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Novel Synthesis of 1,6-Naphthyridin-2(6H)-ones, Quinolin-2(1H)-ones, and Quino [7,8-f] quinoline-2,9(1H,10H)-dione from Common Precursors

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Received 3 June 1991; revised 31 July 1991

1,6-Naphthyridin-2(6H)-ones 3, quinolin-2(1H)-ones 5, and quino[7,8-f]quinoline-2,9(1H,10H)-dione 10 have been synthesized from 5-acyl-6-[2-(dimethylamino)ethenyl]pyridin-2(1H)-ones 1.

In a recent paper, we described the synthesis of 1,6-naphthyridin-2(1H)-ones 2 by the condensation of ammonium acetate with 5-acyl-6-[2-(dimethylamino)-ethenyl]pyridin-2(1H)-ones 1. We now wish to report on an extension of the use of 1 in the synthesis of other heterocyclic systems: namely, 1,6-naphthyridin-2(6H)-ones, quinolin-2(1H)-ones, and quino[7,8-f]quinoline-2,9(1H,10H)-diones.

The condensation of 4-methoxybenzenamine with 1a in refluxing dimethylformamide led to the formation of 1,6-naphthyridin-2(6H)-one (3a) in 51% yield. Similar reaction of 1a with dimethylhydrazine and hydrazine in refluxing methanol gave 3b and 3c in 90% and 73% yield, respectively. The alternative structure 4 for the product resulting from the reaction between 1a and hydrazine was ruled out by the spectral evidence. The pattern in the 1H NMR spectrum of 3c for 7-H and 8-H is similar to those of 3a and 3b whereas that of 4 would be different. The NMR spectrum of 3c in (DMSO- d_6) showed a singlet at $\delta = 6.80$ for the NH₂ group. Furthermore, the mass spectrum of 3c gave a strong peak at m/z = 160 (MH⁺ - NH₂). This type of fragmentation due to the loss of NH₂ supports structure 3c.

Treatment of 1a with concentrated hydrochloric acid at room temperature gave 5-hydroxyquinolin-2(1H)-one $(5a)^2$, a key intermediate³ in the synthesis of β -blockers, in 90% yield. Similar treatment of 1b gave quinolone 5b in 80% yield.

The reaction of 1c was carried out in acetic acid due to the instability of the nitrile group in concentrated hydrochloric acid. Treatment of 1c with refluxing acetic acid resulted in the formation of quinolines 6 and 7 in 34% and 13% yield, respectively. A possible mechanism for the formation of 6 requires enamine 8 which is formed by the reaction between 1c and dimethylamine which in turn is produced by the initial reaction of acetic acid with 1c or during the formation of 7. The acid-catalyzed intramolecular cyclization of 8 leads to the formation of 6 and dimethylamine. The process is repeated until the starting material 1c is consumed.

Treatment of 1a with acetic acid under similar conditions, however, led to the formation of 10 (57%) instead of the expected product 9. The structure of 10 is supported by the spectral data. The mass spectrum gave a MH⁺ peak at 305; ¹³C NMR spectrum indicated eight proton bearing carbon atoms, and ¹H NMR spectrum confirmed the presence of an aldehyde group, two pairs of ortho coupled protons, two uncoupled aromatic protons, and a methyl group. The initial step in the formation of 10 involves the acid-catalyzed dimerization of 1a. The resulting intermediate undergoes further transformations to yield 10.

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6-(4-Methoxyphenyl)-5-methyl-1,6-naphthyridin-2(6H)-one (3a):

A mixture of **1a** (10.3 g, 0.05 mol), 4-methoxybenzenamine (7 g, 0.05 mol), and DMF (50 mL) was heated under reflux for 3.5 h and then concentrated under reduced pressure. The dark brown solid residue was recrystallized from 2-propanol after treating with charcoal; yield: 6.8 g (51 %); mp 212-215 °C dec.

C₁₆H₁₄N₂O₂ calc. C 72.17 H 5.30 N 10.52 (266.30) found 72.08 5.43 10.45

¹H NMR (DMSO- d_6 /TMS): δ = 2.46 (s, 3 H, CH₃), 3.91 (s, 3 H, OCH₃), 6.45 (d, $J_{3,4}$ = 9.6 Hz, 1 H, H-3), 6.95 (d, $J_{7,8}$ = 7.3 Hz, 1 H, H-8), 7.16, 7.58 (A₂B₂, J = 8.8 Hz, C₆H₄O), 7.84 (d, $J_{3,4}$ = 9.6 Hz, 1 H, H-4), 7.90 (d, $J_{7,8}$ = 7.3 Hz, 1 H, H-7).

6-(Dimethylamino)-5-methyl-1,6-naphthyridin-2(6H)-one (3b):

A mixture of 1a (10.3 g, 0.05 mol), N,N-dimethylhydrazine (6 g, 0.1 mol), and MeOH (200 mL) was heated under reflux for 4 h and then concentrated to dryness. The residue was crystallized from 2-propanol to afford 3b; yield: 9.1 g (90%); mp 228-230°C dec.

C₁₁H₁₃N₃O calc. C 65.01 H 6.45 N 20.67 (283.24) found 64.82 6.68 20.56

¹H NMR (CF₃CO₂D/TMS): δ = 3.11 [s, 6 H, N(CH₃)₂], 3.35 (s, 3 H, CH₃), 7.25 (d, $J_{3,4}$ = 9.7 Hz, 1 H, H-3), 7.95 (d, $J_{7,8}$ = 7.3 Hz, 1 H, H-8), 8.48 (d, $J_{3,4}$ = 9.7 Hz, 1 H, H-4), 8.86 (d, $J_{7,8}$ = 7.3 Hz, 1 H, H-7).

6-Amino-5-methyl-1,6-naphthyridin-2(6H)-one (3c):

Following the procedure for the preparation of 3b, 3c was prepared in 77% yield; mp 262-264°C dec.

C₉H₉N₃O calc. C 61.70 H 5.18 H 23.99 (175.19) found 61.82 5.31 23.97

MS (CI/CH₄): m/z = 176 (MH⁺), 175 (M⁺), 160 (MH⁺ - NH₂). ¹H NMR (DMSO- d_6 /TMS): $\delta = 2.76$ (s, 3 H, CH₃), 6.38 (d, $J_{3,4} = 9.7$ Hz, 1 H, H-3), 6.80 (br s, 2 H, NH₂, 6.91 (d, $J_{7,8} = 7.3$ Hz, 1 H, H-8), 7.83 (d, $J_{3,4} = 9.7$ Hz, 1 H, H-4), 7.97 (d, $J_{7,8} = 7.3$ Hz, 1 H, H-7).

5-Hydroxyquinolin-2(1H)-one (5a):

To a solution of conc. HCl (100 mL) cooled in an ice bath was added 1a (10.3 g, 0.05 mol). The resulting solution was taken out of the ice bath, left at r.t. overnight, and concentrated to dryness under vacuum. The residue was treated first with a slight excess of 10% aq K_2CO_3 and then reacidified with AcOH. The resulting precipitate was collected and recrystallized from MeOH to afford 5a: yield: 7.25 g (90%); mp > 300°C (Lit.4 mp 336-341°C).

¹H NMR (DMSO- d_6 /TMS): δ = 6.35 (d, $J_{3,4}$ = 9.7 Hz, 1 H, H-3), 6.55 (d, J = 9.7 Hz, 1 H), 6.73 (d, J = 8.2 Hz, 1 H), 7.23 (t, J = 8.1 Hz, 1 H, H-7), 8.05 (d, $J_{3,4}$ = 9.7 Hz, H-4), 10.3 (br s, 1 H), 11.55 (br s, 1 H).

5-Hydroxy-6-methylquinolin-2(1H)-one (5b):

Following the procedure for the preparation of **5a**, **5b** was prepared in 77% yield: mp 297-300°C.

C₁₀H₉NO₂ calc. C 68.56 H 5.18 N 8.00 (175.19) found 68.68 5.26 8.12

¹H NMR (CF₃CO₂D/TMS): δ = 2.49 (s, 3 H, CH₃), 7.28 (d, $J_{3,4}$ = 9.6 Hz), 1 H, H-3), 7.40 (d, $J_{7,8}$ = 8.3 Hz, 1 H), 7.79 (d, $J_{7,8}$ = 8.3 Hz, 1 H), 9.08 (d, $J_{3,4}$ = 9.6 Hz, 1 H, H-4).

1,2-Dihydro-5-(dimethylamino)-2-oxoquinoline-3-carbonitrile and 1,2-Dihydro-5-hydroxy-2-oxoquinoline-3-carbonitrile (7):

A stirred mixture of 1c (81.5 g, 0.35 mol) and glacial AcOH (1 L) was heated under reflux for 45 h and then concentrated to dryness under reduced pressure. The reddish residue was treated with water (300 mL) and filtered off. Recrystallization from DMF (250 mL) gave 6; yield: 25.4 g (34%); mp 295-298°C dec.

IR (KBr): v = 2230 (CN) cm⁻¹.

¹H NMR (DMSO- d_6 /TMS): $\delta = 2.77$ (s, 6 H, N(CH₃)₂), 6.77 (d, J = 8.0 Hz, 1 H), 6.89 (d, J = 7.2 Hz, 1 H), 7.48 (t, J = 8.1 Hz, H-7), 8.59 (s, 1 H, H-4), 11.25 (s, 1 H, NH).

The mother liquor from above was treated with charcoal, concentrated to ≈ 100 mL and then allowed to stand at r.t. whereupon the second, less polar, component 7 crystallized as orange needles; yield: 8.4 g (13%); mp > 300 °C.

C₁₀H₆N₂O₂ calc. C 64.52 H 3.25 N 15.05 (186.17) found 64.34 3.42 15.09

IR (KBr): v = 2235 (CN) cm⁻¹.

¹H NMR (DMSO- d_6 /TMS): $\delta = 6.59$ (d, J = 8.1 Hz, 1 H), 6.71 (d, J = 8.1 Hz, 1 H), 7.40 (t, J = 8.2 Hz, 1 H, H-7), 8.59 (s, 1 H, H-4), 10.91 (br s, 1 H), 11.25 (br s, 1 H).

1,2,9,10-Tetrahydro-5-methyl-2,9-dioxoquino[7,8-f]quinoline-11-carboxaldehyde (10):

A stirred mixture of 1a (10.3 g, 0.05 mol) and glacial AcOH (100 mL) was heated under reflux for 24 h and then concentrated to dryness. The yellow solid residue was treated with water (100 mL). The product was collected and recrystallized from DMF to afford 10; yield: 4.4 g (57%); mp > 300 °C.

C₁₈H₁₂N₂O₃·0.25H₂O calc. C 70.01 H 4.00 N 9.07 (308.81) found 70.01 4.08 9.07

IR (KBr): v = 1732 (CHO) cm⁻¹.

 $MS (CI/CH_4): m/z = 305 (MH^+).$

¹H NMR (CF₃CO₂D/TMS): δ = 3.06 (s, 3 H, CH₃), 7.46 (d, 1 H, J = 10 Hz), 7.7 (d, 1 H, J = 10 Hz), 8.63 (s, 1 H), 8.83 (d, 1 H, J = 10 Hz), 9.53 (d, J = 10 Hz, 1 H), 9.90 (s, 1 H), 10.46 (s, 1 H, CHO). ¹³C NMR (CF₃CO₂D/TMS): δ = 198.97, 183.64, 168.25, 166.30, 145.84, 145.54, 145.08, 141.06, 138.66, 137.59, 123.34, 123.11, 123.04, 121.75, 121.49, 120.50, 21.83.

The author is thankful to the Department of Molecular Characterization for the ¹H NMR spectra.

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