A Concise Stereoselective Approach to β-Alkylaspartates from β-Lactams via [2 + 2] Cycloaddition Reaction of Ketenes to Glyoxylic Ester-Derived Imines.

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Abstract: syn β -Alkyl aspartates were obtained with high stereoselectivities from *cis* 4-alkoxycarbonyl- β -lactams prepared via [2 + 2] cycloaddition reaction of monoalkylketenes and glyoxalate derived imines.

Alkylation of the β -anion derived from aspartic acid or its derivatives is one of the most direct and extended procedures for the synthesis of β -alkylaspartates, but it often leads to mixtures of syn and anti diastereomers¹. This problem can be solved by the use of β -lactams derived from aspartic acid, followed by stereospecific α -alkylation of their lithium enolates to give exclusively the corresponding trans-isomers², which at once can be directly converted into anti- β -alkylaspartates through an appropriate N₁-C₂ β -lactam ring cleavage³. However, the synthesis of syn diastereomers of β -alkylaspartates, still remains a challenging synthetic task⁴. In this paper we disclose our efforts towards a highly stereoselective synthesis of such a class of α -amino acids by the cycloaddition reaction of ketenes, generated from acid chlorides and triethylamine, to glyoxylic ester derived imines. We first reported the synthesis of 4-methoxycarbonyl β lactams from alkanoyl chlorides and imines derived from p-anisidine in an approach to carbapenem compounds⁵. In parallel with our studies, a Merck group has also reported the synthesis of 4ethoxycarbonyl β -lactams following the same approach⁶. However, in both cases, a mixture of *cis* and *trans* isomers was produced⁷. Therefore, we investigated the possibility of improving the stereochemical outcome of this reaction in order to synthesize $syn \beta$ -alkylaspartates in a completely stereoselective fashion. We found that the treatment of alkanovl chlorides 1 with the imine 2, derived from methyl glyoxalate and di-panisylmethylamine (DAM-NH₂)⁸, in refluxing hexane, benzene or methylene chloride in the presence of triethylamine, afforded *cis*-3-alkyl-4-methoxycarbonyl β -lactams 3 as single isomers.



Scheme 1. Reagents and conditions i NEt₃, CH₂Cl₂ or C₆H₆ reflux, 12-14h ii (NH₄)₂Ce(NO₃)₆, MeCN-H₂O, 0-5^oC, 30min, iii ClSiMe₃, MeOH, r.t. iv PhCOCl, NEt₃, CH₂Cl₂ r.t.

The choice of DAM-NH₂ was guided according to the general parameters that influence the *cis* stereoselectivity of the Staudinger reaction⁹, by our previous observations on related [2 + 2] cycloaddition reactions¹⁰ and by the easy transformation of the resulting β -lactams into the N-unsubstituted ones in a single step⁸. Moreover, during the course of our work, we were encouraged by the recent publication of Sasaki et al. showing the usefulness of such imines in the synthesis of 3-vinylazetidin-2-ones¹¹. The deprotection of N-DAM- β -lactams was easily accomplished by treating them with cerium (IV) ammonium nitrate (CAN) in

acetonitrile-water; for instance, compound 3a gave the expected NH- β -lactam 4a in 83% yield (Scheme 1). Similarly, 3b-e afforded 4b-e in yields ranging from 80% to 88%. Finally, the β -lactam ring opening, achieved by means of trimethylchlorosilane in methanol¹², afforded the expected syn 3-alkylaspartates 5a-e, which were isolated as the N-benzoyl derivatives 6¹³. In view of the above results the next logical aspect we examined was an asymmetric version of this reaction¹⁴.

Asymmetric synthesis from the ketene partner. To test this approach β -dimethylphenylsilyl- β -phenyl propanoyl chloride¹⁵ was selected for development. Two reasons led us to explore this proposal. First, as the silyl moiety is a masked form of the hydroxy group¹⁶, the resulting aspartates could be transformed into β , γ -unsaturated α -amino acids^{3a} and, second, these aspartates might be atractive precursors of optically active β -benzyl aspartates through a simple desilylation step.



Scheme 2. Reagents and conditions i 2, NEt₃, CH_2Cl_2 , reflux, 12-14h ii $(NH_4)_2Ce(NO_3)_6$ MeCN,-H₂O, 0-5°C, 30min iii ClSiMe₃, MeOH, r.t. iv BuN₄F, CH_2Cl_2 , r.t. v (BOC)₂O, CH_2Cl_2 , r.t.

Treatment of 7 with 2 in the presence of triethylamine in methylene chloride as solvent gave 8 in 78 % yield, [m.p. 146-147°C (MeOH), $[\alpha]^{25}_{D}=-30.6^{\circ}(c: 1, CH_2Cl_2)]$ together with its *cis*-diastereomer in a ratio of 91:9 respectively¹⁷. N-Dearylation of 8 and further β -lactam opening led to the formation of the expected aspartate 9 which was isolated as its N-BOC derivative 10 in 81 % overall yield. Desilylation of 10 by means of TBAF¹⁸ afforded 11 in nearly quantitative yield ($[\alpha]^{25}_{D}=-23.4^{\circ}(c: 0.8, CH_2Cl_2)]$). Subsequent HPLC analysis of 11 provided the overall diastereomeric purity for the desilylation step and derivatization sequence. Additionally, to check the enantiomeric purity of the product 11, the corresponding (+)-MTPA amide derivative was prepared by deprotection of the N-BOC group and further acylation using Mosher acid chloride and triethylamine¹⁹. From this result it is clear that the silicon assisted asymmetric synthesis of β -benzylaspartates should be general in scope²⁰.

Asymmetric synthesis from the imine partner. We first attempted to obtain asymmetric induction using the methyl glyoxalate imine derived from (R)- α -naphtylethylamine but with no success⁵. In general, the use of chiral amines as components of the imine partner produced lower levels of asymmetric induction than those obtained by changing the aldehyde component^{14b}. Therefore, we studied the influence of glyoxalate imines 12, derived from some representative chiral glyoxalates²¹, on the stereochemical course of the reaction (Scheme 3). When the imine 12a, derived from menthyl glyoxalate^{21a} and DAM-NH₂, was allowed to react with butanoyl chloride and triethylamine a mixture of *cis*-diastereomers 13a and 14a was obtained in a 1.5: 1 ratio respectively.



Scheme 3. Reagents and conditions: i EtCH₂COCl, NEt₃, CH₂Cl₂ or hexane, reflux, 12-14h

The replacement of DAM-NH₂ as the amino component of the imine partner with p-anisidine (PMP) caused formation of *cis* and *trans* isomers in a total ratio of 3:1, and virtually no diastereoselection was observed for any of the four diastereomers formed. A similar level of reaction diastereoselection was produced when imine **12b** was employed as the source of chirality and only an slight increase in diastereoselection was achieved starting from the imine **12c** derived from (+)-8-phenylmenthyl glyoxalate which produced **13c** and **14c** in a ratio of 1 : 2 respectively²².

Taking into account the reaction mechanism of the asymmetric Staudinger reaction²³, the results disclosed in this preliminary study open up the way to prepare optically active β -alkylaspartates using either a chiral ketene partner or from glyoxalate esters derived from bulky homochiral alcohols. Further studies of this chemistry, are now in progress.

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- 13. Representative data for selected compounds: 6a: Oil. I.R. (film) v (cm⁻¹): 3309, 2943, 1736, 1645, 1525. ¹H-NMR (CDCl₃) δ: 7.82-7.41 (m, 5H, arom.), 7.14 (d, 1H, J= 8.4 Hz, NH), 5.07 (dd, 1H, J= 8.4 Hz, J= 4.8 Hz, CH), 3.78 (s, 3H, OCH₃), 3,72 (s, 3H, OCH₃). 3.11 (dq, 1H, J= 7.5Hz, J= 4.8Hz, CH), 1.32 (d, 3H, J= 7.5Hz, CH₃). ¹³C-NMR (CDCl₃) δ: 173.7, 171.0, 166.9, 133.5, 131.8, 128.5, 127.0, 54.2, 52.6, 52.2, 42.0, 13.2 MS (m/z): 222 (M⁺ 57). 6b: Oil. I.R. (film) v (cm⁻¹): 3300, 2946, 1733, 1641, 1524. ¹H-NMR (CDCl₃) δ: 7.82-7.41 (m, 5H, arom.), 7.11 (d, 1H, J= 8.1 Hz, NH), 5.07 (dd, 1H, J= 8.1 Hz, J= 5.4 Hz, CH), 3.77 (s, 3H, OCH₃), 3,72 (s, 3H, OCH₃), 2.91 (dt, 1H, J= 5.4 Hz, J= 9.3 Hz, CH), 1.97-1.82 (m, 1H, CH), 1.74-1.60 (m, 1H, CH), 1.00 (t, 3H, J= 7.5Hz, CH₃). ¹³C-NMR (CDCl₃) δ: 173.1, 171.0, 166.9, 133.5, 131.8, 128.5, 127.0, 53.2, 52.6, 52.0, 49.5, 21.5, 11.9 MS (m/z): 236 (M⁺ 57). 6c: Oil. I.R. (film) v (cm⁻¹): 3303, 2950, 1734, 1642.¹H-NMR (CDCl₃) δ: 7.81-7.43 (m, 5H, arom.), 6.92 (d, 1H, J= 7.8Hz, NH), 5.10 (dd, 1H, J= 5.1 Hz, J= 7.8 Hz, CH), 3.81 (s, 3H, OCH₃), 3.70 (s, 3H, OCH₃). 2.80 (dd, 1H, J= 5.1 Hz, J= 9.3 Hz, CH), 2.23 (m, 1H, CH), 1.18 (d, 3H, J=6.6Hz, CH₃), 0.97 (d, 3H, J=6.6Hz, CH₃). ¹³C-NMR (CDCl₃) δ: 173.1, 171.2, 166.8, 133.7, 131.9, 128.7, 127.1, 55.7, 52.7, 52.2, 51.8, 27.7, 20.8, 20.7, 11: Oil.¹H-NMR (CDCl₃) δ: 173.1, 171.2, 166.8, 133.7, 131.9, 128.7, 127.1, 55.7, 52.7, 52.2, 51.8, 27.7, 20.8, 20.7, 11: Oil.¹H-NMR (CDCl₃) δ: 7.32-7.28 (m, 5H, arom.), 5.60 (sb, 1H, NH), 4.68 (d, 1H, J=-4.1Hz, CH-N), 3.75 (s, 3H, OCH₃), 3.66 (s, 3H, OCH₃), 3.41 (m, 1H, CHOOMe), 3.02 (m, 1H, CH₂-Ph), 2.83 (m, 1H, CH₂-Ph), 1.46 (s, 9H, ¹Bu).
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