270 Communications SYNTHESIS

showed a molecular ion peak at m/e = 524, which was smaller than the calculated value for the expected adduct 4a by two mass units. The ¹H-N.M.R. spectrum did not show absorptions of methine groups at $\delta = 3.5$ -4.5 ppm, which are typical in Diels-Alder type adducts of dehydrohydantoins. The product showed a high melting point $(325-331^{\circ}\text{C})$ with decomposition). It retained the hydantoin carbonyl absorptions but lacked the imine absorption at $3400-3500 \text{ cm}^{-1}$ in the I.R. spectrum. From these results, it was assigned as 5a, which has a dehydrogenated structure of 4a.

tained in 23% yield. The mass spectrum of this product

Either thermally or in the presence of trifluoroacetic acid, hydantoins 1 proved to react with 3 to give dehydrogenated adducts 5 (Tables 1, 2). Under acid-catalyzed conditions, phenyl- (1a) and benzylhydantoin (1b) gave 5a and 5b quantitatively (Runs 6 and 7).

There are several reports^{7,8} on the Diels-Alder reactions of cyano compounds, in which dehydrogenated products are obtained. However, the addition of imino dienophiles involving oxidative dehydrogenation has not been reported.

The reactivities of the above hydantoins with perylenes were not so high compared with those of cyclic azodienophiles² derived, for example, from 1,4-dioxo-1,2,3,4-tetrahydrophthalazine and lead(IV) acetate. Thus, perylene did not give any adduct with **1a** or **1b** under similar conditions as described in Method A or B in Table 2.

Treatment of 3 with an excess of N-(2,2,2-trichloroethylidene)ethoxy-carbonylamine⁹ in xylene at 130°C for 10 h resulted only in the recovery of 3. The reaction of 3 with N-(2,2,2-trichloroethylidene)-p-to-luene sulfonamide¹⁰ in an autoclave at 160°C for 24 h gave a very complex mixture, in which the corresponding Diels-Alder adduct could not be detected.

Hydrolyses and decarboxylations of adducts 5 were performed according to our previous method. The title compound 6 was obtained in 58-64% yield.

3-Substituted 5-Methoxyhydantoins 1a-e:

Compounds 1a and 1b are prepared from the corresponding hydantoins, bromine, and methanol in acetic acid according to Refs. 15, 12. This procedure is more convenient than the preparation of 1a⁵ from phenylurea and dibutyl oxalate 13, 14. Similarly, the 3-butyl, 3-propyl, and 3-ethyl derivatives 1c-e are prepared from the corresponding alkylhydantoins.

Diels-Alder Reaction of Dehydrohydantoins with Dibenzo[b,n]perylene; The Synthesis of Dibenzo[c,i]phenanthro[1,10,9,8-anmlk]phenanthridine

Sumio Tokita*, Kimihiro Hiruta, Yutaka Yaginuma, Sadayasu Ishikawa, Hisao Nishi

Department of Applied Chemistry, Faculty of Engineering, Saitama University, Shimo-Ohkubo, Urawa-shi, 338 Japan

In continuation of our work ^{1,2,3} on the application of cycloaddition reactions in the synthesis of highly condensed heterocyclic compounds, we report a new preparation of dibenzo[c,i]phenanthro[1,10,9,8-anmlk]phenanthridine (6). The [4+2]-cycloaddition of a conjugated diene to an imine was widely used for the synthesis of nitrogen containing heterocycles⁴. However, the addition of an imine to a fused polycyclic hydrocarbon has hitherto not been reported. 3-Substituted 5methoxyhydantoins (1) are known to lose methanol to form dehydrohydantoins (2) at elevated temperatures^{5,6}.

We selected five hydantoins (1) as precursors of the corresponding dehydrohydantoins (2), and the reaction of 1 with dibenzo[b,n] perylene (3) was studied. When 5-methoxy-3-phenylhydantoin (1a) was heated with 3 in p-xylene in an autoclave at 160° C for 72 h, the cycloaddition product was ob-

Table 1. Compounds 5a-e prepared

Prod- uct	m.p. [°C] (dec.)	Molecular Formula"	M.S. m/e (M+)	I.R. (KBr) ν [cm ⁻¹]	[†] H-N.M.R. (DMSO-d ₆ /TMS) δ [ppm]
5a	325-331°	$C_{37}H_{20}N_2O_2$ (524.6)	524	3070, 1780, 1720, 1400, 755	6.44 (s, 1H); 7.55 (m, 5H); 7.65-8.2 (m, 8H); 8.4-9.1 (m, 6H)
5b	350-356°	$C_{38}H_{22}N_2O_2$ (538.6)	538	3070, 1775, 1710, 1440, 755	4.77 (s, 2 H); 6.30 (s, 1 H); 7.5 (m, 5 H); 7.65-8.0 (m, 8 H); 8.5-9.0 (m, 6 H)
5c	265-268°	$C_{35}H_{24}N_2O_2$ (504.6)	504	3060, 2930, 1775, 1705, 1410, 750	0.96 (t, 3 H); 1.0-1.9 (m, 4 H); 3.55 (t, 2 H); 6.11 (s, 1 H); 7.5-8.1 (m, 8 H); 8.3-9.0 (m, 6 H)
5d	180-182°	$C_{34}H_{22}N_2O_2$ (490.6)	490	3070, 2970, 2940, 1776, 1710, 1415, 754	0.95 (t, 3 H); 1.2-1.95 (m, 2 H); 3.44 (t, 2 H); 6.20 (s, 1 H); 7.6-8.1 (m, 8 H); 8.5-9.0 (m, 6 H)
5e	192-194°	$C_{33}H_{20}N_2O_2$ (476.5)	476	3050, 2910, 1767, 1700, 1419, 747	1.25 (t, 3 H); 3.58 (q, 2 H); 6.16 (s, 1 H); 7.6-8.0 (m, 8 H); 8.5-9.0 (m, 6 H)

^a Satisfactory microanalyses obtained: C ± 0.43 , H ± 0.27 , N ± 0.33 .

Table 2. Reaction of 3 with 1a-e

Run	Hydantoin	Method	Product	Yield [%]
1	la	A	5a	23
2	1b	Α	5b	60
3	1c	Α	5c	62
4	1d	Α	5d	47
5	1e	Α	5e	13
6	1a	В	5a	95
7	1 b	. B	5b	96
8	1c	В	5c	74
9	1 d	В	5d	51
10	1e	В	5e	48

N-Benzyl-15,16-dihydrodibenzo[*c,i*]phenanthro[1,10,9,8-*anmlk*]phenanthridine-dicarboximide (5b); Typical Procedure:

Method A, Dibenzo[b,n]perylene (3; 100 mg, 0.28 mmol), 5-methoxy-3-benzylhydantoin (1b; 625 mg, 2.8 mmol), and p-xylene (2 ml) are heated at 160°C for 72 h in a stainless-steel autoclave. After cooling, the mixture is filtered and washed with methanol. The product is then chromatographed on an alumina column with xylene/ethanol (200/1) as eluent; yield: 92 mg (60%).

Method B: To a stirred, boiling solution of dibenzo[b,n]perylene (3; 100 mg, 0.28 mmol) and 5-methoxy-3-benzylhydantoin (1b; 625 mg, 2.8 mmol) in benzene (60 ml), trifluoroacetic acid (1.5 g, 13 mmol) is added dropwise. Stirring is continued under reflux for 24 h. The solvent is then removed on a rotary evaporator and the residue is chromatographed as above; yield: 142 mg (95%).

$\label{eq:discontinuity} Dibenzo[c,i] phenanthrol 1, 10, 9, 8-anmlk] phenanthridine (6); Typical Procedure:$

Sodium hydroxide pellets (93% pure; 175 mg, 4.0 mmol) are added in one portion to a stirred solution of compound 5b (108 mg, 0.20 mmol) in benzyl alcohol (30 ml) at 90°C. Stirring is continued at 90–100°C for 30 min. After cooling, the mixture is acidified with a solution of concentrated hydrochloric acid (2.5 ml) in methanol (7.5 ml), and then evaporated. The precipitated product is filtered with methanol (10 ml), washed with water and methanol. Sublimation at 300–350°C/2 torr (6 h) gives a yellow powder (59 mg). Recrystallization of the sublimate from a 1500-fold amount of bis[2-hydroxyethyl] ether affords 6 as pale yellow needles; yield: 48 mg (64%): m.p. 396–400°C (dec.).

$C_{29}H_{15}N$	calc.	C 92.28	H 4.01	N 3.71
(377.4)	found	91.90	3.77	4.13

M.S.: $m/e = 377 \text{ (M}^+\text{)}.$

¹ S. Tokita, K. Hiruta, K. Kitahara, H. Nishi, Synthesis 1982, 229.

² S. Tokita et al., Synthesis 1982, 854.

- ³ S. Tokita, K. Hiruta, K. Kitahara, H. Nishi, Bull. Chem. Soc. Jpn. 55, 3933 (1982).
- S. M. Weinreb, J. I. Levin, Heterocycles 12, 949 (1979).
- ⁵ D. Ben-Ishai, E. Goldstein, Tetrahedron 27, 3119 (1971).
- ⁶ E. Goldstein, D. Ben-Ishai, Tetrahedron Lett. 1969, 2631.
- ⁷ R. Bergamasco, Q. N. Porter, Aust. J. Chem. **30**, 1061 (1977).
- 8 Y. Butsugan, S. Yoshida, M. Muto, T. Bito, *Tetrahedron Lett.* 1971, 1129.
- ⁹ T. Imagawa, K. Sisido, M. Kawanishi, Bull. Chem. Soc. Jpn. 46, 2922 (1973).
- ¹⁰ G. Kresze, R. Albrecht, Chem. Ber. 97, 490 (1964).
- D. Ben-Ishai, G. Ben-Et, A. Warshawsky, J. Heterocyclic Chem. 1970, 1289.
- ¹² H. Finkbeiner, J. Org. Chem. 30, 3414 (1965).
- ¹³ T. L. Davis, K. C. Blanchard, Org. Synth. Coll. Vol. 1, 453 (1932).
- ¹⁴ F. J. Wolf, J. Weijlard, Org. Synth. Coll. Vol. IV, 124 (1963).

I.R. (KBr): v = 3070, 1600, 1410, 750 cm⁻¹.

¹H-N.M.R. (FSO₃H/TMS): $\delta = 8.1 - 9.6$ ppm (m).