

## CAMTSCHATCANIDINE, AN ALKALOID FROM *FRITILLARIA CAMTSCHATCENSIS*

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**Key Word Index**—*Fritillaria camtschatcensis*; Liliaceae; bulb; solanidanine alkaloids; 18-nor-17 $\beta$ -methyl-22,26-epiminocholestane.

**Abstract**—From the hydrolysed glycoalkaloid fraction from bulbs of mature *Fritillaria camtschatcensis* in addition to already reported alkaloids a new solanidanine alkaloid, 22 *R*,25 *S*-solanid-5-ene-3 $\beta$ ,27-diol (camtschatcanidine), was isolated and its structure elucidated by spectral analysis and its conversion to solanidine. Also veralkamine was identified from the same plant.

### INTRODUCTION

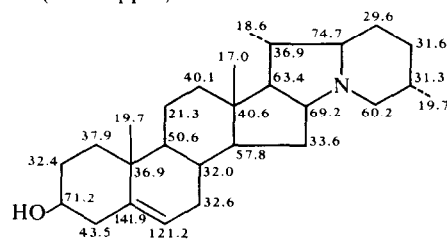
Solanidine (3) was isolated from the bulb [1] and the aerial part [2, 3] of mature *Fritillaria camtschatcensis* as the main alkaloid as well as the *N*-methyl-22,26-epiminocholestene derivatives, hapepunine and anrakorinine, in addition to tomatidenol and solasodine, as minor alkaloids [2, 3]. In continuation of our work on the alkaloids of this plant, a new alkaloid, camtschatcanidine (1a) has been isolated from the bulb as well as the known 18-nor-17 $\beta$ -methyl-22,26-epiminocholestane alkaloid, veralkamine (2) originally isolated from *Veratrum album* var. *lobelianum* [4,5]. This paper described both the isolation of these alkaloids and the structure elucidation of 1a.

### RESULTS AND DISCUSSION

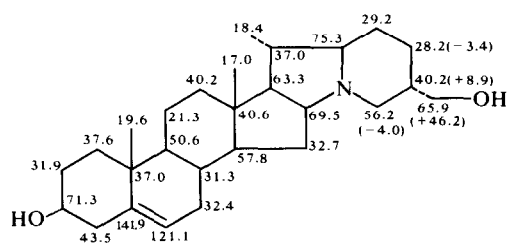
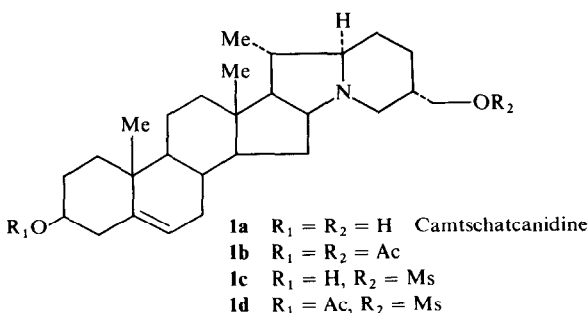
Employing the extraction and separation procedure previously reported [2,3], 7.03 g of alkamines were obtained after hydrolysis of the glycosidic fraction isolated from 3.65 kg of bulbs of *F. camtschatcensis*. This crude alkaloid was fractionated by column chromatography on alumina and afforded five components, 1.75 g of solanidine (3), 1.4 mg of hapepunine, 44 mg of camtschatcanidine (1a), 3 mg of veralkamine (2), and 3.4 mg of a dihydroxylated spirosolane alkaloid.

The mass spectrum of 1a revealed the  $M^+$  at  $m/e$  413, the base peak at  $m/e$  166 and a diagnostic peak at  $m/e$  220. These three peaks correspond to the typical fragment ions of solanidanine alkaloids hydroxylated at ring E or F. This is also supported by the  $^1H$  NMR spectrum which contains signals for one secondary Me group at  $\delta$  1.04, two tertiary Me groups at  $\delta$  0.96 and 1.05, one hydroxymethyl group at  $\delta$  3.82 (downfield shift to  $\delta$  3.92 on acetylation), a secondary OH group at  $\delta$  3.44 (downfield shift to  $\delta$  4.60 on acetylation), and an olefinic proton at  $\delta$  5.44.

To determine the position of the hydroxymethyl group, the  $^{13}C$  NMR spectrum of 1a was compared with that of 3 [6]. Both showed a very similar pattern, thus assignment of many carbon atoms could be made. Substitution of 4 by OH at C-27 as in 1a caused significant shifts at C-24 (−3.4 ppm), C-25 (+18.9 ppm), C-26 (−4.0 ppm) and C-27 (+46.2 ppm).



Solanidine (3)

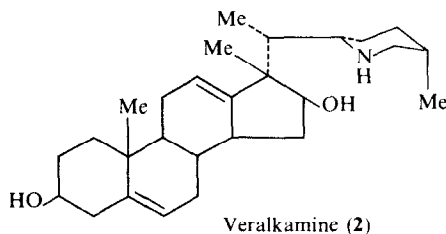


Camtschatcanidine (1a)

In order to establish the detailed stereochemistry of camtschatcanidine (**1a**), this 27-hydroxy-solanidine derivative has been converted into **3**. **1a** was treated with three equivalents of mesyl chloride in anhydrous pyridine affording, after high performance low pressure chromatography the monomesylate (**1c**) [MS *m/e*: 491 ( $M^+$ ), 298, 244;  $^1\text{H-NMR}$  (100 MHz,  $\text{CDCl}_3$ , TMS):  $\delta$  3.0 (3H, s,  $-\text{SO}_2\text{CH}_3$ ), 4.06 (2H, m,  $-\text{CH}_2\text{OMs}$ )]. **1c** was transformed into the monoacetate (**1d**) which was reduced with lithium aluminium hydride in anhydrous ether leading to **3** (4.6 mg, 63.8% yield), identical in every respects with an authentic specimen of solanidine. Therefore, the structure of camtschatcanidine was established as (22*R*,25*S*)-solanid-5-ene-3 $\beta$ ,27-diol (**1a**).

The  $^1\text{H-NMR}$  spectrum of **2** exhibited six tertiary Me protons at  $\delta$  0.93, each three secondary Me protons at 0.81 and 0.96, two protons bearing an oxygen substituent at 3.50 (1H, *m*) and 4.05 (1H, *q*,  $J = 6$  Hz), and two olefinic protons at 5.32.

The chemical shifts and coupling constants of these signals agree closely with those reported for veralkamine from *V. album* var. *lobelianum* [4, 5], and the mass spectrum of **2** also supported the shown structure, revealing the  $M^+$  at *m/e* 413 and the base peak at *m/e* 98.2 melted at 129–134° (plates from methanol) or 120–124° (plates from acetone), and did not show a double mp reported by Tomko *et al.* [4, 5] (mp 117.5–121 /167–169.5°). However, an authentic sample of veralkamine kindly provided by Professor Tomko and recrystallized from acetone, also did not show a double mp and melted at 120–123°; this mp was not depressed by admixture with **2**.



130 (100), 129;  $^1\text{H NMR}$  (100 MHz,  $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$  0.91 (3H, s, 18-H), 1.05 (3H, s, 19-H), 1.13 (3H, d,  $J = 7$  Hz, 21-H), 3.74 (3H, m, 3 $\alpha$ -H and 27-H), 4.64 (1H, m, 16-H), 5.36 (1H, m, 6-H).

*Acetylation of 3a to triacetate (3b).* Acetylation of 2 mg of **3a** in 0.5 ml of pyridine with 0.3 ml of  $\text{Ac}_2\text{O}$  at room temp. overnight, followed by usual work-up, afforded 1.5 mg of *N,O,O*-triacetate (**3b**), mp 128–129°; MS  $m/e$ : 597 ( $\text{M}^+$ ), 583 ( $\text{M}^+ - \text{CH}_3$ ), 554 ( $\text{M}^+ - \text{Ac}$ ), 537 ( $\text{M}^+ - \text{HOAc}$ ), 522 ( $537 - \text{CH}_3$ ), 494 ( $537 - \text{Ac}$ ), 434 ( $494 - \text{HOAc}$ ), 224 (100);  $^1\text{H NMR}$  (100 MHz,  $\text{CDCl}_3$ , TMS):  $\delta$  0.92 (3H, s, 18-H), 1.04 (3H, s, 19-H), 1.08 (3H, d,  $J = 7$  Hz, 21-H), 2.03 (3H, s,  $-\text{OAc}$ ), 2.07 (3H, s,  $-\text{OAc}$ ), 2.22 (3H, s, NAc), 4.04 (3H, m, 27-H and 16-H), 4.64 (1H, m, 3 $\alpha$ -H), 5.38 (1H, m, 6-H).

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