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stereochemical information is ultimately of enzymatic origin.

New aminocyclitols with quaternary stereocentres via acylnitroso cycloaddition with an *ipso,ortho* arene dihydrodiol^{\ddagger}

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ABSTRACT

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A R T I C L E I N F O

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Dedicated to the memory of Professor J. Grant Buchanan (1926–2012)

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1. Introduction

The dearomatisation-dihydroxylation of an arene by a microorganism was first reported by Gibson in 1968.¹ In the ensuing period, the arene cis-diol products of such biotransformations have been shown to be exceedingly versatile starting materials for synthesis.² For example, many natural products,³ drug molecules,⁴ polymers⁵ and dyes⁶ have all been synthesised by routes, which take advantage of the diverse functionality in these building blocks. Such syntheses are often of single enantiomers, as the metabolism of substituted arenes gives rise to enantiopure diols in most instances. The microorganisms employed for the production of arene cis-diols are usually those expressing benzene dioxygenase (BDO), toluene dioxygenase (TDO), naphthalene dioxygenase (NDO) and biphenyl dioxygenase (BPDO) enzymes. The regio- and stereochemical outcome of these biotransformations may be predicted by Boyd's model;⁷ the sense of enantioinduction is conserved across organisms and substrates (Scheme 1a, ortho, meta-dihydroxylation).

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However, organisms expressing benzoate dioxygenase (BZDO)⁸ oxidise benzoic acids with both different regioselectivity and also the opposite absolute sense of enantioinduction (Scheme 1b, *ipso,ortho*-dihydroxylation). Certain substituted benzoic acids also undergo dearomatisation—dihydroxylation.⁹ The synthetic exploitation of *ipso,ortho* arene *cis*-diols such as **4** has been infrequent compared to *ortho,meta* arene *cis*-diols of type **2**. Nevertheless, we^{3a,j,9a,10} and others^{3m,4,n,0,11} have reported uses of **4** in various synthetic contexts.

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(a) Arene ortho, meta dihydroxylation

Microbial ipso.ortho-dihydroxylation of benzoic acid by the B9 mutant strain of Ralstonia eutropha per-

mits rapid construction of aminocyclitols containing a quaternary stereocentre. Installation of the amine

functionality is achieved by use of an acylnitroso dienophile for a hetero-Diels-Alder reaction. Both

aminotetrols and aminohexols are accessible as single enantiomers by this route. Both NOESY spectro-

scopic and X-ray crystallographic analyses were required to distinguish cycloadduct isomers. Notably, subsequent to the biooxidation step, all new stereocentres are installed under substrate control. Thus, all



Scheme 1. Regio- and stereoselectivity of dioxygenases.







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Aminocyclitols (or 'azacarbasugars') are a privileged class of structures for drug development as they can serve as effective mimics of natural carbohydrates. The amino functionality can impart modified biological activity with respect to the parent carbohydrate and the lack of an endocyclic oxygen leads to enhanced hydrolytic stability.¹² Various aminocyclitols are currently in clinical use:¹³ most notably, voglibose 5^{14} and acarbose 6^{15} are α -glucosidase inhibitors used to treat type II diabetes, as is the structurally related iminosugar miglitol 7^{16} (Fig. 1). Voglibose is an *N*-alkylated derivative of a natural product, valiolamine **8**, which itself possesses α -glucosidase inhibitory activity.¹⁷ Aminocyclitols are also being used or evaluated as therapies for other diseases of carbohydrate metabolism, such as Noctylvalienamine 9,¹⁸ amino derivatives 10-11 of *myo*-inositol,¹⁹ bicyclic analogues 12-13,²⁰ triazolylamino derivatives 14^{21} of scyllo-inositol and homologated calystegine 15²² for Gaucher disease and 4-*epi-N*-octylvalienamine $16^{13b,18,23}$ for G_{M1}-gangliosidosis. Aminocyclitols are core structural motifs of aminoglycoside antibiotics^{12e,24} such as validamycin A 17 and antifungal activity of aminocyclitols such as salbostatin **18** has also been reported.^{24e,25}

purpose. The first report^{36a} dates from 1991, wherein Hudlický reported introducing the amino functionality by means of an acylnitroso cycloaddition (Scheme 2a). This approach has also been used on two subsequent occasions.^{36b,C} Other reported methods for introduction of the nitrogen(s) are (i) alkene epoxidation and ring-opening with azide,^{36b,37} phthalimide,^{37d} ammonia³⁸ or other amines^{36b,39} (Scheme 2b), (ii) alkene aziridination^{37b,e,38b,40} (Scheme 2c), including aziridine opening with nitrogen nucleophiles to access vicinal diaminocyclitols,^{38b,40b} (iii) displacement of a triflate by azide^{34c} (Scheme 2d) and (iv) addition of trimethylsilylazide to an enone⁴¹ (Scheme 2e).

In contrast to *ortho,meta* arene *cis*-diols of type **2**, the *ipso,ortho* arene *cis*-diols had not been used for aminocyclitol synthesis until our report in 2011 on the preparation from **4** of 'inosaminoacids' (**44**), zwitterionic aminocyclitols bearing a *C*-carboxy substituent.⁴² In the present paper we describe the synthesis from **4** of aminocyclitols in which the side chain is in a lower oxidation state (**45**); these aminocyclitols were anticipated to have markedly different solubilities (due to their non-zwitterionic nature) and



Fig. 1. Aminocyclitols of medicinal relevance.

In view of the many current and potential therapeutic applications of aminocyclitols, they have attracted a great degree of attention from the synthetic community. For example, syntheses of natural and non-natural aminocyclitols have been reported starting from carbohydrates by the Ferrier reaction,²⁶ from tricarbonyl(η⁵cyclohexadienyl)iron complexes,²⁷ from inositols²⁸ and quercitols (deoxy-inositols),²⁹ from quinic acid³⁰ and by ring-closing metathesis.^{30e,31} The impetus to discover new glycosidase inhibitors has led chemists to target not only naturally occurring aminocyclitols but also novel structural variants such as medium rings^{31h,32} (e.g., **19–21**), polycyclics^{30e,32d,33} (e.g., **22–25**) and fluorinated derivatives^{33a,34} (e.g., **26–27**, Fig. 2).

Arene *cis*-diols are ideal starting materials for the synthesis of aminocyclitols³⁵—the first two hydroxyl groups are introduced in the biotransformation and additional hydroxyls may be introduced by a variety of further oxidative transformations. The *ortho,meta* arene *cis*-diols of type **2** have been used extensively for this

glycosidase inhibitory activities to those we have previously reported (Fig. 3).

2. Results and discussion

Primary alcohol **46** is accessible from *ipso,ortho* arene *cis*-diol **4** by previously reported transformations:^{11a,b,e} ketalisation, esterification and reduction (we have found LiBH₄ to be the most effective reductant^{3a}). We intended to introduce nitrogen functionality by means of an acylnitroso Diels–Alder reaction whereby the dienophile is generated in situ under oxidative conditions (cf. Scheme 2a). Thus, protection of the free hydroxyl group was needed; introduction of a silyl ether proceeded to give **47** without any competing rearomatisation (a side reaction that can occur upon exposure of arene *cis*-diols to extremes of pH⁴³). Fully protected diene **47** underwent cycloaddition to afford two isomeric products, **48** and **49** (Scheme 3).



Fig. 2. Selected non-natural aminocyclitols.

The selectivity of cycloadditions employing dienes derived from arene cis-diols has been studied previously;⁴⁴ precedent suggested that approach of the dienophile to the diene face opposite the acetonide would be favoured.^{11e,36a,45} NOESY spectra for both adducts showed correlations between the olefinic protons and the acetonide *endo* methyl protons, confirming the approach of the dienophile to the upper face. Distinguishing the two structures in which the Cbz group is distal (**48**) or proximal (**49**) to the silyl ether was not possible by NMR methods and required crystallographic analysis of a derivative (see below).

The residual alkenes in 48 and 49 were dihydroxylated under Upjohn conditions; cycloadduct 48 afforded a single cis-diol 50 in good yield, but surprisingly the corresponding transformation of 49 to cis-diol 51 was lower yielding and hampered by the formation of tetracyclic carbonate side-product 52. The structure of 52 was assigned on the basis of its molecular mass, a characteristic $\nu_{(C=0)}$ absorbance at 1808 cm⁻¹ and a ¹³C NMR resonance at δ =154.4 ppm. We propose that 52 is formed from 51 by attack of a newly introduced hydroxyl group on the carbamate carbonyl and C-N bond cleavage, followed by attack of the other hydroxyl group and extrusion of benzyl alcohol (Scheme 4). Also in support of the structure of 52, treatment of cis-diol 50 with TBAF did not give the expected desilylated structure 54, but instead gave cyclic carbonate 55, the structure of which was secured by X-ray crystallographic analysis (Fig. 4). Cyclic carbonate **55** has $v_{(C=0)}=1800$ cm⁻¹ and a ¹³C NMR resonance at δ =154.6 ppm, comparable with **52**.

Hydrogenolysis of diols **50** and **51**, as well as their less highly oxygenated precursors **48** and **49** proceeded smoothly over palladium on carbon, effecting multiple reductive transformations in one-pot (N–O bond scission, Cbz deprotection, alkene reduction; Scheme 5). The more functionalised products **58** and **59** were isolated in lower yield than **56** and **57**, which we ascribe to their unavoidable partial retention on Celite and/or silica. The final removal of acid-labile protecting groups could be executed in a succinct manner simply by exposing **56–59** to aqueous hydrochloric acid, followed by an organic extraction to remove the lipophilic silanol byproduct. Concentration of the aqueous phase gave pure novel aminocyclitols **60–63** (Scheme 6). Aminotetrols **61** and **62** and aminohexols **63** and **64** were evaluated for the inhibition of glycosidase activity⁴⁶ against α -glucosidase (type I from *Saccharomyces cerevisiae*), β -glucosidase (almond), β -galactosidases (from *Aspergillus oryzae* and *Escherichia coli*) and β -glucuronidases (from bovine liver, *E. coli* and *Patella vulgata*); no inhibitory activity at 100 μ M was observed.

3. Conclusion

We have described herein the synthesis of novel, densely functionalised aminocyclitols from a simple aromatic precursor, using a biotransformation and a hetero-Diels—Alder reaction. The brevity of this route (nine steps from benzoic acid to products bearing six contiguous stereocentres, including a quaternary stereocentre) underscores the utility of microbial arene oxidation for the rapid introduction of complexity.

4. Experimental section

4.1. General

For general experimental methods, please see the Supplementary data.

4.1.1. tert-Butyl((((3aR,7aR)-2,2-dimethyl-3a,7a-dihydrobenzo[d][1,3] dioxol-3a-yl)methoxy)dimethylsilane (**47**). To alcohol **46** (548 mg, 3.01 mmol, 1.00 equiv) dissolved in dichloromethane (30 mL), triethylamine (1.06 mL, 7.52 mmol, 2.50 equiv) was added and

(a) Acylnitroso cycloaddition (Ref. 36a)



(b) Epoxidation / opening with azide (Ref. 37i)



(c) Aziridination (Ref. 38b)



(d) Displacement of triflate with azide (Ref. 34c)



(e) Addition of TMSN₃ to an enone (Ref. 41)



Scheme 2. Representative examples of reported strategies for introduction of nitrogen in the synthesis of aminocyclitols from ortho, meta arene cis-diols.



Fig. 3. Contrasting current and previous work.

stirred at 0 °C. *tert*-Butyldimethylsilyl trifluoromethanesulfonate (0.829 mL, 3.61 mmol, 1.20 equiv) was added dropwise at 0 °C over 5 min. The resulting mixture was stirred at 0 °C for 1 h. The reaction mixture was transferred to a separating funnel, saturated brine (10 mL) was added then extracted with ethyl acetate (3×10 mL). The organic phase was dried over magnesium sulfate and filtered. The filtrate was concentrated under reduced pressure and purified via column chromatography (15% ethyl acetate—petrol) to give **47** (636 mg, 72%) as a colourless oil: R_f 0.56 (15% ethyl acetate—petrol); $[\alpha]_D^{25}$ –137.5 (*c* 9.4, CH₂Cl₂); ν_{max} (neat)/cm⁻¹ 2987, 2954, 2930, 2898, 2857, 1472, 1463, 1414, 1378, 1368, 1251, 1214, 1172, 1151, 1130, 1093, 1076, 1038, 1006, 938, 915, 834, 775, 709, 664, 641; ¹H NMR (250 MHz, CDCl₃); δ 6.10–5.94 (m, 3H), 5.67 (d, *J*=10.0 Hz, 1H), 4.52 (d, *J*=5.0 Hz, 1H), 3.55 (d, *J*=12.0 Hz, 1H), 3.44 (d, *J*=12.0 Hz, 1H), 1.43



Scheme 3. Cycloaddition of diene derived from *ipso,ortho* arene *cis*-diol **4.** Reagents and conditions: (a) TBDMSOTF, Et_3N , CH_2Cl_2 , 0 °C, 1 h, 72%; (b) BnOC(O)NHOH **30**, Bu₄NIO₄ CH₂Cl₂, -78 °C, 20 h.

(s, 3H), 1.36 (s, 3H), 0.87 (s, 9H), 0.05 (s, 3H), 0.04 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 129.6, 125.1, 124.5, 122.6, 105.7, 80.1, 71.2, 65.8, 27.1, 26.5, 25.7, 18.2, -5.3, -5.5; HRMS *m*/*z* (ES⁺) [found (M+Na)⁺ 319.1705, C₁₆H₂₈NNaO₃Si requires M⁺, 319.1688].

4.1.2. (3aR,4R,7S,7aR)-Benzyl 3a-(((tert-butyldimethylsilyl)oxy)methyl)-2,2-dimethyl-3a,4,7,7a-tetrahydro-4,7-(epoxyimino)benzo[d]



Scheme 4. Dihydroxylation of cycloadducts. Reagents and conditions: (a) 2 mol % OsO₄, 1 equiv NMO, acetone/H₂O 4:1, rt, 48 h, 81% (50), 44% (51), 17% (52); (b) 1 equiv TBAF, THF, 0 °C, 12 h, 24%.

[1,3]dioxole-8-carboxylate (**48**) and (3aR,4S,7R,7aR)-benzyl 7a-(((tert-butyldimethylsilyl)oxy)methyl)-2,2-dimethyl-3a,4,7,7a-tetrahydro-4,7-(epoxyimino)benzo[d][1,3]dioxole-8-carboxylate (**49**). To a solution of diene **47** (446 mg, 1.50 mmol, 1.00 equiv) and



Fig. 4. ORTEP diagram of 55 showing ellipsoids at 50% probability. H atoms are shown as spheres of arbitrary radius.

tetrabutylammonium periodate (1.30 g, 3.01 mmol, 2.00 equiv) in dichloromethane (30 mL) at -78 °C was added N-(benzyloxycarbonyl)hydroxylamine (503 mg, 3.01 mmol, 2.00 equiv) in dichloromethane (10 mL) dropwise via cannula over 5 min. The reaction mixture was stirred at -78 °C under N₂ for 20 h, then diluted with ethyl acetate (10 mL) and washed with saturated aqueous sodium thiosulfate solution (5 mL) and then brine (5 mL). The organic layer was separated and dried over magnesium sulfate, then concentrated under reduced pressure and purified by column chromatography (5% ethyl acetate-petrol) to give **48** (217 mg, 31%) as a colourless oil and 49 (403 mg, 58%) as a colourless oil. Compound **48**: R_f 0.19 (15% ethyl acetate-petrol); $[\alpha]_D^{25}$ +1.53 (*c* 0.66, CH₂Cl₂); *v*_{max} (neat)/cm⁻¹ 3515, 2929, 2857, 1747, 1713, 1497, 1455, 1379, 1312, 1284, 1248, 1211, 1184, 1151, 1097, 1061, 1024, 984, 929, 910, 776, 751, 731, 696, 616; ¹H NMR (300 MHz, CDCl₃); 7.25 (s, 5H), 6.46 (br t, J=5.0 Hz, 1H), 6.28 (ddd, J=7.5, 2.5, 1.5 Hz, 1H), 5.10 (d, *J*=10.0 Hz, 1H), 5.06 (dd, *J*=6.0, 3.0 Hz, 1H), 5.02 (d, *J*=10.0 Hz, 1H), 4.82 (ddd, J=6.0, 3.0, 1.5 Hz, 1H), 4.07 (d, J=3.0 Hz, 1H), 3.80 (d, J=12.0 Hz, 1H), 3.76 (d, J=12.0 Hz, 1H), 1.30 (s, 3H), 1.21 (s, 3H), 0.08 (s, 9H), 0.01 (s, 3H), -0.01 (s, 3H); ¹³C NMR (75 MHz, CDCl₃); δ 158.2, 135.5, 132.4, 129.0, 128.4, 128.2, 128.0, 112.3, 84.0, 75.7, 72.2, 68.0, 66.2, 53.6, 28.0, 27.0, 25.9, 18.4, -5.5, -5.6; HRMS m/z (ES⁺) [found (M+Na)⁺ 484.2140, C₂₄H₃₅NaO₆Si requires M⁺, 484.2131]. Compound **49**: R_f 0.28 (15% ethyl acetate-petrol); $[\alpha]_D^{25}$ +12.6 (*c* 0.87, CH₂Cl₂); ν_{max} (neat)/cm⁻¹ 2930, 2857, 1748, 1709, 1498, 1462, 1380, 1370, 1327, 1295, 1248, 1214, 1170, 1150, 1096, 1080, 1065, 1026, 1005, 934, 904, 836, 776, 736, 696, 683, 670; ¹H NMR (300 MHz, CDCl₃) δ 7.34 (s, 5H), 6.53 (ddd, J=7.5, 2.5, 1.5 Hz, 1H), 6.42 (br t, J=5.0 Hz, 1H), 5.22 (d, J=12.0 Hz, 1H), 5.19 (d, J=12.0 Hz,



Scheme 5. Hydrogenolysis. Reagents and conditions: (a) H₂, Pd/C, EtOAc, rt, 24 h, 99% (56), 98% (57), 50% (58), 45% (59).

1H), 5.06 (ddd, *J*=5.0, 5.0, 1.5 Hz, 1H), 4.89 (dd, *J*=5.0, 2.5 Hz, 1H), 4.18 (d, *J*=2.5 Hz, 1H), 3.90 (s, 2H), 1.39 (s, 3H), 1.28 (s, 3H), 0.90 (s, 9H), 0.09 (s, 3H), 0.08 (s, 3H); 13 C NMR (75 MHz, CDCl₃); δ 157.5, 135.5, 131.1, 129.8, 128.4, 128.3, 128.1, 112.5, 84.1, 75.1, 71.5, 68.1, 65.6, 53.6, 28.3, 27.1, 26.0, 18.5, -5.3, -5.4; HRMS *m*/*z* (ES⁺) [found (M+Na)⁺ 484.2108, C₂₄H₃₅NaO₆Si requires M⁺, 484.2131].

4.1.3. (3aR,4R,5S,6R,7S,7aR)-Benzyl 3a-(((tert-butyldimethylsilyl) oxy)methyl)-5,6-dihydroxy-2,2-dimethylhexahydro-4,7-(epoxyimino)benzo[d][1,3]dioxole-8-carboxylate (**50**). To alkene **48** (195 mg, 0.422 mmol, 1.00 equiv) in acetone–water (4/1, 20 mL) was added *N*-methylmorpholine *N*-oxide (49 mg, 0.422 mmol,

1.00 equiv) as a solid. Osmium tetroxide (2.5% in *tert*-butanol. 80 µL. 8.4 µmol, 2.0 mol %) was added via syringe and the reaction mixture was stirred at room temperature for 48 h. A colour change from colourless to pale yellow was observed. The reaction mixture was transferred to a separating funnel, diluted with ethyl acetate (15 mL) and washed with saturated aqueous sodium thiosulfate (5 mL) and brine (5 mL). The organic phase was separated and dried over magnesium sulfate, concentrated under reduced pressure and purified by column chromatography (50% ethyl acetate-petrol) to give **50** (172 mg, 81%), as a colourless oil. R_f 0.38 (40% ethyl aceta-te-petrol); $[\alpha]_D^{25}$ -27 (*c* 8.2, CHCl₃); ν_{max} (neat)/cm⁻¹ 3419, 2959, 2929, 2886, 2857, 1708, 1553, 1498, 1460, 1408, 1384, 1258, 1212, 1071, 836, 779; ¹H NMR (250 MHz, CDCl₃) δ 7.36–7.30 (m, 5H), 5.18 (s, 2H), 4.66 (br s, 1H), 4.38 (br s, 1H), 4.24-4.17 (m, 3H), 3.82 (d, *I*=12.5 Hz, 1H), 3.75 (d, *I*=12.5 Hz, 1H) 3.75–3.63 (m, 2H), 1.43 (s, 3H), 1.40 (s, 3H), 0.87 (s, 9H), 0.04 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 156.8, 135.4, 128.2, 128.1, 127.9, 111.1, 82.0, 78.0, 76.4, 72.3, 67.9, 64.9, 61.8, 61.2, 26.5, 26.4, 25.7, 18.2, -5.6, -5.7; HRMS m/z (ES⁺) [found (M+Na)⁺ 496.2400, C₂₄H₃₇NNaO₈Si requires M⁺, 496.2367].

4.1.4. (3aR,4S,5S,6R,7R,7aR)-Benzyl 7*a*-(((tert-butyldimethylsilyl) oxy)methyl)-5,6-dihydroxy-2,2-dimethylhexahydro-4,7-(epoxvimino)benzo[d][1,3]dioxole-8-carboxylate (51) and (3aS,4S,4aR,7aR,8R,8aR)-7a-(((tert-butyldimethylsilyl)oxy)methyl)-6,6dimethylhexahydro-4,8-(epoxyimino)benzo[1,2-d:4,5-d']bis([1,3]dioxole)-2-one (52). To alkene 49 (489 mg, 1.06 mmol, 1.00 equiv) in acetone-water (4/1, 20 mL) was added N-methylmorpholine Noxide (124 mg, 1.06 mmol, 1.00 equiv) as a solid. Osmium tetroxide (2.5% in tert-butanol, 160 uL, 0.011 mmol, 2.0 mol %) was added via syringe and the reaction mixture was stirred at room temperature for 48 h. A colour change from colourless to pale yellow was observed and the reaction mixture was transferred to a separating funnel, diluted with ethyl acetate (10 mL) and washed with saturated aqueous sodium thiosulfate (5 mL) then brine (5 mL). The organic phase was separated and dried over magnesium sulfate, concentrated under reduced pressure and purified by column chromatography (50% ethyl acetate-petrol) to give 51 (230 mg, 44%) and 52 (70 mg, 17%) both as colourless oils. Also isolated was recovered 49 (97 mg, 19%). Compound 51: Rf 0.36 (50% ethyl acetate-petrol); $[\alpha]_D^{25}$ +17 (*c* 11.5, CHCl₃); ν_{max} (neat)/cm⁻¹ 3428, 2929, 2856, 1710, 1498, 1454, 1383, 1256, 1102, 1064, 835, 777; ¹H NMR (250 MHz, CDCl₃) δ 7.38–7.32 (m, 5H), 5.20 (s, 2H), 4.67 (br s, 1H), 4.39 (br s, 1H), 4.25-4.21 (m, 3H), 3.84 (d, J=10.0 Hz, 1H), 3.78 (d, *J*=10.0 Hz, 1H) 3.54 (br s, 1H), 3.32 (d, *J*=5.0 Hz, 1H) 1.44 (s, 3H), 1.42 (s, 3H), 0.89 (s, 9H), 0.06 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 158.9, 135.5, 128.5, 128.3, 128.3, 111.7, 83.1, 76.2, 73.2, 68.3, 66.0, 62.5, 61.4, 58.9, 26.7, 26.6, 25.9, 18.4, -5.5, -5.6; HRMS m/z (ES⁺) [found (M+Na)⁺ 518.2210, C₂₄H₃₇NNaO₈Si requires M⁺, 518.2186].



Scheme 6. One-pot deprotection. Reagents and conditions: (a) 1 M HCl_(aq), rt, 24 h; EtOAc extraction, 80% (62), 94% (63), 88% (64), 99% (65).

Compound **52**: $R_f 0.57$ (50% ethyl acetate—petrol); $[\alpha]_D^{25} - 6.5$ (*c* 3.5, CHCl₃); ν_{max} (neat)/cm⁻¹ 3270, 2955, 2930, 2857, 1808, 1463, 1361, 1254, 1166, 1077, 836, 779; ¹H NMR (300 MHz, CDCl₃) δ 6.06 (br s, 1H), 5.03–5.02 (m, 2H), 4.33 (d, *J*=6.0 Hz, 1H), 4.21 (d, *J*=6.0 Hz, 1H), 4.08 (d, *J*=12.0 Hz, 1H), 3.90 (d, *J*=12.0 Hz, 1H), 3.67 (br s, 1H) 1.45 (s, 3H), 1.44 (s, 3H), 0.90 (s, 9H), 0.09 (s, 3H), 0.08 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 154.4, 110.9, 81.8, 73.7, 71.7, 69.9, 68.3, 66.2, 53.4, 26.9, 26.6, 25.8, 18.3, -5.4, -5.5; HRMS *m*/*z* (ES⁺) [found (M+Na)⁺ 410.1664, C₁₇H₂₉NNaO₇Si requires M⁺, 410.1611].

4.1.5. (3aS,4R,4aR,7aR,8S,8aR)-4a-(Hydroxymethyl)-6,6dimethylhexahydro-4,8-(epoxyimino)benzo[1,2-d:4,5-d']bis([1,3]dioxole)-2-one (55). To silyl ether 50 (165 mg, 0.333 mmol, 1.00 equiv) in THF (20 mL) at 0 °C was added tetrabutylammonium fluoride (0.33 mL, 1.0 M in THF, 1.00 equiv) dropwise over 5 min. The reaction mixture was stirred at 0 °C for 12 h, then transferred to a separating funnel and diluted with ethyl acetate (20 mL) and washed with water (2×10 mL). The organic phase was then washed further with brine (5 mL), dried over magnesium sulfate, concentrated under reduced pressure and purified by column chromatography (50% ethyl acetate-petrol) to give 55 (34 mg, 24%), as a colourless oil. R_f 0.20 (50% ethyl acetate-petrol); $[\alpha]_D^{25}$ -1.5 (*c* 1.7, CHCl₃); *v*_{max} (neat)/cm⁻¹ 3268, 2992, 2923, 1800, 1780, 1454, 1369, 1165, 1067, 770; ¹H NMR (400 MHz, $(CD_3)_2CO)$ δ 6.88 (br s, 1H), 5.16 (dd, *J*=8.0, 4.0 Hz, 1H), 5.09 (d, *J*=8.0 Hz, 1H), 4.42 (d, *J*=4.0 Hz, 1H), 4.27 (t, *J*=4.0 Hz, 1H), 4.19 (br s, 1H), 3.97–3.84 (m, 3H), 1.50 (s, 3H), 1.48 (s, 3H); 13 C NMR (100 MHz, (CD₃)₂CO) δ 154.5, 110.1, 81.7, 73.7, 72.0, 70.6, 69.4, 64.5, 51.6, 25.8, 25.6; HRMS m/z (ES⁺) [found (M+Na)⁺ 296.0787. C₁₁H₁₆NNaO₇Si requires M⁺. 296.0746].

4.1.6. (3aS,4R,7S,7aR)-7-Amino-3a-(((tert-butyldimethylsilyl)oxy) methyl)-2,2-dimethylhexahydrobenzo[d][1,3]dioxol-4-ol (56). To alkene 48 (61 mg, 0.13 mmol, 1.0 equiv) in ethyl acetate (20 mL) was added palladium on carbon (6.0 mg, 10 mass %). The reaction mixture was stirred under an atmosphere of H₂ at room temperature for 24 h, then filtered through Celite. The filtrate was concentrated under reduced pressure and purified by column chromatography (50% ethyl acetate-petrol) to give 56 as a colourless oil (43 mg, 99%). R_f 0.85 (50% ethyl acetate-petrol); $[\alpha]_D^{25}$ -15 (c 0.55, CHCl₃); v_{max} (neat)/cm⁻¹ 2961, 2927, 2854, 1590, 1519, 1463, 1392, 1300, 1251, 1215, 1151, 1033, 747; ¹H NMR (300 MHz, CDCl₃) δ 4.05 (d, J=3.0 Hz, 1H), 3.97 (d, J=12.0 Hz, 1H), 3.83 (d, J=12.0 Hz, 1H), 3.80 (app q, J=3.0 Hz, 1H), 3.46-3.42 (m, 1H), 2.76 (br s, 3H), 2.02-1.87 (m, 1H), 1.83-1.72 (m, 2H), 1.61-1.51 (m, 1H), 1.46 (s, 3H), 1.37 (s, 3H), 0.91 (s, 9H), 0.10 (s, 3H), 0.09 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 108.1, 83.6, 80.1, 73.2, 65.1, 47.4, 28.1, 26.7, 25.9, 25.4, 24.5, 18.3, -5.5, -5.6; HRMS m/z (ES⁺) [found (M+Na)⁺ 354.2054, C₁₆H₃₃NNaO₄Si requires M⁺, 354.20761.

4.1.7. (3*a*R,4*S*,7*R*,7*a*R)-7-*Amino*-7*a*-(((tert-butyldimethylsilyl)oxy) methyl)-2,2-dimethylhexahydrobenzo[d][1,3]dioxol-4-ol (**57**). To alkene **49** (253 mg, 0.548 mmol, 1.00 equiv) in ethyl acetate (20 mL) was added palladium on carbon (26 mg, 10 mass %). The reaction mixture was stirred under an atmosphere of H₂ at room temperature for 24 h, then filtered through Celite. The filtrate was concentrated under reduced pressure and purified by column chromatography (20% methanol–79% ethyl acetate–1% triethylamine) to give **57** as a colourless oil (179 mg, 98%). *R*_f 0.43 (20% methanol–79% ethyl acetate–1% triethylamine); $[\alpha]_D^{25}$ –37 (*c* 3.5, CHCl₃); ν_{max} (neat)/cm⁻¹ 3282, 2998, 2930, 2856, 1472, 1378, 1250, 1216, 1087, 963, 835; ¹H NMR (300 MHz, CDCl₃) δ 4.03 (br s, 1H), 3.96–3.93 (m, 2H), 3.70 (d, *J*=11.0 Hz, 1H), 3.36 (br s, 3H), 3.12 (br s, 1H), 2.00–1.88 (m, 1H), 1.82–1.72 (m, 2H), 1.62–1.58 (m, 1H), 1.41 (s, 3H), 1.32 (s, 3H), 0.88 (s, 9H), -0.09 (s,

6H); ¹³C NMR (75 MHz, CDCl₃) δ 108.4, 83.1, 79.8, 67.3, 65.8, 52.3, 27.5, 26.3, 26.0, 25.9, 23.7, 18.3, -5.3, -5.5; HRMS *m/z* (ES⁺) [found (M+Na)⁺ 354.2076, C₁₆H₃₃NNaO₄Si requires M⁺, 354.2076].

4.1.8. (3aS.4R.5R.6R.7S.7aR)-7-Amino-3a-(((tert-butyldimethylsilyl) oxy)methyl)-2.2-dimethylhexahydrobenzo[d][1.3]dioxole-4.5.6-triol (58). To diol 50 (28 mg, 0.056 mmol, 1.0 equiv) in ethyl acetate (20 mL) was added palladium on carbon (5.0 mg, 20 mass %). The reaction mixture was stirred under an atmosphere of H₂ at room temperature for 24 h, then filtered through Celite. The filtrate was concentrated under reduced pressure and purified by column chromatography (10% methanol-89% chloroform-1% triethylamine) to give 58 as a colourless oil (11 mg, 50%). R_f 0.25 (15%) methanol-84% ethyl acetate-1% triethylamine); $[\alpha]_D^{25} + 43$ (c 0.6, H₂O); *v*_{max} (neat)/cm⁻¹ 3324, 3256, 2965, 2813, 2787, 1432, 1376, 1244, 1156, 1020, 978, 836, 774; ¹H NMR (300 MHz, CDCl₃) δ 4.46–3.93 (m, 10H), 1.45 (s, 3H), 1.39 (s, 3H), 0.92 (s, 9H), 0.12 (s, 6H); 13 C NMR (75 MHz, CDCl₃) δ 108.9, 83.8, 74.3, 70.6, 67.6, 65.3, 64.7, 51.9, 29.6, 27.9, 26.6, 26.0, 25.9, 18.4, -5.3, -5.4; HRMS m/z (ES⁺) [found (M+Na)⁺ 386.1968, C₁₆H₃₃NNaO₆Si requires M⁺, 386.1907].

4.1.9. (3aR,4S,5S,6S,7R,7aR)-7-Amino-7a-(((tert-butyldimethylsilyl) oxy)methyl)-2,2-dimethylhexahydrobenzo[d][1,3]dioxole-4,5,6-triol (59). To diol 51 (120 mg, 0.242 mmol, 1.00 equiv) in ethyl acetate (20 mL) was added palladium on carbon (12 mg, 10 mass %). The reaction mixture was stirred under an atmosphere of H₂ at room temperature for 24 h, then filtered through Celite. The filtrate was concentrated under reduced pressure and purified by column chromatography (50% ethyl acetate-petrol) to give 59 as a colourless oil (39 mg, 45%). $R_f 0.35$ (50% ethyl acetate-petrol); $[\alpha]_D^{25} + 33$ (c 0.39, CHCl₃); *v*_{max} (neat)/cm⁻¹ 3404, 2958, 2930, 2857, 1463, 1382, 1255, 1210, 1101, 1060, 863, 779; ¹H NMR (500 MHz, CDCl₃) δ 4.25–4.20 (m, 2H), 4.18–4.15 (m, 2H), 4.03 (d, *J*=15.0 Hz, 1H), 3.91 (d, J=15.0 Hz, 1H), 3.30 (s, 1H), 1.46 (br s, 3H), 1.43 (s, 3H), 0.92 (s, 9H), 0.11 (s, 3H), 0.10 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 110.3, 81.9, 73.8, 72.9, 66.7, 62.7, 60.2, 58.3, 26.8, 26.8, 25.9, 18.4, -5.3, -5.4; HRMS m/z (ES⁺) [found (M+H)⁺ 364.2093, C₁₆H₃₃NO₆Si requires M⁺, 364.2155].

4.1.10. (1*S*,2*R*,3*S*,4*R*)-2,3,4-*Trihydroxy*-3-(*hydroxymethyl*)*cyclohexylammonium chloride* (**60**). Hydroxyamine **56** (43 mg, 0.13 mmol) was stirred in aqueous hydrochloric acid (1.0 M, 20 mL) at room temperature for 24 h. The aqueous phase was washed with ethyl acetate (2×10 mL) to remove the silanol byproduct. The aqueous phase was concentrated under reduced pressure to give **60** as a colourless oil (26 mg, 94%). $[\alpha]_D^{25}$ +22 (*c* 1.3, H₂O); ν_{max} (neat)/cm⁻¹ 3336, 2981, 2482, 1602, 1383, 1233, 1156, 1069, 1021, 956, 797; ¹H NMR (300 MHz, D₂O) δ 3.85 (s, 1H), 3.64 (d, *J*=12.0 Hz, 1H), 3.57 (d, *J*=12.0 Hz, 1H), 3.55 (d, *J*=9.0 Hz, 1H), 3.37–3.18 (m, 1H), 1.80–1.55 (m, 4H); ¹³C NMR (75 MHz, D₂O) δ 76.0, 69.9, 68.6, 63.5, 52.1, 25.8, 22.7; HRMS *m/z* (ES⁺) [found (M+Na)⁺ 200.0906, C₇H₁₅NNaO₄ requires M⁺, 200.0898].

4.1.11. (1*R*,2*R*,3*R*,4*S*)-2,3,4-Trihydroxy-2-(hydroxymethyl)cyclohexylammonium chloride (**61**). Hydroxyamine **57** (30 mg, 0.090 mmol) was stirred in aqueous hydrochloric acid (1.0 M, 20 mL) at room temperature for 24 h. The aqueous phase was washed with ethyl acetate (2×10 mL) to remove the silanol byproduct. The aqueous phase was concentrated under reduced pressure to give **61** as a pale yellow oil (17 mg, 88% yield). [α] $_{D}^{55}$ +35 (c 0.85, H₂O); ν_{max} (film)/cm⁻¹ 3317, 2940, 2508, 1400, 1071, 1027, 972, 799; ¹H NMR (300 MHz, D₂O) δ 3.86 (d, *J*=12.0 Hz, 1H), 3.81 (td, *J*=7.5, 4.0 Hz, 1H), 3.55 (d, *J*=12.0 Hz, 1H), 3.49 (d, *J*=6.0 Hz, 1H), 3.38–3.34 (m, 1H), 1.92–1.63 (m, 3H), 1.52–1.40 (m, 1H); ¹³C NMR (75 MHz, D₂O) δ 74.1, 72.3, 69.8, 63.4, 54.0, 26.0, 22.8; HRMS *m*/*z* (ES⁺) [found (M+H)⁺ 179.1129, C₇H₁₆NO₄ requires M⁺ 179.1157].

4.1.12. (1*S*,2*R*,3*R*,4*R*,5*R*,6*R*)-2,3,4,5,6-Pentahydroxy-3-(hydroxymethyl)cyclohexylammonium chloride (**62**). Aminotriol **58** (11 mg, 0.030 mmol) was stirred in aqueous hydrochloric acid (1.0 M, 20 mL) at room temperature for 24 h. The aqueous phase was washed with ethyl acetate (2×10 mL) to remove the silanol byproduct. The aqueous phase was concentrated under reduced pressure to give **62** as a colourless oil (8 mg, 99%). [α]_D²⁵ –47 (*c* 0.4, H₂O); ν_{max} (neat)/cm⁻¹ 3302, 2958, 2511, 1629, 1508, 1077, 1028, 808, 723; ¹H NMR (300 MHz, D₂O) δ 4.24 (q, *J*=3.0 Hz, 1H), 4.04–3.99 (m, 3H), 3.89 (d, *J*=12.0 Hz, 1H), 3.75 (d, *J*=12.0 Hz, 1H), 3.57 (dd, *J*=10.5, 3.0 Hz, 1H); ¹³C NMR (75 MHz, D₂O) δ 75.5, 72.6, 70.4, 65.6, 65.1, 62.8, 53.3; HRMS *m/z* (ES⁺) [found (M+Na)⁺ 232.0783, C₇H₁₅NNaO₆ requires M⁺ 232.0797].

4.1.13. (1*R*,2*R*,3*R*,4*S*,5*S*,6*S*)-2,3,4,5,6-Pentahydroxy-2-(hydroxymethyl)cyclohexylammonium chloride (**63**). Triol **59** (35 mg, 0.096 mmol) was stirred in aqueous hydrochloric acid (1.0 M, 20 mL) at room temperature for 24 h. The aqueous phase was washed with ethyl acetate (2×10 mL) to remove the silanol byproduct. The aqueous phase was concentrated under reduced pressure to give **63** as a pale yellow oil (21 mg, 80%). [α]_D²⁵ +32 (*c* 0.5, H₂O); ν_{max} (neat)/cm⁻¹ 3272, 2943, 2507, 1622, 1496, 1398, 1184, 1155, 1074, 1025, 928, 892, 814, 712; ¹H NMR (250 MHz, D₂O) δ 4.16–4.14 (m, 2H), 3.91 (d, *J*=12.0 Hz, 1H), 3.83–3.70 (m, 3H), 3.59–3.56 (br t, *J*=2.5 Hz, 1H); ¹³C NMR (75 MHz, D₂O) δ 75.9, 75.6, 72.4, 69.3, 66.3, 66.1, 59.9; HRMS *m*/*z* (ES⁺) [found (M+H)⁺ 210.0965, C₇H₁₆NO₆ requires M⁺, 210.0978].

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Supplementary data

¹H and ¹³C NMR spectra for all novel compounds, as well as selected 2D NMR spectra. X-ray crystallographic data for **55** (CCDC: #908420). Supplementary data associated with this article can be found in the online version, at http://dx.doi.org/10.1016/ j.tet.2013.04.033.

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