

REFERENCES

1. Lakshmi, M. V., Ratram, C. V. and Subba Rao, N. V. (1972) *Indian J. Chem.* **10**, 564.
2. Chakraborty, D. P. and Chowdhury, B. K. (1967) *Tetrahedron Letters* 3472.
3. Ramstrad, E., Lin, W. N. C., Lin, T. J. and Koo, W. Y. (1968) *Tetrahedron Letters* 811.
4. Gupta, G. L. and Nigam, S. S. (1970) *Planta Med.* **19**, 83.
5. Sanyal, P. K. and Bose, P. K. (1977) *Sci. Cult.* **35**, 332.
6. Bhattacharya, P., Roy, S., Biswas, A., Bhattacharya, L. and Chakraborty, D. P. (1978) *J. Indian Chem. Soc.* **55**, 308.
7. Roy, S. and Bhattacharya, L. (1981) *J. Indian Chem. Soc.* **58**, 1212.
8. Joshi, B. B. and Kamat, N. V. (1969) *Indian J. Chem.* **7**, 636.
9. Chowdhury, S. K. and Chakraborty, D. P. (1971) *J. Indian Chem. Soc.* **48**, 80.
10. Stanley, W. L., Waiss, A. C., Lundin, R. E., Jr. and Vannier, S. H. (1965) *Tetrahedron* **21**, 89.
11. Horeau, A. (1961) *Tetrahedron Letters* 506.
12. Horeau, A. and Kagan, H. B. (1964) *Tetrahedron* **20**, 2431.
13. Kikuchi, T., Yokoi, T., Umemoto, K. and Shingu, T. (1974) *Yakugaku Zasshi* **94**, 1616.

Phytochemistry, Vol. 22, No. 3, pp. 794–795, 1983.
Printed in Great Britain.

0031-9422/83/030794-02\$03.00/0
© 1983 Pergamon Press Ltd.

CARPUSIN: A NOVEL 2-HYDROXY-2-BENZYLCOUMARANONE FROM *PTEROCARPUS MARSUPIUM*

JAMES MATHEW and A. V. SUBBA RAO

Department of Chemistry, Osmania University, Hyderabad, 500007, India

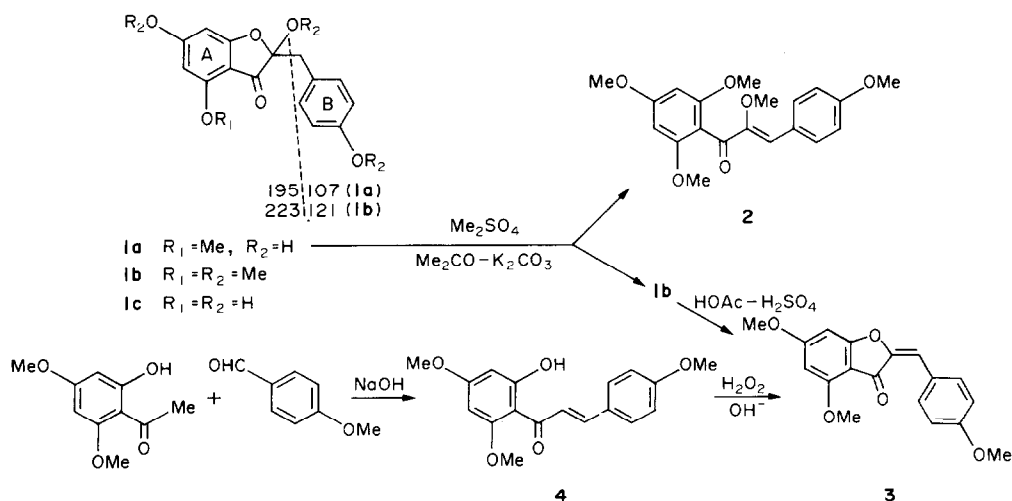
(Revised received 12 July 1982)

Key Word Index—*Pterocarpus marsupium*; Leguminosae; heartwood; 2-hydroxy-2-benzylcoumaranone.

Abstract—The structure of carpusin, an extractive of the heartwood of *Pterocarpus marsupium*, has been established as 2-benzyl-2, 4', 6-trihydroxy-4-methoxybenzo(b)furan-3(2H)one, on the basis of spectral evidence and its conversion to tetra-*O*-methylmaesopsin and 4,6,4'-trimethoxyaurone.

In continuation of our earlier work [1] extraction of the heartwood of *Pterocarpus marsupium* and chromatography of the ether solubles over Si gel using chloroform-ethyl acetate (3:2) as eluent, afforded a colourless crystalline compound, $C_{16}H_{14}O_6$, mp 215° , $[\alpha]_D^{25} \pm 0^\circ$, M^+ 302, designated as carpusin (**1a**).

Compound **1a** showed phenolic properties and gave a cherry-red colour with acetic anhydride and concentrated sulphuric acid characteristic of 2-hydroxy-2-benzylcoumaranones [2]. Functional analysis of **1a** showed a carbonyl (IR ν_{\max} 1675 cm^{-1}), one phenolic methoxyl and one benzylic- CH_2 - ($^1\text{H NMR}$ 60 MHz singlets at δ 3.9



and 3.1, respectively) and two phenolic and one alcoholic hydroxyl groups [formation of a trimethyl ether (**1b**), M^+ 344, showing additional ^1H NMR signals at δ 3.82, 3.85 (phenolic methoxyl) and 3.35 (alcoholic methoxyl)].

The 4,6,4'-substitution pattern of the benzene rings in **1a** was supported by the ^1H NMR spectrum. The protons of the A-ring appeared as a pair of *meta*-coupled doublets showing an AB pattern ($J = 2.5$ Hz) at δ 6.05 (H-7) and 5.95 (H-5); that of the B-ring appeared as a pair of *ortho*-coupled doublets ($J = 8.5$ Hz) showing an A_2B_2 pattern at δ 7.15 (H-2', H-6') and 6.7 (H-3', H-5').

The UV spectrum of **1a** showed $\lambda_{\text{max}}^{\text{MeOH}}$ nm(log ϵ): 290 (4.41) and 324 sh comparable to that of maesopsin (**1c**) [3]. In the presence of sodium acetate the shorter wavelength band in the UV spectrum showed a bathochromic shift of 27 nm supporting the presence of a hydroxyl group at the C-6 position [4]. The methoxyl group must be at the C-4 position as shown by the negative ferric reaction and the absence of a bathochromic shift in the UV spectrum with aluminium chloride [4].

The mass spectrum of **1a** as well as that of the trimethyl ether **1b** showed prominent ions at m/z 107 and 121 (both 100%), respectively, arising from the benzylic moiety; the fragmentation being shown by the dotted line in structures **1a** and **1b**. Prominent ions at m/z 195 for **1a** and at 223 for **1b** were also observed.

In the ^{13}C NMR (90 MHz) spectrum of **1a** the signals may be tentatively assigned as follows: δ 171.91 (C-3), 168.33 (C-6), 158.80 (C-4'), 155.50 (C-4, C-7a), 130.95 (C-2', C-6'), 123.91 (C-1'), 114.37 (C-3', C-5'), 105.44 (C-3a), 101.32 (C-2), 92.27 (C-5), 90.43 (C-7), 55.26 (OMe) and 40.15 ($\alpha\text{-CH}_2$).

The physical data of the trimethyl ether **1b** [mp 129° , UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 292 (4.35)] are similar to that reported for the tetramethyl ether of maesopsin (**1c**) [3].

Methylation (dimethyl sulphate, acetone, potassium carbonate) of **1a** also produced α -4, 2', 4', 6'-pentamethoxy-chalcone (**2**) which was separated from **1b** by CC over Si gel. Compound **2** was reported earlier as a methylation product of maesopsin and its physical data agreed with that reported [3].

Compound **1b** on treatment with boiling acetic acid containing 5% sulphuric acid, gave 4, 6, 4'-trimethoxyaurone (**3**), mp $169\text{--}170^\circ$, UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 340 (4.21), 392 (4.45), which was found to be identical (mp, mmp and superimposable IR) with a sample prepared by the A.F.O. oxidation [5] of the required chalcone **4** obtained by the condensation [6] of 2-hydroxy-4,6-dimethoxyacetophenone and anisaldehyde. Thus structure **1a** for carpusin was confirmed.

This appears to be the first report of **1a** from natural sources and the first instance of a compound of its class from a *Pterocarpus* species.

Acknowledgements—We wish to express our thanks to Dr. G. Srimanarayana and Professor C. V. Ratnam, Department of Chemistry, Osmania University, for helpful discussions.

REFERENCES

1. James Mathew., Subba Rao, A. V. and Subba Rao, N. V. (1977) *Curr. Sci.* **46**, 337.
2. King, H. G. C. and White, T. (1961) *J. Chem. Soc.* 3539.
3. Janes, N. F., King, F. E. and Morgan, J. W. W. (1963) *J. Chem. Soc.* 1356.
4. L. Jurd (1962) in *The Chemistry of Flavonoid Compounds* (Geissman, T. A., ed.) p. 107. Macmillan, New York.
5. Geissman, T. A. and Fukushima, D. K. (1948) *J. Am. Chem. Soc.* **70**, 1686.
6. Shinoda, J. and Sato, T. (1930) *J. Pharm. Soc. Jpn.* **50**, 265.