

## EXPERIMENTAL

**Plant material.** Fronds of *Asplenium adiantum-nigrum* L. were collected on Mount Etna, Sicily.

**Isolation procedure.** Fresh fronds of *Asplenium adiantum-nigrum* were homogenized and extrd  $3 \times$  with hot EtOH. The combined extracts were filtered, concd to small vol. *in vacuo* and re-filtered. The xanthone was isolated by successive prep. PC in BAW, 5% HOAc and BEW.  $R_f$  data are: BAW 0.45, 5% HOAc 0.25, BEW 0.17. Total acid hydrolysis was carried out with 2 N HCl (1 hr at 100° under  $N_2$ ); controlled acid hydrolysis was carried out with 10% aq. HOAc (2.5 hr under reflux). Treatment with  $\beta$ -glucosidase was carried out in citrate-phosphate buffer, pH 4.5, at 37° for 20 hr. 1,3,7,8-Tetrahydroxyxanthone was identified by UV spectral analysis with shift reagents [2,4], MS and comparison with an authentic sample (TLC on Si gel: 3 solvents); D-glucose and laminaribiose were identified by Co-PC (4 solvents), TLC on Si gel (*n*-BuOH-HOAc-Et<sub>2</sub>O-H<sub>2</sub>O, 9:6:3:1) and GLC of their trimethylsilyl derivatives [13].

**Methylation of xanthone.** The xanthone was methylated ( $Me_2SO_4$ - $K_2CO_3$ - $Me_2CO$ ) and hydrolysed with 0.3 N HCl (4 hr under reflux). The partially methylated aglycone was identified as 1-hydroxy-3,7,8-trimethoxyxanthone by UV spectral analysis with shift reagents [2,4], MS and comparison with an authentic sample (TLC on Si gel; 3 solvents); methylated sugars were identified by PC according to ref. [14] and TLC on Si gel (EtOAc- $CHCl_3$ , 1:1).

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glucoside, Dr. S. Occhipinti (University of Catania) for the mass spectra and Mr. A. D'Urso (Botanic Institute, University of Catania) for help in acquiring the plant material.

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## AN UNSYMMETRICAL DIARYLHEPTANOID FROM *CURCUMA LONGA*

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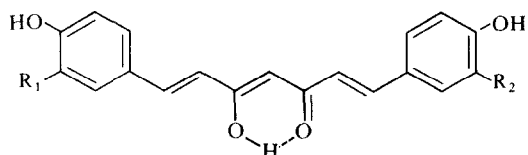
(Revised received 1 February 1980)

**Key Word Index**—*Curcuma longa*; Zingiberaceae; turmeric; diarylheptanoid; dihydrocurcumin; 1,7-bis-(4-hydroxy-3-methoxyphenyl)-hept-1-en-3,5-dione.

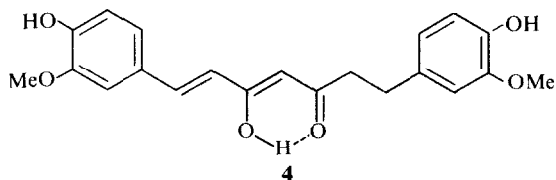
Earlier work on the pigments of turmeric (*Curcuma longa* L.) revealed the presence of three major phenolic diarylheptanoids, namely curcumin (1), feruloyl-(4-hydroxycinnamoyl)methane (2) and bis-(4-hydroxycinnamoyl)methane (3) [1,2]. Biosynthesis and metabolism of diarylheptanoids have recently attracted attention [2,3]. In the course of our work on the metabolism of curcumin [4], we have isolated a new diarylheptanoid from the benzene extract of *Curcuma longa* rhizomes.

This compound,  $C_{21}H_{22}O_6$  ( $M^+$  370), had a UV

maximum at 375 nm and its IR spectrum showed bands at 3400 (OH), 1630 ( $COCH_2CO$ ), 1600 and 1510  $cm^{-1}$  ( $C=C$  and aromatic). It gave the rubrocurcumin reaction [5] with boric acid and oxalic acid (with visible max shifting to 420 nm), a characteristic reaction of  $\beta$ -diketones. The 270 MHz  $^1H$  NMR spectrum in  $DMSO-d_6$  showed signals for two phenolic hydroxyl protons at  $\delta$ 9.65 and 8.76 (disappearing on exchange with  $D_2O$ ), two singlets at 3.74 (3 H) and 3.83 (3 H) (OMe), two doublets at 7.48 and 6.64 ( $J = 14$  Hz) for two *trans*-related olefinic protons, clusters of aromatic proton signals centred around 6.65,



- 1  $R_1 = R_2 = \text{OMe}$   
 2  $R_1 = \text{H}; R_2 = \text{OMe}$   
 3  $R_1 = R_2 = \text{H}$



6.80 and 7.11 for 3, 2 and 1 protons, respectively, and a one-proton singlet at 5.88 attributable to the proton on the central carbon of the  $\beta$ -diketone in its enol form. In addition, it showed two triplets at  $\delta$  2.70 and 2.90 for two protons each ( $J = 6$  Hz) indicating the presence of two methylene groups, possibly attached to each other. These data indicate the compound to be dihydrocurcumin (**4**). This was supported by the mass spectral fragmentation which showed the base peak at  $m/e$  137 for hydroxy-(methoxy)benzyl ion and peaks at 233 for feruloylacetyl-methyl, 219 for feruloylacetyl, 191 for feruloylmethyl and 177 for feruloyl ions.

Structure **4** for this compound was confirmed by the oxidation of the compound with dichlorodicyanobenzoquinone in tetrahydrofuran, which yielded curcumin (**1**). The isolation of an unsymmetrical diarylheptanoid from *Curcuma longa* may be of biogenetic significance [2].

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Mps are uncorr. UV spectra were measured in EtOH and IR as KBr discs.  $^1\text{H}$  NMR spectra were recorded in  $\text{DMSO}-d_6$  at 270 MHz using TMS as int. standard. MS were recorded at 70 eV.

**Isolation.** Dry turmeric powder (7 kg, Allepey variety) was extracted in cold with hexane for 48 hr followed by  $\text{C}_6\text{H}_6$  (48 hr). The  $\text{C}_6\text{H}_6$  extract was coned and the mixture of the three major pigments (**1**, **2** and **3**) which separated was filtered off. The mother liquor was transferred to a Si gel column and eluted with hexane followed by hexane- $\text{C}_6\text{H}_6$  (1:1). The latter fraction was then chromatographed on a column of Sephadex LH-20 in hexane- $\text{CHCl}_3$ -MeOH (2:1:1). Repeated chromatography over Sephadex LH-20 in the above solvent mixture gave **4** as yellow crystals, mp  $178^\circ$  ( $\text{CHCl}_3$ ) (0.4 g). (Found: C, 66.38, H, 5.9.  $\text{C}_{21}\text{H}_{22}\text{O}_6 \cdot 0.5\text{H}_2\text{O}$  requires: C, 66.4; H, 6.0%). UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm (log  $\epsilon$ ): 375 (4.39), 305 (3.73), 280 (3.84), 260 sh, 250 (4.10).  $^1\text{H}$  NMR:  $\delta$  2.69 (2 H, t,  $J = 6$  Hz), 2.78 (2 H, t,  $J = 6$  Hz), 3.74 (3 H, s), 3.82 (3 H, s), 5.88 (1 H, s), 6.65 (3 H, m and 1 H, d,  $J = 14$  Hz), 6.80 (2 H, m), 7.11 (1 H, dd,  $J = 7, 1$  Hz), 7.30 (1 H, s), 7.48 (1 H, d,  $J = 14$  Hz), 8.76 (1 H, s), 9.65 (1 H, s). MS  $m/e$  (rel. int.): 370 (14.4), 233 (6), 219 (28), 191 (28), 177 (88), 150 (32), 137 (100).

**Oxidation of dihydrocurcumin.** The compound (10 mg) was stirred with recrystallized DDQ (6 mg) in THF at room temp. for 30 min. The product was purified by column chromatography over Sephadex LH-20 in hexane- $\text{CHCl}_3$ -MeOH (2:1:1) yielding curcumin, mp and mmp  $181-182^\circ$  (IR superimposable with that of authentic sample).

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