 15.75. Found: C, 45.58; H, 5.13; F, 5.30; N, 15.80.

7-(2,4,6-Tri-O-benzoyl-3-deoxy-3-fluoro- $\beta$ -D-allopyranosyl)theophylline (**13**). — Conventional treatment of **12** with benzoyl chloride–pyridine gave **13** (80%), m.p.  $126$ – $127^\circ$  (from MeOH),  $[\alpha]_D^{20} -35^\circ$  (c 0.1, methanol).

*Anal.* Calc. for  $C_{34}H_{29}FN_4O_9$ : C, 62.20; H, 4.40; F, 2.90; N, 8.50. Found: C, 62.20; H, 4.34; F, 2.83; N, 8.40.

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