Preliminary communication

A facile synthesis of benzyl 2-acetamido-6-*O*-acetyl-4-*O*-(6-*O*-acetyl-3,4-*O*-isopropylidene- β -D-galactopyranosyl)-2-deoxy- β -D-glucopyranoside, a key intermediate for the synthesis of *O*- α -L-fucopyranosyl-(1 \rightarrow 2)-*O*- β -D-galactopyranosyl-(1 \rightarrow 4)-*O*-[α -L-fucopyranosyl-(1 \rightarrow 3)]-2-acetamido-2-deoxy-D-glucopyranose*

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(Received December 3rd, 1982; accepted for publication, December 13th, 1982)

Convenient and practical syntheses of complex saccharides having well defined structures that are generally a part of the carbohydrate moiety of glycoconjugates have been a challenge to carbohydrate chemists. Interest in the chemical synthesis of these molecules has greatly increased, as these synthetic oligosaccharides can be effectively employed in the study of glycosidases, glycosyltransferases, and lectins. Moreover, they can play an important role as synthetic antigens. In recent years, a battery of new catalysts along with different glycosylating agents² have proved to be effective in the chemical synthesis of various oligosaccharides.

For sequential synthesis of oligosaccharides, certain glycosylating agents having temporary and persistent protecting-groups have now become available. For rapid synthesis of saccharides, it is also important to develop facile methods for the preparation of appropriately protected sugar alcohols that are to be further glycosylated. In our laboratory, we have developed a practical and convenient methodology for the synthesis of O- α -L-fucopyranosyl- $(1\rightarrow 2)$ -O- β -D-galactopyranosyl- $(1\rightarrow 3)$ -O- $[\alpha$ -L-fucopyranosyl- $(1\rightarrow 2)$ -O- β -D-galactopyranosyl- $(1\rightarrow 3)$ -O- $[\alpha$ -L-fucopyranosyl- $(1\rightarrow 2)$ -O- β -D-galactopyranose³, the Lewis b blood-group antigenic determinant, from the readily accessible benzyl 2-acetamido-2-deoxy-3-O- β -D-galactopyranosyl- β -D-glucopyranosyl- $(1\rightarrow 4)$ -O- $[\alpha$ -L-fucopyranosyl- $(1\rightarrow 2)$ -O- β -D-galactopyranosyl- $(1\rightarrow 4)$ -O- $[\alpha$ -L-fucopyranosyl- $(1\rightarrow 3)$]-2-acetamido-2-deoxy-D-glucopyranosyl- $(1\rightarrow 4)$ -O- $[\alpha$ -L-fucopyranosyl- $(1\rightarrow 3)$]-2-aceta-mido-2-deoxy-D-glucopyranosyl- $(1\rightarrow 4)$ -O- $[\alpha$ -L-fucopyranosyl- $(1\rightarrow 3)$]-2-aceta-mido-2-deoxy-D-glucopyranose-4-5, which has been found to occur as part of the carbo-hydrate moiety of blood-group substances.

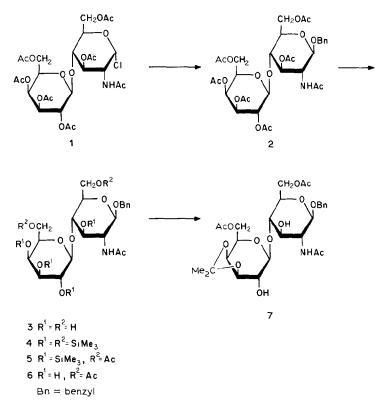
Alais and Veyrières⁶ have now described a practical technique for the preparation of 2-acetamido-2-deoxy-4-O- β -D-galactopyranosyl-D-glucopyranose (*N*-acetyllactosamine) from the commercially available 3-O- β -D-galactopyranosyl-D-arabinose. The

^{*}Synthetic Studies in Carbohydrates, Part XXXI. For Part XXX, see ref. 1.

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purified *N*-acetyllactosamine was conveniently converted into the acetylated sugar halide 1, as described by Kaifu and Osawa⁷. Treatment of halide 1 (5 g, 7.64 mmol) in benzene (20 mL) with benzyl alcohol (3.5 mL), in the presence of mercuric cyanide (3 g), afforded benzyl 2-acetamido-3,6-di-*O*-acetyl-2-deoxy-4-*O*-(2,3,4,6-tetra-*O*-acetyl- β -D-galactopyranosyl)- β -D-glucopyranoside (2) in 74% yield (4.1 g); amorphous; $[\alpha]_{\rm D}$ -34.1° (c 1, chloroform); ¹H-n.m.r. data (CDCl₃): δ 1.90 (s, 3 H, NAc), 1.93, 2.03 and 2.10 (s each, 18 H, 6 OAc), and 7.25 (m, 5 H, aromatic). On *O*-deacetylation, compound 2 (3.5 g) gave benzyl 2-acetamido-2-deoxy-4-*O*- β -D-galactopyranosyl- β -D-glucopyranoside (3) in 83% yield (1.9 g); m.p. 265–266° (from methanol), $[\alpha]_{\rm D}$ -23.2° (c 1.6, Me₂SO).

Anal. Calc. for C₂₁H₃₁NO₁₁: C, 53.27; H, 6.60; N, 2.96. Found: C, 53.24; H, 6.45; N, 2.76.



A solution of 3 (2 g) in absolute pyridine (60 mL) was treated with hexamethyldisilazane (24 mL), followed by the addition of chlorotrimethylsilane⁸ (10 mL). After being stirred for 24 h at 60°, the mixture was processed, to give the completely silylated derivative 4 (3.68 g) in 96% yield; amorphous; $[\alpha]_D$ -17.6° (c 2.6, chloroform); the i.r. spectrum showed the absence of hydroxyl group. Compound 4 (4.53 g, 5 mmol) in absolute pyridine (10 mL) and acetic anhydride (7.5 mL) was treated with glacial acetic acid (1.2 g, 20 mmol) at room temperature^{9,10}; the course of the reaction was monitored by t.l.c. in 4:1 (v/v) ether-hexane. After 40 h, the mixture was processed, to give a solid mass that was purified by chromatography on a column of silica gel, with elution with 4:1 (v/v) ether- hexane, to afford amorphous 5 (3.64 g, 86%): $[\alpha]_D$ -13.7° (c 2, chloroform). A mixture of compound 5 (4.35 g, 5.14 mmol) in methanol (25 mL) was exposed to the action of 30% acetic acid for 6 h at room temperature⁹. to give benzyl 2-acetamido-6-*O*-acetyl-4-*O* (6-*O*-acetyl- β -D-galactopyranosyl)-2-deoxy β -D-glucopyranoside (6) in 84% yield (2.4 g); amorphous; $[\alpha]_D$ -15° (c 1.4, Me₂SO), ¹H-n.m.r. data (Me₂SO-d₆): δ 1.80 (s, 3 H, NAc), 1.97 and 2.03 (s each, 2 × 3 H, 2 OAc), 7.3 (m, 5 H, aromatic), and 7.77 (1 H, NH).

Anal. Calc. for C₂₅H₃₅NO₁₃•H₂O: C, 52.16; H, 6.48; N, 2.43. Found: C, 51.96, H, 6.24; N, 2.42.

TABLE I

Atom	Compound			
	3	6	7	
C-1	100.38	100.19	98.99	
C-2	54.65	54.45	56.58	
C-3	72.03	71.74	72.56	
C-4	81.27	80.46	80.77	
C-5	74.94	72.64	72.99	
C-6	60.39	62.98	63.43	
CH, Ph	69.41	69.56	70.65	
NCOCH.	22.93	22.82	23.39	
осо <i>с</i> н,		20.37, 20.60	20.62, 20.95	
NHCOCH,	168.56	168.72	170.67	
OCOCH₃ ⊂		169.90, 170.05	170.92	
C-1′	103.83	103.48	102.83	
C-2′	70.49	69.83	71.46	
C-3'	73.08	71.74	78.95	
C-4'	68.06	68.01	72.99	
C-5′	75.45	72.29	72.99	
C-6′	60.39	63.21	63.18	

¹³C-N.M.R. CHEMICAL SHIFTS ^a

^{*a*} Solvent was $Me_2 SO-d_6$, except for $CDCl_3$ for 7. Chemical shifts in p.p.m. downfield from $Me_4 Si$ (internal) at 25.2 MHz.

Isopropylidenation¹¹ of compound 6 (2 g, 3.6 mmol) with 2.2-dimethoxypropane (4 mL) in *N*,*N*-dimethylformamide in the presence of *p*-toluenesulfonic acid (30 mg) for 1 h at 60° gave a solid mass which was purified by chromatography on a column of silica gel with elution with 19:1 (v/v) chloroform -methanol, to give amorphous 7 (1.4 g) in 65.3% yield; $[\alpha]_D$ +3.1° (*c* 1.2, chloroform); ¹H-n.m.r. data (CDCl₃): δ 1.3 and 1.47 (2 s, 2 × 3 H, isopropylidene methyls), 1.93 (s, 3 H, NAc), 2.03 and 2.07 (s each, 2 × 3 H, 2 OAc), 5.70 (d, 1 H, *J*_{NH,2} 7 Hz, NH), and 7.3 (m, 5 H, aromatic). The ¹³C-n.m.r. spectrum exhibited resonance for the acetal carbon atom at 110.45 p.p.m., and the chemical shifts for the methyl groups (26.21 and 28.01 p.p.m.) were slightly separated, thereby supporting the presence of a five-membered¹², cyclic ring.

Anal. Calc. for C₂₈H₃₉NO₁₃: C, 56.27; H, 6.58; N, 2.34. Found: C, 56.06; H, 6.47; N, 2.13.

Starting from benzyl 2-acetamido-3-*O*-allyl-6-*O*-benzyl-2-deoxy- α -D-glucopyranoside as the key intermediate, Jacquinet and Sinäy⁵ accomplished the synthesis of *O*- α -L-fucopyranosyl-(1 \rightarrow 2)-*O*- β -D-galactopyranosyl-(1 \rightarrow 4)-*O*-[α -L-fucopyranosyl-(1 \rightarrow 3)]-2-acetamido-2-deoxy- α -D-glucopyranose. As this "aglycon", as well as the glycosylating reagent 3,4,6-tri-*O*-benzyl-1,2-*O*-(*tert*-butoxyethylidene)- α -D-galactopyranose employed in their synthetic methodology, involve multisteps for their preparation, we consider that synthetic diol 7 may prove to be a suitable, key intermediate for the facile synthesis of *O*- α -L-fucopyranosyl-(1 \rightarrow 2)-*O*- β -D-galactopyranosyl-(1 \rightarrow 4)-*O*-[α -L-fucopyranosyl-(1 \rightarrow 3)]-2-acetamido-2-deoxy-D-glucopyranose. According to our preliminary investigations, L-fucosylation of 7 with 2,3,4-tri-*O*-benzyl- α -L-fucopyranosyl bromide, under catalysis by bromide ion¹³, afforded mono-fucosylated product.

ACKNOWLEDGMENTS

We thank C. F. Piskorz for his excellent technical assistance, and Mrs. Onda Simmons for recording the n.m.r. spectra. We also thank Miss Marie Fox for kindly typing the manuscript. The n.m.r. studies were supported by National Cancer Institute Core Grant CA-16056. This investigation was supported by Grant No. CA-24051, awarded by the National Institutes of Health.

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